Construction of Activated Carbons-Supported B₃N₃ Doped Carbons as Metal-free Catalyst for Dehydrochlorination of 1,2-Dichloroethane to Produce Vinyl Chloride

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*Corresponding author: Tel:+86 021 64250643 E-mail: qihm@ecust.edu.cn. (H. Qi), jiangb@mail.sioc.ac.cn. (B. Jiang) The FTIR spectra was shown in Figure S1 (a), the absorption bands at 3290 cm⁻¹ and 2110 cm⁻¹ are the C=C-H and -Ph-C=CH stretch vibrations, respectively. The band at 2180 cm⁻¹ is the -B-C=C-Ph- stretch vibration¹. The typical absorption bands at 3430 cm⁻¹ and 1459 cm⁻¹ are assigned to the N-H and B-N stretching vibration, respectively. And an out-of-plane bending vibration at 684 cm⁻¹ of B₃N₃ ring is also observed. The ¹H NMR spectra was shown in Figure S1 (b), the aromatic and ethynyl protons appear as multiple at 7.20-7.80 and 3.11 ppm², respectively. And the characteristic peak at 5.60 ppm was assigned to the NH- on B₃N₃.



Figure S1. FTIR spectra (a) and ¹H NMR spectra of PBSZ.



Figure S2. (a) the conversion of 1,2-DCE over the B, N-ACs catalysts; (b) the selectivity of VCM over the B, N-ACs catalysts. Reaction conditions: temperature= 250° C, 0.1 MPa, LHSV (1,2-DCE) = 0.67 h⁻¹.



Figure S3. AFM images of PBSZ precursor after carbonized at 700°C.



Figure S4. SEM micrograph of (a)(b) fresh PBSZ-AC-1 catalyst and (c)(d) used PBSZ-AC-1 catalyst; (e) elemental distribution mapping of C, N, B in fresh PBSZ-AC-1



Figure S5. SEM micrograph of (a)(b) fresh PBSZ-AC-2 catalyst and (c)(d) used PBSZ-AC-2 catalyst; (e) elemental distribution mapping of C, N, B in fresh PBSZ-AC-2



Figure S6. SEM micrograph of (a)(c) fresh PBSZ-AC-3 catalyst and (b)(d) used PBSZ-AC-3 catalyst; (e) elemental distribution mapping of C, N, B in fresh PBSZ-AC-3



Figure S7. TEM images of (a) PBSZ-AC-1, (b) PBSZ-AC-2 and (c) PBSZ-AC-3 catalyst.



Figure S8. N₂ sorption isotherm and pore size distributions of the used B, N-ACs

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