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ESI for

Synergetic catalysis of Se and Cu allowing diethoxylation of halomethylene ketones using O_2 as the mild oxidant

Zhicheng Cao,^{ab} Xin Deng,^a Chao Chen,^a Yonghong Liu,^{*a} Lei Yu,^{*a} and Xuefeng Jiang^{*b}

^a School of Chemistry and Chemical Engineering, Yangzhou University, Yangzhou, Jiangsu

225002, China. E-mails: <u>yhliu@yzu.edu.cn</u>(Y. Liu), <u>yulei@yzu.edu.cn</u> (L. Yu).

^b Shanghai Key Laboratory of Green Chemistry and Chemical Process, Department of Chemistry,

East China Normal University, Shanghai 200062, China. E-mail: xfjiang@chem.ecnu.edu.cn.

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1. Experimental details

General: Reagents were purchased from Sinopharm Chemical Reagent and Energy Chemical with their purities more than 98% and were directly used as received. Solvents were analytical pure (AR) and directly used without any special treatment. Melting points were measured by WRS-2A digital instrument. IR spectra were determined on Bruker Tensor 27 Infrared spectrometer. NMR spectra were recorded on a Bruker Avance 400 instrument (400 MHz for ¹H NMR and ¹³C NMR) using CDCl₃ as the solvent and Me₄Si as the internal standard. Chemical shifts for ¹H NMR were referred to internal Me₄Si (0 ppm) and *J*-values were shown in Hz. Mass spectra were determined on a Thermo Trace DSQ II or a Shimadzu GCMS-QP2010 Ultra spectrometer (EI). Elemental analysis was performed on an Elementar Vario EL cube instrument.

General procedure for the synthesis of α -carbonyl acetals: To a 100 mL reaction tube, a piece of magnetic bar, 1 mmol of bromomethylene ketone 1, 5 mol% of PhSe(O)OH (vs. 1), 2 mol% of Cu(OAc)₂:H₂O (vs. 1) and 3 mL of EtOH were added. The tube was then charged with O₂ and sealed. After being magnetically stirred at 100 °C for 24 h, the solvent was removed by evaporation under reduced pressure on a rotary evaporator. The residue was separated by preparative thin-layer chromatography (silica plate with eluent petroleum ether/EtOAc = 40/1) to afford the related product **2**.

Synthesis of 2a at kilogram scale: To a 10 L reaction kettle equipped with water condenser, 1.2 kg of 1a (10 mol), 94.5 g of PhSe(O)OH, 40 g of Cu(OAc)₂·H₂O and 3 L EtOH were added. Oxygen was then introduced into the system via pipe below the liquid level of the system at the velocity of 0.9-1.0 cm³/s. The reaction system was mechanically stirred (500-550 rpm) and heated to reflux for 48 h. Excess EtOH in the system was evaporated and recycled, while the residue was then separated via distillation under reduced pressure (130-136 °C/10 mmHg) to give 1.58 kg of 2a (in ca 76% yield).

2. Detailed conditional optimizations

 Table S1 Condition optimizations^a



Entry	Cat. (/%) ^b	T/ºC; t/h	2a /% ^c
1	PhSe(O)OH (5)	130; 48	25 ^d
2	$Cu(OAc)_2(2)$	130; 24	32 ^d
3	PhSe(O)OH (2), $Cu(OAc)_2$ (2)	130; 7	57 ^d
4	PhSe(O)OH (5), $Cu(OAc)_2$ (2)	130; 7	76
5	PhSe(O)OH (10), Cu(OAc) ₂ (2)	130; 7	78
6	PhSe(O)OH (5), $Cu(OAc)_2$ (2)	100; 7	50^d
7	PhSe(O)OH (5), Cu(OAc) ₂ (2)	100; 16	71^d
8	PhSe(O)OH (5), $Cu(OAc)_2$ (2)	100; 24	84
9	PhSe(O)OH (5), Cu(OAc) ₂ (2)	100; 32	80
10	PhSe(O)OH (5), $Cu(OAc)_2$ (2)	90; 24	76^d
11	PhSe(O)OH (5), Cu(OAc) ₂ (2)	90; 30	80^d
12	PhSe(O)OH (5), $Cu(OAc)_2$ (2)	80; 24	71^d
13	PhSe(O)OH (5), $Cu(OAc)_2$ (2)	70; 24	63 ^{<i>d</i>}
14	PhSe(O)OH (5), $Cu(OAc)_2$ (2)	60; 24	45^d
15	3,5-(CF ₃) ₂ C ₆ H ₃ Se(O)OH (5), Cu(OAc) ₂ (2)	100; 24	76
16	4-MeOC ₆ H ₄ Se(O)OH (5), Cu(OAc) ₂ (2)	100; 24	58 ^d
17	$(PhSe)_2$ (2.5), $Cu(OAc)_2$ (2)	100; 24	78
18	$(PhSe)_{2}(5), Cu(OAc)_{2}(2)$	100; 24	81
19	Se powder (5), $Cu(OAc)_2$ (2)	100; 24	38 ^d
20	$SeO_{2}(5), Cu(OAc)_{2}(2)$	100; 24	48^d
21	$Na_2SeO_3(5), Cu(OAc)_2(2)$	100; 24	45^d
22	PhSe(O)OH (5), CuCl ₂ (2)	100; 24	76
23	PhSe(O)OH (5), CuBr ₂ (2)	100; 24	74
24	PhSe(O)OH (5), CuBr (2)	100; 24	79
25	PhSe(O)OH (5), CuO (2)	100; 24	70^d

26	PhSe(O)OH (5), Cu_2O (2)	100; 24	67 ^{<i>d</i>}
27	PhSe(O)OH (5), Cu powder (2)	100; 24	36 ^d

^{*a*} 1 mmol of **1a** and 3 mL of alcohol were employed; The reaction was performed in a sealed tube. ^{*b*} Catalyst molar ratios based on **1a** amount inside the parentheses. ^{*c*} Isolated yield based on **1a**. ^{*d*} Reaction uncompleted.

3. Characterization data of the products



2,2-Diethoxy-1-phenylethan-1-one (2a): Oil, 84% yield, 174.9 mg; IR (film): 3060, 2977, 2929, 2882, 1685, 1625, 1596, 1467, 1386, 1294, 1115, 1060 cm⁻¹; ¹H NMR (400 MHz, CDCl₃, TMS, ppm): δ 8.09 (d, *J* = 8.4 Hz, 2H), 7.47 (t, *J* = 7.9 Hz, 1H), 7.36 (t, *J* = 7.8 Hz, 2H), 5.20 (s, 1H), 3.75–3.64 (m, 2H), 3.63–3.53 (m, 2H), 1.17 (t, *J* = 7.7 Hz, 6H); ¹³C NMR (100 MHz, CDCl₃, TMS, ppm): δ 193.88, 133.73, 133.34, 129.63, 128.24, 102.44, 63.15, 15.12; MS (EI, 70 eV) m/z (%) 208 (26) [M⁺], 105 (100); Known compound.¹



2,2-Diethoxy-1-(*p***-tolyl)ethen-1-one (2b):** Oil, 81% yield, 180.0 mg; IR (film): 3034, 2980, 2930, 2883, 1686, 1608, 1572, 1446, 1311, 1293, 1114, 1064 cm⁻¹; ¹H NMR (400 MHz, CDCl₃, TMS, ppm): δ 7.77 (d, *J* = 8.0 Hz, 2H), 6.94 (d, *J* = 8.0 Hz, 2H), 4.97 (s, 1H), 3.45 (dt, *J* = 14.3, 7.1 Hz, 2H), 3.35 (p, *J* = 7.1 Hz, 2H), 2.09 (s, 3H), 0.94 (t, *J* = 7.1 Hz, 6H). ¹³C NMR (100 MHz, CDCl₃, TMS, ppm): δ 193.69, 144.40, 131.26, 129.83, 129.08, 102.23, 63.03, 21.77, 15.22; MS (EI, 70 eV) m/z (%) 222 (21) [M⁺], 119 (100); Known compound.¹



2,2-Diethoxy-1-(*m***-tolyl)ethenone (2c):** Oil; 83% yield, 184.5 mg; IR (KBr film): 3035, 2978, 2928, 2883, 1693, 1595, 1447, 1381, 1292, 1255, 1120, 1062, 903, 794, 737, 652, 490 cm⁻¹; ¹H NMR (400 MHz, CDCl₃, TMS, ppm): δ 7.87 (s, 1H), 7.89 (d, *J* = 8.0 Hz, 1H), 7.32–7.24 (m, 2H), 5.23 (s, 1H), 3.71–3.54 (m, 4H), 2.33 (s, 3H), 1.17 (t, *J* = 8.0 Hz, 6H); ¹³C NMR (100 MHz, CDCl₃, TMS, ppm): δ 194.18, 138.16, 134.32, 133.86, 130.00, 128.26, 126.98, 101.99, 63.01, 21.38, 15.23; MS (EI, 70 eV) m/z (%) 222 (20) [M⁺], 119 (100); Known compound.¹



2,2-Diethoxy-1-(*o***-tolyl)ethenone (2d):** Oil; 84%, 186.7 mg; IR (film): 3065, 2975, 2927, 1697, 1600, 1570, 1452, 1381, 1294, 1226, 1114, 1062, 911, 735, 647, 476 cm⁻¹; ¹H NMR (400 MHz, CDCl₃, TMS, ppm): δ 7.92 (d, *J* = 8.0 Hz, 1H), 7.32 (dd, *J*₁ = 4.0 Hz, *J*₂ = 8.0 Hz, 1H), 7.20–7.17 (m, 2H), 5.15 (s, 1H), 3.71–3.54 (m, 4H), 2.45 (s, 3H), 1.16 (t, *J* = 6.0 Hz, 6H); ¹³C NMR (100 MHz, CDCl₃, TMS, ppm): δ 197.34, 139.83, 134.16, 131.85, 130.15, 125.33, 102.31, 63.07, 21.48, 15.21; MS (EI, 70 eV) m/z (%) 222 (18) [M⁺], 119 (100); Known compound.¹



2,2-Diethoxy-1-(3-methoxyphenyl)ethenone (2e): Oil, 50% yield, 119.1 mg; IR (film): 3077, 2976, 2933, 2883, 1688, 1596, 1582, 1486, 1432, 1324, 1288, 1115, 1060 cm⁻¹; ¹H NMR (400 MHz, CDCl₃, TMS, ppm): δ 7.71 (d, *J* = 8.0 Hz, 1H), 7.59 (s, 1H), 7.29 (dd, *J*₁ = 8.0 Hz, *J*₂ = 8.0 Hz 1H), 7.05 (d, *J* = 8.0 Hz, 1H), 5.22 (s, 1H), 3.78 (s, 3H), 3.73–3.51 (m, 4H), 1.18 (t, *J* = 8.0 Hz, 6H); ¹³C NMR (100 MHz, CDCl₃, TMS,

ppm): δ 193.83, 159.55, 135.07, 129.39, 122.48, 120.23, 113.59, 102.11, 63.11, 55.44, 15.25; MS (EI, 70 eV) m/z (%) 238 (9) [M⁺], 135 (100); Known compound.¹



2,2-Diethoxy-1-(4-chlorophenyl)ethan-1-one (2f): Oil, 80%, 194.2 mg; IR (KBr film): 3075, 2980, 2932, 2884, 1694, 1590, 1570, 1489, 1402, 1288, 1118, 1092, 1063, 1016 cm⁻¹; ¹H NMR (400 MHz, CDCl₃, TMS, ppm): δ 8.10 (d, *J* = 8.0 Hz, 2H), 7.40 (d, *J* = 8.0 Hz, 2H), 5.15 (s, 1H), 3.78–3.57 (m, 4H), 1.21 (d, *J* = 8.0 Hz, 6H). ¹³C NMR (100 MHz, CDCl₃, TMS, ppm): δ 193.05, 139.90, 131.92, 131.34, 128.65, 103.20, 63.59, 15.19; MS (EI, 70 eV) m/z (%) 242 (8) [M⁺], 139 (100); Known compound.¹



2,2-Diethoxy-1-(3-chlorophenyl)ethan-1-one (2g): Oil, 83%, 201.4 mg; IR (KBr film): 3070, 2981, 2929, 2884, 1688, 1590, 1572, 1485, 1403, 1289, 1114, 1092, 1065, 1011 cm⁻¹; ¹H NMR (400 MHz, CDCl₃, TMS, ppm): δ 8.05 (s, 1H), 7.98 (d, *J* = 7.9 Hz, 1H), 7.43 (d, *J* = 8.0 Hz, 1H), 7.29 (t, *J* = 7.9 Hz, 1H), 5.10 (s, 1H), 3.74–3.66 (m, 2H), 3.61–3.51 (m, 2H), 1.16 (t, *J* = 7.1 Hz, 6H). ¹³C NMR (100 MHz, CDCl₃, TMS, ppm): δ 192.68, 135.14, 134.39, 133.17, 129.65, 127.91, 102.92, 63.51, 15.09; MS (EI, 70 eV) m/z (%) 242 (8) [M⁺], 139 (100); Known compound.¹



1-(4-Bromophenyl)-2,2-diethoxyethanone (2h): Oil, 80%, 229.8 mg; IR (KBr film): 2978, 2931, 2883, 1693, 1586, 1484, 1400, 1287, 1118, 1071, 1012 cm⁻¹; ¹H NMR (400 MHz, CDCl₃, TMS, ppm): δ 8.02 (d, *J* = 8.4 Hz, 2H), 7.56 (d, *J* = 8.4 Hz, 2H), 5.14 (s, 1H), 3.81–3.68 (m, 2H), 3.67–3.55 (m, 2H), 1.22 (t, *J* = 7.1 Hz, 6H). ¹³C NMR (100

MHz, CDCl₃, TMS, ppm): δ 193.19, 131.60, 131.38, 130.38, 128.70, 103.14, 63.55, 15.16; MS (EI, 70 eV) m/z (%) 286 (5) [M⁺], 183 (100); Known compound.¹



2,2-Diethoxy-1-(4-nitrophenyl)ethan-1-one (2i): Oil, 69%, 174.7 mg; IR (KBr film): 3108, 3075, 3050, 2978, 2939, 2902, 1701, 1603, 1526, 1480, 1371, 1345, 1278, 1115, 1053, 1014 cm⁻¹; ¹H NMR (400 MHz, CDCl₃, TMS, ppm): δ 8.29 (d, *J* = 8.6 Hz, 2H), 8.22 (d, *J* = 8.6 Hz, 2H), 5.10 (s, 1H), 3.82–3.72 (m, 2H), 3.65–3.56 (m, 2H), 1.20 (t, *J* = 7.1 Hz, 6H). ¹³C NMR (100 MHz, CDCl₃, TMS, ppm): δ 192.72, 150.38, 138.08, 131.09, 123.36, 103.85, 64.19, 15.19; MS (EI, 70 eV) m/z (%) 253 (11) [M⁺], 105 (100); Known compound.¹



2,2-Diethoxy-1-(1H-indol-3-yl)ethenone (2j): Yellow soild, m. p.: 131-132 °C, 56%, 138.4 mg; IR (KBr film): 3423, 3197, 2978, 2888, 1635, 1513, 1416, 1305, 1246, 1160, 1105, 1058, 956, 914, 805, 753, 634, 592, 523 cm⁻¹; ¹H NMR (400 MHz, CDCl₃, TMS, ppm): δ 9.08 (s, 1H), 8.43 (d, *J* = 8.0 Hz, 1H), 8.34 (d, *J* = 3.0 Hz, 1H), 7.42 (d, *J* = 8.0 Hz, 1H), 7.31–7.25 (m, 2H), 5.12 (s, 1H), 3.84–3.58 (m, 4H), 1.24 (t, *J* = 8.0 Hz, 6H); ¹³C NMR (100 MHz, CDCl₃, TMS, ppm): δ 190.94, 135.77, 134.33, 126.32, 123.69, 122.79, 122.41, 113.46, 111.37, 103.44, 62.89, 15.24; MS (EI, 70 eV) m/z (%) 247 (10) [M⁺], 116 (100); Anal. Calcd for C₁₄H₁₇NO₃: C, 68.00; H, 6.93. Found: C, 68.05; H, 6.88.



2,2-Diethoxy-1-(naphthalen-2-yl)ethan-1-one (2k): Oil; 57%, 147.2 mg; IR (KBr film): 3070, 2980, 2932, 2889, 1689, 1598, 1580, 1449, 1396, 1285, 1229, 1117, 1063 cm⁻¹; ¹H NMR (400 MHz, CDCl₃, TMS, ppm): δ 8.77 (s, 1H), 8.14 (dd, J = 8.6, 1.6 Hz, 1H), 7.96 (d, J = 8.1 Hz, 1H), 7.84 (t, J = 8.8 Hz, 2H), 7.61–7.49 (m, 2H), 5.40 (s, 1H), 3.82–3.55 (m, 4H), 1.25 (t, J = 7.1 Hz, 6H); ¹³C NMR (100 MHz, CDCl₃, TMS, ppm): δ 194.09, 135.81, 132.45, 132.02, 131.08, 129.92, 128.64, 128.09, 127.72, 126.57, 124.94, 102.56, 63.22, 15.25; MS (EI, 70 eV) m/z (%) 258 (16) [M⁺], 155 (100); Known compound.¹

References

C. Chen, Z. Cao, X. Zhang, Y. Li, L. Yu and X. Jiang, *Chin. J. Chem.*, 2020, 38, 1045.

4. NMR spectra of products





230 220 210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 f1 (ppm)







¹H NMR spectrum of **2c**



¹³C NMR spectrum of **2c**









¹H NMR spectrum of 2e



¹³C NMR spectrum of **2e**



¹H NMR spectrum of **2f**



















¹H NMR spectrum of **2**j



¹³C NMR spectrum of **2**j



¹H NMR spectrum of **2**k



¹³C NMR spectrum of **2**k

