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A highly sensitive and selective on-off fluorescent sensor based on complex of Polyschiff-Fe $^{2+}$ for Cr(VI) detection in the aqueous medium

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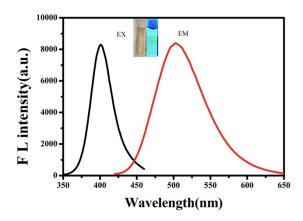


Fig. S1. The Excitation and emission spectra of PS-Fe²⁺

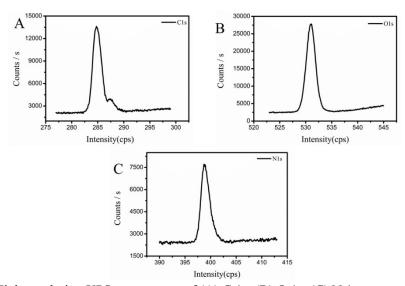


Fig. S2. High-resolution XPS survey scan of (A) C 1s, (B) O 1s, (C) N 1s, spectra of PS-Fe²⁺

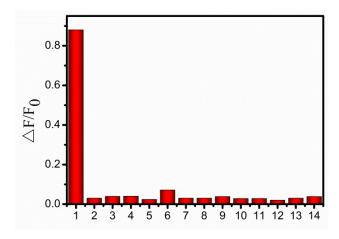


Fig. S3. Selectivity of determination method of PS-Fe²⁺

 $Cr_{2}O_{7}{}^{2\text{-}},\ Cu^{2+},\ Mn^{2+},\ Pb^{2+},\ Ag^{+},\ Hg^{+},\ Ba^{2+},\ Cd^{2+},\ Cr^{3+},\ SO_{4}{}^{2\text{-}},\ NO_{2}{}^{\text{-}},\ SCN^{\text{-}},\ PO_{4}{}^{3\text{-}}\ and\ C_{2}O_{4}{}^{2\text{-}}.$

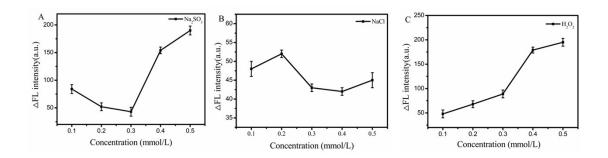


Fig. S4. Fluorescence intensity stability of the PS-Fe² of NaCl, Na₂SO₃ and H₂O₂

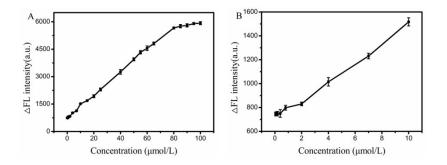


Fig. S5. Correlation analysis of Cr(VI) using **PS-Fe**²⁺. (A) Fluorescence quenching of **PS-Fe**²⁺ containing system upon different concentrations of $Cr_2O_7^{2-}$ (0.01--100 μ M) addition; (B) Fluorescence quenching of **PS-Fe**²⁺ containing system upon different concentrations of $Cr_2O_7^{2-}$ (0.01--10 μ M) addition.