## **Supplementary Material**

## Degradation of norfloxacin by sulfate radical-based visible light-Fenton by copper-doped Bi2WO6

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Figure S1 Schematic diagram of Cu-Bi<sub>2</sub>WO<sub>6</sub> synthesis process.



Figure S2 (a)  $N_2$  adsorption–desorption isotherms and (b) pore size distribution of  $Bi_2WO_6$  and  $5Cu-Bi_2WO_6$ .



Figure S3 XRD spectra of  $5Cu-Bi_2WO_6$  before and after reaction.



Figure S4 SEM images of  $5Cu-Bi_2WO_6$  after reaction.



Figure S5 Degradation of different contaminant by 5Cu-Bi<sub>2</sub>WO<sub>6</sub>.

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No.	Systems	Light	Target	Reaction time	Operation parameters
		source	pollutant	and efficiency	
1	PMS/CuBi <sub>2</sub> O <sub>4</sub> /VL	300W	ТС	Nearly complete	[PMS]= 0.125 mg/mL
		xenon lamp	50 mg/L	removal	[catalyst]=0.5 g/L
				60 min	
2	PMS/Co-BiVO <sub>4</sub> /VL	xenon lamp	Tetracycline	Nearly complete	[PMS]= 5 mM
			hydrochloride	removal	[catalyst]=0.2 g/L
			40 mg/L	25 min	
3	MIL-53(Fe)/PS/VL	300W	tetracycline	99.6% TC	[PMS]= 8 mM
		xenon lamp	hydrochloride	60 min	[catalyst]=0.2 g/L
			300 mg/L		
4	solar/MOFs@COFs/PS	500 W	bisphenol A	82%	[PMS]= 0.5g /L
		Xenon lamp	50 mg/L	120 min	[catalyst]=0.25 g/L
5	Vis/TiO <sub>2</sub> /PMS	300 W	Perfluorooctano	100%	[PMS]= 0.75 g L <sup>-1</sup>
		Xenon lamp	ic	9 h	[TiO <sub>2</sub> ] = 0.25 g L-1
			50 mg/L		
6	Vis/Fe(II)/V(IV) self-	300 W	Sulfamethoxazo	96.6%	$[catalyst] = 0.5 \text{ g } \text{L}^{-1}, [PMS]$
	doped FeVO <sub>4</sub> /PMS	Xenon lamp	le	60 min	= 0.406 mM
			0.02 mM		
7	Vis/MoS <sub>2</sub> /Ag/g-	300 W	Tetracycline	79.7%	[PMS]= 0.1 mM
	C <sub>3</sub> N <sub>4</sub> /PMS	Xenon lamp	20 mg/L	50 min	[catalyst]=0.2 g/L
8	Vis/ Fe/C <sub>3</sub> N <sub>4</sub> /PS	350 W	rhodamine B	100%	[catalyst] = 400 ppm, [PS] =
		Xenon lamp	20 mg/L	40 min	3 mM
					pH = 3.5
This	Cu-Bi <sub>2</sub> WO <sub>6</sub> +Vis+PMS	30 W LED	Norfloxacin 20	89.27%	[catalyst]=0.5 g/L
work		lamp	mg/L	60 min	[PS]=0.4 mM
					Neutral pH

Table S1 Studies on of photocatalysts in the catalyst/PMS/PS/Vis reactions

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Table S2 Possible intermediates of NOF during the reaction Structure Name m/zÖ QН F Ò NOF m/z=320HN QН O N1 m/z=350ΗN QН Ò N2 m/z=336 HN QН Ò N3 m/z=322 H<sub>2</sub>N °0



