Supporting information

for

Photocatalytic degradation of diphenhydramine in aqueous solution by natural dolomite

Lihong Song^a, Chunlin Yi^a, Qingfeng Wu^{a,*}, Zhaohui Li^{b,*}, Weibin Zhang^a, Hanlie Hong^c

^aSchool of Physics and Optoelectronic Engineering, Yangtze University, 1 Nanhuan Road, Jingzhou, Hubei 434023, China

^bDepartment of Geosciences, University of Wisconsin-Parkside, 900 Wood Road, Kenosha, WI 53144, USA

^cFaculty of Earth Sciences, China University of Geosciences, 388 Lumo Road, Wuhan, Hubei 430074, China



Fig. S1. SEM image of the natural dolomite at different magnifications



Fig. S2. XRD pattern of the natural dolomite sample.



Fig. S3. The optimized structures of (a) the $2 \times 2 \times 1$ dolomite supercell, and (b) the supercell with 3 Fe²⁺substitution for 3 Mg²⁺.



Fig. S4. The UV-Vis reflectance spectra of natural dolomite (a), and estimated band gap from UV-Vis spectra (b).







Fig. S5. HRMS spectra of major byproducts after 45 min irradiation in the presence of natural dolomite.