# On single-electron magnesium bonding formation and the effect of methyl substitution 

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Figure S1. The optimized geometries of the $\mathrm{MgX}_{2}(\mathrm{X}=\mathrm{F}, \mathrm{H})$ molecules and $\mathrm{CH}_{3}, \mathrm{CH}_{2} \mathrm{CH}_{3}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ and $\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}$ radicals at the MP2/aug-cc-pVTZ level.


Fig S2. (a) Potential energy scan for $\mathbf{I}$, (b) Potential energy scan for $\mathbf{I}^{\prime}$. SC1 and SC 2 represent the distance between Mg and C atoms and the $\mathrm{X}-\mathrm{Mg}-\mathrm{C}(\mathrm{X}=\mathrm{F}, \mathrm{H})$ angle, respectively. Distances in $\AA$ and angles in degrees.


Fig. $\mathbf{S 3}$ (a) The relationship between interaction energy and the number of methyl substituents, (b) the relationship between electron density $(\rho)$ at the $\mathrm{Mg} \cdots \mathrm{C} 1 \mathrm{BCP}$ and the number of methyl substituents of the $\mathrm{X}_{2} \mathrm{Mg} \cdots \mathrm{Y}\left[\mathrm{X}=\mathrm{F}, \mathrm{H} ; \mathrm{Y}=\mathrm{CH}_{3}, \mathrm{CH}_{2} \mathrm{CH}_{3}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}, \mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right] \text { complexes. }}\right.$

Table S1 The symmetry, distance between Mg and C 1 atoms ( $d_{\mathrm{Mg}-\mathrm{Cl}}$, in $\AA$ ), $\mathrm{X}-\mathrm{Mg}-\mathrm{X}$ angle ( $\alpha$, in degree), $\mathrm{Mg}-\mathrm{X} 1 / \mathrm{X} 2$ bond length ( $R_{\mathrm{Mg}-\mathrm{X} 1}$, in $\AA$ ), and the $D_{\mathrm{X} 1-\mathrm{Mg}-\mathrm{C} 1-\mathrm{H} 1}$ dihedral angle (in degree) of the $\mathrm{X}_{2} \mathrm{Mg} \cdots \mathrm{Y}\left[\mathrm{X}=\mathrm{F}, \mathrm{H} ; \mathrm{Y}=\mathrm{CH}_{3}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ ] complexes obtained by using different methods with the same aug-cc-pVTZ basis set.

| Complex | Method | Symmetry | $d_{\mathrm{Mg}-\mathrm{C} 1}$ | $\alpha$ | $R_{\mathrm{Mg}-\mathrm{X1} 1}$ | $R_{\mathrm{Mg}-\mathrm{X} 2}$ | $D_{\mathrm{X1} 1-\mathrm{Mg}-\mathrm{Cl}-\mathrm{H} 1}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| I | MP2 | $C_{s}$ | 2.573 | 163.0 | 1.778 | 1.779 | 0.0 |
|  | wB97XD | $C_{s}$ | 2.527 | 164.9 | 1.767 | 1.768 | 0.0 |
|  | M06-2X | $C_{s}$ | 2.506 | 165.1 | 1.752 | 1.754 | 0.0 |
|  | B3LPY-D3(BJ) | $C_{s}$ | 2.514 | 162.2 | 1.767 | 1.768 | 0.0 |
|  | B2PLYPD3 | $C_{s}$ | 2.525 | 162.5 | 1.768 | 1.769 | 0.0 |
| II-1 |  |  |  |  |  |  |  |
|  | MP2 | $C_{s}$ | 2.509 | 160.1 | 1.781 | 1.781 | 32.2 |
|  | wB97XD | $C_{s}$ | 2.475 | 161.5 | 1.770 | 1.770 | 32.0 |
|  | M06-2X | $C_{s}$ | 2.443 | 161.5 | 1.756 | 1.756 | 30.2 |
|  | B3LPY-D3(BJ) | $C_{s}$ | 2.458 | 158.6 | 1.770 | 1.770 | 32.2 |
|  | B2PLYPD3 | $C_{s}$ | 2.466 | 159.3 | 1.771 | 1.771 | 32.1 |
|  |  |  |  |  |  |  |  |
|  | MP2 | $C_{1}$ | 2.521 | 160.1 | 1.780 | 1.783 | 12.3 |
|  | B3LPY-D3(BJ) | $C_{l}$ | 2.465 | 158.3 | 1.769 | 1.772 | 6.0 |
|  | B2PLYPD3 | $C_{l}$ | 2.473 | 159.0 | 1.771 | 1.773 | 8.8 |
|  |  |  |  |  |  |  |  |
|  | MP2 | $C_{s}$ | 2.707 | 166.6 | 1.717 | 1.717 | 89.9 |
|  | wB97XD | $C_{s}$ | 2.590 | 165.4 | 1.719 | 1.719 | 89.9 |
|  | M06-2X | $C_{s}$ | 2.608 | 165.2 | 1.704 | 1.704 | 89.8 |
|  | B3LPY-D3(BJ) | $C_{s}$ | 2.551 | 162.8 | 1.716 | 1.716 | 89.9 |
|  | B2PLYPD3 | $C_{s}$ | 2.629 | 165.1 | 1.712 | 1.712 | 89.9 |
|  |  |  |  |  |  |  |  |
|  | MP2 | $C_{s}$ | 2.644 | 164.3 | 1.719 | 1.719 | 31.8 |
|  | wB97XD | $C_{s}$ | 2.583 | 163.3 | 1.719 | 1.719 | 31.9 |
|  | M06-2X | $C_{s}$ | 2.579 | 163.4 | 1.706 | 1.706 | 31.8 |
|  | B3LPY-D3(BJ) | $C_{s}$ | 2.581 | 161.7 | 1.715 | 1.715 | 32.1 |
|  | B2PLYPD3 | $C_{s}$ | 2.607 | 162.9 | 1.713 | 1.713 | 32.0 |

