Supplementary Information

Visible Light Responsive TiO₂ Photocatalysts for

Degradation of Indoor Acetaldehyde

Suzuko Yamazaki*, Keisuke Kozasa, Kohshiro Okimura, Kensuke Honda

Department of Chemistry, College of Science, Graduate School of Sciences and Technology for Innovation, Yamaguchi University, Yamaguchi 753-8512, Japan.

*Corresponding author. Tel.: +81-83-933-5763, E-mail address: yamazaki@yamaguchi-u.ac.jp

Experimental

Cleaning method of the dialysis tubing

The dialysis tubing (Spectra/Por[®] 3, molecular weight cut-off: 3500) was cut into 23 cm length and immersed into 1 L of an aqueous solution containing 1.0 x 10^{-3} mol dm⁻³ ethylenediaminetetraacetic acid and 2% NaHCO₃ under stirring at 80°C for 30 min. After cooling to room temperature, it was washed thoroughly with water. Then, it was immersed in water under stirring at 80°C for 30 min followed by cooling and washing with water thoroughly.

Additive	Doped amount of metal ions (atom%)									
amount (atom%)	Cr(III)	Pt(IV)	V(III)	Fe(III)	Cu(II)	Nb(V)	Mo(VI)	W(VI)	Mn(II)	Ru(III)
0.10		0.10		0.10					0.01	
0.20	0.20				0.11					
0.30			0.30							
0.50	0.50	0.50	0.50	0.50	0.23	0.50	0.50	0.50	0.20	0.42
0.80	0.80									
1.0		1.0	1.0	1.0	0.35	1.0	1.0	1.0	0.30	0.90
2.0	1.7	1.9	1.8		0.70	1.0			0.47	1.8
5.0	2.9									
10	4.2									

Table S1. Amounts of metal ions used in the synthesis and those doped in the finally obtained samples.



Figure S1. XRD patterns of Cr-TiO₂, Pt-TiO₂, V-TiO₂ and TiO₂ powder sintered at 500 °C.



Figure S2. Effect of sintering temperature on the r_0 values for 0.8 atom% Cr-TiO₂, 0.1 atom% Pt-TiO₂, and 1.0 atom% V-TiO₂.



Figure S3. UV-vis diffuse reflectance spectra of (a) Cr-TiO₂, (b) Pt-TiO₂, and (c) V-TiO₂ powder sintered at 500 °C. Black line (0 atom%) shows the spectra of TiO₂.



Figure S4. XPS spectra of (a) Cr-TiO₂, (b) Pt-TiO₂, and (c) V-TiO₂ powder sintered at 500°C.



Figure S5. Mott-Schottky plots of TiO₂-DC film at different frequencies (250 – 10000 Hz).



Figure S6. Photocatalytic degradation of CH₃CHO on 0.8 atom% Cr-TiO₂ synthesized with and without dialysis.



Figure S7. Time course of the concentrations of (a) CH₃CHO and (b) 2-propanol in the reaction vessel containing 0.8 atom% Cr-TiO₂ and various amounts of water vapor.