Supporting Information (3 pages)

## Pre-oxidization-induced Change of Physicochemical Characteristics and Removal Behaviours in Conventional Drinking Water Treatment Processes for Polyethylene Microplastics

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**Text 1S.** The electron paramagnetic resonance (EPR) spin trapping experiments were performed using a Bruker A200 ESP 300E instrument at 300 K. The EPR spectrometer settings in the spin trapping experiments were: center field, 351.194 mT; sweep width, 10.00 mT; modulation amplitude, 0.1 mT; sweep time, 41 s; microwave frequency, 9.858 GHz; microwave power, 2.25 mW; receiver gain,  $1.42 \times 10^4$ . The sample of ozonation system was draw at the reaction time of 3 min, followed by DMPO addition ([DMPO]<sub>0</sub> = 12.5 mg/L) and subsequent filtration. The filtrated sample was absorbed in the capillaries before transferring into the EPR spin trapping apparatus.



**Fig. 1S**. Evolution of  $UV_{254}$  for mixed solution of HA acid and PE MPs. (Experimental conditions:  $[HA]_0= 8 \text{ mg}\cdot\text{L}^{-1}$ , PE MPs dose = 32 mg·L<sup>-1</sup>, 20 NTU turbidity, pH=7.0, ionic strength 10 mM, 25 °C)



**Fig. 2S**. The EPR spectra of DMPO-HO• sample collected from pre-ozonation system in the presence of DMPO. (Experimental conditions:  $[O_3]_0= 2.0 \text{ mg}\cdot\text{L}^{-1}$ , NOM-preloading PE MPs dose = 5 mg·L<sup>-1</sup>, pH=7.0, ionic strength 10 mM, 25 °C, pH 7.0.)