

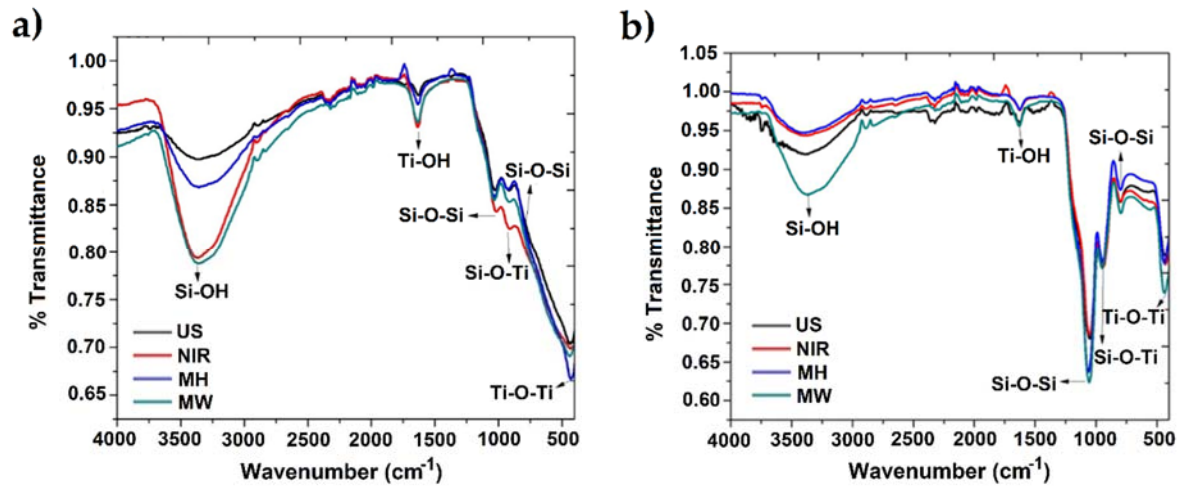
Content

- 1.- Table 1** FTIR data of SiO₂-TiO₂ catalyst, most important bands.
- 2.- Figure 1** FTIR of SiO₂-TiO₂ for a) 1:4, b) 4:1 molar ratio catalyst obtained by MH, US, NIR and MW.
- 3.- Figure 2** Diffractograms of SiO₂-TiO₂ for a) 1:1, b) 4:1 molar ratio catalyst.
- 4.- Table 2** Band gap energy values calculated from DRS-UV-vis spectra.
- 5.- Figure 3** Curves of Kubelka-Munk of SiO₂-TiO₂ for a) 1:4, b) 4:1 molar ratio catalyst.
- 6.-** Determination of Band gap energy as function of Kubelka-Munk method.

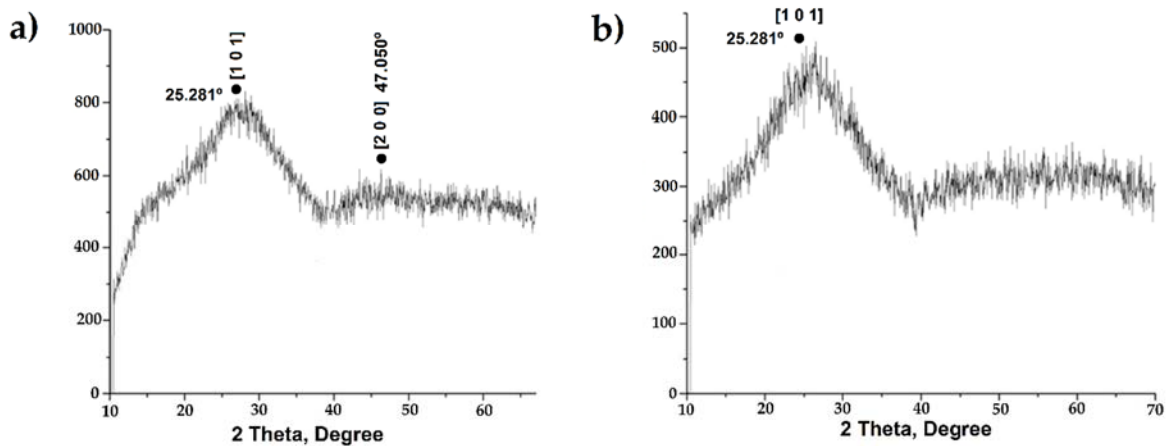
1.-

Table 1 FTIR data of SiO₂-TiO₂ catalyst, most important bands

Bonds	Molar ratio SiO ₂ -TiO ₂			
	1:2	1:4	4:1	2:1
Si-O-Si	1043-1025	1021-1047	1060-1046	1064-1049
Si-O-Si	801-794	801-798	808-794	794-797
Si-O-Ti	927-905	920-902	952-937	944-939
Ti-O-Ti	436-421	435-416	440-422	435-423



2.- Figure 1 FTIR of SiO₂-TiO₂ for a) 1:4, b) 4:1 molar ratio catalyst obtained by MH, US, NIR and MW.

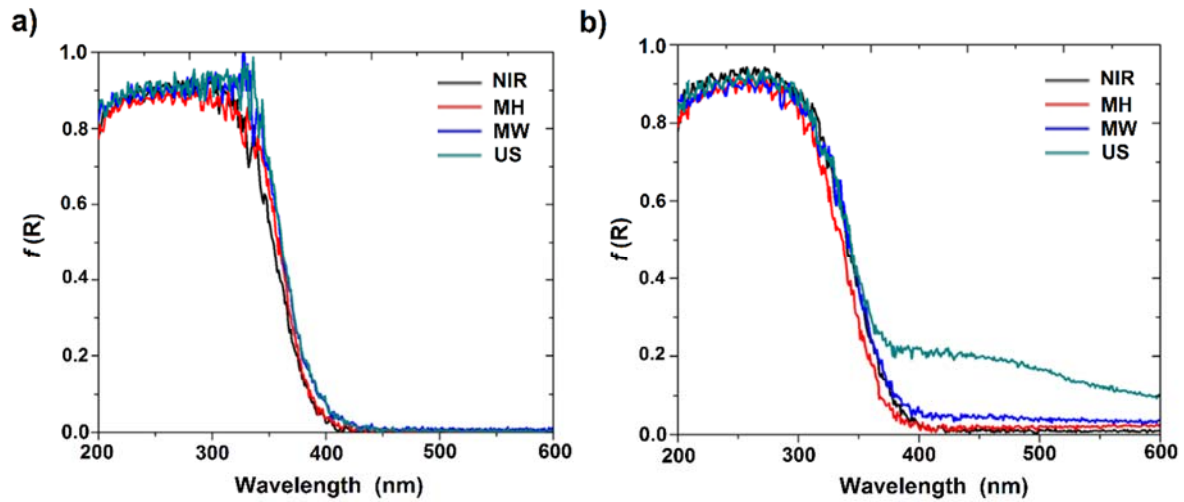


3.- Figure 2 Diffractograms of SiO₂-TiO₂ for a) 1:1, b) 4:1 molar ratio catalyst.

4.-

Table 2 Band gap energy values calculated from DRS-UV-vis spectra

Molar ratio SiO ₂ -TiO ₂	MH	NIR	US	MW
	λ (nm)/E _g (eV)			
1:1	368/3.3	363/3.4	378/3.2	375/3.3
1:2	380/3.2	364/3.4	372/3.3	370/3.3
1:4	377/3.3	375/3.3	381/3.2	382/3.2
2:1	360/3.4	359/3.4	358/3.4	361/3.4
4:1	355/3.5	360/3.4	366/3.4	361/3.4



5.- **Figure 3** Curves of Kubelka-Munk of SiO₂-TiO₂ for a) 1:4, b) 4:1 molar ratio catalyst.

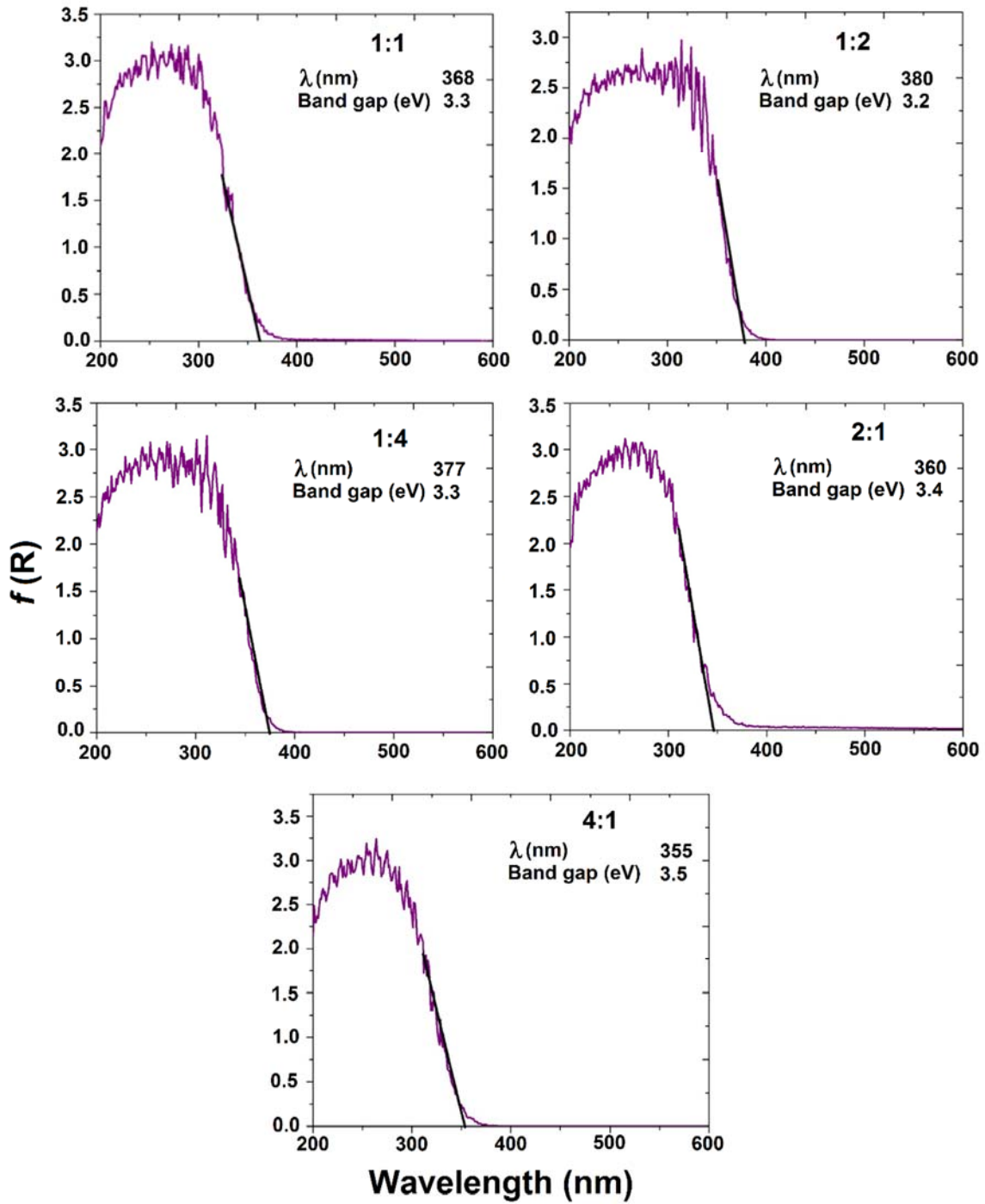
6.- Determination of band gap energy as function of Kubelka-Munk method.

$$E_g = \frac{hc}{\lambda_g}$$

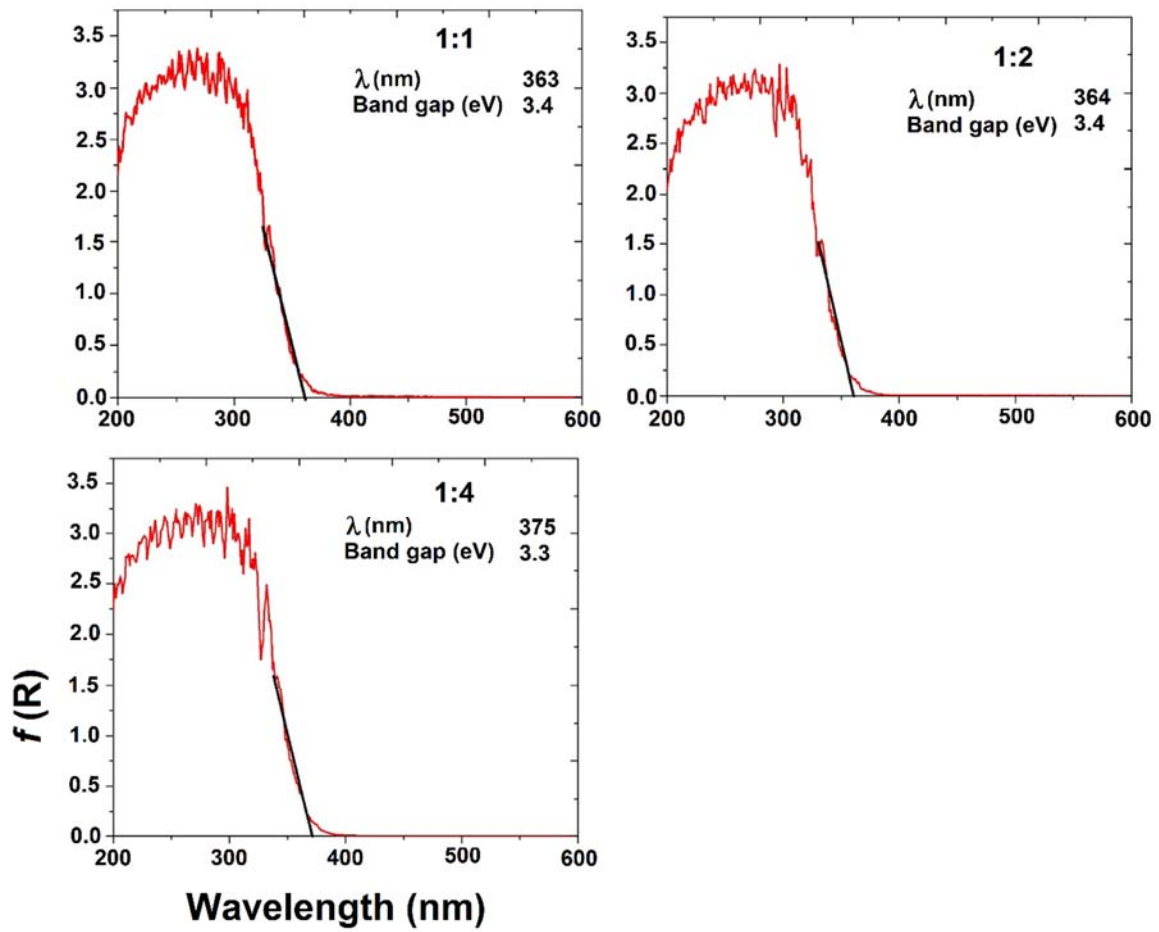
$$E_g = \frac{(4.09 \times 10^{-15} \text{ eV}) \left(3 \times 10^{17} \frac{\text{nm}}{\text{s}}\right)}{\lambda_g}$$

$$E_g \text{ (eV)} = \left(\frac{1226.2}{\lambda_g}\right)$$

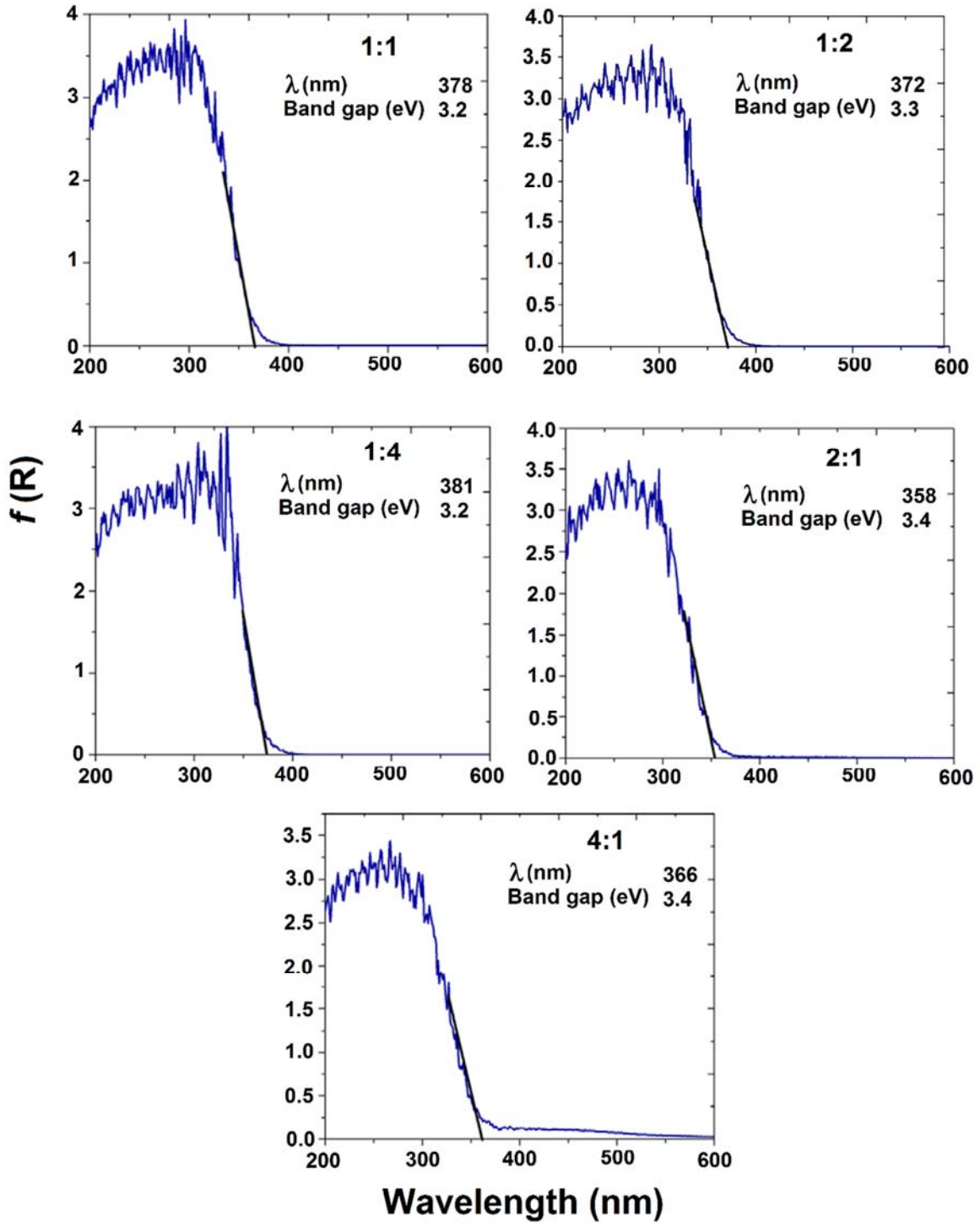
MH



NIR



US



MW

