

Electronic Supplementary information

Highly crystalline anatase TiO₂ nanocuboids as an efficient photocatalyst for hydrogen generation

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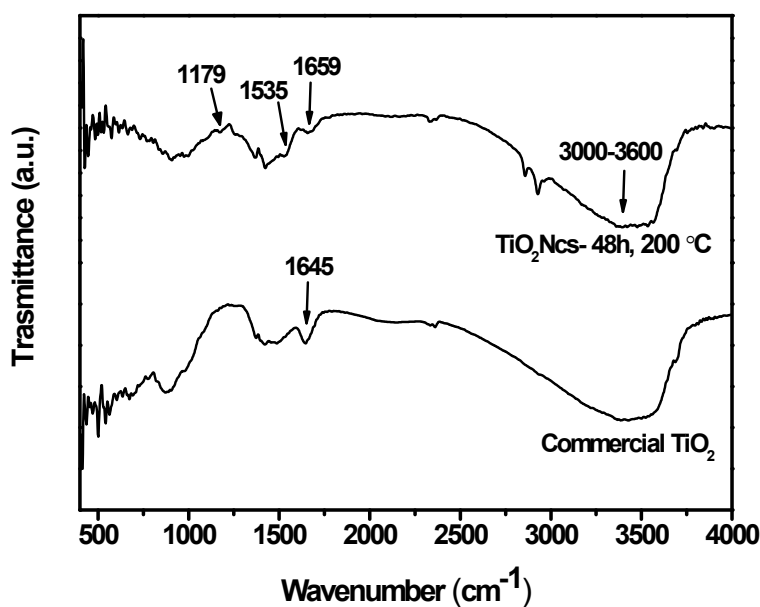


Figure S1. FTIR spectra of commercial TiO₂ and TiO₂ nanocuboids prepared at 200 °C for 48 h

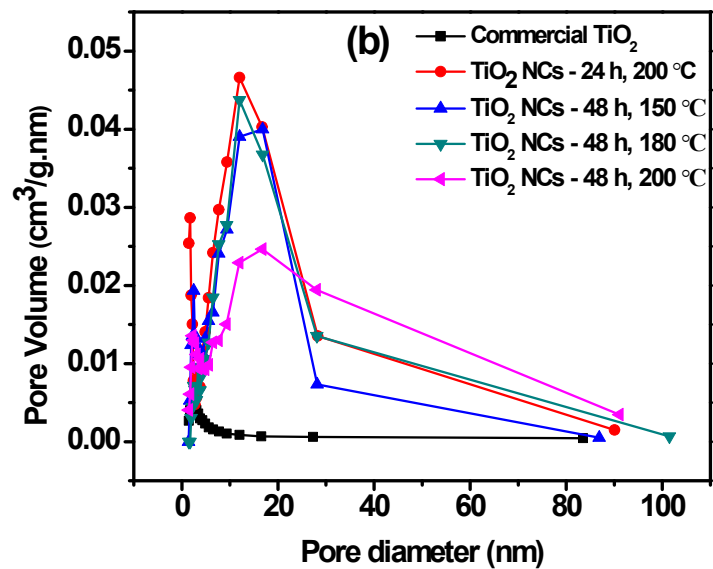
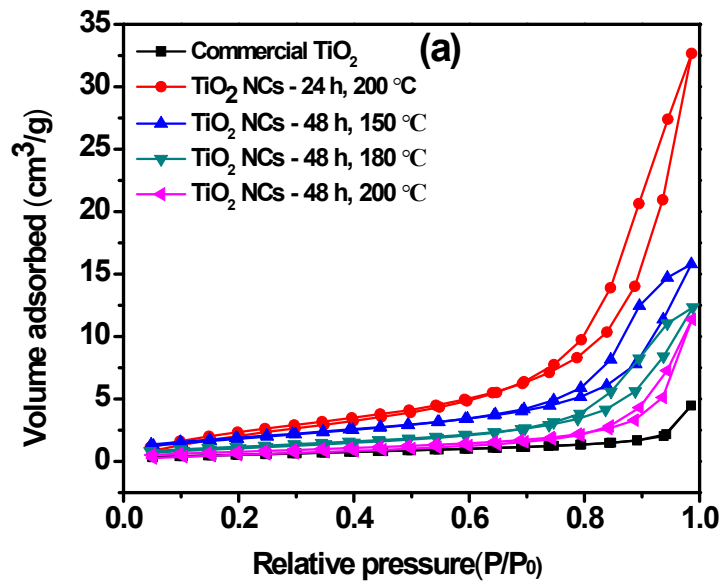
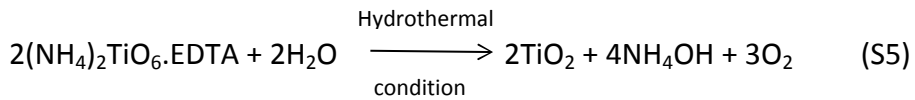
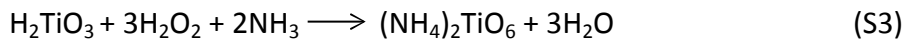
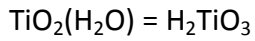
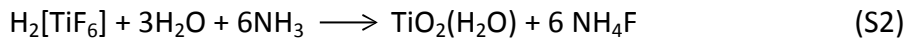


Figure S2. (a) N_2 adsorption-desorption isotherms and (b) corresponding pore size distribution curves of commercial TiO_2 and as prepared TiO_2 nanocuboids

Possible reactions involved in the formation of TiO₂ nanocuboids



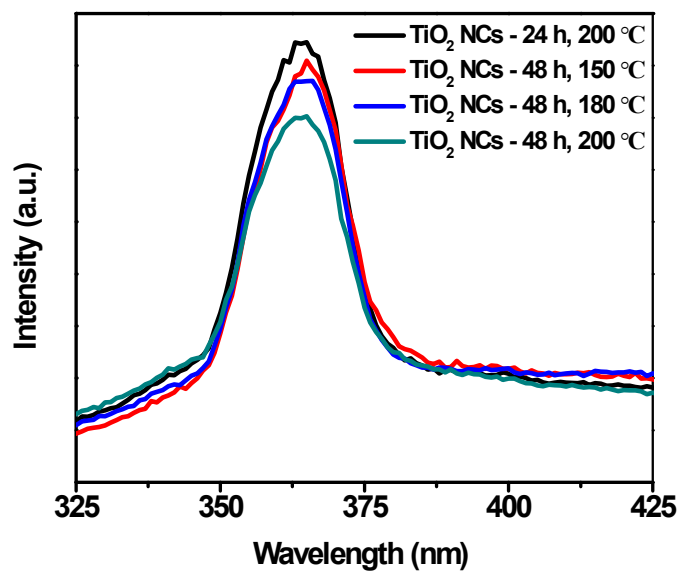


Figure S3. Photoluminescence spectra of TiO₂ nanocuboids prepared under different reaction time and temperature conditions

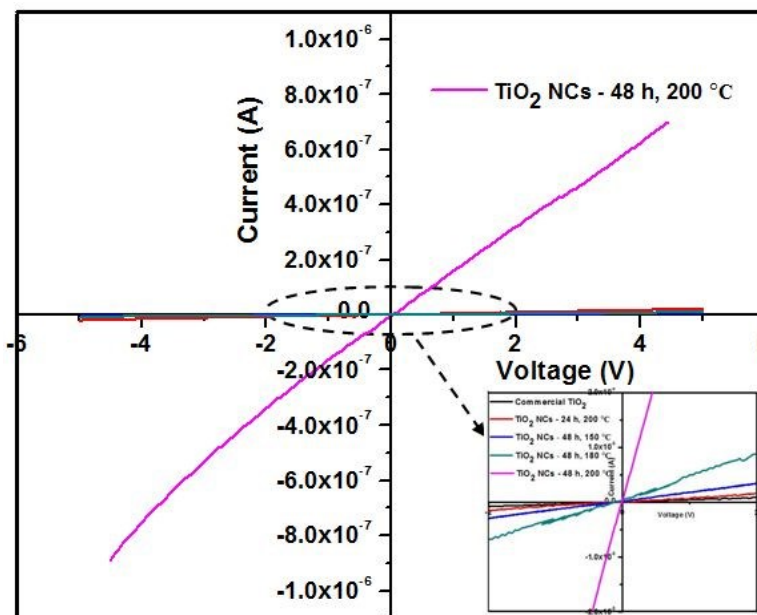


Figure S4. I-V plots of commercial TiO₂ and as prepared TiO₂ nanocuboids with illumination (1.5AM) of light.

Table S1. Specific surface area, pore volume and average pore size of commercial TiO₂ and as prepared TiO₂ nanocuboids

Sample	Specific surface area (m² g⁻¹)	Pore volume (cm³ g⁻¹)	Average pore radius (nm)
Commercial TiO ₂	29.3	0.038	6.64
TiO ₂ -NCs 24 h, 200 °C	122.9	0.459	9.91
TiO ₂ -NCs 48 h, 150 °C	111.6	0.346	6.99
TiO ₂ -NCs 48 h, 180 °C	103.2	0.402	8.85
TiO ₂ -NCs 48 h, 200 °C	85.7	0.495	12.04

Table S2. Comparison of the photocatalytic H₂ generation activity of anatase TiO₂ nanostructures reported by various researchers with as prepared anatase TiO₂ nanocuboids.

Sr. No.	Photocatalyst	Amount (g)	Light Source	H ₂ generation (μmol h ⁻¹ g ⁻¹)	Reference no.
1	TiO ₂ nanocuboids	0.02	400W Hg lamp	3866.44	This work
2	TiO ₂ (3D-flowerlike nanosheets)	1	450W Hg lamp	117.6	1
3	TiO ₂ (nanosheets)	0.1	300W Xe lamp	2036	2
4	TiO ₂ (cube-like morphology)	0.05	300W Xe lamp	104.54	3
5	TiO ₂ (twinned nanocrystals)	0.05	300W Xe lamp	1272.66	3
6	TiO ₂ (mesoporous nanofibers)	0.1	300W Xe lamp	231.7	4
7	TiO ₂ (hollow spheres)	0.015	300W Xe lamp	21.2	5

References:

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