Supporting Information

First Cycloruthenation of 2-Alkenylpyridines: Synthesis,

Characterization and Properties

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Empirical formula	$C_{33}H_{26}F_6N_5PRu\cdot 2CH_3OH$	
Formula weight	802.71	
Temperature	293(2) K	
Wavelength	0.71073 Å	
Crystal system	triclinic	
Space group	P-1	
Unit cell dimensions	<i>a</i> = 7.5639(4) Å	<i>α</i> =99.165(3)°
	<i>b</i> = 12.0236(5) Å	β=93.603(3)°
	<i>c</i> = 19.0129(9) Å	<i>γ</i> = 91.597(3)°
Volume	1702.43(14) Å ³	
Ζ	2	
Density (calculated)	1.566 Mg/m ³	
Absorption coefficient	0.581 mm ⁻¹	
F(000)	816.0	
20 range for data collection	3.434 to 49.992	
Index ranges	$-8 \le h \le 8, 14 \le k \le 14, 22 \le l \le 22$	
Reflections collected	29410	
Independent reflections	29410 [R _{int} = -, R _{sigma} = 0.0866]	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	29410/0/416	
Goodness-of-fit on F ²	1.090	
Final R indices [I>2 σ (I)]	$R_1 = 0.0671, wR_2 = 0.1864$	
R indices (all data)	$R_1 = 0.0889, wR_2 = 0.2026$	
Largest diff. peak and hole	0.857/-0.770 eÅ ⁻³	

Table S1. Crystal data and structure refinement for $2 \cdot 2CH_3OH$.



Fig. S2¹ HNMR spectrum of (E)-2-(2-(thiophen-2-yl)vinyl)pyridine in CDCl₃.



Fig. S4¹ HNMR spectrum of (E)-N,N-dimethyl-4-(2-(pyridin-2-yl)vinyl)aniline in CDCl₃.









Fig. S6¹ HNMR spectrum of complex **2** in CD₃CN. Inset: The MS spectrum of complex **2**.









Fig. S8¹ HNMR spectrum of complex **4** in CD₃CN. Inset: The MS spectrum of complex **4**.



Fig. S10¹³ CNMR spectrum of complex 1 in CD₃CN.



Fig. S12¹³ CNMR spectrum of complex **3** in CD₃CN.



Fig. S14¹³ CNMR spectrum of complex **5** in d₆-DMSO.



Fig. S15 FT-IR spectra of complexes 1–5 in KBr pellets.



Fig. S16 Absorption spectra of 2-alkenylpyridine ligands (20 µM) of complexes 1–5 in CH₃CN.



Fig. S17 Emission spectra of complexes 1–5 in 2-methyltetrahydrofuran glasses at 77 K ($\lambda_{ex} = 500$ nm).



Fig. S18 Cyclic voltammetric responses for complexes 1-5 (1 ×10⁻⁴ M) dissolved in acetonitrile containing 0.1 M [*n*-Bu₄N]PF₆.



Fig. S19 Absorption spectral changes of complexes 1-4 (20 μ M) after being incubated for 4 h in ethanol/Briton–Robinson buffer solutions (v/v=1:2, pH 11.95–1.82), respectively. Arrows show spectral changes upon decreasing pH. Inset: Changes of absorption intensity at different pHs.



Fig. S20 Absorption spectral changes of complexes (20 μ M) after being incubated for 4 h in CH₃CN/ethanol/Briton-Robinson buffer solutions (v/v/v = 2:8:90), respectively. Arrows show spectral changes upon decreasing pH. Inset: Changes of absorption intensity at different pHs. (a) pH=11.95-1.82; (b)pH = 11.95-1.82 (c) pH=11.02-1.91.



Fig. S21 MS spectrum of complex 1 incubated in CH₃OH/HAc/H₂O (v/v/v=1:1:1) for 4 h.



Fig. S22 MS spectrum of complex **2** incubated in CH₃OH/HAc/H₂O (v/v/v=1:1:1) for 4 h.



Fig. S23 MS spectrum of complex **3** incubated in CH₃OH/HAc/H₂O (v/v/v=1:1:1) for 4 h.



Fig. S24 MS spectrum of complex 4 incubated in CH₃OH/HAc/H₂O (v/v/v=1:1:1) for 4 h.



Fig. S25 MS spectrum of complex **5** incubated in CH₃OH/HAc/H₂O (v/v/v=1:1:1) for 4 h.



Fig. S26 MS spectrum of complex 1 incubated in CH₃OH/HAc/H₂O (v/v/v=1:1:1) for 20 h



Fig. S27 MS spectrum of complex **2** incubated in CH₃OH/HAc/H₂O (v/v/v=1:1:1) for 20 h.



Fig. S28 MS spectrum of complex **3** incubated in CH₃OH/HAc/H₂O (v/v/v=1:1:1) for 20 h.



Fig. S29 MS spectrum of complex 4 incubated in CH₃OH/HAc/H₂O (v/v/v=1:1:1) for 20 h.



Fig. S30 MS spectrum of complex **5** incubated in CH₃OH/HAc/H₂O (v/v/v=1:1:1) for 20 h.



Fig. S31 MS spectrum of complex 1 incubated in $CH_3OH/HCl/H_2O$ (v/v =1:1:1) for 20 h.



Fig. S32 MS spectrum of complex 2 incubated in $CH_3OH/HCl/H_2O$ (v/v =1:1:1) for 20 h



Fig. S33 MS spectrum of complex **3** incubated in CH₃OH/HCl/H₂O (v/v =1:1:1) for 20 h



Fig. S34 MS spectrum of complex 4 incubated in $CH_3OH/HCl/H_2O$ (v/v =1:1:1) for 20 h



Fig. S35 MS spectrum of complex **5** incubated in CH₃OH/HCl/H₂O (v/v =1:1:1) for 20 h



Fig. S36 MS spectrum of Ru(bpy)₂Cl₂ with ligand of complex **5** in CH₃OH treated with AgBF₄ and then NaOH.