

Supporting Information

Asymmetric total synthesis of (+)-xestoquinone and (+)-adociaquinones A and B

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General experimental procedures

All reactions were carried out under an inert nitrogen atmosphere with dry solvents under anhydrous conditions unless otherwise noted. Anhydrous dichloromethane, tetrahydrofuran and toluene were purified by the PS-MD-5 (Innovative Technology) solvent purification system. Dimethyl sulfoxide used for IBX oxidation was purchased from commercially available anhydrous solvent. Anhydrous diethyl ether was distilled from sodium. The solvents used for condition screening of desymmetric intramolecular Michael addition are all commercially available analytical-grade solvents. TLC analyses were performed on EMD 250 μm Silica Gel HSGF₂₅₄ plates and visualized by quenching of UV fluorescence ($\lambda_{\text{max}} = 254 \text{ nm}$), or by staining phosphomolybdic acid, or potassium permanganate. Flash column chromatography was performed as described by Still^[1], employing SiliCycle UltraPure Silica Gels: SilicaFlash[®] P60 40 – 63 μm (230 – 400 mesh). ¹H and ¹³C NMR spectra were recorded on a Bruker-500, 400 spectrometers. Chemical shifts for ¹H and ¹³C NMR spectra are reported in ppm (δ) relative to residue protium in the solvent (CDCl₃: δ 7.26, 77.0 ppm; DMSO: δ 2.50, 40.4 ppm) and the multiplicities are presented as follows: s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, br = broad. High-resolution mass spectra (HRMS) were acquired on Waters Micromass GCT Premier or Bruker Daltonics, Inc. APEXIII 7.0 TESLA FTMS. Mass spectra were acquired on Agilent 5975C. Infrared (IR) spectra was obtained using a Shimadzu IRTracer-100 fourier transform infrared spectroscopy (FTIR). The $[\alpha]_{\text{D}}^{20}$ was recorded at 365 nm using Anton Paar MCP 5500.

General experimental procedure A for desymmetric intramolecular Michael addition without AcOH

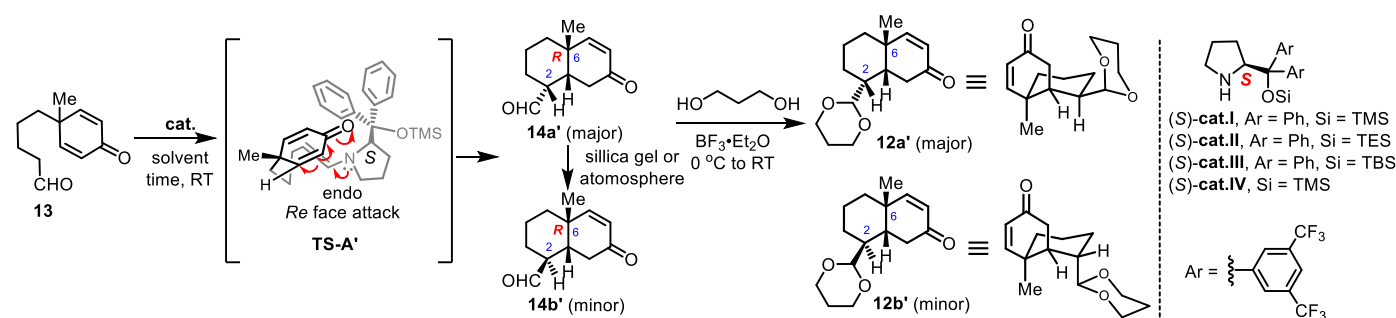
A solution of **13**, catalyst in analytical-grade solvent was stirred at room temperature for indicated time without inert nitrogen atmosphere. Then the solution was evacuated and purified by flash chromatography or preparation lamella chromatography (20% ethyl acetate – petroleum ether).

General experimental procedure B for desymmetric intramolecular Michael addition with AcOH

A solution of **13**, catalyst and additive AcOH in analytical-grade solvent was stirred at room temperature for indicated time without inert nitrogen atmosphere. Then the solution was quenched with saturated NaHCO₃ and extracted with ethyl acetate. The combined organic layers were washed with brine, dried over anhydrous sodium sulfate, concentrated and purified by silica gel flash chromatography or preparation lamella chromatography (20% ethyl acetate – petroleum ether).

Detail of the screening conditions of desymmetric intramolecular Michael addition

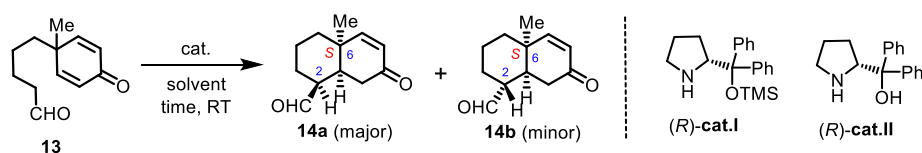
Table S1. Screening conditions of desymmetric intramolecular Michael addition to obtain **14'** using *S*-catalysts.



entry	cat. (equiv.)	solvent	time	yield/d.r. at C2 ^a	e.e. ^b
1 ^c	Et ₂ NH (10.0)	MeOH/H ₂ O (20:1)	6.0 h	64%	rac.
2	(<i>S</i>)-cat.I (0.5)	MeOH/H ₂ O (20:1)	10.5 h	48% / 3.0:1	14a' : -87%; 14b' : -31%
3	(<i>S</i>)-cat.II (0.5)	MeOH/H ₂ O (20:1)	10.5 h	56% / 2.8:1	14a' : -85%; 14b' : -5%
4	(<i>S</i>)-cat.III (0.5)	MeOH/H ₂ O (20:1)	48.0 h	trace	14a' : -80%; 14b' : -15%
5	(<i>S</i>)-cat.IV (0.5)	MeOH/H ₂ O (20:1)	48.0 h	trace	14a' : -84%; 14b' : -6%
6	(<i>S</i>)-cat.I (0.5)	MeOH	2.0 h	63% / 5.0:1	14a' : -85%; 14b' : -37%
7	(<i>S</i>)-cat.I (0.5)	DCM	24.0 h	51% / 1.1:1	14a' : -93%; 14b' : -86%
8	(<i>S</i>)-cat.I (0.5)	Et ₂ O	24.0 h	50% / 2.8:1	14a' : -96%; 14b' : -93%
9	(<i>S</i>)-cat.I (0.5)	MeCN	24.0 h	47% / 1.0:1	14a' : -93%; 14b' : -88%
10	(<i>S</i>)-cat.I (0.5)	toluene	5.5 h	42% / 22.0:1	14a' : -97%; 14b' : -89%
11 ^c	(<i>S</i>)-cat.I (0.5)	toluene	11.0 h	52% / 4.5:1	14a' : -97%; 14b' : -89%
12 ^d	(<i>S</i>)-cat.I (0.3)	toluene	16.0 h	73% ^e / 1.7:1 ^f	14a' : -97%; 14b' : -94%
13 ^g	(<i>S</i>)-cat.I (0.2)	toluene	16.0 h	86% ^g / 2.0:1 ^f	14a' : -96%; 14b' : -89%
14 ^h	(<i>S</i>)-cat.I (0.4)	toluene	12.0 h	62% ⁱ / 14.0:1 ^j	14a' : -96%; 14b' : -89%

All reactions were performed using **13** (5.8 mg, 0.03 mmol, 1.0 equiv., 0.05 M) and catalyst at room temperature in analytical-grade solvents, unless otherwise noted. ^aThe yields and diastereoisomeric ratios (d.r.) were determined from the crude ¹H NMR spectrum of **14'** using CH₂Br₂ as an internal standard, unless otherwise noted. ^bThe enantiomeric excess (e.e.) values were determined by chiral high-performance liquid chromatography (Chiralpak IG-H). ^cCompound **13**: 9.6 mg, 0.05 mmol, 0.1 M. ^dCompound **13**: 96 mg, 0.5 mmol, 0.1 M. ^eIsolated yield of **14a'+14b'**. ^fThe d.r. values were determined from the ¹H NMR spectrum of purified **14'** after purification by silica gel column chromatography. ^gCompound **13**: 1.0 g, 5.2 mmol, 0.1 M. ^hCompound **13**: 288 mg, 1.5 mmol, 0.1 M. ⁱIsolated yield of **12a'+12b'**. ^jThe d.r. values were determined from the crude ¹H NMR spectrum of **12'** obtained from the one-pot process.

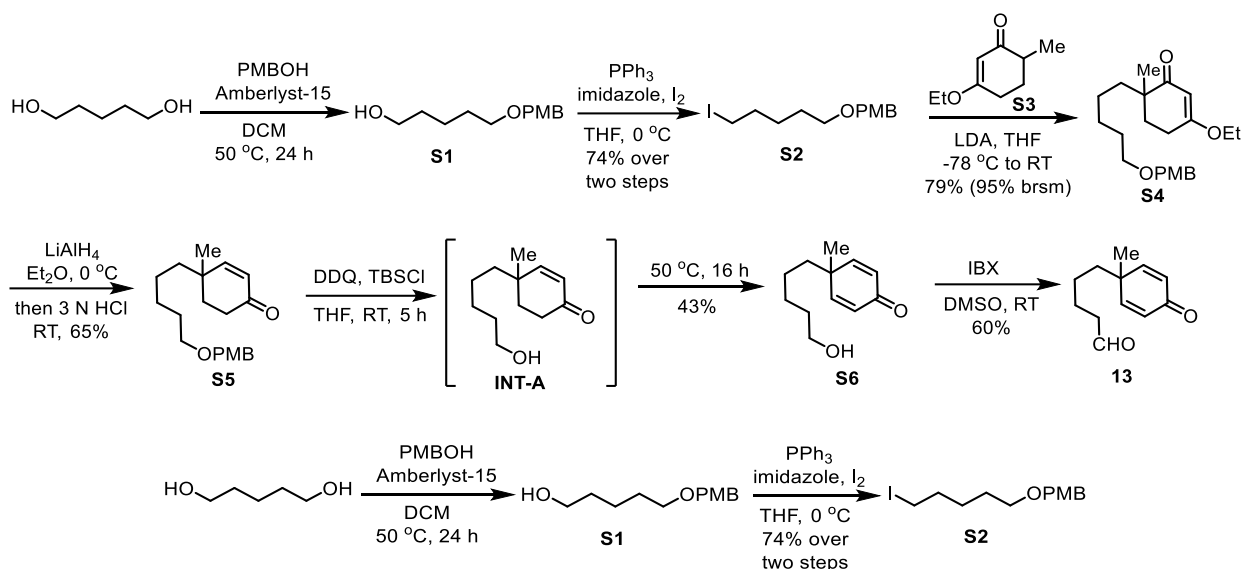
Table S2. Supplementary experiments as reviewer's suggestions.

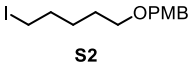


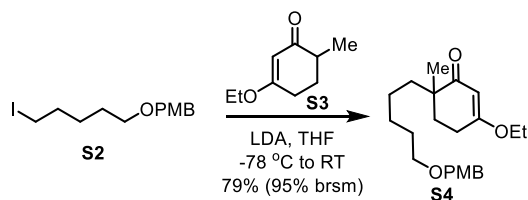
entry	condition	result
1	without 13 (<i>R</i>)-cat.I (0.05 mmol, 16.3 mg, 1.0 equiv.) AcOH (0.25 mmol, 15.0 mg, 5.0 equiv.) toluene (0.5 mL), RT, 9.0 h	(<i>R</i>)-cat.I : stable TMS group: not deprotected
2	13 (0.05 mmol, 9.6 mg, 1.0 equiv.) (<i>R</i>)-cat.II (0.05 mmol, 16.3 mg, 1.0 equiv.) toluene (0.5 mL), RT, 9.0 h	N.R.
3	13 (0.05 mmol, 9.6 mg, 1.0 equiv.) (<i>R</i>)-cat.II (0.05 mmol, 16.3 mg, 1.0 equiv.) AcOH (0.25 mmol, 15.0 mg, 5.0 equiv.) toluene (0.5 mL), RT, 9.0 h	N.R.

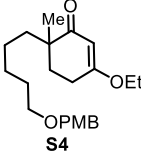
The entry 1 in **Table S2** indicate that the Hayashi–Jørgensen catalyst (*R*)-cat.I was stable under the stirred solution of 0.5 N AcOH/toluene for more than 9 hours. Besides, we conclude that (*R*)-cat.I was the effective catalyst in this desymmetrization reaction based on entries 2, 3 in **Table S2**.

Experimental procedures and spectroscopic data

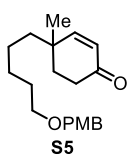
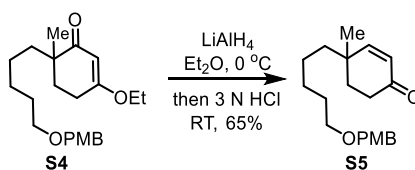



S2 To a stirred solution of 4-methoxybenzyl alcohol (55.26 g, 49.6 mL, 0.40 mol, 1.0 equiv.) and 1,5-pentanediol (62.5 g, 62.9 mL, 0.60 mol, 1.5 equiv.) in anhydrous DCM (500 mL) at room temperature was added Amberlyst-15 resin (8.29 g, 15% w/w). The mixture was refluxed at 50 °C for 24 hours then filtered through silica gel and washed with DCM (6×50 mL). The filtrate was concentrated and the obtained crude compound **S1** was dissolved in anhydrous THF (800 mL). Then imidazole (81.7 g, 1.20 mol, 3.0 equiv.), PPh₃ (157.37 g, 0.60 mol, 1.5 equiv.), I₂ (152.29 g, 0.6 mol, 1.5 equiv.) was successively added to the solution at 0 °C. After stirred at 0 °C for 1 hour, the solution was quenched with saturated Na₂S₂O₃ (100 mL) then THF was evacuated. The mixture was extracted with ethyl acetate (3×200 mL). The combined organic layers were washed with water, brine, dried over anhydrous sodium sulfate, concentrated, and added 200 mL Et₂O. Then the undissolved triphenylphosphine oxide was filtered and washed with Et₂O (3×50 mL). The filtrate was concentrated and purified by silica gel flash chromatography (5% to 10% ethyl acetate – petroleum ether) to obtain **S2** as a light yellow viscous oil (98.87 g, 74% over two steps). R_f = 0.42 (10% ethyl acetate – petroleum ether). The NMR spectra of **S2** were consistent with the previous report [2].



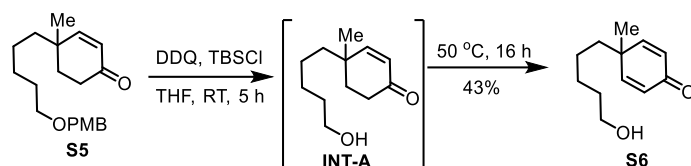

S4 To a stirred solution of diisopropylamine (28 mL, 200 mmol, 2.0 equiv.) in anhydrous THF (300 mL) at 0 °C under nitrogen atmosphere was slowly added *n*-BuLi (80 mL, 200 mmol, 2.0 equiv., 2.5 M in hexane). After stirring at 0 °C for 30 minutes, the compound **S3** (15.4 g, 100 mmol, 1.0 equiv.) in anhydrous THF (50 mL) was slowly added at -78 °C. Then the mixture was stirred for 30 minutes at -78 °C and stirred for 30 minutes at room temperature. After that, compound **S2** (66.8 g, 200 mmol, 2.0

equiv.) in anhydrous THF (50 mL) was slowly added to the mixture at $-78\text{ }^{\circ}\text{C}$ and slowly warmed to room temperature. After stirring at room temperature overnight, the mixture was quenched with water (100 mL) and extracted with ethyl acetate ($3\times 100\text{ mL}$). The combined organic layers were washed with brine, dried over anhydrous sodium sulfate, concentrated, and purified by silica gel flash chromatography (10% to 20% ethyl acetate – petroleum ether) to obtain **S4** as a light yellow viscous oil (27.35 g, 79%) and 2.46 g recycling starting material **S3**. $R_f = 0.24$ (10% ethyl acetate – petroleum ether); Light yellow viscous oil; $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 7.24 (d, $J = 8.6\text{ Hz}$, 2H), 6.86 (d, $J = 8.6\text{ Hz}$, 2H), 5.22 (s, 1H), 4.41 (s, 2H), 3.87 (q, $J = 7.0\text{ Hz}$, 2H), 3.79 (s, 3H), 3.41 (t, $J = 6.6\text{ Hz}$, 2H), 2.48 – 2.31 (m, 2H), 1.94 – 1.82 (m, 1H), 1.70 (ddd, $J = 13.4, 7.5, 5.7\text{ Hz}$, 1H), 1.64 – 1.54 (m, 2H), 1.54 – 1.46 (m, 1H), 1.46 – 1.36 (m, 1H), 1.34 (t, $J = 7.1\text{ Hz}$, 3H), 1.32 – 1.27 (m, 2H), 1.27 – 1.17 (m, 2H), 1.05 (s, 3H) ppm; $^{13}\text{C NMR}$ (100 MHz, Chloroform-*d*) δ 204.2, 175.6, 159.1, 130.8, 129.2 (2C), 113.7 (2C), 101.3, 72.5, 70.1, 64.1, 55.2, 43.2, 36.9, 32.2, 29.7, 26.9, 26.1, 23.8, 22.3, 14.2 ppm; IR ν_{max} 2935, 2858, 1653, 1608, 1512, 1458, 1375, 1359, 1246, 1190, 1099, 1035, 896, 846, 821 cm^{-1} ; HRMS–EI (m/z): $[\text{M}]^+$ calculated for $\text{C}_{22}\text{H}_{32}\text{O}_4$, 360.2301, found, .360.2299.

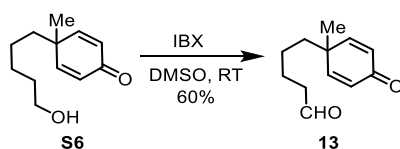


To a stirred mixture of LiAlH_4 (5.73 g, 151.2 mmol, 1.2 equiv.) in anhydrous Et_2O (400 mL) at $0\text{ }^{\circ}\text{C}$ under nitrogen atmosphere was slowly added compound **S4** (43.62 g, 126 mmol, 1.0 equiv.) in anhydrous Et_2O (100 mL). After stirred at $0\text{ }^{\circ}\text{C}$ for 30 minutes, 300 mL saturated NH_4Cl was slowly added to the mixture and then added 300 mL 3 N HCl at $0\text{ }^{\circ}\text{C}$. The mixture was slowly warmed to room temperature and stirred for another 1 hour. Separated the Et_2O layer and washed with water, saturated NaHCO_3 , brine. Then the separated total aqueous layer was extracted with ethyl acetate ($3\times 100\text{ mL}$) and the separated ethyl acetate layer was washed with water, saturated NaHCO_3 , brine. The combined organic layers were dried over anhydrous sodium sulfate, concentrated, and purified by silica gel flash chromatography (10% to 20% ethyl acetate – petroleum ether) to obtain **S5** as a light yellow viscous oil (26.0 g, 65%). $R_f = 0.30$ (10% ethyl acetate – petroleum ether); Light yellow viscous oil; $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 7.25 (d, $J = 8.4\text{ Hz}$, 2H), 6.87 (d, $J = 8.6\text{ Hz}$, 2H), 6.65 (d, $J = 10.2\text{ Hz}$, 1H), 5.86 (d, $J = 10.2\text{ Hz}$, 1H), 4.43 (s, 2H), 3.80 (s, 3H), 3.43 (t, $J = 6.5\text{ Hz}$, 2H), 2.47 – 2.38 (m, 2H), 1.94 (ddd, $J = 14.5, 8.9, 5.9\text{ Hz}$, 1H), 1.74 (dt, $J = 12.8, 6.0\text{ Hz}$, 1H), 1.66 – 1.58 (m, 2H), 1.50 – 1.22 (m, 6H), 1.11 (s, 3H) ppm; $^{13}\text{C NMR}$ (100 MHz, Chloroform-*d*) δ 199.7, 159.4, 159.1, 130.7, 129.2 (2C), 127.3, 113.8 (2C), 72.6, 69.9, 55.3, 41.0, 35.6, 34.2, 33.5, 29.7, 26.9, 24.9, 24.0 ppm; IR ν_{max} 2933, 2859, 1682, 1612, 1512, 1463, 1301, 1265, 1246, 1097, 1035, 804 cm^{-1} ; HRMS–EI

(*m/z*): [*M*]⁺ calculated for C₂₀H₂₈O₃, 316.2038, found, 316.2034.

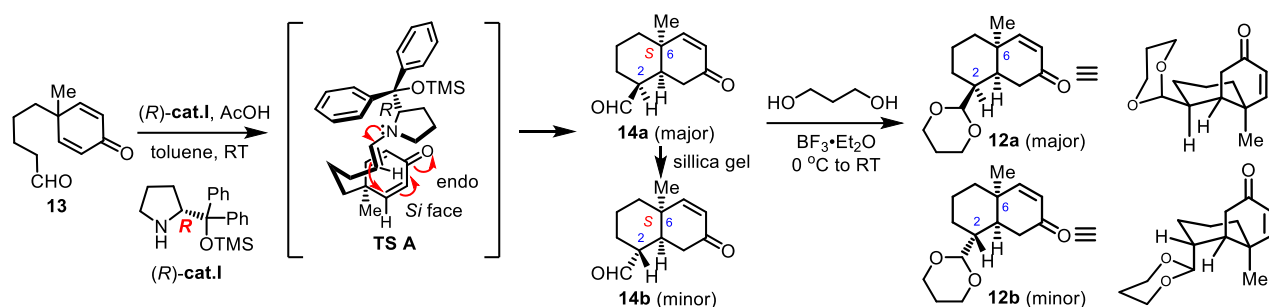


To a stirred solution of **S5** (13.0 g, 41.08 mmol, 1.0 equiv.) in anhydrous THF (400 mL) at room temperature was successively added DDQ (28.0 g, 123.24 mmol, 3.0 equiv.), TBSCl (6.81 g, 45.19 mmol, 1.1 equiv.) portion-wisely. After stirring at room temperature for 5 hours, the brown mixture was stirred at 50 °C for further 16 hours. The mixture was concentrated and diluted with ethyl acetate (150 mL) then quenched with saturated Na₂S₂O₃/NaHCO₃ (150 mL, v/v = 1:1). Separated the ethyl acetate layer and washed with water, brine. Then the separated total aqueous layer was extracted with ethyl acetate (3×100 mL). The combined organic layers were dried over anhydrous sodium sulfate, concentrated, and purified by silica gel flash chromatography (30% to 60% ethyl acetate – petroleum ether) to obtain **S6** as a brown viscous oil (3.43 g, 43%). *R_f* = 0.20 (30% ethyl acetate – petroleum ether); Brown viscous oil; ¹H NMR (500 MHz, Chloroform-*d*) δ 6.76 (d, *J* = 10.1 Hz, 2H), 6.26 (d, *J* = 10.1 Hz, 2H), 3.60 (t, *J* = 6.5 Hz, 2H), 1.66 – 1.59 (m, 2H), 1.55 – 1.47 (m, 2H), 1.47 (br s, 1H), 1.36 – 1.26 (m, 2H), 1.24 (s, 3H), 1.19 – 1.09 (m, 2H) ppm; ¹³C NMR (125 MHz, Chloroform-*d*) δ 186.4, 155.9 (2C), 128.8 (2C), 62.7, 42.0, 40.5, 32.4, 26.1, 26.0, 24.8 ppm; IR *v*_{max} 3051, 2934, 2863, 1741, 1662, 1616, 1459, 1267, 1246, 1037, 862, 704 cm⁻¹; HRMS–EI (*m/z*): [*M*]⁺ calculated for C₁₂H₁₈O₂, 194.1307, found, 194.1305.

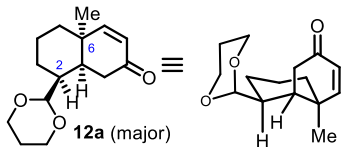


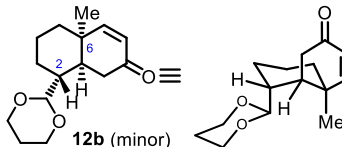
To a stirred solution of **S6** (3.30 g, 17.00 mmol, 1.0 equiv.) in anhydrous DMSO (50 mL) at room temperature was added IBX (9.52 g, 34.00 mmol, 2.0 equiv.) portion-wisely. After stirring at room temperature for 2 hours, the mixture was quenched with saturated Na₂S₂O₃/NaHCO₃ (50 mL, v/v = 1:1) and extracted with ethyl acetate (3×50 mL). The combined organic layers were washed with water, brine, then dried over anhydrous sodium sulfate, concentrated, and purified by silica gel flash chromatography (30% to 40% ethyl acetate – petroleum ether) to obtain **13** as a light brown oil (1.96 g, 60%). *R_f* = 0.20 (20% ethyl acetate – petroleum ether); Light brown oil; ¹H NMR (500 MHz, Chloroform-*d*) δ 9.69 (s, 1H), 6.72 (d, *J* = 10.0 Hz, 2H), 6.22 (d, *J* = 10.0 Hz, 2H), 2.36 (td, *J* = 7.3, 1.5 Hz, 2H), 1.63 – 1.57 (m, 2H), 1.57 – 1.49 (m, 2H), 1.21 (s, 3H), 1.16 – 1.06 (m, 2H) ppm; ¹³C NMR (125 MHz, Chloroform-*d*) δ 202.0, 186.1, 155.5 (2C), 128.8 (2C), 43.5, 41.8, 40.2, 26.0, 24.5, 22.0 ppm; IR *v*_{max} 3053, 2937, 2862, 1718, 1664, 1624, 1458,

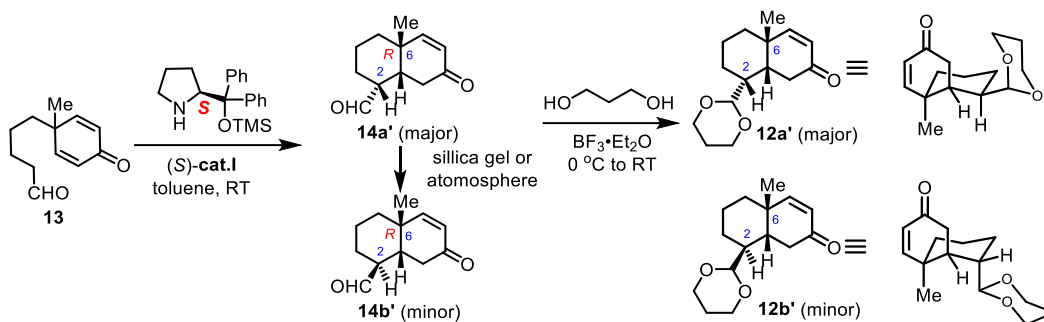
1404, 1267, 1242, 1105, 864 cm^{-1} ; HRMS–EI (m/z): $[\text{M}]^+$ calculated for $\text{C}_{12}\text{H}_{16}\text{O}_2$, 192.1150, found, 192.1151.



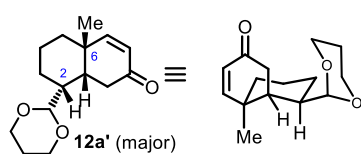
A solution of **13** (1.54 g, 8.00 mmol, 1.0 equiv.), (*R*)-**cat.I** (520 mg, 1.60 mmol, 0.2 equiv.) and AcOH (96 mg, 1.60 mmol, 0.2 equiv.) in analytical-grade toluene (80 mL) was stirred at room temperature for 9.5 hours with inert nitrogen atmosphere. Then 1,3-propanediol (2.44 g, 2.31 mL, 32.00 mol, 4.0 equiv.), 46.5% $\text{BF}_3 \cdot \text{Et}_2\text{O}$ (1.14 g, 0.99 mL, 8.0 mmol, 1.0 equiv.) was added to the solution at 0 °C and stirred at room temperature for further 2 hours. The solution was quenched with saturated NaHCO_3 (50 mL) and extracted with ethyl acetate (3×50 mL). The combined organic layers were washed with brine, then dried over anhydrous sodium sulfate, concentrated and purified by flash chromatography (20% ethyl acetate – petroleum ether) to obtain pure **12a** (1.05 g), pure **12b** (0.15 g) and the mixture of **12a+12b** (0.40 g) all as light yellow viscous oil (total 1.6 g, total 80% for one pot synthesis, d.r. = 5.5:1).

 **12a** (major) $R_f = 0.30$ (20% ethyl acetate – petroleum ether); Light yellow viscous oil; $[\alpha]_{\text{D}}^{20} = -10.5$ ($c = 1.00$ in DCM); $^1\text{H NMR}$ (500 MHz, Chloroform-*d*) δ 6.62 (d, $J = 10.1$ Hz, 1H), 5.84 (d, $J = 10.1$ Hz, 1H), 4.24 (d, $J = 7.4$ Hz, 1H), 4.12 – 4.02 (m, 2H), 3.76 – 3.63 (m, 2H), 2.46 – 2.35 (m, 1H), 2.30 – 2.20 (m, 2H), 2.10 – 2.02 (m, 1H), 2.01 – 1.94 (m, 1H), 1.81 – 1.69 (m, 1H), 1.67 – 1.60 (m, 1H), 1.54 – 1.44 (m, 3H), 1.33 (d, $J = 13.5$ Hz, 1H), 1.16 (s, 3H), 1.14 – 1.10 (m, 1H) ppm; $^{13}\text{C NMR}$ (125 MHz, Chloroform-*d*) δ 200.6, 161.5, 126.0, 103.1, 67.00, 66.97, 40.1, 38.3, 36.1, 34.7, 31.3, 25.9, 24.3, 21.3, 20.3 ppm; IR ν_{max} 2937, 2860, 1676, 1614, 1375, 1267, 1240, 1143, 1114, 1086, 1014, 935, 852 cm^{-1} ; HRMS–EI (m/z): $[\text{M}]^+$ calculated for $\text{C}_{15}\text{H}_{22}\text{O}_3$, 250.1569, found, 250.1571.

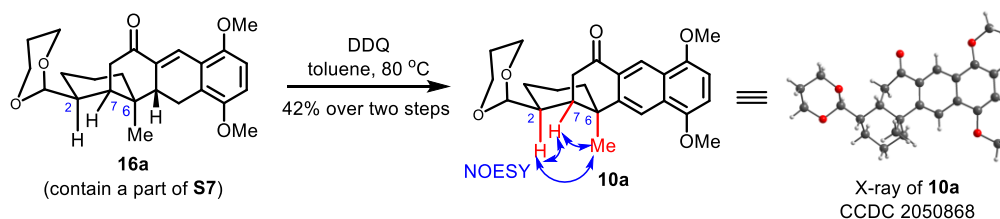
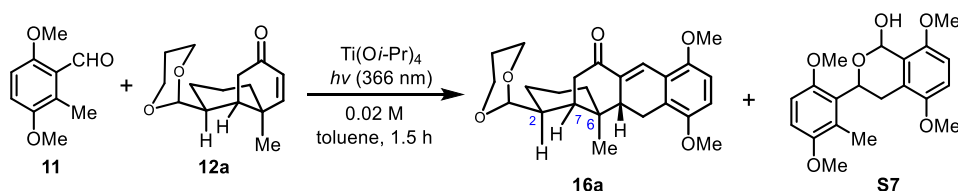
 **12b** (minor) $R_f = 0.34$ (20% ethyl acetate – petroleum ether); Light yellow viscous oil; $[\alpha]_{\text{D}}^{20} = -64.9$ ($c = 0.60$ in DCM); $^1\text{H NMR}$ (500 MHz, Chloroform-*d*) δ 6.53 (dd, $J = 10.1, 2.1$ Hz, 1H), 5.88 (d, $J = 10.1$ Hz, 1H), 4.57 (d, $J = 2.4$ Hz, 1H), 4.09 – 4.01 (m, 2H), 3.77 – 3.62 (m, 2H), 2.77 – 2.60 (m, 2H), 2.06 – 1.92 (m, 2H), 1.89 – 1.79 (m, 1H), 1.69 – 1.58 (m, 3H), 1.35 – 1.25 (m, 2H), 1.22 (s, 3H), 1.21 – 1.11 (m, 2H) ppm; $^{13}\text{C NMR}$ (125 MHz, Chloroform-*d*) δ 199.4, 159.1, 128.3, 102.2, 67.1, 67.0, 42.0, 40.9, 40.2, 37.53, 37.52, 27.8, 25.8, 25.3, 22.6 ppm; IR ν_{max} 2936, 2858, 1676, 1610, 1377, 1238, 1151, 1122, 1101, 1016, 997, 941, 893 cm^{-1} ; HRMS–EI (m/z): $[\text{M}]^+$ calculated for $\text{C}_{15}\text{H}_{22}\text{O}_3$, 250.1569, found, 250.1567.



A solution of **13** (288 mg, 1.50 mmol, 1.0 equiv.), (*S*)-**cat.I** (195 mg, 0.60 mmol, 0.4 equiv.) in analytical-grade toluene (15 mL) was stirred at room temperature for 12 hours with inert nitrogen atmosphere. Then 1,3-propanediol (457 mg, 0.43 mL, 6.00 mol, 4.0 equiv.), 46.5% $\text{BF}_3 \cdot \text{Et}_2\text{O}$ (213 mg, 0.185 mL, 1.5 mmol, 1.0 equiv.) was added to the solution at 0 °C and stirred at room temperature for further 2 hours. The solution was quenched with saturated NaHCO_3 (10 mL) and extracted with ethyl acetate (3×10 mL). The combined organic layers were washed with brine, then dried over anhydrous sodium sulfate, concentrated and purified by flash chromatography (20% ethyl acetate – petroleum ether) to obtain pure **12a'** as a light yellow viscous oil (233 mg, 62% for one pot synthesis, d.r. = 14.0:1).



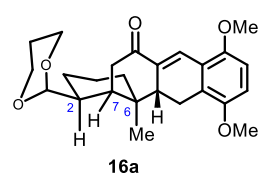
$R_f = 0.30$ (20% ethyl acetate – petroleum ether); Light yellow viscous oil; $[\alpha]_D^{20} = +9.4$ (c = 0.60 in DCM); $^1\text{H NMR}$: the same to compound **12a**; $^{13}\text{C NMR}$: the same to compound **12a**; IR ν_{max} 2928, 2863, 1684, 1676, 1376, 1275, 1240, 1145, 1116, 1015, 934, 852, 764 cm^{-1} ; HRMS–EI (m/z): $[\text{M}]^+$ calculated for $\text{C}_{15}\text{H}_{22}\text{O}_3$, 250.1569, found, 250.1571.



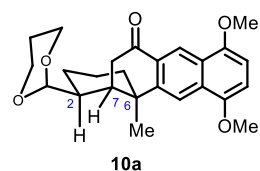
To a solution of dienophile **12a** (2.00 g, 8.00 mmol, 1.0 equiv.), aromatic aldehyde **11** (2.16 g, 12.0 mmol, 1.5 equiv.) in anhydrous and degassed toluene (400 mL) (concentration for dienophile **12a** is 0.02 M) was added titanium(IV) isopropoxide (6.82 g, 7.10 mL, 24.0 mmol, 3.0 equiv.) under N_2 . After homogeneous mixing, the solution was divided into 10 parallel reactions in 10 quartz tubes (10×40 mL). 5 parallel reactions were conducted with 5 quartz tubes once. The solution was photolyzed at room temperature in a Rayonet chamber reactor (16 lamps) at $\lambda_{\text{max}} = 366 \text{ nm}$ for 1.5 hours (**Note**: the atmosphere temperature among quartz tubes was 35 to 40 °C). After the above 5 parallel reactions were over, the reaction mixture was quenched with

saturated NaHCO₃ (50 mL). Then another 5 parallel reactions were conducted for another 1.5 hours, the reaction mixture was also quenched with saturated NaHCO₃ (50 mL). The total mixture was filtered through silica gel and washed with ethyl acetate (6×50 mL). The combined organic layers were washed with brine, then dried over anhydrous sodium sulfate, concentrated and purified by flash chromatography (10% to 30% ethyl acetate – petroleum ether) to obtain 1.80 g **16a** (contain ~ 9% **S7**) as a yellow viscous oil.

To a solution of above obtained 1.80 g **16a** (contain ~ 9% **S7**) in anhydrous toluene (40 mL) was added DDQ (1.97 g, 8.69 mmol, 2.0 equiv.) at room temperature. After stirred at 80 °C for 3 hours, the mixture was quenched with saturated Na₂S₂O₃/NaHCO₃ (40 mL, v/v = 1:1) and extracted with ethyl acetate (3×50 mL). The combined organic layers were washed with water, brine, then dried over anhydrous sodium sulfate, concentrated, and purified by silica gel flash chromatography (20% ethyl acetate – petroleum ether) to obtain pure **10a** as a yellow foam solid (1.38 g, 42% over two steps).

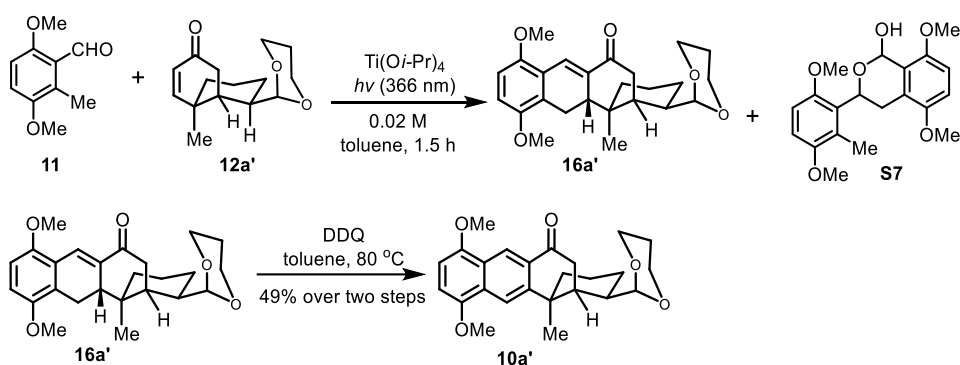


16a $R_f = 0.36$ (30% ethyl acetate – petroleum ether); Yellow viscous oil; $[\alpha]_D^{20} = +179.0$ ($c = 1.10$ in DCM); ¹H NMR (400 MHz, Chloroform-*d*) δ 7.81 (d, $J = 2.4$ Hz, 1H), 6.79 (d, $J = 8.9$ Hz, 1H), 6.65 (d, $J = 8.9$ Hz, 1H), 4.41 (d, $J = 7.6$ Hz, 1H), 4.02 (dd, $J = 11.4, 5.0$ Hz, 1H), 3.92 (dd, $J = 11.5, 5.0$ Hz, 1H), 3.80 (s, 3H), 3.78 (s, 3H), 3.64 (td, $J = 12.3, 2.6$ Hz, 1H), 3.41 (td, $J = 12.0, 2.6$ Hz, 1H), 2.99 (dd, $J = 15.7, 6.9$ Hz, 1H), 2.72 (ddd, $J = 11.5, 6.9, 2.5$ Hz, 1H), 2.66 – 2.60 (m, 1H), 2.57 (d, $J = 7.1$ Hz, 1H), 2.49 (dd, $J = 17.6, 5.7$ Hz, 1H), 2.04 – 1.88 (m, 3H), 1.76 (ddd, $J = 13.8, 7.0, 3.0$ Hz, 1H), 1.57 – 1.50 (m, 3H), 1.47 – 1.35 (m, 2H), 1.28 – 1.21 (m, 1H), 0.96 (s, 3H) ppm; ¹³C NMR (100 MHz, Chloroform-*d*) δ 200.5, 151.7, 150.0, 134.5, 127.0, 126.4, 122.4, 112.5, 108.6, 103.6, 66.9, 66.0, 55.93, 55.91, 41.7, 40.6, 39.1, 38.3, 35.8, 35.5, 25.6, 24.8, 24.0, 21.8, 19.4 ppm; IR ν_{\max} 2957, 2925, 2863, 1675, 1599, 1576, 1484, 1465, 1275, 1267, 1110, 1098, 1003, 763, 759 cm⁻¹; HRMS–EI (m/z): $[M]^+$ calculated for C₂₅H₃₂O₅, 412.2250, found, 412.2255.



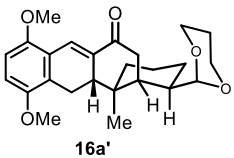
10a $R_f = 0.42$ (30% ethyl acetate – petroleum ether); Yellow foam solid, m.p. 110-112 °C; Compound **10a** was recrystallized from dichloromethane at room temperature to obtain yellow crystals, CCDC 2050868; $[\alpha]_D^{20} = +1.5$ ($c = 1.00$ in DCM); ¹H NMR (400 MHz, Chloroform-*d*) δ 8.93 (s, 1H), 8.23 (s, 1H), 6.69 (d, $J = 8.3$ Hz, 1H), 6.57 (d, $J = 8.3$ Hz, 1H), 4.30 (d, $J = 7.3$ Hz, 1H), 4.14 – 4.04 (m, 2H), 3.912 (s, 3H), 3.908 (s, 3H), 3.77 – 3.64 (m, 2H), 2.78 (dd, $J = 16.6, 13.9$ Hz, 1H), 2.62 (dd, $J = 16.7, 4.2$ Hz, 1H), 2.49 (dt, $J = 13.9, 4.2$ Hz, 1H), 2.29 – 2.19 (m, 1H), 2.11 – 1.98 (m, 1H), 1.80 – 1.69 (m, 2H), 1.69 – 1.61 (m, 3H), 1.60 (s, 3H), 1.36 – 1.20 (m, 2H) ppm; ¹³C NMR (100 MHz, Chloroform-*d*) δ 198.9, 150.8, 148.9, 148.6, 129.1, 128.6, 124.2, 122.5, 119.4, 105.6, 103.4, 102.7, 66.93, 66.91, 55.6, 55.5, 40.0, 38.2, 37.7, 35.9, 35.8, 25.9, 25.2, 21.4, 21.2 ppm; IR ν_{\max} 3055, 2937, 2862, 2837,

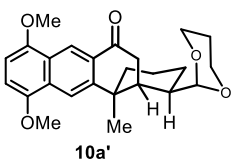
1684, 1626, 1589, 1458, 1267, 1240, 1143, 1103, 1091, 1008, 985 cm^{-1} ; HRMS–EI (m/z): $[\text{M}]^+$ calculated for $\text{C}_{25}\text{H}_{30}\text{O}_5$, 410.2093, found, 410.2089.

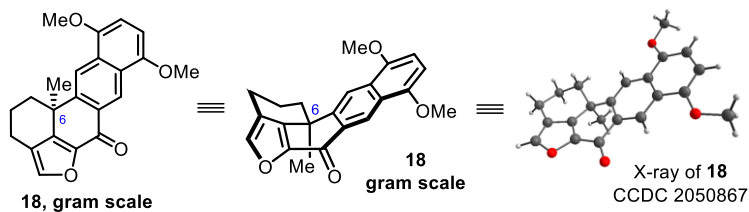
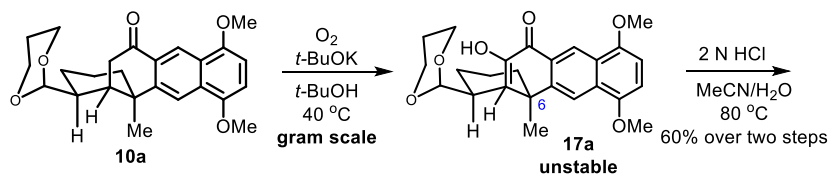


To a solution of dienophile **12a'** (200 mg, 0.80 mmol, 1.0 equiv.), aromatic aldehyde **11** (216 mg, 1.20 mmol, 1.5 equiv.) in anhydrous and degassed toluene (40 mL) (concentration for dienophile **12a'** is 0.02 M) was added titanium(IV) isopropoxide (682 mg, 0.71 mL, 2.40 mmol, 3.0 equiv.) under N_2 . After homogeneous mixing, the solution was photolyzed at room temperature in a Rayonet chamber reactor (16 lamps) at $\lambda_{\text{max}} = 366 \text{ nm}$ for 1.5 hours (**Note**: the atmosphere temperature among quartz tubes was 35 to 40 $^\circ\text{C}$). After that, the mixture was quenched with saturated NaHCO_3 (10 mL) and filtered through silica gel and washed with ethyl acetate (6 \times 10 mL). The combined organic layers were washed with brine, then dried over anhydrous sodium sulfate, concentrated and purified by flash chromatography (10% to 30% ethyl acetate – petroleum ether) to obtain 190 mg **16a'** as a yellow viscous oil.

To a solution of above obtained **16a'** (190 mg, 0.46 mmol, 1.0 equiv.) in anhydrous toluene (10 mL) was added DDQ (218 mg, 0.96 mmol, 2.0 equiv.) at room temperature. After stirred at 80 $^\circ\text{C}$ for 3 hours, the mixture was quenched with saturated $\text{Na}_2\text{S}_2\text{O}_3/\text{NaHCO}_3$ (10 mL, v/v = 1:1) and extracted with ethyl acetate (3 \times 10 mL). The combined organic layers were washed with water, brine, then dried over anhydrous sodium sulfate, concentrated, and purified by silica gel flash chromatography (20% ethyl acetate – petroleum ether) to obtain **10a'** as a yellow foam solid (161 mg, 49 % over two steps).

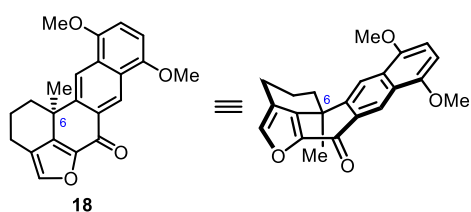
 **16a'** $R_f = 0.36$ (30% ethyl acetate – petroleum ether); Yellow viscous oil; $[\alpha]_{\text{D}}^{20} = -144.9$ ($c = 0.40$ in DCM); $^1\text{H NMR}$: the same to **16a**; $^{13}\text{C NMR}$: the same to **16a**; IR ν_{max} 2960, 2925, 2857, 1691, 1675, 1598, 1570, 1483, 1464, 1275, 1260, 1144, 1096, 1016, 802, 799 cm^{-1} ; HRMS–EI (m/z): $[\text{M}]^+$ calculated for $\text{C}_{25}\text{H}_{32}\text{O}_5$, 412.2250, found, 412.2248.

 **10a'** $R_f = 0.42$ (30% ethyl acetate – petroleum ether); Yellow foam solid, m.p. 57-59 $^\circ\text{C}$; $[\alpha]_{\text{D}}^{20} = -4.4$ ($c = 0.60$ in DCM); $^1\text{H NMR}$: the same to **10a**; $^{13}\text{C NMR}$: the same to **10a**; IR ν_{max} 2954, 2930, 2927, 2857, 1683, 1627, 1590, 1458, 1435, 1343, 1333, 1268, 1239, 1142, 1119, 1104, 1092, 1009, 934 cm^{-1} ; HRMS–EI (m/z): $[\text{M}]^+$ calculated for $\text{C}_{25}\text{H}_{30}\text{O}_5$, 410.2093, found, 410.2097.

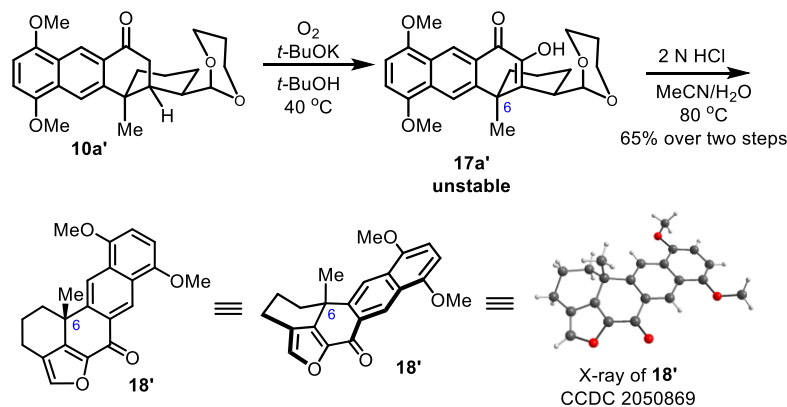


To a solution of 10a (1.31 g, 3.2 mmol, 1.0 equiv.) in analytical-grade *t*-BuOH (60 mL) was added *t*-BuOK (3.60 g, 32.0 mmol, 10.0 equiv.). Then the mixture was stirred with bubbling O_2 into the mixture at 40 °C for 3 hours. After that, *t*-BuOH was evacuated and the mixture was diluted with ethyl acetate (50 mL) and water (30 mL). Separated the organic layer and the aqueous layer was washed with ethyl acetate (3×50 mL). The combined organic layers were washed with water, brine, dried over anhydrous sodium sulfate, then concentrated to obtain 1.12 g crude 17a as a yellow foam solid for the next step without further purification.

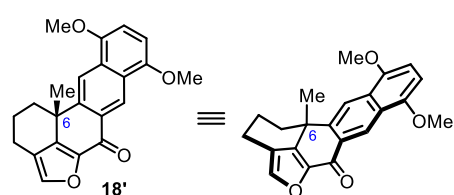
To a solution of above obtained crude 17a in analytical-grade MeCN (60 mL) was added 2 N HCl (10 mL) at room temperature. Then the solution was stirred at 80 °C for 5 hours and quenched with saturated NaHCO_3 (50 mL). The mixture was extracted with DCM (3×60 mL) and washed with brine. The combined organic layers were dried over anhydrous sodium sulfate, concentrated, and purified by silica gel flash chromatography (20% to 30% ethyl acetate – petroleum ether) to obtain 18 as a yellow solid (668 mg, 60% over two steps). **Note:** The xestoquinol dimethyl ether 18 was easier to dissolve in DCM than ethyl acetate.



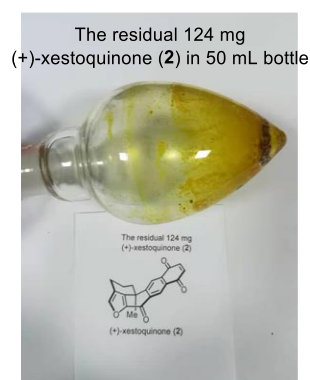
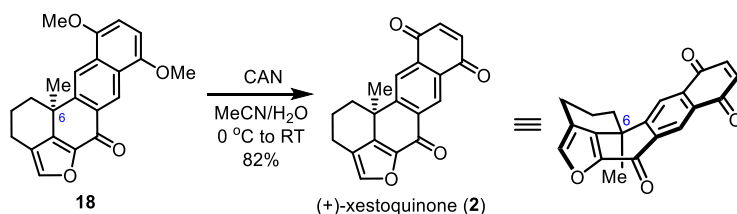
$R_f = 0.38$ (30% ethyl acetate – petroleum ether); Yellow solid, m.p. 243–245 °C; Compound 18 was recrystallized from THF/hexane (*v/v* = 1/2) at room temperature to obtain yellow crystals, CCDC 2050867; $[\alpha]_D^{20} = +99.6$ (*c* = 1.00 in DCM); $^1\text{H NMR}$ (400 MHz, Chloroform-*d*) δ 9.23 (s, 1H), 8.23 (s, 1H), 7.42 (t, *J* = 1.4 Hz, 1H), 6.73 (d, *J* = 8.3 Hz, 1H), 6.63 (d, *J* = 8.4 Hz, 1H), 3.93 (s, 3H), 3.92 (s, 3H), 2.81 (ddt, *J* = 16.9, 7.8, 1.7 Hz, 1H), 2.61 – 2.50 (m, 2H), 2.29 – 2.14 (m, 1H), 2.13 – 2.02 (m, 1H), 1.76 (td, *J* = 13.2, 4.3 Hz, 1H), 1.49 (s, 3H) ppm; $^{13}\text{C NMR}$ (100 MHz, Chloroform-*d*) δ 172.7, 150.6, 148.6, 146.7, 146.6, 144.6, 143.6, 131.2, 127.2, 124.5, 123.9, 121.2, 117.4, 105.9, 103.2, 55.6, 55.5, 36.2, 33.6, 31.9, 18.6, 17.1 ppm; IR ν_{max} 2938, 2835, 1670, 1614, 1541, 1471, 1465, 1423, 1356, 1267, 1244, 1192, 1144, 1091, 1043, 904, 862, 806 cm^{-1} ; HRMS–EI (*m/z*): $[\text{M}]^+$ calculated for $\text{C}_{22}\text{H}_{20}\text{O}_4$, 348.1362, found, 348.1360.



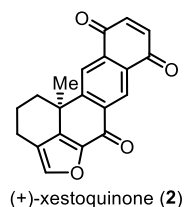
The enantiomer compound **18'** (136 mg, 65% over two steps) was synthesized according to the above similar procedures using **10a'** (246 mg, 0.60 mmol, 1.0 equiv.) as starting material. **Note:** The xestoquinol dimethyl ether **18'** was easier to dissolve in DCM than ethyl acetate.



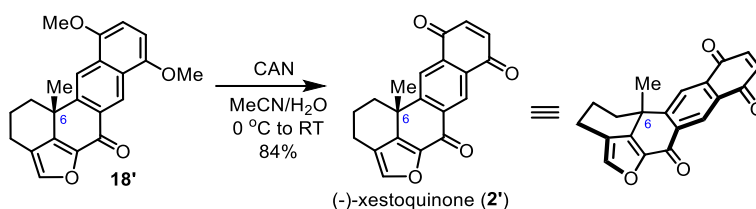
$R_f = 0.38$ (30% ethyl acetate – petroleum ether); Yellow solid, m.p. 248–250 $^\circ\text{C}$; Compound **18'** was recrystallized from THF/hexane ($v/v = 1/2$) at room temperature to obtain yellow crystals, CCDC 2050869; $[\alpha]_D^{20} = -95.4$ ($c = 1.00$ in DCM); $^1\text{H NMR}$: the same to **18**; $^{13}\text{C NMR}$: the same to **18**; IR ν_{max} 3102, 2944, 2836, 1666, 1628, 1613, 1533, 1464, 1423, 1357, 1267, 1245, 1190, 1145, 1089, 1044, 905, 865, 802 cm^{-1} ; HRMS–EI (m/z): $[\text{M}]^+$ calculated for $\text{C}_{22}\text{H}_{20}\text{O}_4$, 348.1362, found, 348.1360.



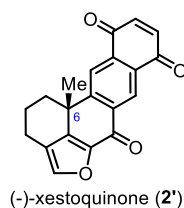
To stirred solution of **18** (208 mg, 0.6 mmol, 1.0 equiv.) in $\text{MeCN}/\text{H}_2\text{O}$ (50 mL, $v/v = 4:1$) was added CAN (987 mg, 1.8 mmol, 3.0 equiv.) at $0\text{ }^\circ\text{C}$. Then the solution was stirred at room temperature for 1 hours. After that, the mixture was extracted with ethyl acetate ($3 \times 30\text{ mL}$). The combined organic layers were washed with water, brine, dried over anhydrous sodium sulfate, concentrated, and purified by silica gel flash chromatography (30% ethyl acetate – petroleum ether) to obtain (+)-xestoquinone (**2**) as a yellow-brown solid (156 mg, 82%).



$R_f = 0.26$ (30% ethyl acetate – petroleum ether); Yellow-brown solid, m.p. 213-216 °C; $[\alpha]_D^{20} = +11.2$ ($c = 1.00$ in DCM); $^1\text{H NMR}$ (500 MHz, Chloroform- d) δ 8.98 (s, 1H), 8.22 (s, 1H), 7.54 (br t, $J = 1.5$ Hz, 1H), 7.05 (d, $J = 10.0$ Hz, 1H), 7.02 (d, $J = 10.0$ Hz, 1H), 2.89 (ddt, $J = 17.1$, 8.0, 1.8 Hz, 1H), 2.66 (dddd, $J = 17.0$, 10.0, 8.5, 1.5 Hz, 1H), 2.58 (dt, $J = 12.9$, 3.6 Hz, 1H), 2.35 – 2.24 (m, 1H), 2.23 – 2.15 (m, 1H), 1.76 (td, $J = 13.1$, 4.4 Hz, 1H), 1.53 (s, 3H) ppm; $^{13}\text{C NMR}$ (125 MHz, Chloroform- d) δ 184.6, 183.8, 170.1, 156.2, 147.3, 145.0, 144.0, 139.3, 138.7, 137.9, 133.2, 130.3, 126.9, 123.2, 121.6, 37.4, 32.6, 31.2, 18.4, 16.9 ppm; IR ν_{max} 3105, 2862, 1670, 1614, 1456, 1444, 1319, 1267, 1238, 1134, 1095, 898, 846, 796 cm^{-1} ; HRMS–EI (m/z): $[\text{M}]^+$ calculated for $\text{C}_{20}\text{H}_{14}\text{O}_4$, 318.0892, found, 318.0889.

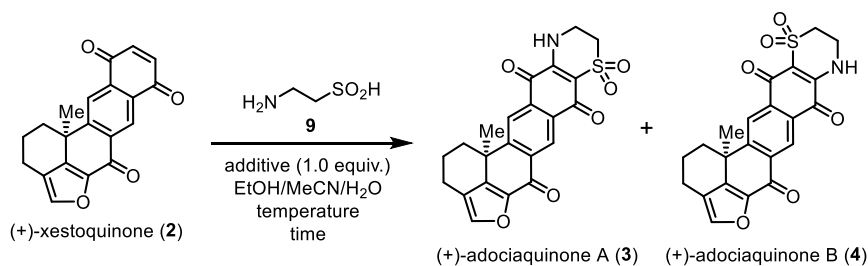


The (-)-xestoquinone (**2'**) (40 mg, 84%) was synthesized according to the above similar procedures using **18'** (52 mg, 0.15 mmol, 1.0 equiv.) as starting material.



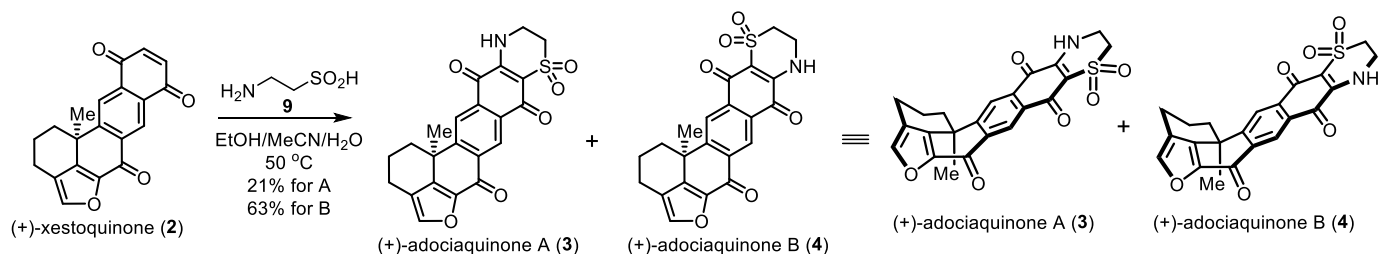
$R_f = 0.26$ (30% ethyl acetate – petroleum ether); Yellow-brown solid, m.p. 96 - 98 °C; $[\alpha]_D^{20} = -8.2$ ($c = 1.00$ in DCM); $^1\text{H NMR}$: the same to **2**; $^{13}\text{C NMR}$: the same to **2**; IR ν_{max} 2953, 2926, 2856, 1669, 1602, 1539, 1444, 1430, 1318, 1274, 1236, 1134, 1092, 1058, 986, 845, 764 cm^{-1} ; HRMS–EI (m/z): $[\text{M}]^+$ calculated for $\text{C}_{20}\text{H}_{14}\text{O}_4$, 318.0892, found, 318.0894.

Table S3. Screening conditions of the late-stage cyclization.

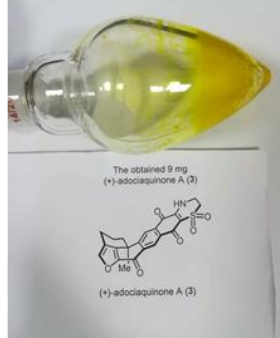


entry	additive	temperature/time	ratio of 3 : 4 ^a	combined yield ^a
1	--	-20 °C / 6 h	1:2.5	28%
2	--	0 °C / 3 h	1:2.2	28%
3	--	20 °C / 1 h	1:2.3	33%
4 ^b	--	50 °C / 3 h	1:3.0	84% ^c
5	CeCl ₃ ·7H ₂ O	50 °C / 1 h	1:2.2	52%
6	TFA	50 °C / 1 h	1:1.1	13%
7	Cs ₂ CO ₃	50 °C / 1 h	--	messy, N.D.
8	NEt ₃	50 °C / 1 h	1:2.5	18%

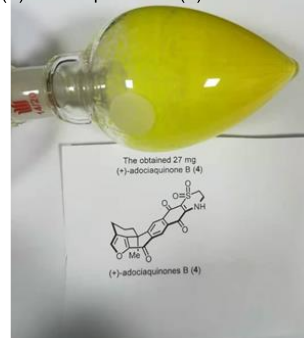
All reactions were performed using (+)-xestoquinone (**2**) (3.2 mg, 0.01 mmol, 1.0 equiv.) and hypotauroine **9** (1.7 mg, 0.015 mmol, 1.5 equiv.) as starting materials in EtOH/MeCN/H₂O (0.5 mL, v/v/v = 2:2:1), unless otherwise noted. ^aThe ratios of **3**:**4** and combined yields were determined from crude $^1\text{H NMR}$ spectrum of **3**+**4** using CH₂Br₂ as an internal standard, ^b(+)-xestoquinone (**2**) (32 mg, 0.1 mmol, 1.0 equiv.) and hypotauroine **9** (17 mg, 0.15 mmol, 1.5 equiv.) in EtOH/MeCN/H₂O (5 mL, v/v/v = 2:2:1); ^c isolated combined yield of **3**+**4**.



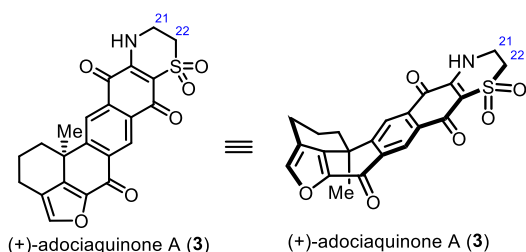
The obtained 9 mg
(+)-adociaquinone A (3) in 25 mL bottle



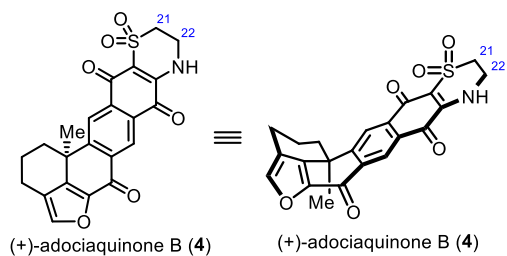
The obtained 27 mg
(+)-adociaquinone B (4) in 50 mL bottle



To a stirred solution of (+)-xestoquinone (2) (32 mg, 0.1 mmol, 1.0 equiv.) in EtOH/MeCN/H₂O (5 mL, v/v/v = 2:2:1) was added hypotaurine 9 (17 mg, 0.15 mmol, 1.5 equiv.) at room temperature. Then the solution was stirred at 50 °C for 3 hours. TLC analysis showed all (+)-xestoquinone (2) was consumed. The solvent was evacuated under vacuum, and the residue was purified by preparation lamella chromatography (7% MeOH-DCM, washed with 10% MeOH-DCM) directly to afford yellow solid (+)-adociaquinone A (3) (9.0 mg, 21%), (+)-adociaquinone B (4) (27.0 mg, 63%).



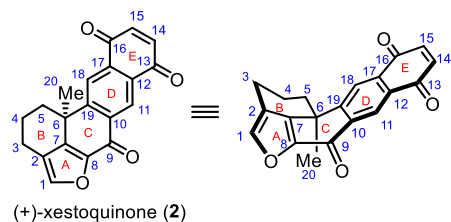
$R_f = 0.52$ (10% methanol-dichloromethane); Yellow solid, m.p. >320 °C (decomposed); $[\alpha]_D^{20} = +65.2$ (c = 0.1 in chloroform-methanol (v/v = 2:1)); ¹H NMR (500 MHz, DMSO-*d*₆) δ 9.35 (br s, 1H, NH), 8.68 (s, 1H), 8.26 (s, 1H), 8.00 (s, 1H), 3.88 (t, *J* = 6.0 Hz, 2H), 3.40 (t, *J* = 6.0 Hz, 2H), 2.84 (dd, *J* = 17.0, 8.0 Hz, 1H), 2.64 – 2.60 (m, 1H), 2.60 – 2.57 (m, 1H), 2.30 – 2.16 (m, 1H), 2.13 – 2.03 (m, 1H), 1.65 (td, *J* = 13.0, 4.4 Hz, 1H), 1.50 (s, 3H) ppm; ¹³C NMR (125 MHz, DMSO-*d*₆) δ 177.9, 173.8, 169.4, 157.4, 147.8, 147.2, 146.2, 143.0, 136.0, 134.4, 128.6, 125.4, 123.0, 121.7, 111.7, 48.2, 39.4 (C21, buried in the peak of DMSO-*d*₆), 37.2, 31.8, 30.3, 17.9, 16.3 ppm; IR ν_{max} 3270, 2928, 2858, 1668, 1655, 1589, 1508, 1458, 1344, 1282, 1238, 1116, 1028, 864 cm⁻¹; HRMS–ESI (*m/z*): [M+Na]⁺ calculated for C₂₂H₁₇NO₆SN⁺, 446.0669, found, 446.0663.



$R_f = 0.48$ (10% methanol-dichloromethane); Yellow solid, m.p. >320 °C (decomposed); $[\alpha]_D^{20} = +100.4$ ($c = 0.05$ in chloroform-methanol (v/v = 2:1)); $^1\text{H NMR}$ (500 MHz, $\text{DMSO-}d_6$) δ 9.24 (br s, 1H, NH), 8.72 (s, 1H), 8.28 (s, 1H), 8.00 (s, 1H), 3.88 (t, $J = 5.9$ Hz, 2H), 3.40 (t, $J = 5.9$ Hz, 2H), 2.84 (dd, $J = 17.1, 7.9$ Hz, 1H), 2.65 (dd, $J = 13.2, 4.1$ Hz, 1H), 2.59 (dd, $J = 17.4, 8.9$ Hz, 1H), 2.28 – 2.15 (m, 1H), 2.12 – 2.01 (m, 1H), 1.63 (td, $J = 12.9, 4.3$ Hz, 1H), 1.50 (s, 3H) ppm; $^{13}\text{C NMR}$ (125 MHz, $\text{DMSO-}d_6$) δ 178.3, 173.7, 169.4, 154.6, 147.9, 147.1, 146.2, 143.0, 137.9, 131.8, 130.8, 124.7, 123.4, 121.6, 111.3, 48.2, 39.3 (C22, buried in the peak of $\text{DMSO-}d_6$), 36.8, 31.6, 30.3, 17.8, 16.2 ppm; IR ν_{max} 3274, 2935, 2862, 1716, 1664, 1616, 1541, 1508, 1458, 1267, 1240, 1143, 935, 864, 806 cm^{-1} ; HRMS–ESI (m/z): $[\text{M}+\text{Na}]^+$ calculated for $\text{C}_{22}\text{H}_{17}\text{NO}_6\text{SNa}^+$, 446.0669, found, 446.0662.

Note: The (+)-adociaquinones A (3) and B (4) were uneasy to dissolve in organic solution, especially for the major product (+)-adociaquinone B (4), which was uneasy to dissolve in $\text{DMSO-}d_6$.

Comparison of NMR spectroscopic data of natural and synthetic (+)-xestoquinone (2), (+)-adociaquinones A (3) and B (4)



Natural product (+)-xestoquinone: $[\alpha]_D^{25} = +17.2$ ($c = 1.16$ in DCM) ^[3]

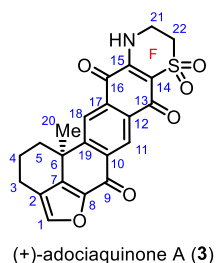
Our Synthetic (+)-xestoquinone: $[\alpha]_D^{20} = +11.2$ ($c = 1.00$ in DCM)

Table S4. Comparison of ¹H NMR spectroscopic data of natural ^[4] and synthetic ^[5] (+)-xestoquinone with this synthetic work.

position	natural (a) (Laurent's work) δ ¹ H [ppm; mult; <i>J</i> (Hz)], 300 MHz, Chloroform- <i>d</i>	synthetic (b) (Harada's work) δ ¹ H [ppm; mult; <i>J</i> (Hz)], 400 MHz, Chloroform- <i>d</i>	synthetic (c) (this work) δ ¹ H [ppm; mult; <i>J</i> (Hz)], 500 MHz, Chloroform- <i>d</i>	deviation (a-c; b-c) $\Delta\delta$ (ppm)
1	7.54; t; 1.5	7.54; br t; 1.5	7.54; br t; 1.5	0.00; 0.00
3a	2.64; dddd; 17.1, 9.8, 8.4, 1.5	2.64; dddd; 17.0, 9.9, 8.4, 1.5	2.66; dddd; 17.0, 10.0, 8.5, 1.5	-0.02; -0.02
3b	2.88; dddd; 17.1, 8.0, 2.5, 1.5	2.88; dddd; 17.0, 8.0, 2.2, 1.5	2.89; ddt; 17.1, 8.0, 1.8	-0.01; -0.01
4	2.22; m, 2H	2.28; m, 1H 2.19; m, 1H	2.29; m, 1H 2.19; m, 1H	-0.07, 0.03; -0.01, 0.00
5a	1.75; ddd; 13.0, 13.0, 4.7	1.76; ddd; 13.0, 13.0, 4.5	1.76; td; 13.1, 4.4	-0.01; 0.00
5b	2.57; ddd; 12.8, 3.6, 3.6	2.58; ddd; 13.0, 4.1, 3.0	2.58; dt; 12.9, 3.6	-0.01; 0.00
11	9.03; s	9.05; s	8.98; s	0.05; 0.07
14	7.02; s, 2H	7.06; d; 10.4	7.05; d; 10.0	-0.03, 0.00;
15		7.03; d; 10.4	7.02; d; 10.0	0.01, 0.01
18	8.23; s	8.25; s	8.22; s	0.01; 0.03
20	1.53; s	1.54; s	1.53; s	0.00; 0.01

Table S5. Comparison of ^{13}C NMR spectroscopic data of natural $^{[4]}$ (+)-xestoquinone with this synthetic work.

position	natural (a) (Laurent's work) $\delta^{13}\text{C}$ [ppm], 75 MHz, Chloroform- <i>d</i>	synthetic (b) (this work) $\delta^{13}\text{C}$ [ppm], 125 MHz, Chloroform- <i>d</i>	deviation (a-b) $\Delta\delta$ (ppm)
1	145.0	145.0	0.0
2	121.5	121.6	-0.1
3	16.9	16.9	0.0
4	18.4	18.4	0.0
5	31.2	31.2	0.0
6	37.4	37.4	0.0
7	147.3	147.3	0.0
8	144.0	144.0	0.0
9	170.3	170.1	0.2
10	137.9	137.9	0.0
11	127.0	126.9	0.1
12	130.3	130.3	0.0
13	183.9	183.8	0.1
14	139.4	139.3	0.1
15	138.7	138.7	0.0
16	184.7	184.6	0.1
17	133.2	133.2	0.0
18	123.2	123.2	0.0
19	156.2	156.2	0.0
20	32.6	32.6	0.0



Natural product (+)-adociaquinone A: $[\alpha]_{\text{D}} = +31.7$ ($c = 4.66$ in MeCN) ^[6]

Harada's Synthetic (+)-adociaquinone A: $[\alpha]_{\text{D}}^{20} = +70$ ($c = 0.107$ in chloroform -methanol
(v/v = 2:1)) ^[7]

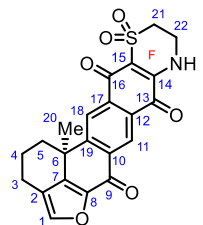
Our Synthetic (+)-adociaquinone A: $[\alpha]_{\text{D}}^{20} = +65.2$ ($c = 0.1$ in chloroform -methanol (v/v =
2:1))

Table S6. Comparison of ¹H NMR spectroscopic data of natural ^[6] and synthetic ^[7] (+)-adociaquinone A with this synthetic work.

position	natural (a) (Ireland's work) $\delta^1\text{H}$ [ppm; mult; J (Hz)], 500 MHz, DMSO- <i>d</i> ₆	synthetic (b) (Harada's work) $\delta^1\text{H}$ [ppm; mult; J (Hz)], 500 MHz, DMSO- <i>d</i> ₆	synthetic (c) (this work) $\delta^1\text{H}$ [ppm; mult; J (Hz)], 500 MHz, DMSO- <i>d</i> ₆	deviation (a-c; b-c) $\Delta\delta$ (ppm)
1	7.97; s	8.00; s	8.00; s	-0.03; 0.00
3a	2.56; dd; 17.0, 9.0	2.60; m	2.59; m	-0.03; 0.01
3b	2.82; dd; 17.0, 8.5	2.84; dd; 17.4, 8.2,	2.84; dd; 17.0, 8.0	-0.02; 0.00
4a	2.06; m	2.08; m	2.08; m	-0.02; 0.00
4b	2.21; m	2.23; m	2.22; m	-0.01; 0.01
5a	1.63; dt; 13.0, 4.5	1.66; ddd; 12.9, 12.9, 4.2	1.65; td; 13.0, 4.4	-0.02; 0.01
5b	2.60; m	2.60; m	2.61; m	-0.01; -0.01
11	8.65; s	8.69; s	8.68; s	-0.03; 0.01
18	8.24; s	8.26; s	8.26; s	-0.02; 0.00
20	1.48; s	1.50; s	1.50; s	-0.02; 0.00
21	3.87; m	3.88; br s	3.88; t; 6.0	-0.01; 0.00
22	3.39; t; 6.0	3.40; t; 6.0	3.40; t; 6.0	-0.01; 0.00
NH	9.33; br s	9.34; br s	9.35; br s	-0.02; -0.01

Table S7. Comparison of ^{13}C NMR spectroscopic data of natural ^[6] and synthetic ^[7] (+)-adociaquinone A with this synthetic work.

position	natural (a) (Ireland's work) $\delta^{13}\text{C}$ [ppm], 125 MHz, DMSO- d_6	synthetic (b) (Harada's work) $\delta^{13}\text{C}$ [ppm], 125 MHz, DMSO- d_6	synthetic (c) (this work) $\delta^{13}\text{C}$ [ppm], 125 MHz, DMSO- d_6	deviation (a-c) $\Delta\delta$ (ppm)
1	146.2	--	146.2	0.0
2	121.6	--	121.7	-0.1
3	16.3	--	16.3	0.0
4	17.8	--	17.9	-0.1
5	30.3	--	30.3	0.0
6	37.2	--	37.2	0.0
7	147.7	--	147.8	-0.1
8	143.0	--	143.0	0.0
9	169.3	--	169.4	-0.1
10	135.9	--	136.0	-0.1
11	125.4	--	125.4	0.0
12	128.6	--	128.6	0.0
13	173.7	--	173.8	-0.1
14	111.7	--	111.7	0.0
15	147.2	--	147.2	0.0
16	177.9	--	177.9	0.0
17	134.4	--	134.4	0.0
18	122.9	--	123.0	-0.1
19	157.3	--	157.4	-0.1
20	31.8	--	31.8	0.0
21	39.4	--	39.4, buried in the peak of DMSO- d_6	0.0
22	48.2	--	48.2	0.0



(+)-adociaquinone B (4)

Natural product (+)-adociaquinone B: $[\alpha]_D = +21.5$ ($c = 1.86$ in MeCN) ^[6]

Harada's Synthetic (+)-adociaquinone B: $[\alpha]_D^{20} = +74$ ($c = 0.0668$ in chloroform -methanol
($v/v = 2:1$)) ^[7]

Our Synthetic (+)-adociaquinone B: $[\alpha]_D^{20} = +100.4$ ($c = 0.05$ in chloroform -methanol (v/v
 $= 2:1$))

Table S8. Comparison of ¹H NMR spectroscopic data of natural ^[6] and synthetic ^[7] (+)-adociaquinone B with this synthetic work.

position	natural (a) (Ireland's work) δ ¹ H [ppm; mult; <i>J</i> (Hz)], 500 MHz, DMSO- <i>d</i> ₆	synthetic (b) (Harada's work) δ ¹ H [ppm; mult; <i>J</i> (Hz)], 500 MHz, DMSO- <i>d</i> ₆	synthetic (c) (this work) δ ¹ H [ppm; mult; <i>J</i> (Hz)], 500 MHz, DMSO- <i>d</i> ₆	deviation (a-c; b-c) $\Delta\delta$ (ppm)
1	7.99; s	8.00; s	8.00; s	-0.01; 0.00
3a	2.58; dd; 16.5, 8.5	2.59; dd; 16.9, 9.0	2.59; dd; 17.4, 8.9	-0.01; 0.00
3b	2.83; dd; 16.5, 7.5	2.84; dd; 16.9, 7.7	2.84; dd; 17.1, 7.9	-0.01; 0.00
4a	2.05; m	2.07; m	2.07; m	-0.02; 0.00
4b	2.20; m	2.21; m	2.21; m	-0.01; 0.00
5a	1.62; dt; 13.0, 4.0	1.63; ddd; 12.8, 12.8, 4.2	1.63; td; 12.9, 4.3	-0.01; 0.00
5b	2.64; m	2.65; ddd; 12.8, 3.0, 3.0	2.65; dd; 13.2, 4.1	-0.01; 0.00
11	8.70; s	8.72; s	8.72; s	-0.02; 0.00
18	8.26; s	8.28; s	8.28; s	-0.02; 0.00
20	1.49; s	1.50; s	1.50; s	-0.01; 0.00
21	3.38; t; 6.0	3.40; buried in the peak of water	3.40; t; 5.9	-0.02; 0.00
22	3.86; m	3.88; t; 5.9	3.88; t; 5.9	-0.02; 0.00
NH	9.23; br s	8.90; br s	9.24; br s	-0.01; -0.34

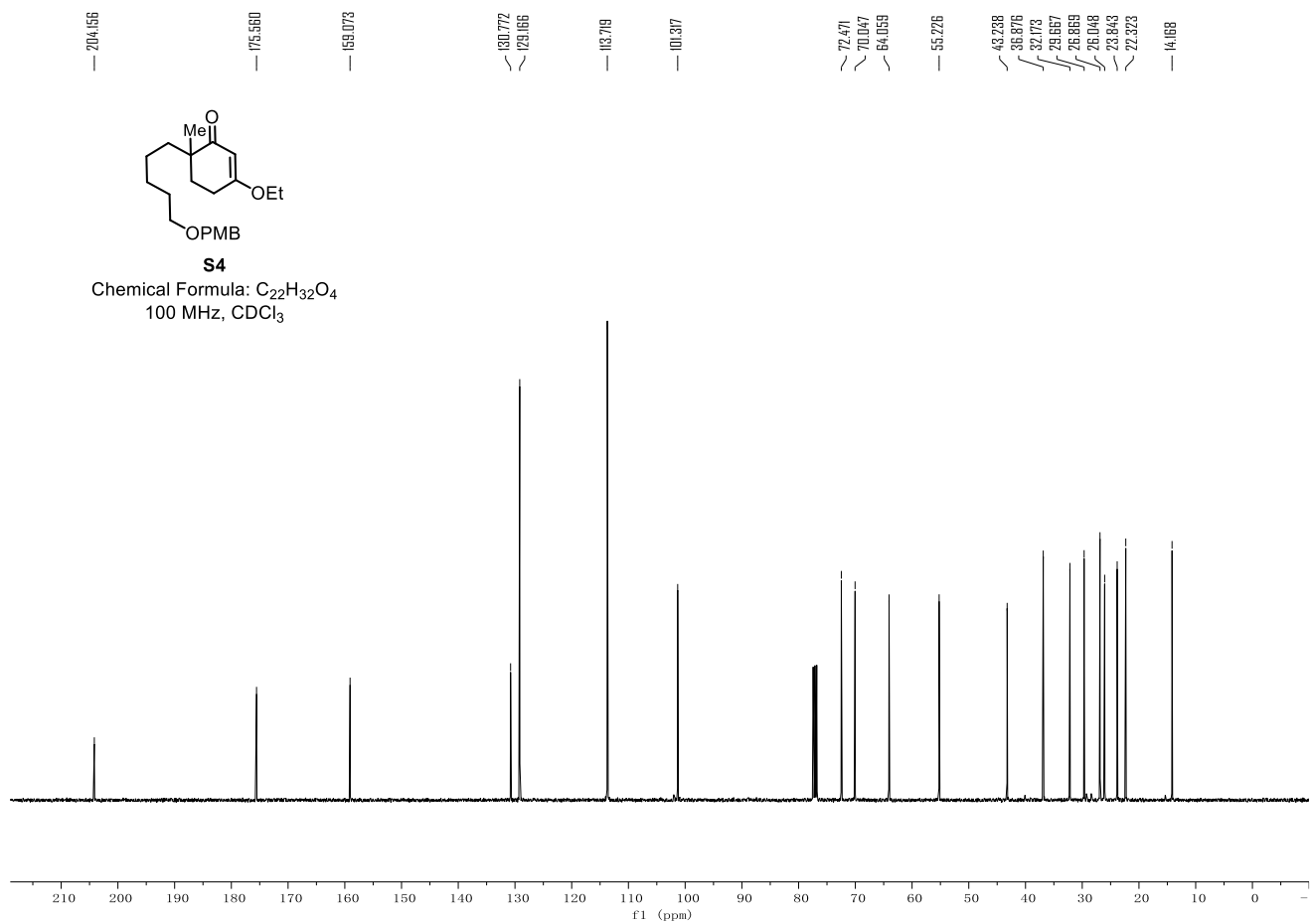
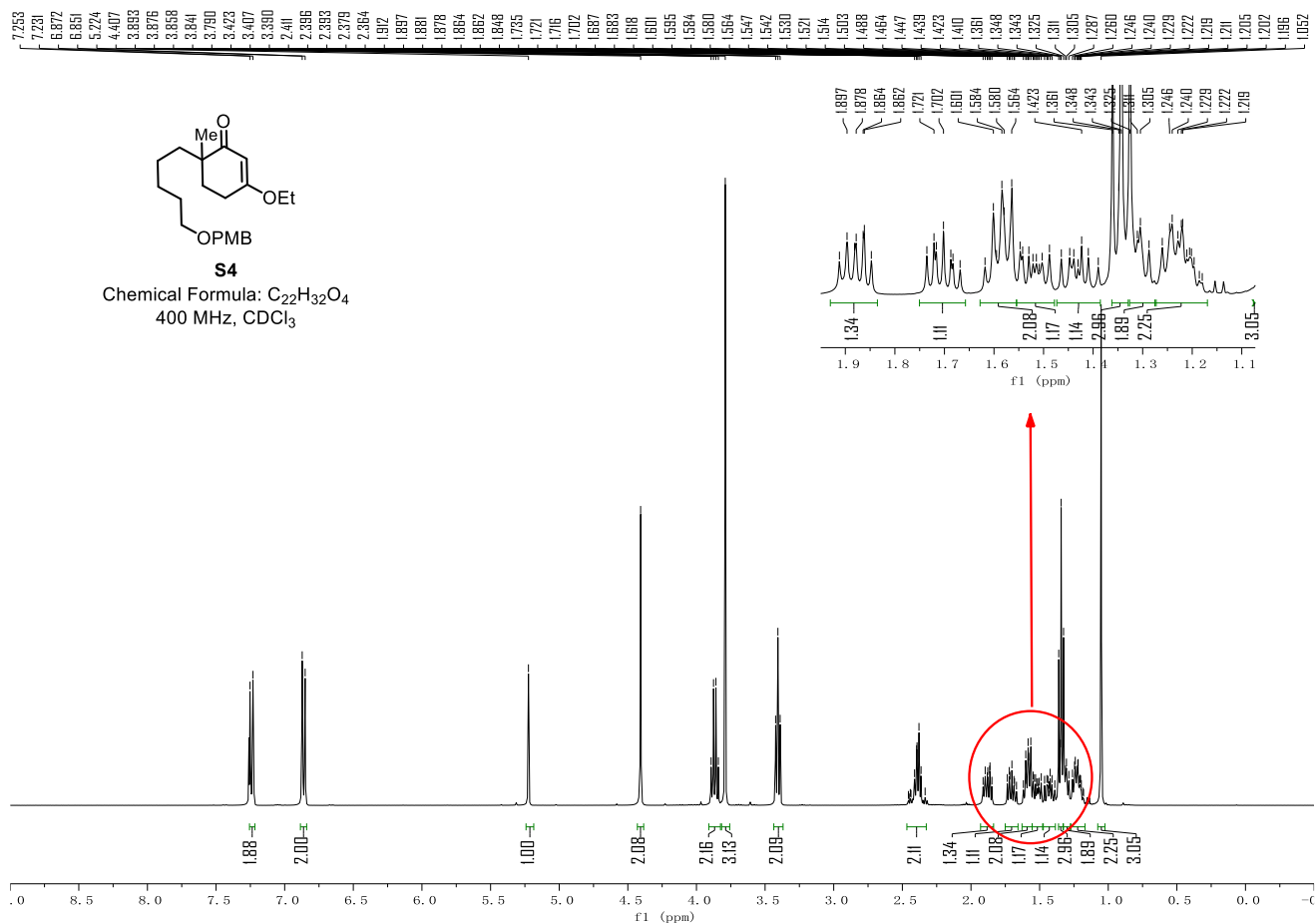
Table S9. Comparison of ^{13}C NMR spectroscopic data of natural ^[6] and synthetic ^[7] (+)-adociaquinone B with this synthetic work.

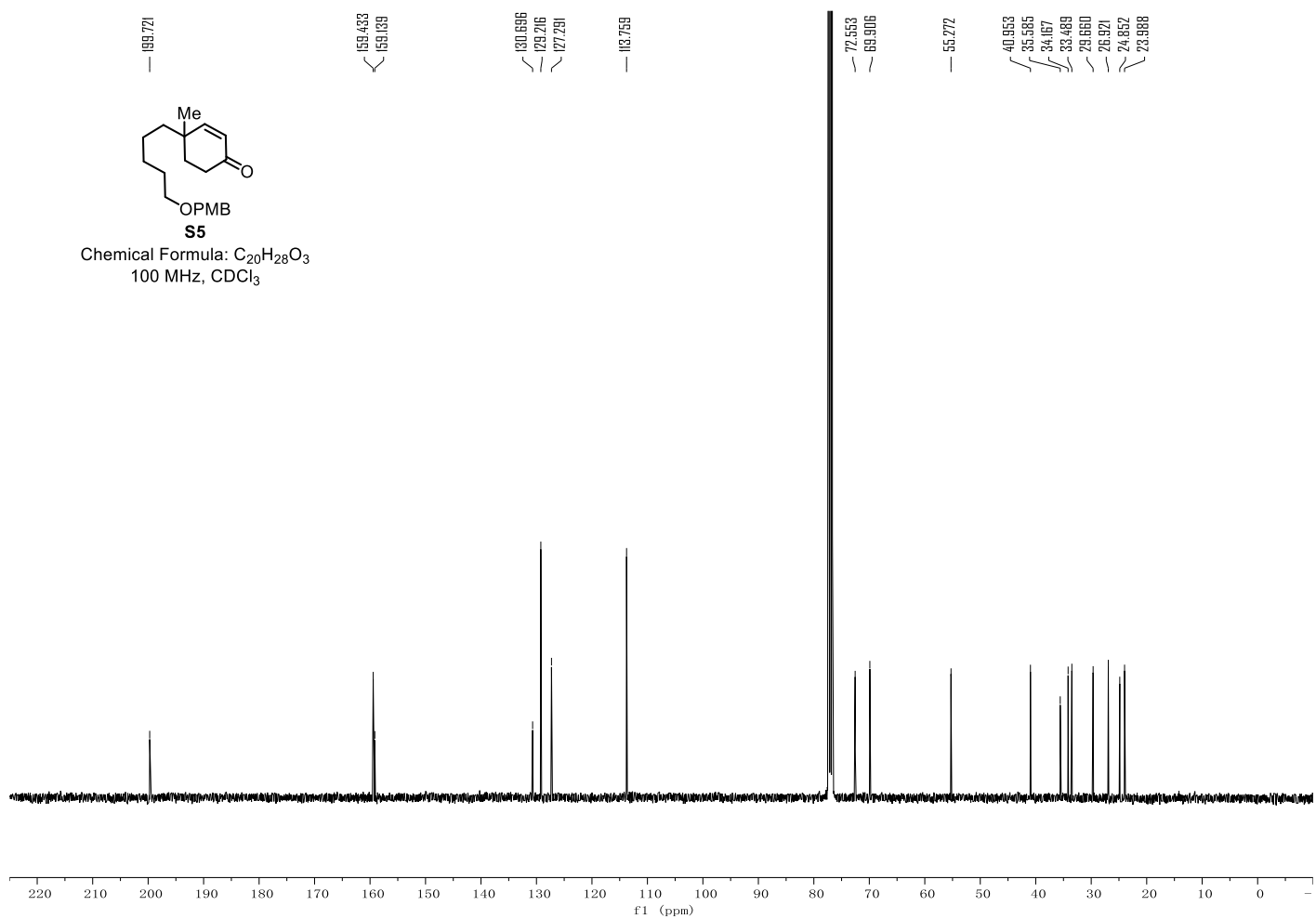
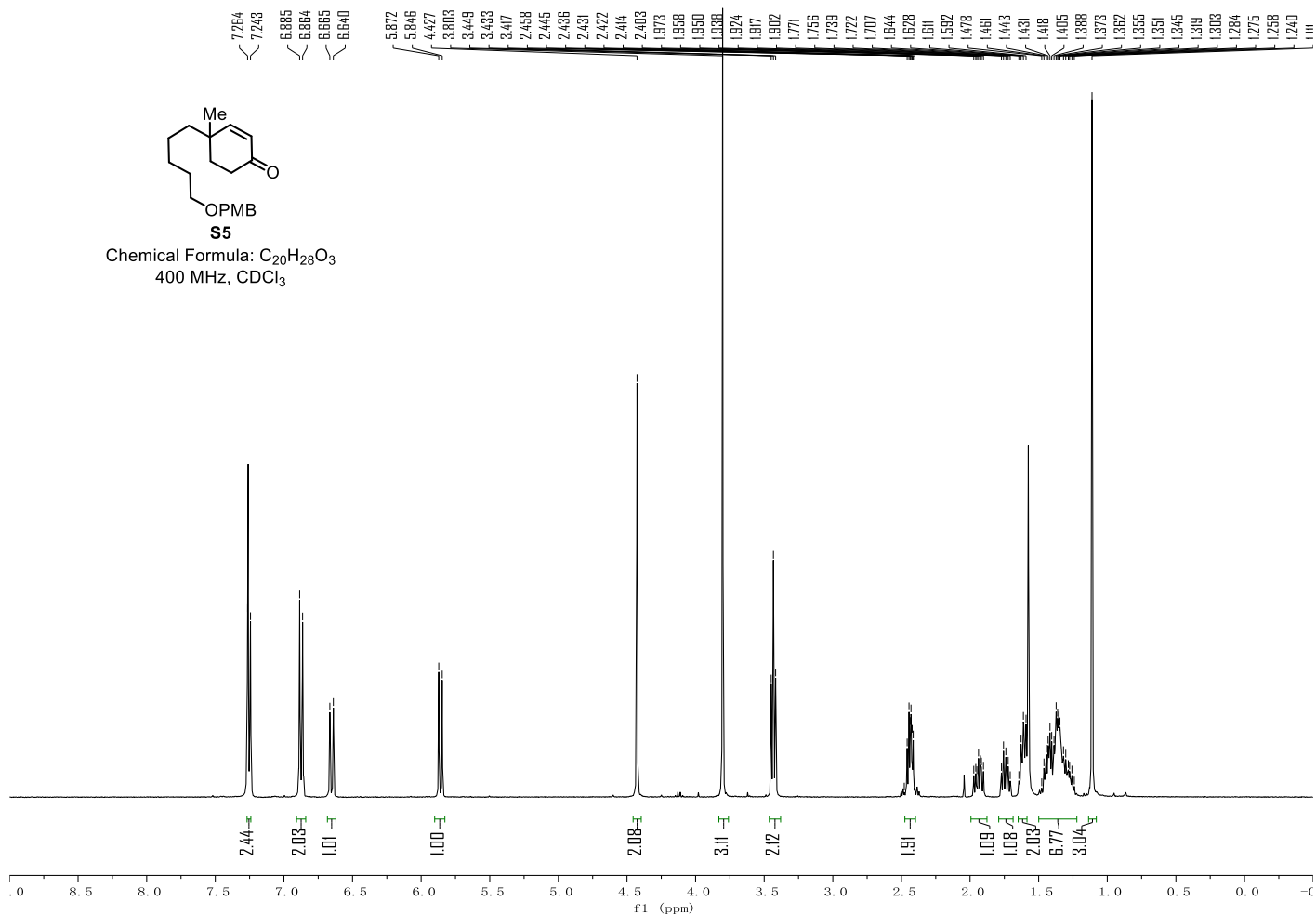
position	natural (a) (Ireland's work) $\delta^{13}\text{C}$ [ppm], 125 MHz, DMSO- <i>d</i> ₆	synthetic (b) (Harada's work) $\delta^{13}\text{C}$ [ppm], 125 MHz, DMSO- <i>d</i> ₆	synthetic (c) (this work) $\delta^{13}\text{C}$ [ppm], 125 MHz, DMSO- <i>d</i> ₆	deviation (a-c; b-c) $\Delta\delta$ (ppm)
1	146.2	146.1	146.2	0.0; -0.1
2	121.6	121.6	121.6	0.0; 0.0
3	16.2	16.2	16.2	0.0; 0.0
4	17.8	17.8	17.8	0.0; 0.0
5	30.3	30.3	30.3	0.0; 0.0
6	36.8	36.8	36.8	0.0; 0.0
7	147.9	147.9	147.9	0.0; 0.0
8	143.1	143.1	143.0	0.1; 0.1
9	169.4	169.4	169.4	0.0; 0.0
10	137.9	137.9	137.9	0.0; 0.0
11	124.7	124.8	124.7	0.0; 0.1
12	130.9	130.9	130.8	0.1; 0.1
13	173.7	173.7	173.7	0.0; 0.0
14	147.1	147.1	147.1	0.0; 0.0
15	111.3	111.4	111.3	0.0; 0.1
16	178.3	178.3	178.3	0.0; 0.0
17	131.8	131.8	131.8	0.0; 0.0
18	123.4	123.4	123.4	0.0; 0.0
19	154.6	154.6	154.6	0.0; 0.0
20	31.6	31.6	31.6	0.0; 0.0
21	48.2	48.3	48.2	0.0; 0.1
22	39.3	around 40, buried in the peak of DMSO- <i>d</i> ₆	39.3, buried in the peak of DMSO- <i>d</i> ₆	0.0; 0.7

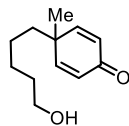
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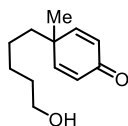
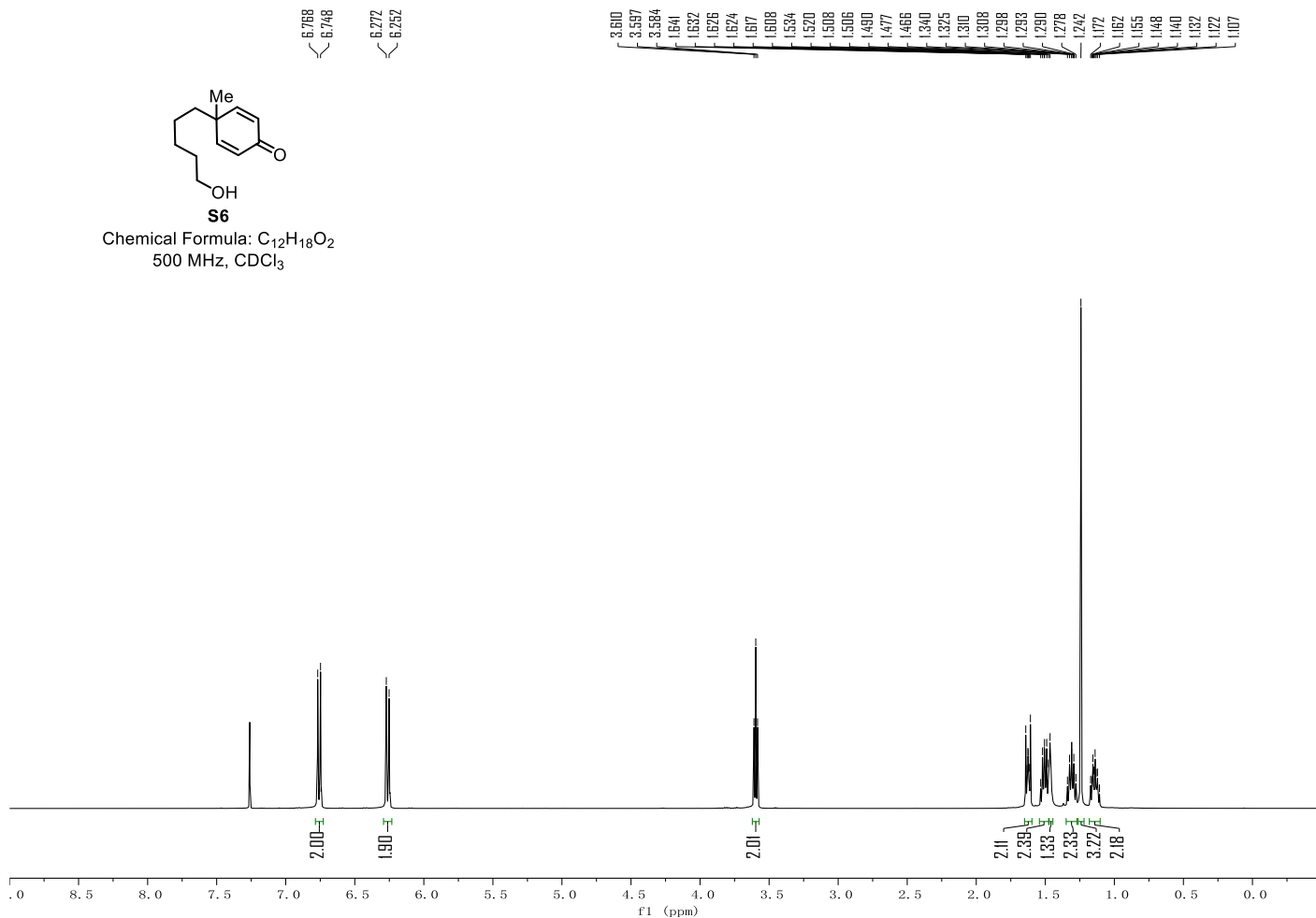
¹H and ¹³C NMR, HPLC spectra of the synthetic intermediates and products



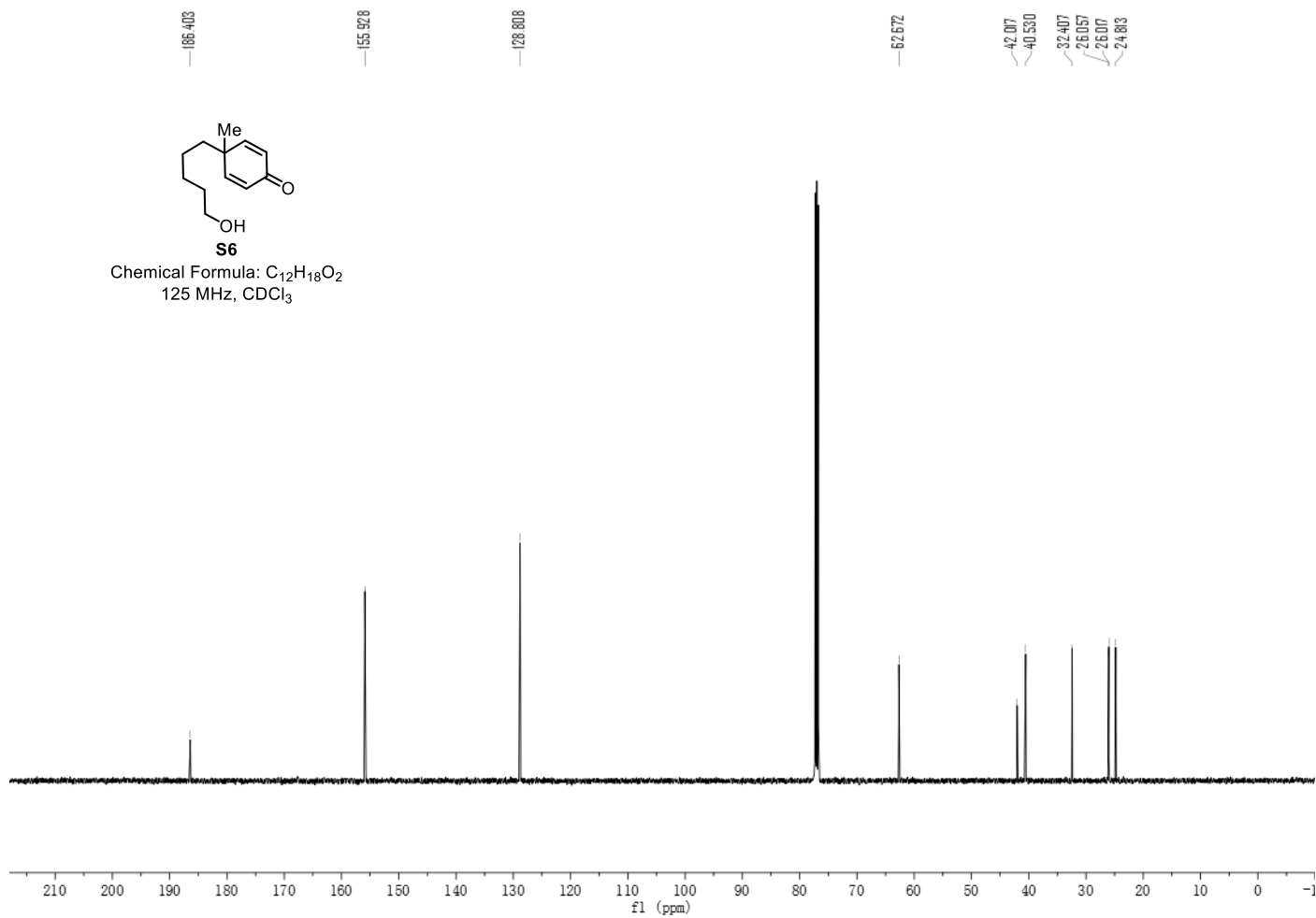


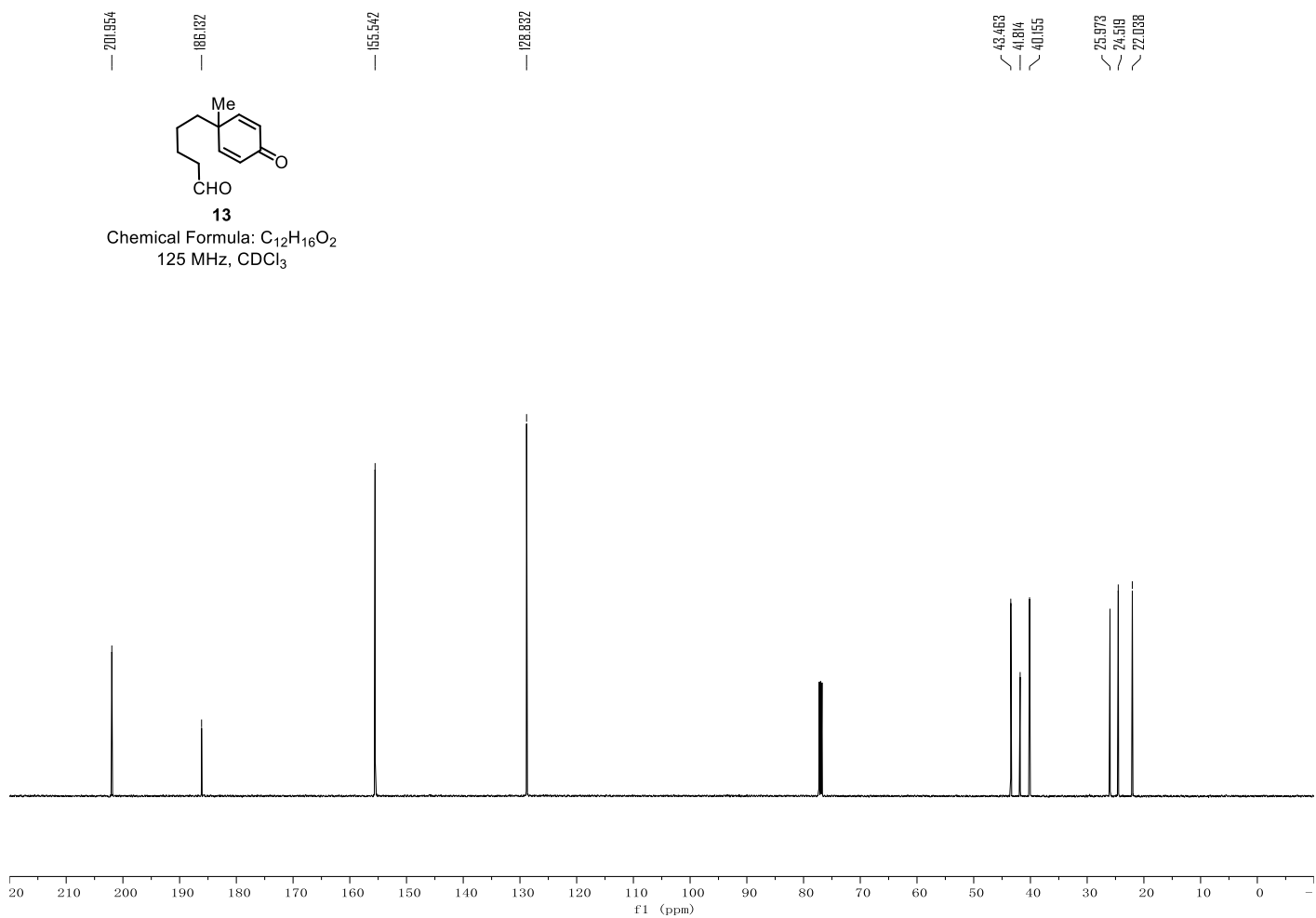
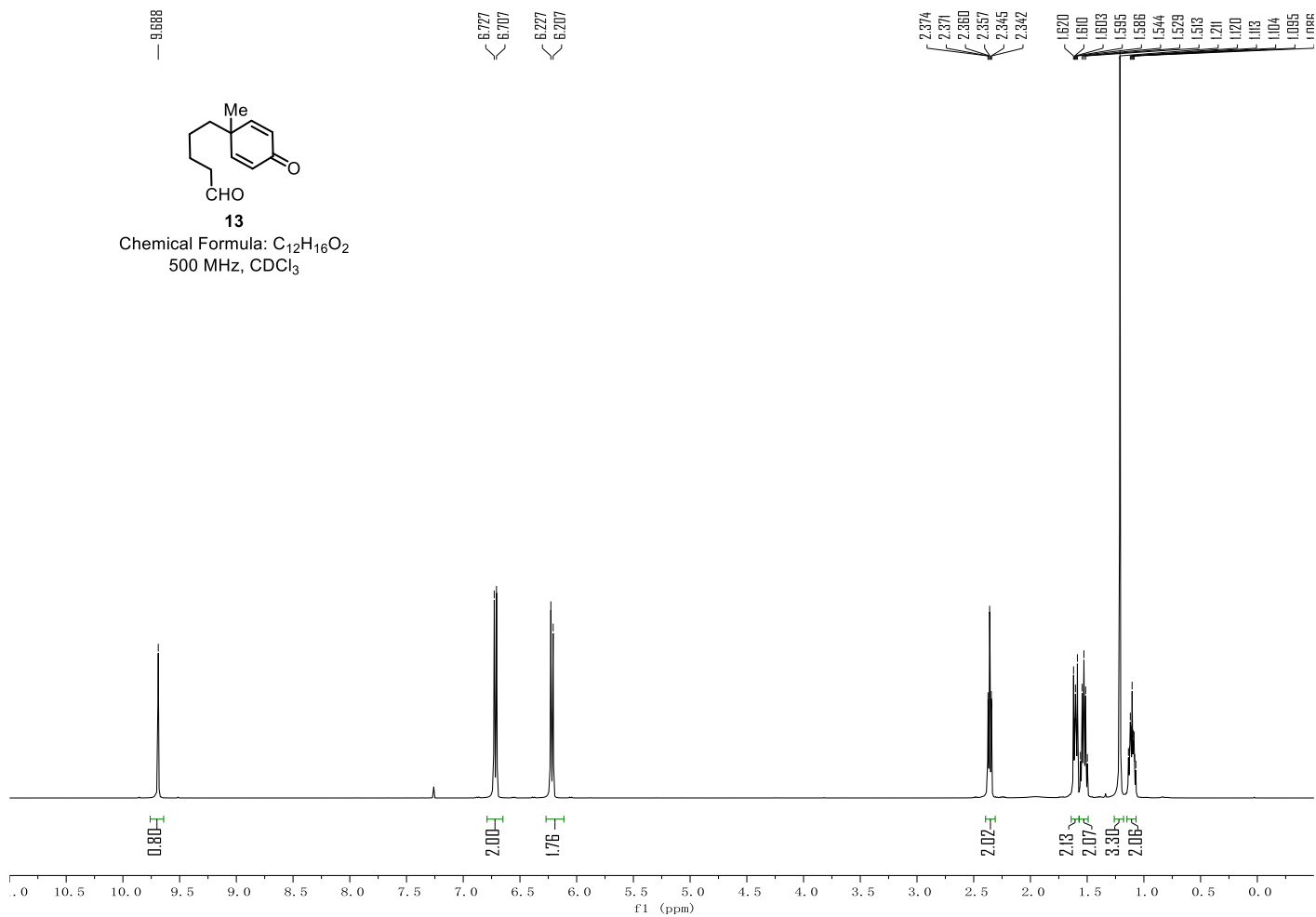


S6
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 500 MHz, CDCl₃

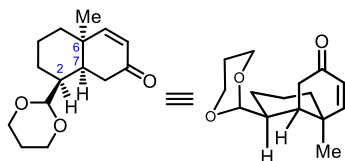


S6
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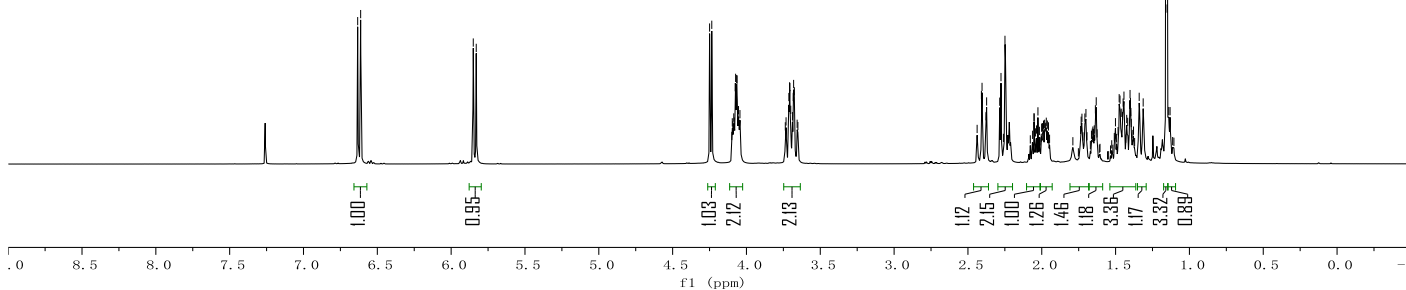




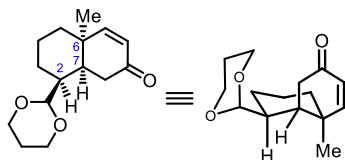
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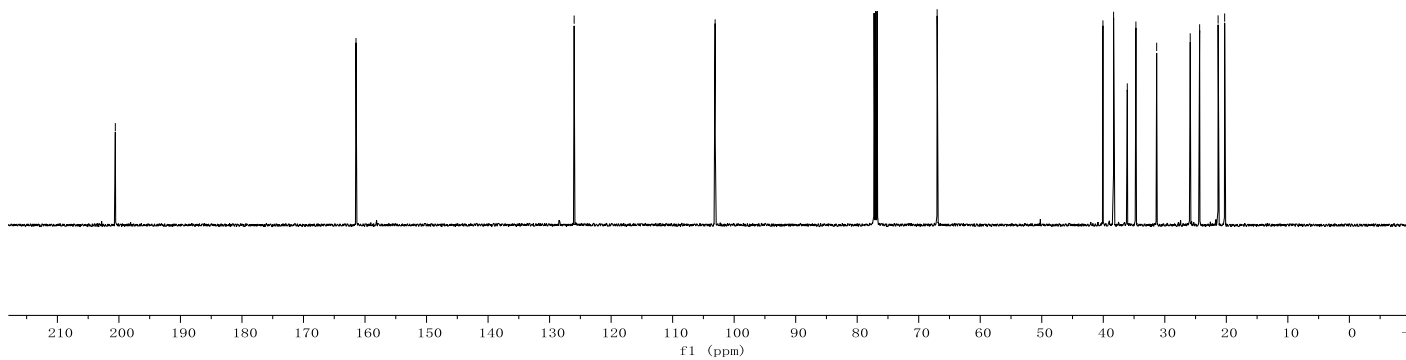
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500 MHz, CDCl₃

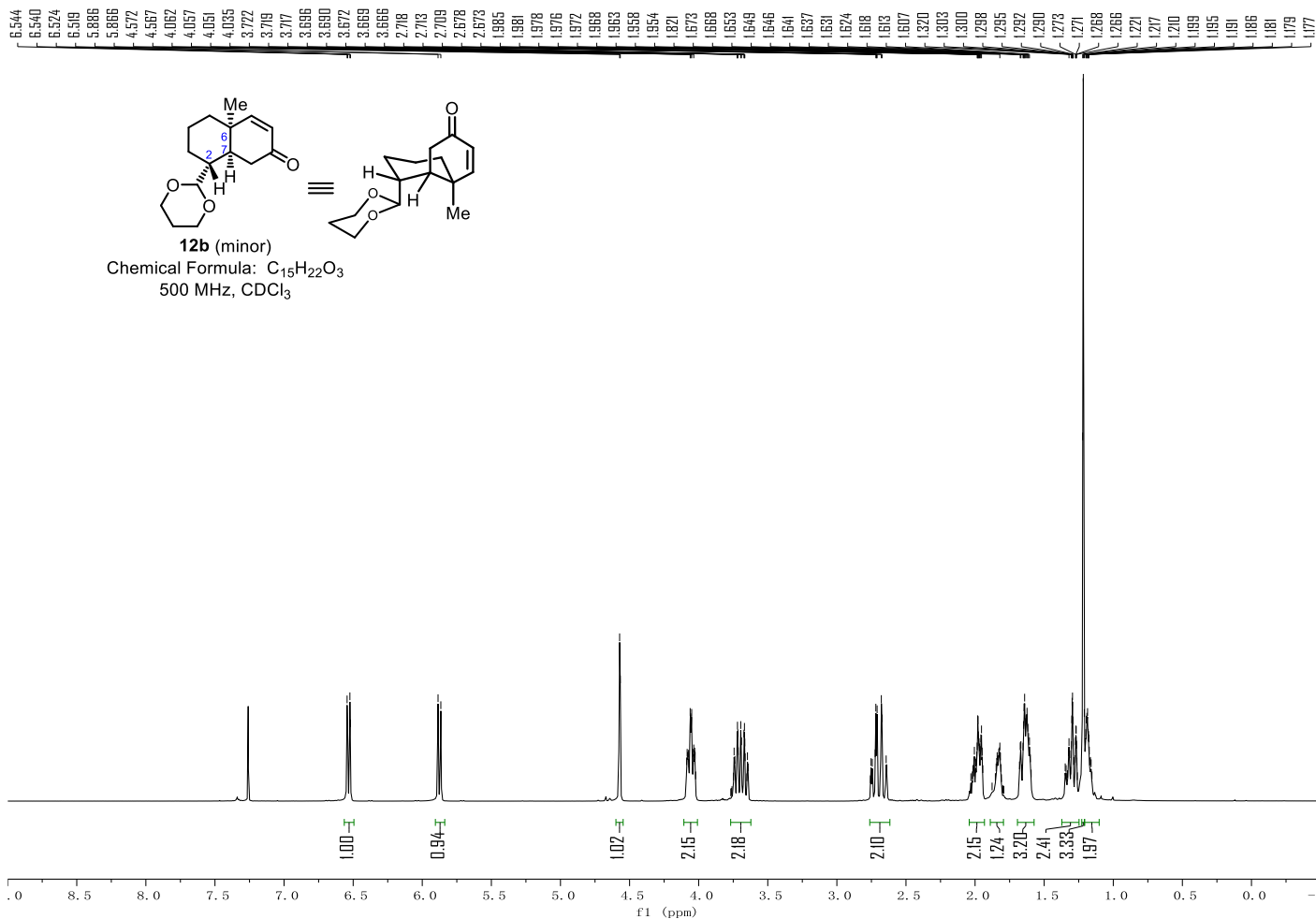


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12a (major)
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125 MHz, CDCl₃





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66.981

42.028

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40.194

37.534

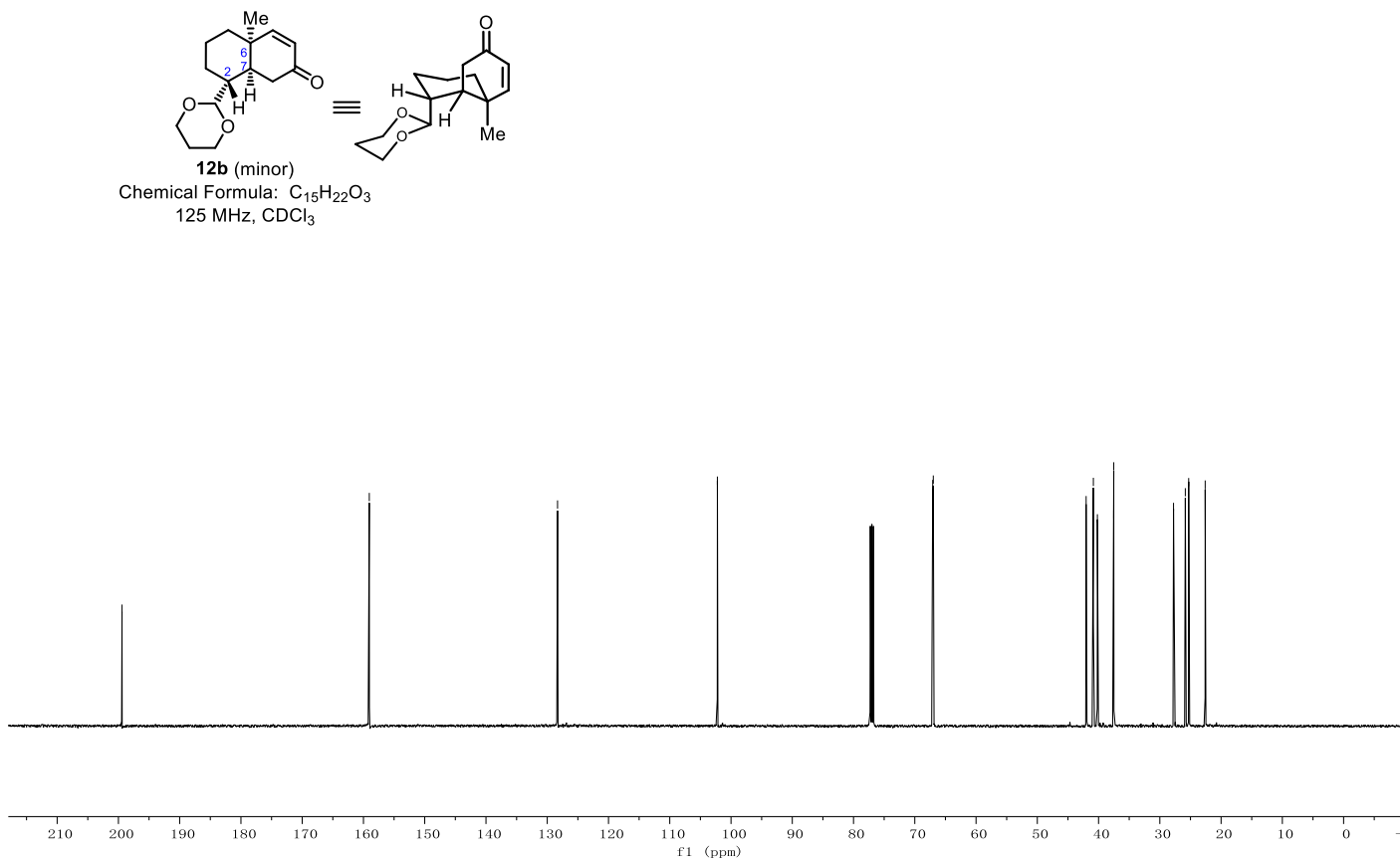
37.520

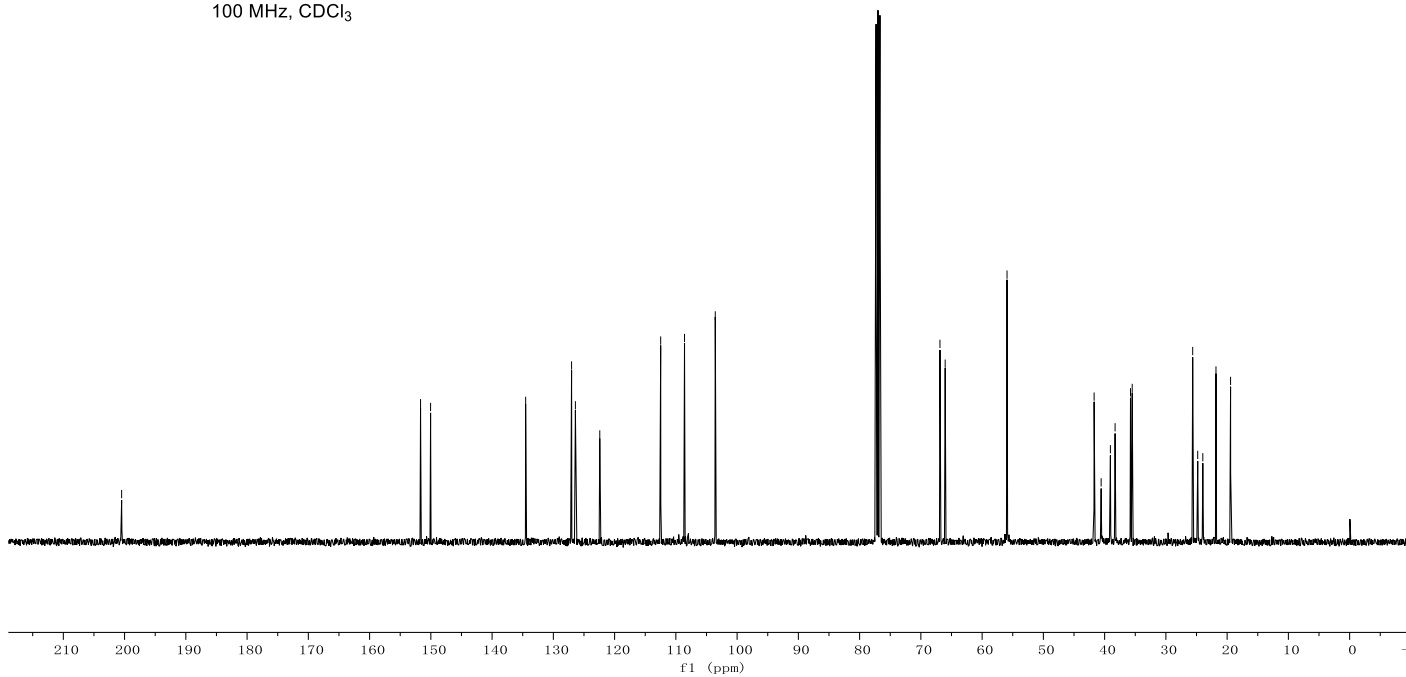
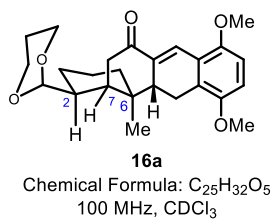
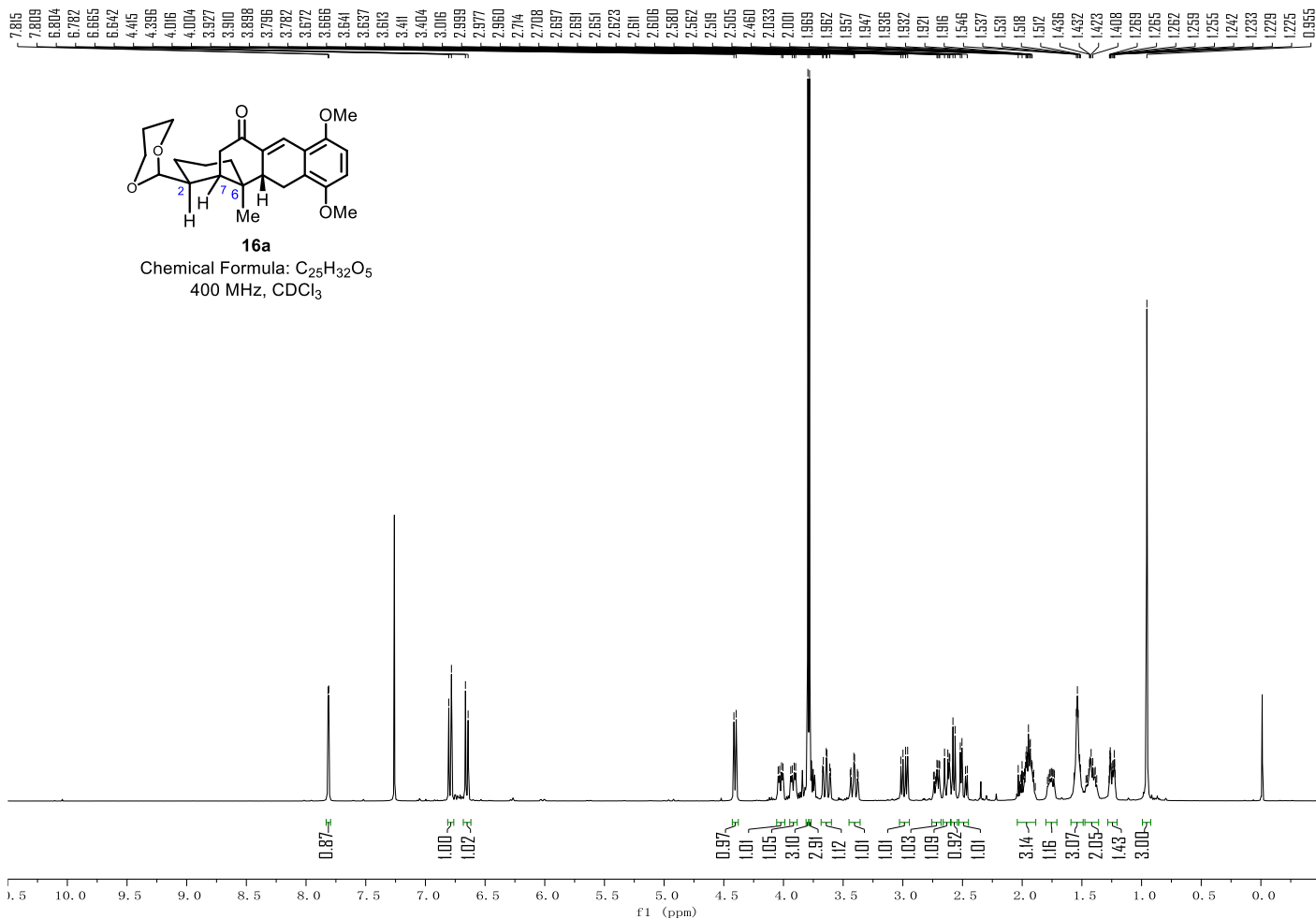
27.766

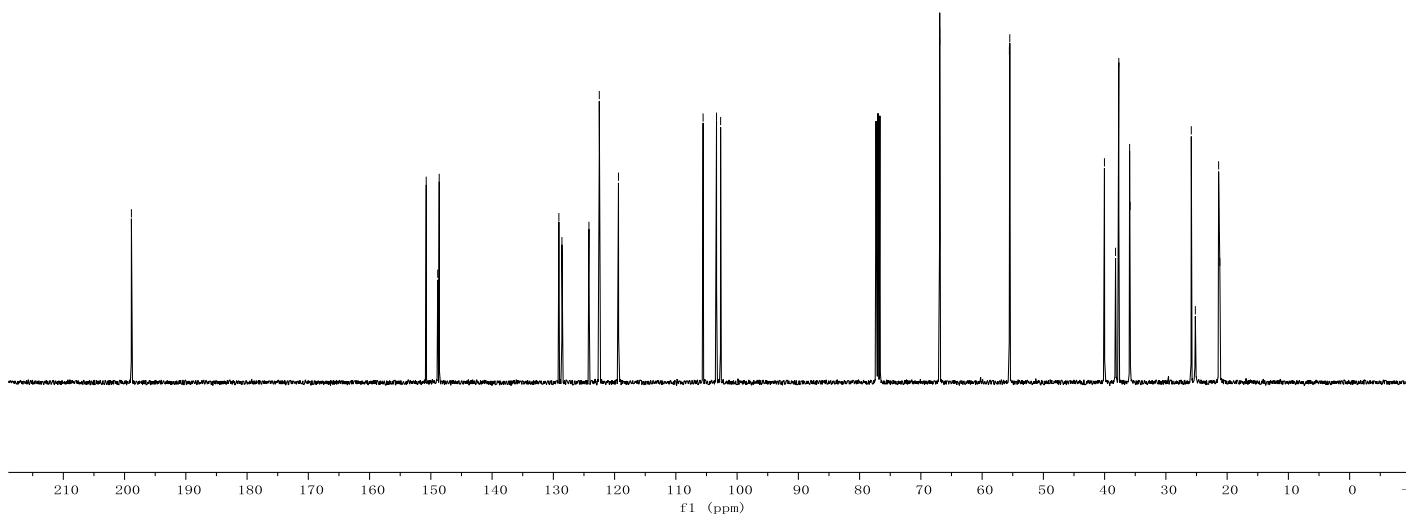
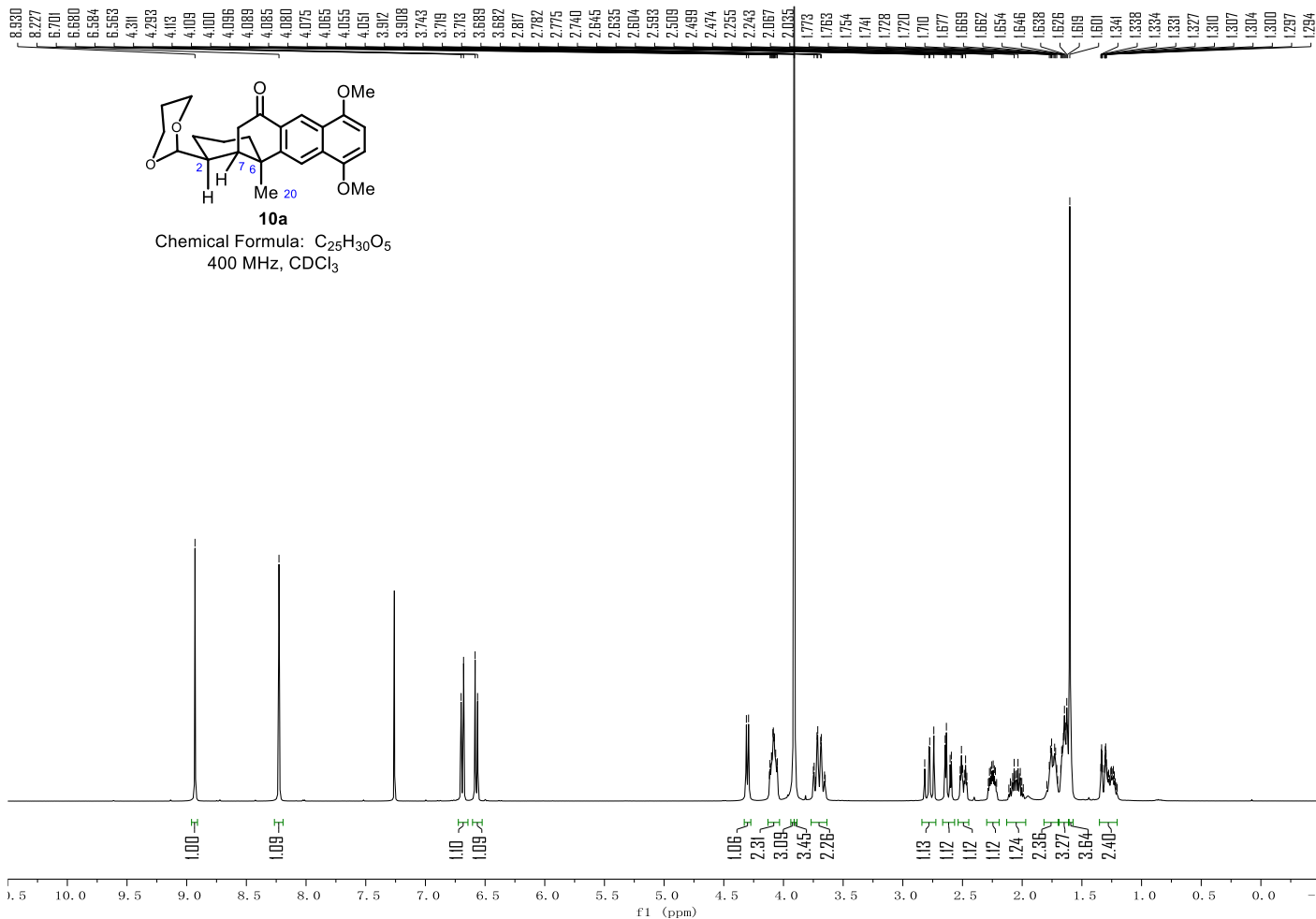
25.839

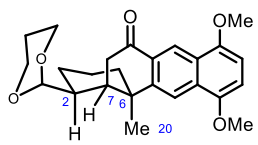
25.290

22.596



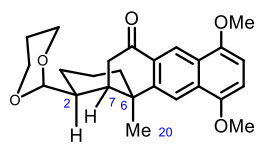
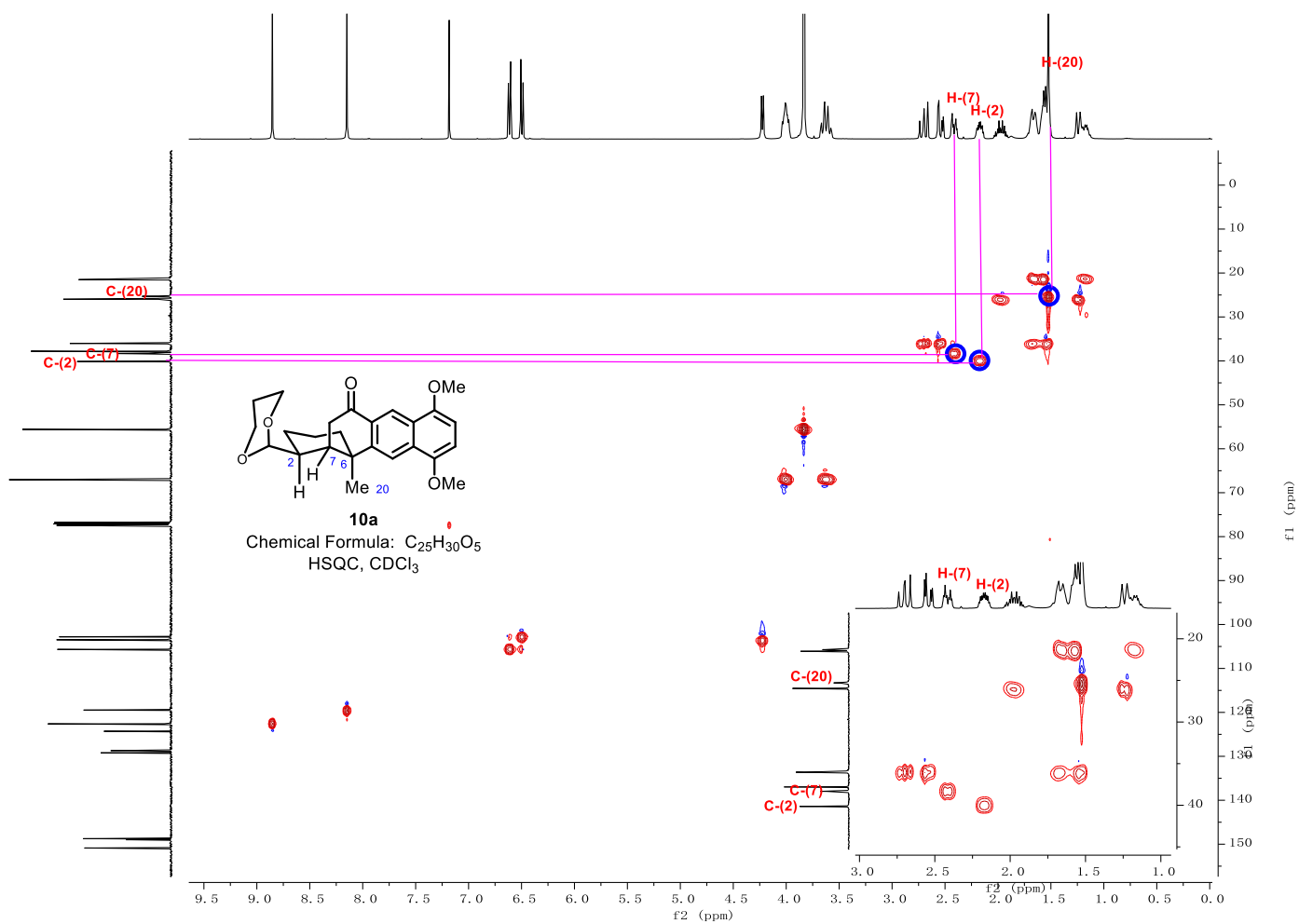
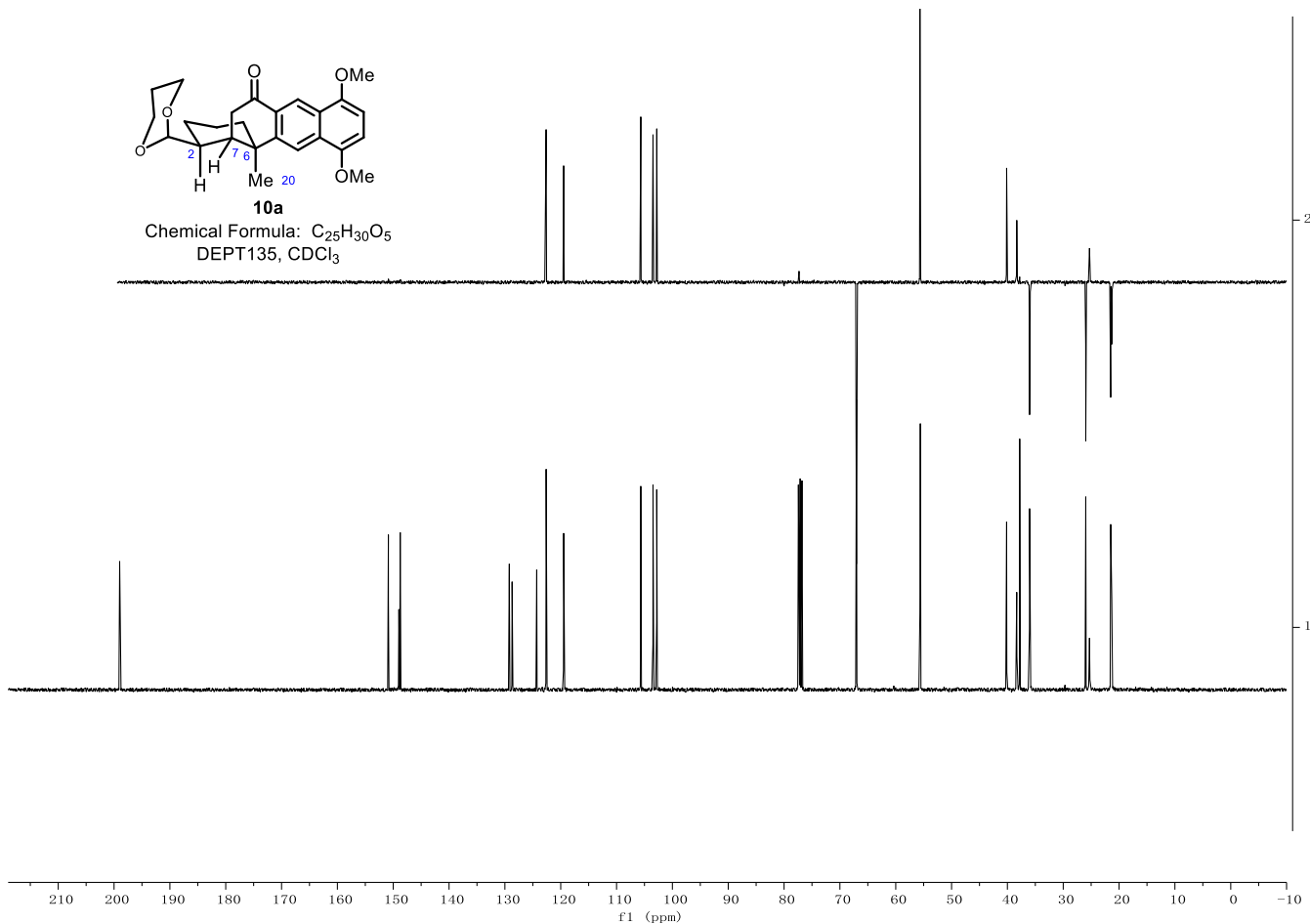






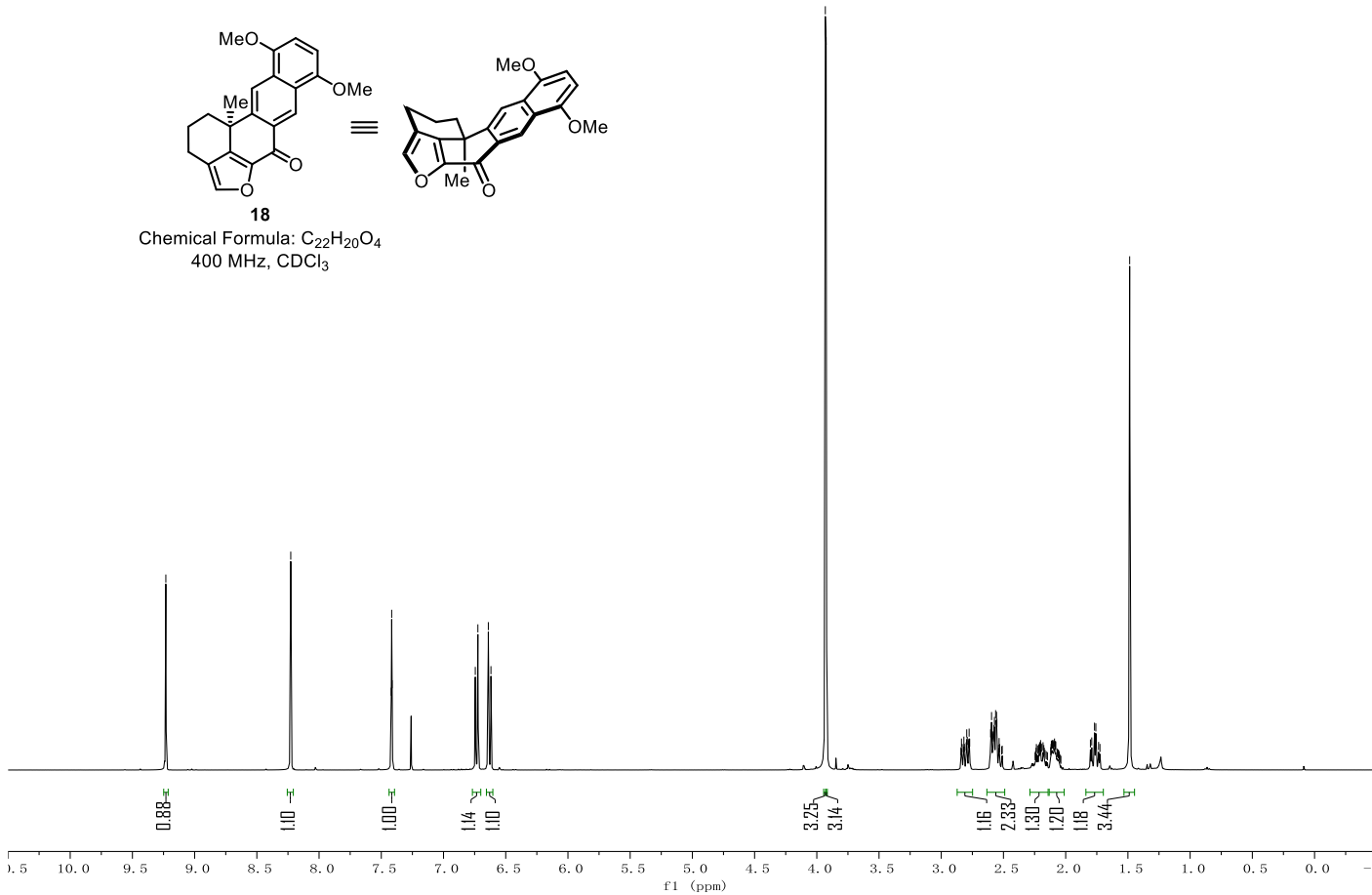
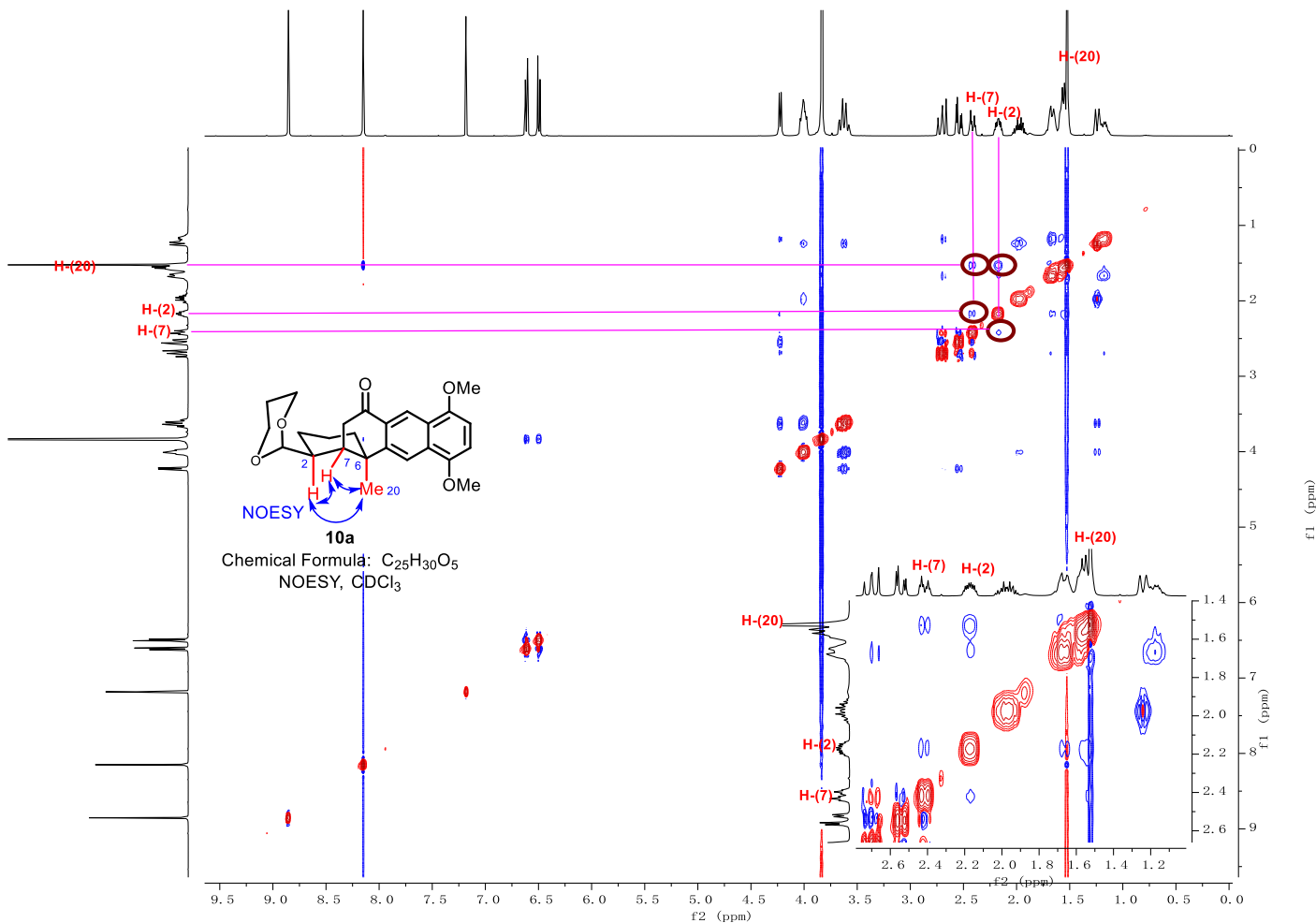
10a

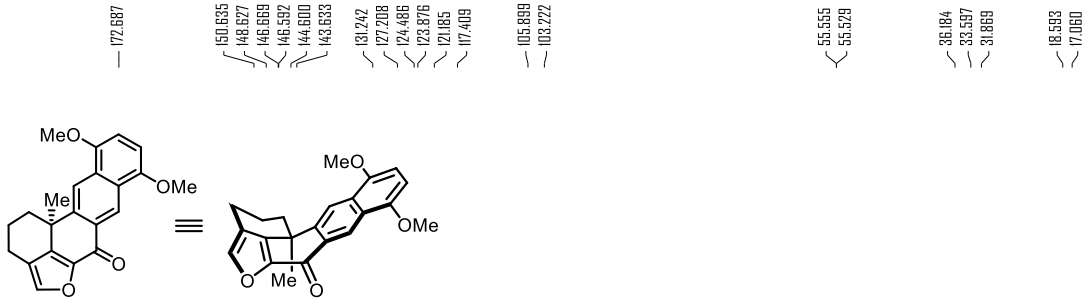
Chemical Formula: $C_{25}H_{30}O_5$
DEPT135, $CDCl_3$



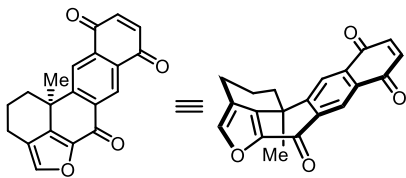
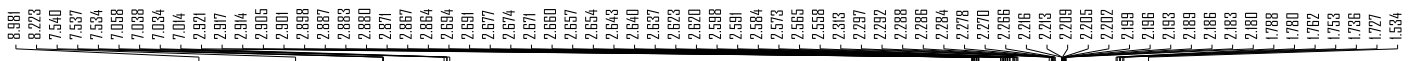
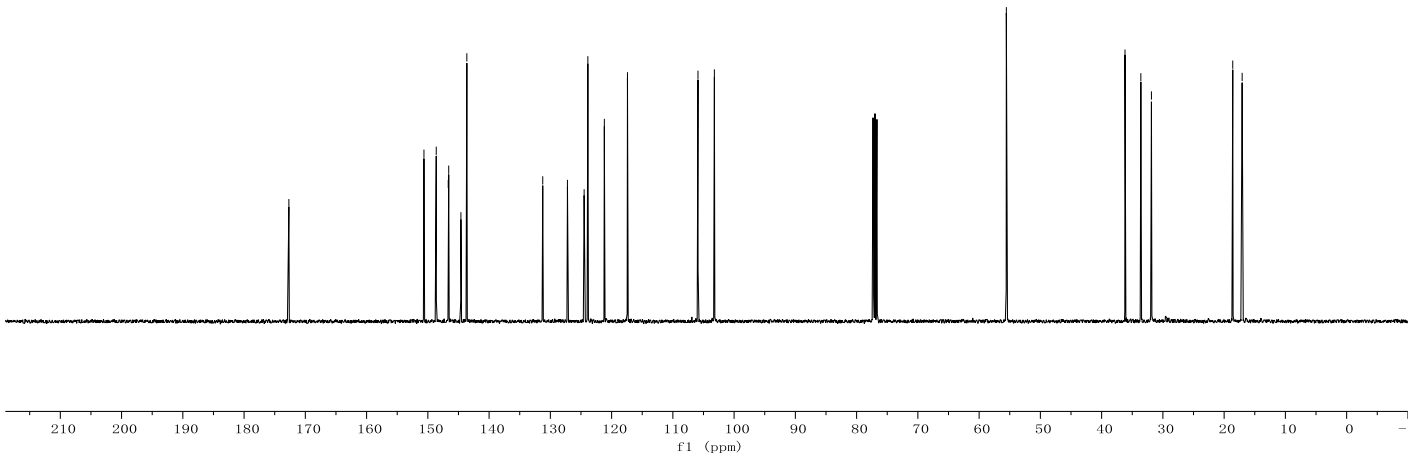
10a

Chemical Formula: $C_{25}H_{30}O_5$
HSQC, $CDCl_3$

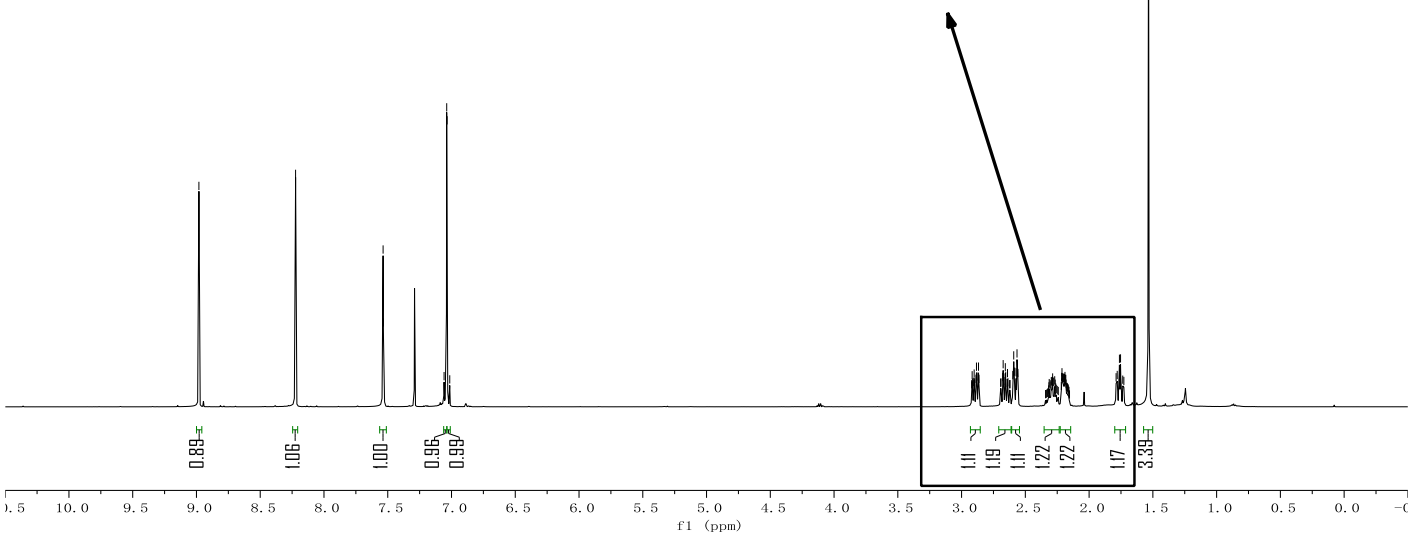
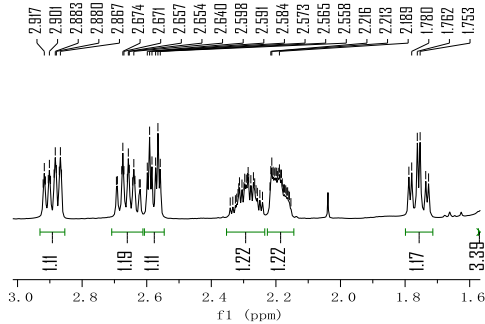




Chemical Formula: $C_{22}H_{20}O_4$
100 MHz, $CDCl_3$

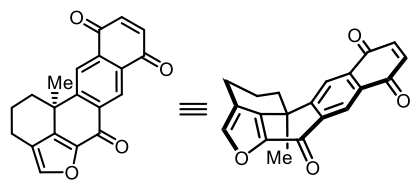


(+)-xestoquinone (2)
Chemical Formula: $C_{20}H_{14}O_4$
500 MHz, $CDCl_3$

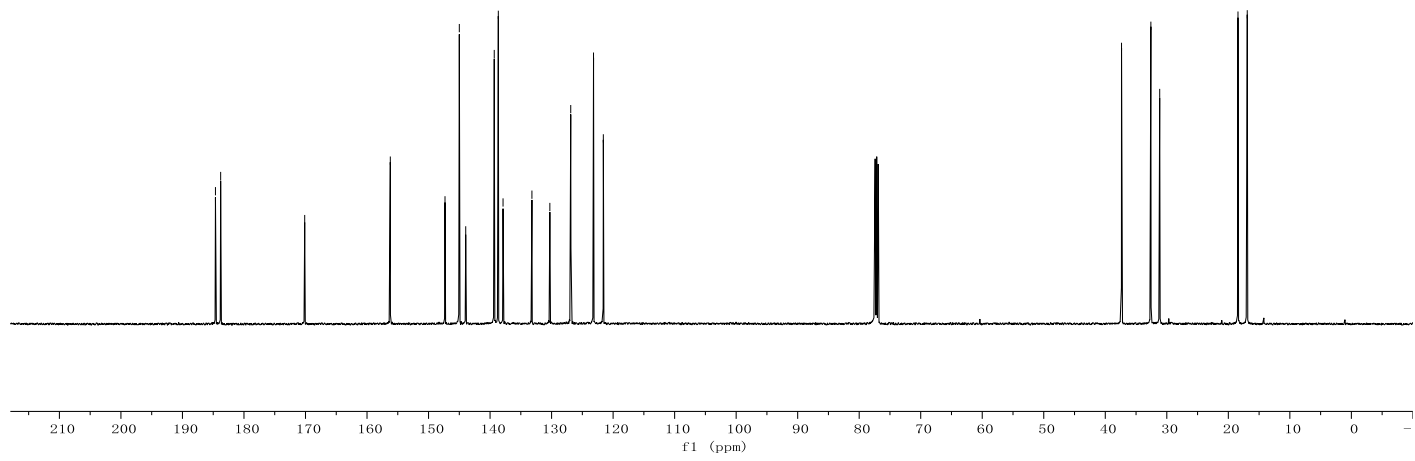


184.636
183.790
170.127
156.215
147.323
144.999
143.556
139.319
138.658
137.889
133.195
130.281
126.886
123.683
121.588

37.360
32.592
31.556
18.431
16.923

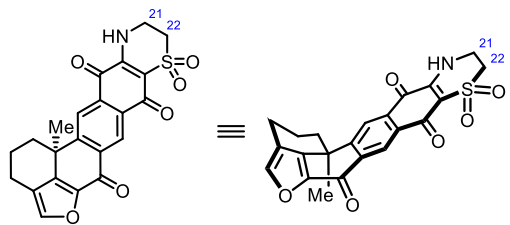


(+)-xestoquinone (2)
Chemical Formula: C₂₀H₁₄O₄
125 MHz, CDCl₃

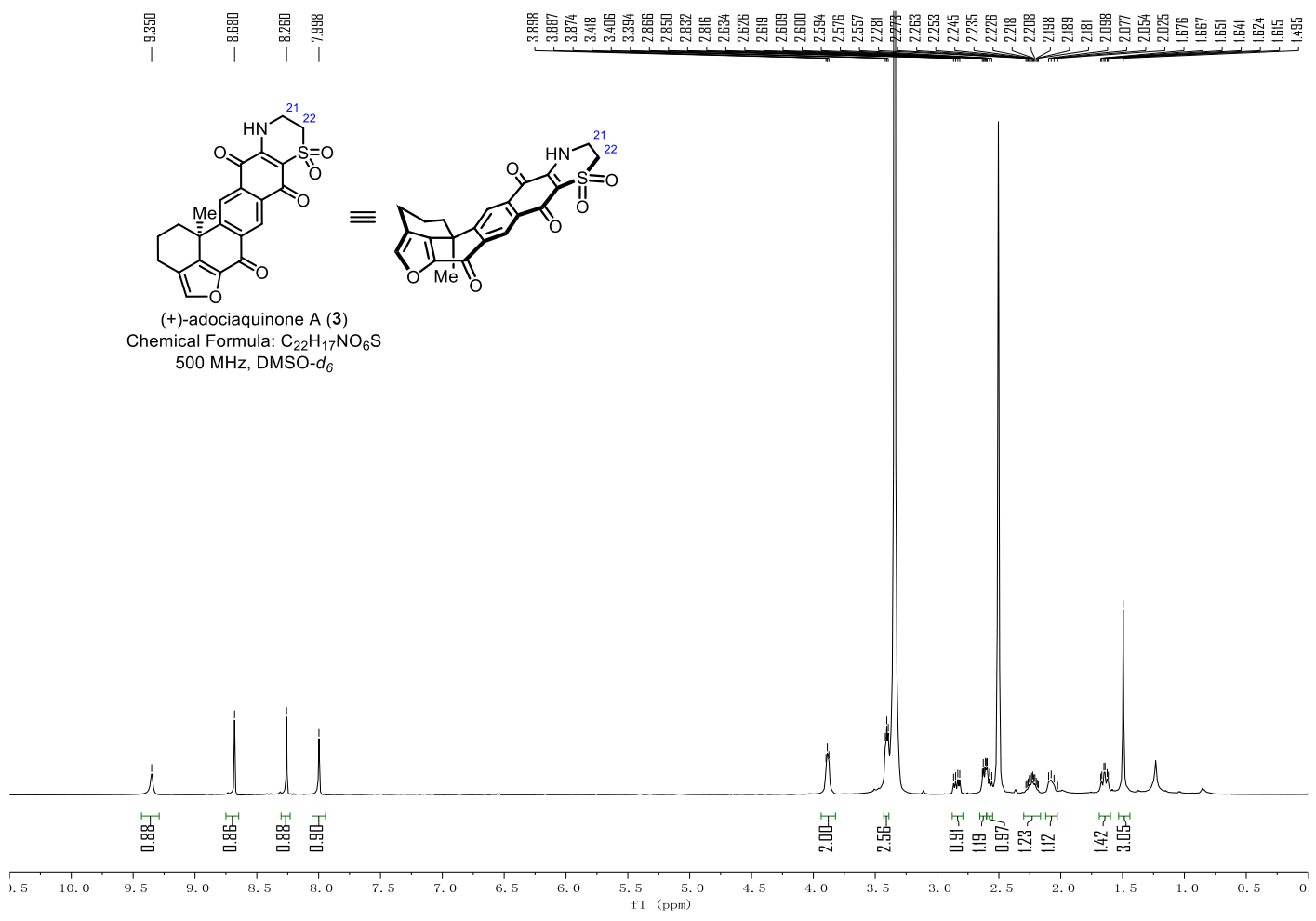


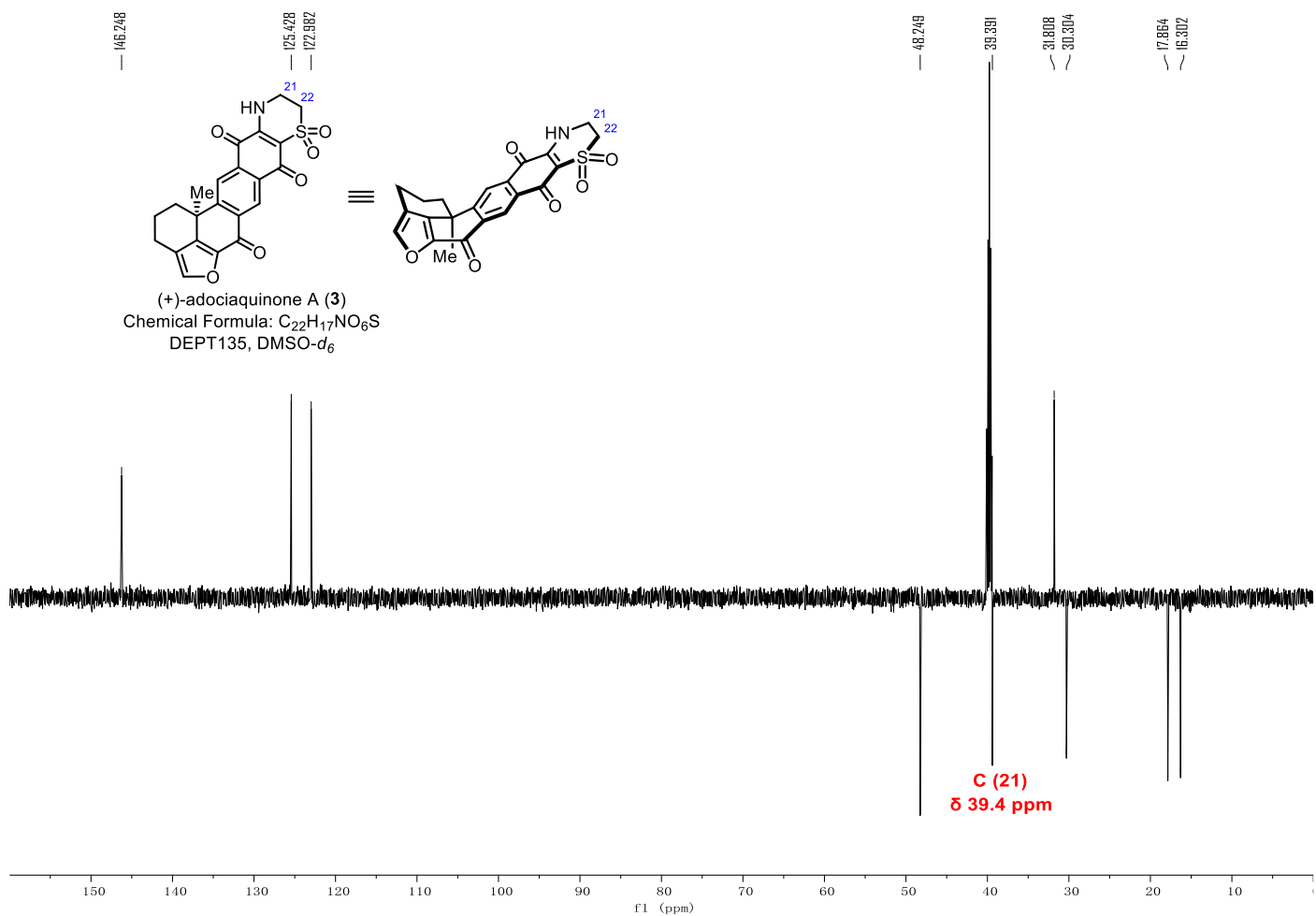
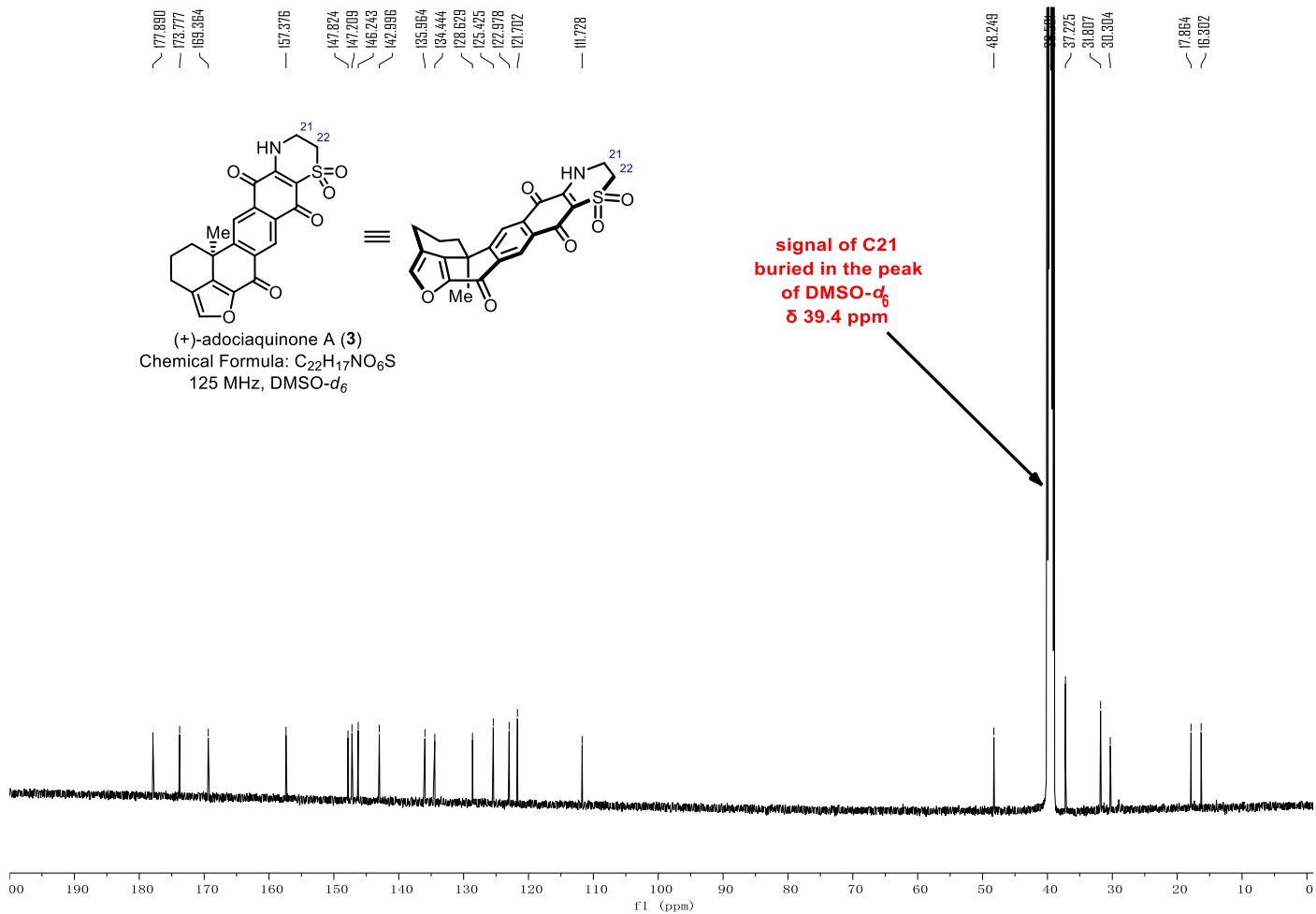
9.350
8.680
8.260
7.998

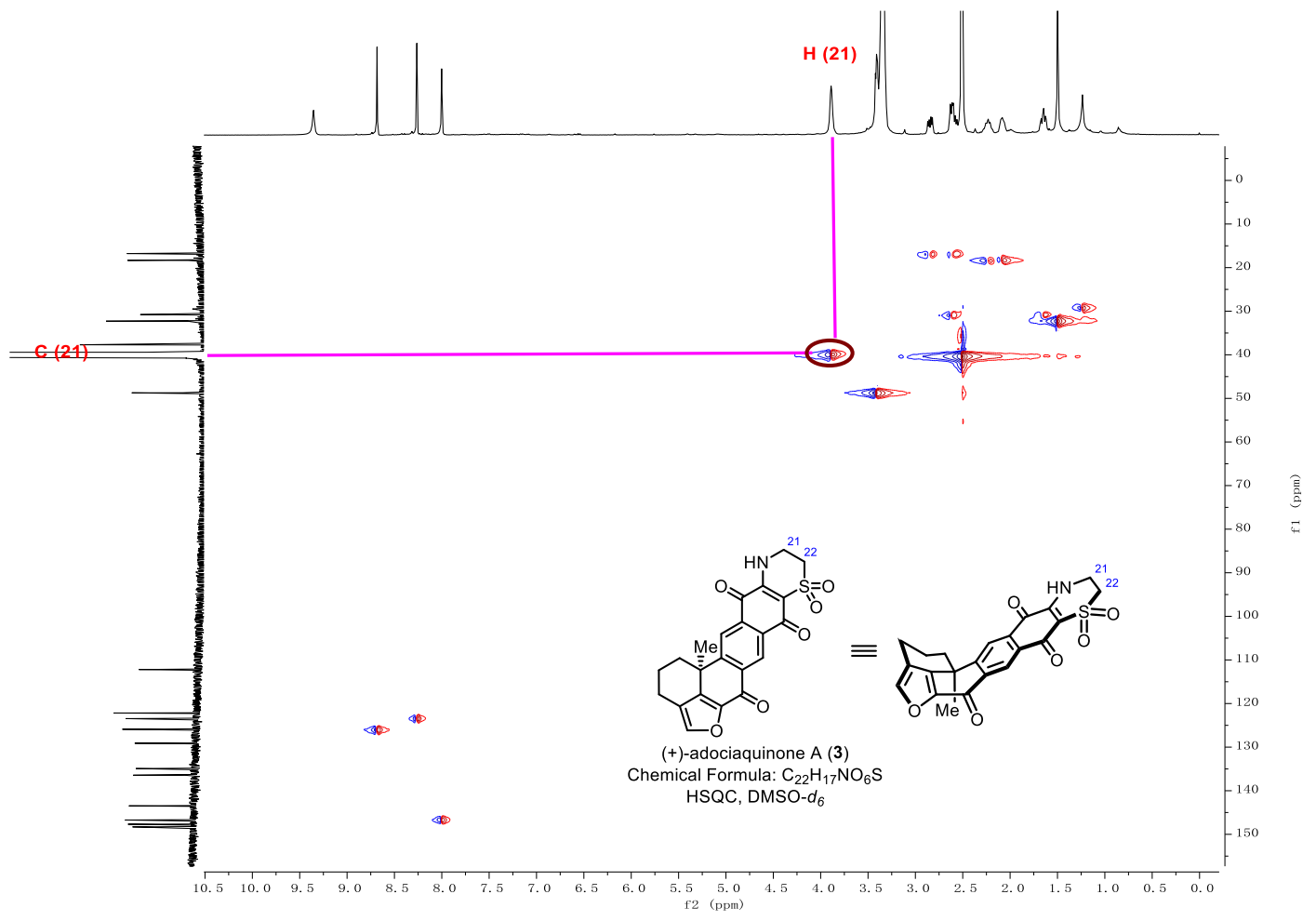
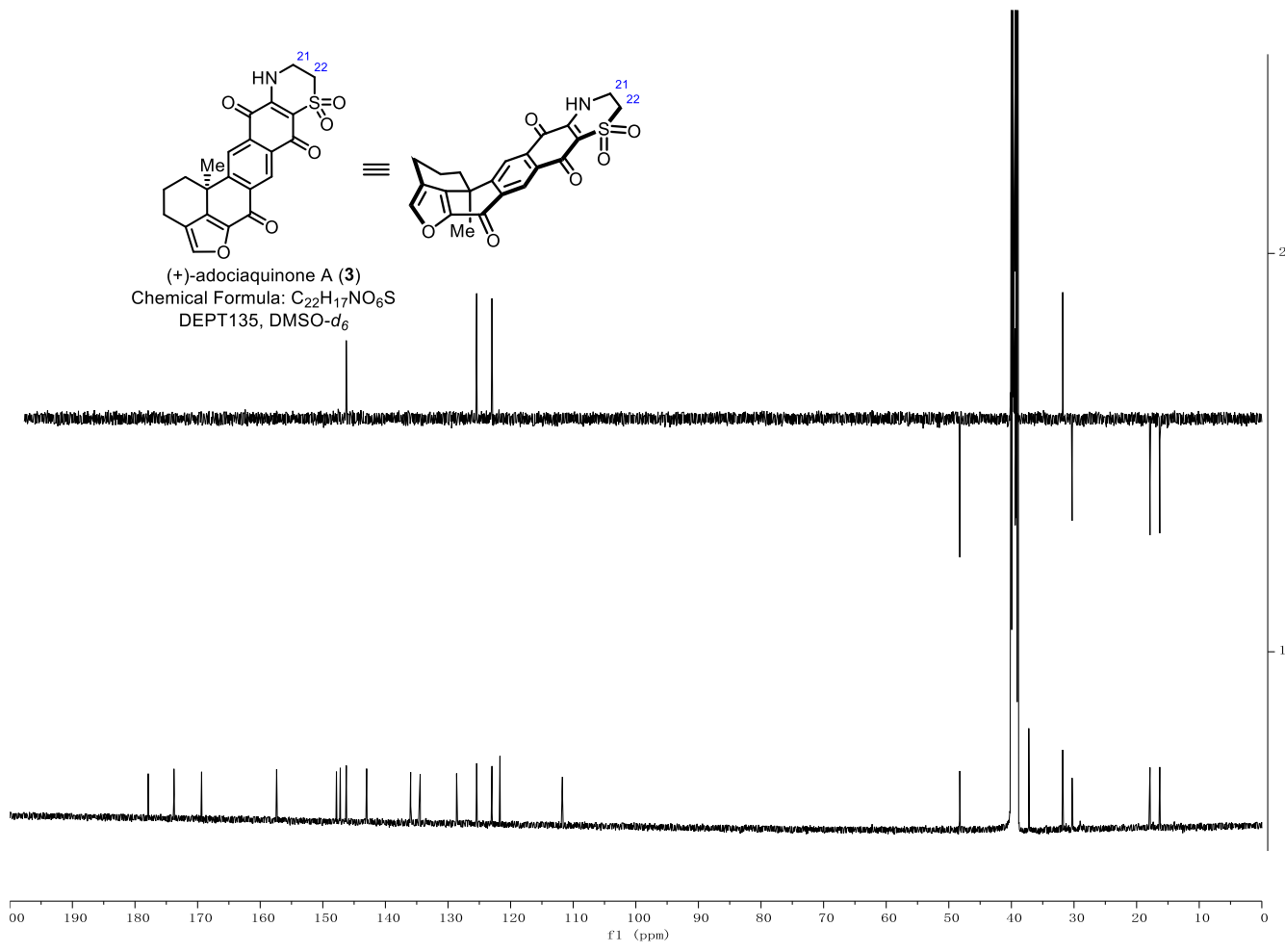
3.888
3.887
3.874
3.418
3.406
3.394
2.866
2.850
2.832
2.816
2.626
2.634
2.618
2.609
2.600
2.594
2.576
2.557
2.281
2.279
2.263
2.253
2.245
2.235
2.226
2.218
2.208
2.198
2.189
2.181
2.098
2.077
2.054
2.025
1.676
1.667
1.651
1.644
1.624
1.615
1.495

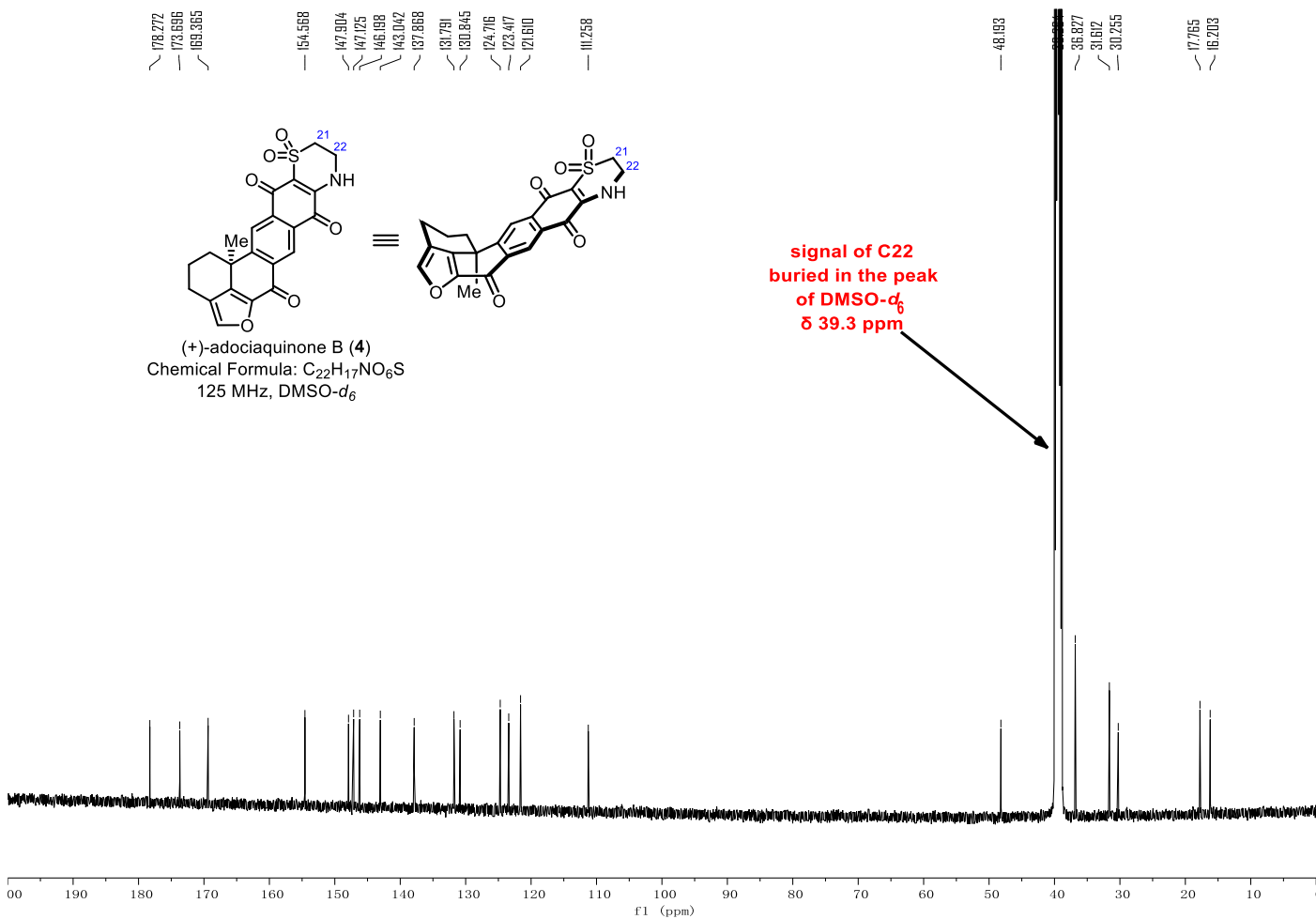
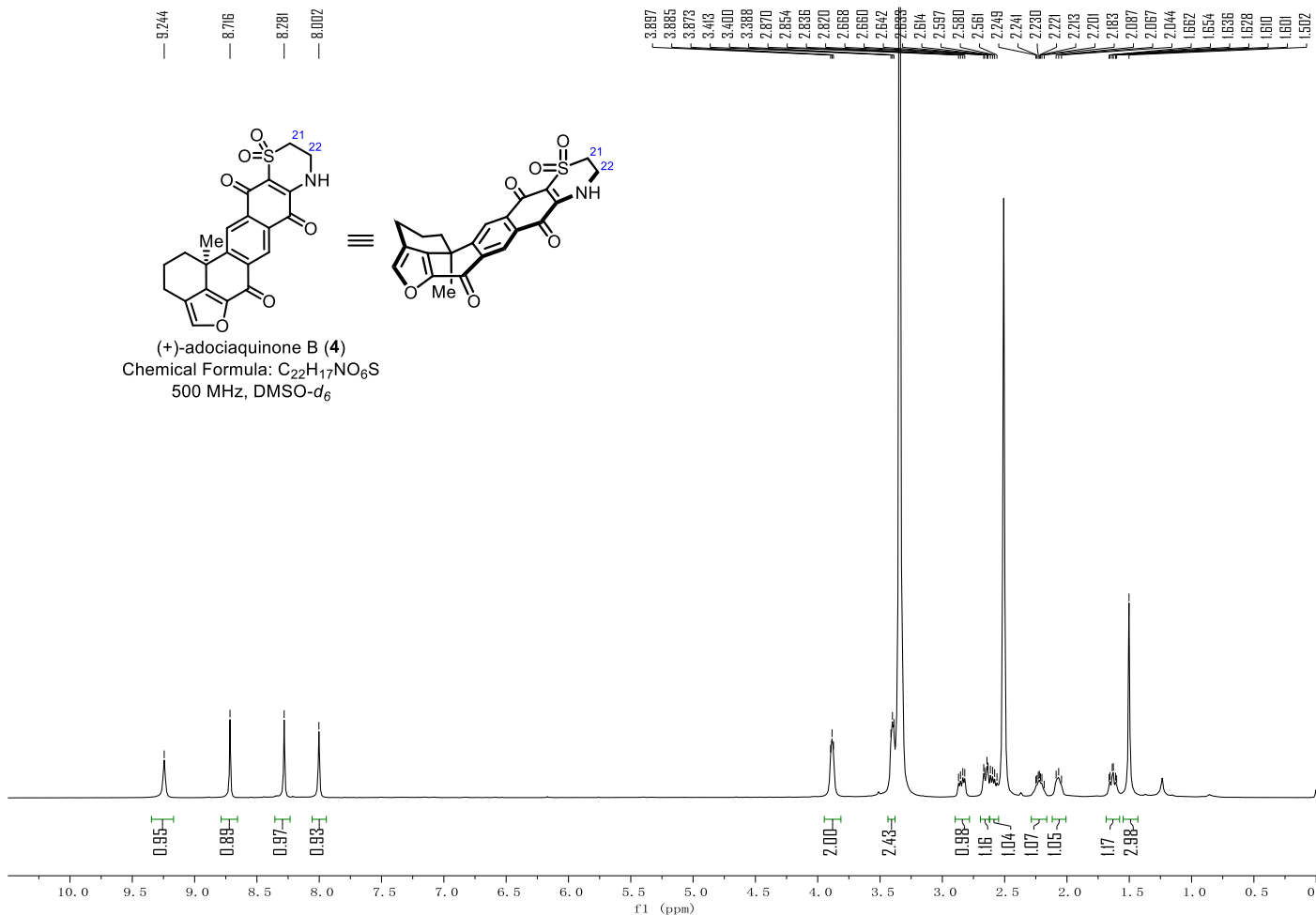


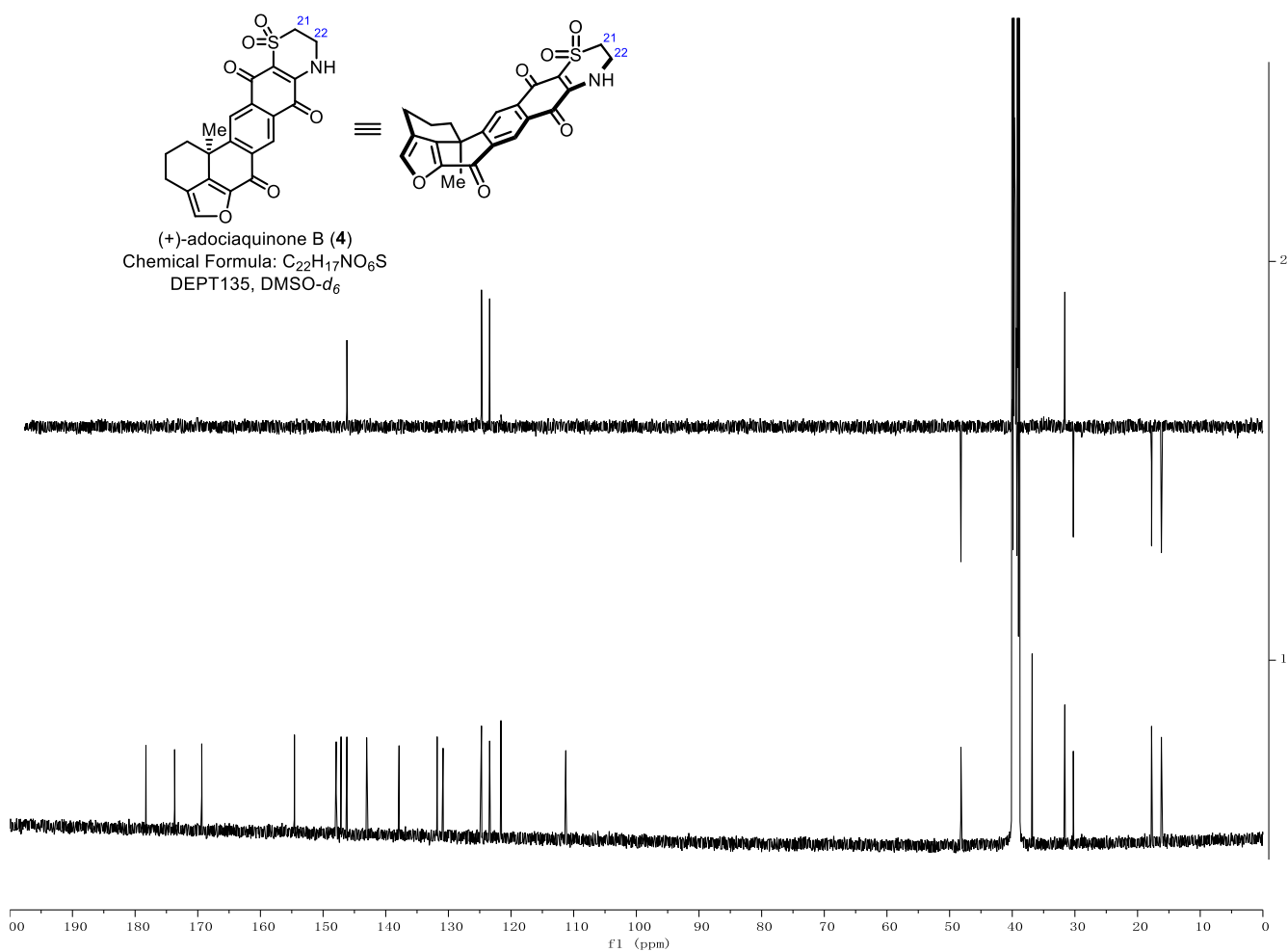
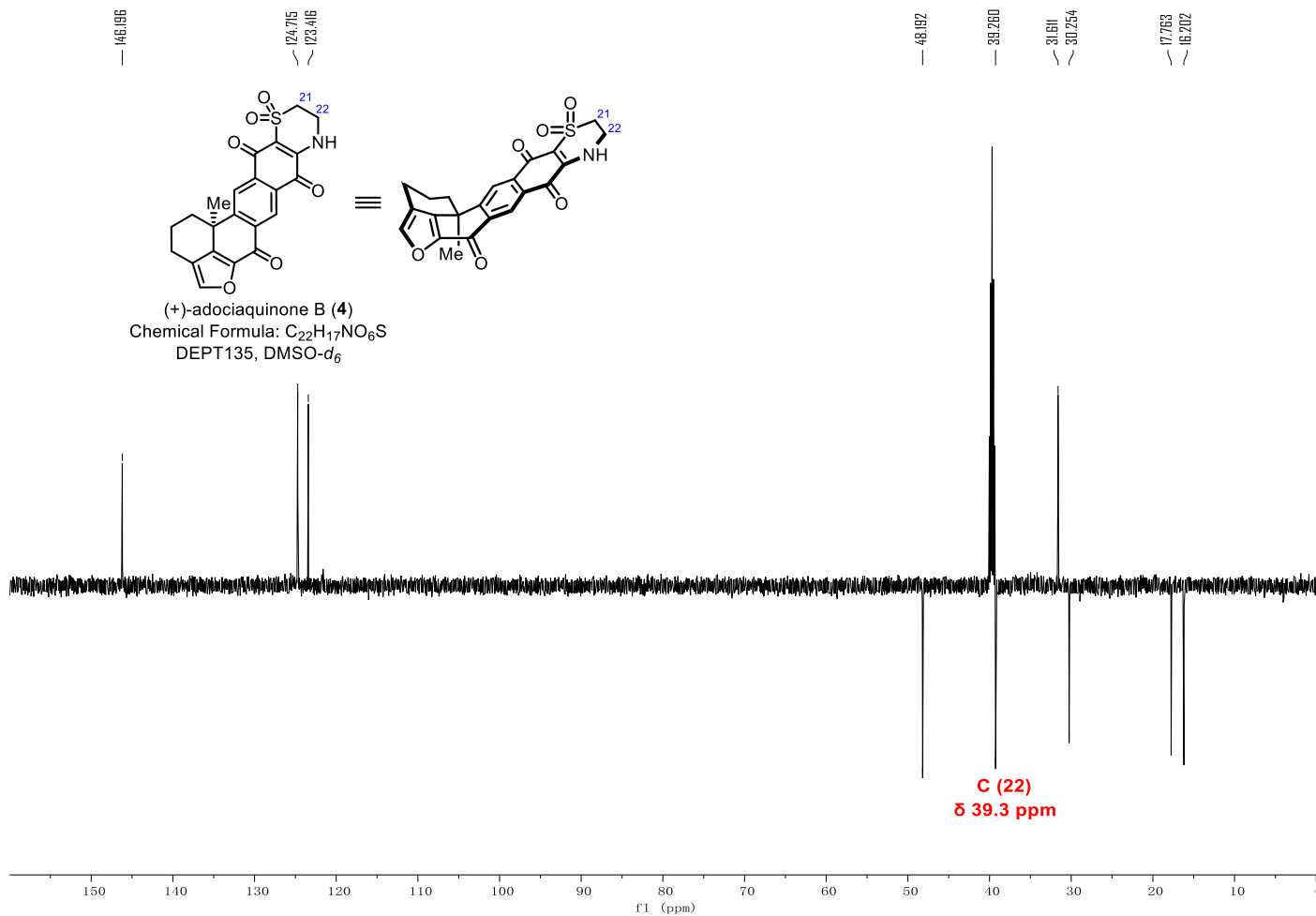
(+)-adociaquinone A (3)
Chemical Formula: C₂₂H₁₇NO₆S
500 MHz, DMSO-*d*₆

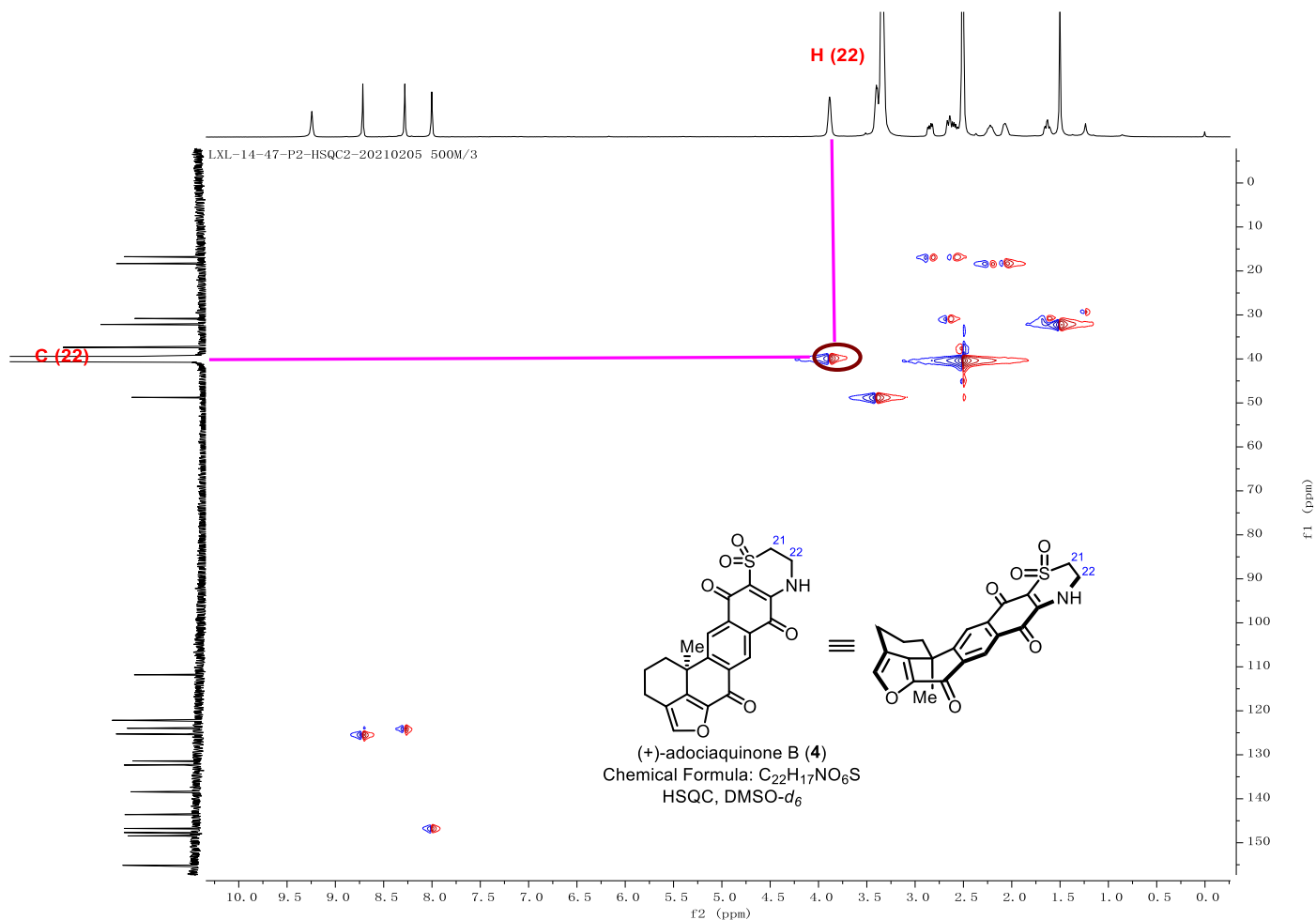






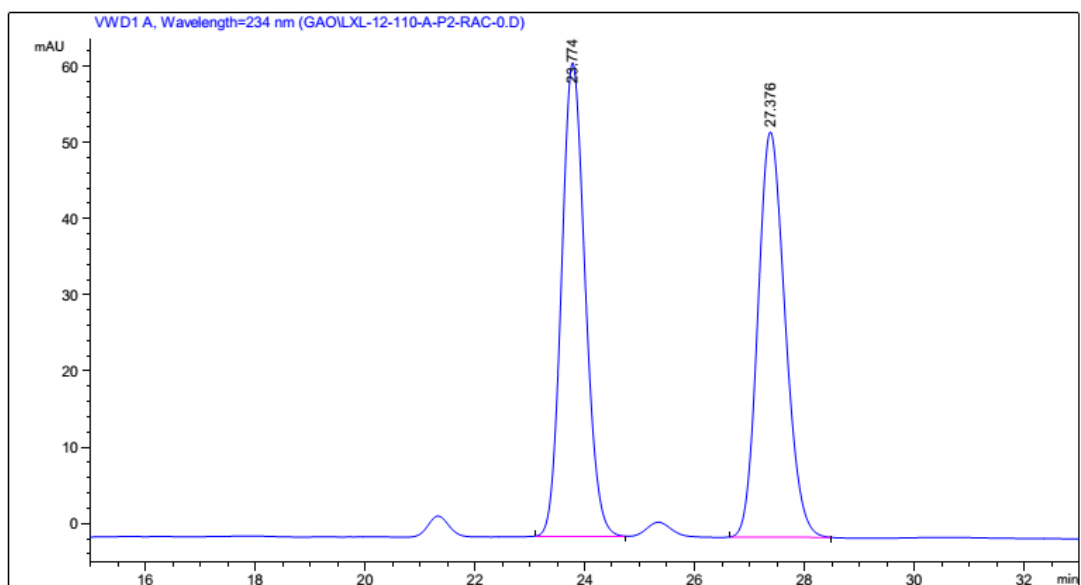






HPLC spectra of product 14.

1. Table S1, entry 1, racemate of product 14a/14a': HPLC analysis, Chiralpak IG-H, *i*-PrOH/hexane = 20/80, 0.8 mL/min, 234 nm; t_{r1} = 23.774 min, t_{r2} = 27.376 min.

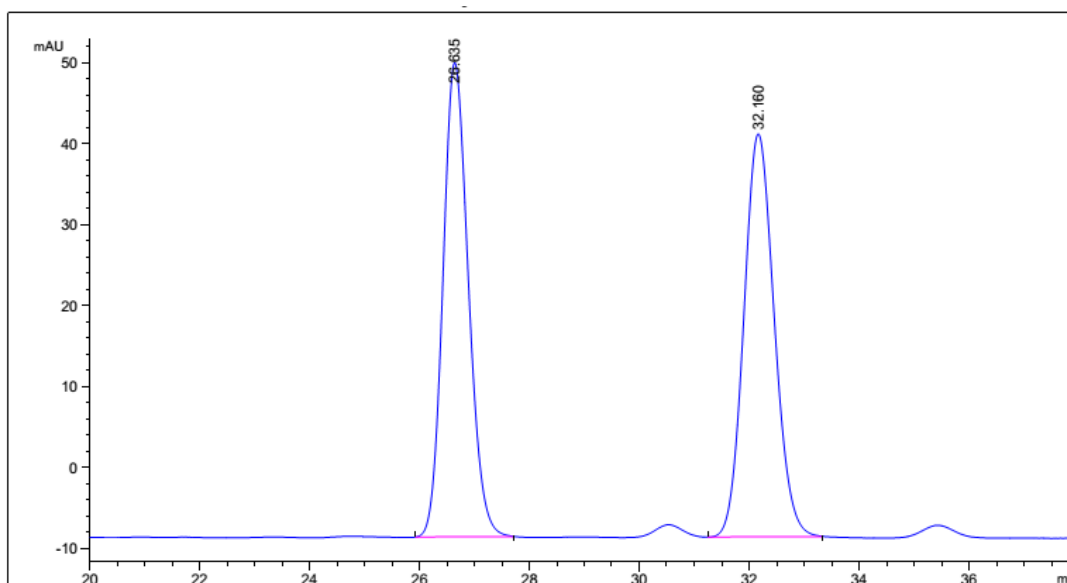


Signal 1: VWD1 A, Wavelength=234 nm

Peak #	RetTime [min]	Type	Width [min]	Area mAU *s	Height [mAU]	Area %
1	23.774	BB	0.4633	1856.48975	62.19220	50.2387
2	27.376	BB	0.5358	1838.84778	53.19823	49.7613

Totals : 3695.33752 115.39043

4. **Table S1**, entry 1, racemate of product **14b/14b'**: HPLC analysis, Chiralpak IG-H, *i*-PrOH/hexane = 15/85, 0.8 mL/min, 234 nm; t_{r1} = 26.635 min, t_{r2} = 32.160 min.

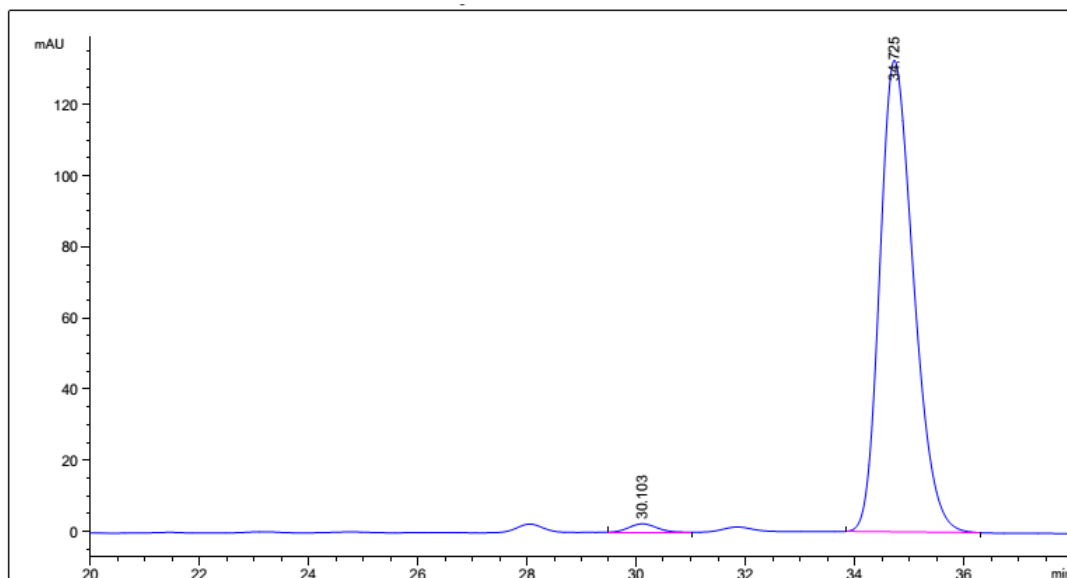


Signal 1: VWD1 A, Wavelength=234 nm

Peak #	RetTime [min]	Type	Width [min]	Area mAU	Area *s	Height [mAU]	Area %
1	26.635	BB	0.4890	1879.38318		58.63712	49.6089
2	32.160	VB	0.5913	1909.01538		49.80637	50.3911

Totals : 3788.39856 108.44349

5. **Table 1**, entry 12, 1.31g scale, 97% e.e. for product **14a**: HPLC analysis, Chiralpak IG-H, *i*-PrOH/hexane = 15/85, 0.8 mL/min, 234 nm; t_{r1} (minor) = 30.103 min, t_{r2} (major) = 34.725 min.

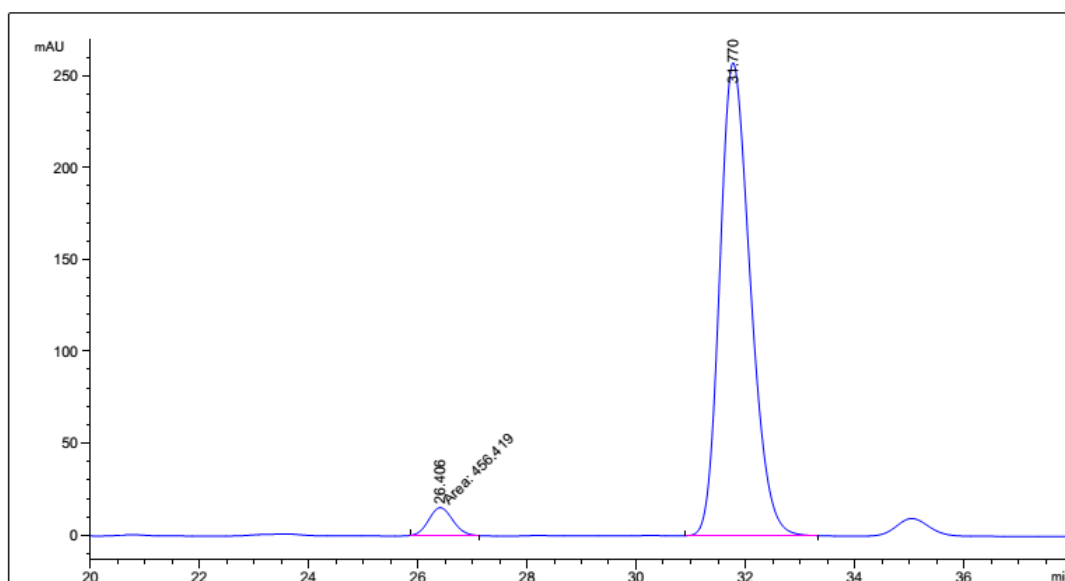


Signal 1: VWD1 A, Wavelength=234 nm

Peak #	RetTime [min]	Type	Width [min]	Area mAU	Area *s	Height [mAU]	Area %
1	30.103	BB	0.4274	83.47428		2.36739	1.4422
2	34.725	BB	0.6506	5704.58008		132.56462	98.5578

Totals : 5788.05436 134.93202

6. **Table 1**, entry 12, 1.31g scale, 91% e.e. for product **14b**: HPLC analysis, Chiralpak IG-H, *i*-PrOH/hexane = 15/85, 0.8 mL/min, 234 nm; t_{r1} (minor) = 26.406 min, t_{r2} (major) = 31.770 min.

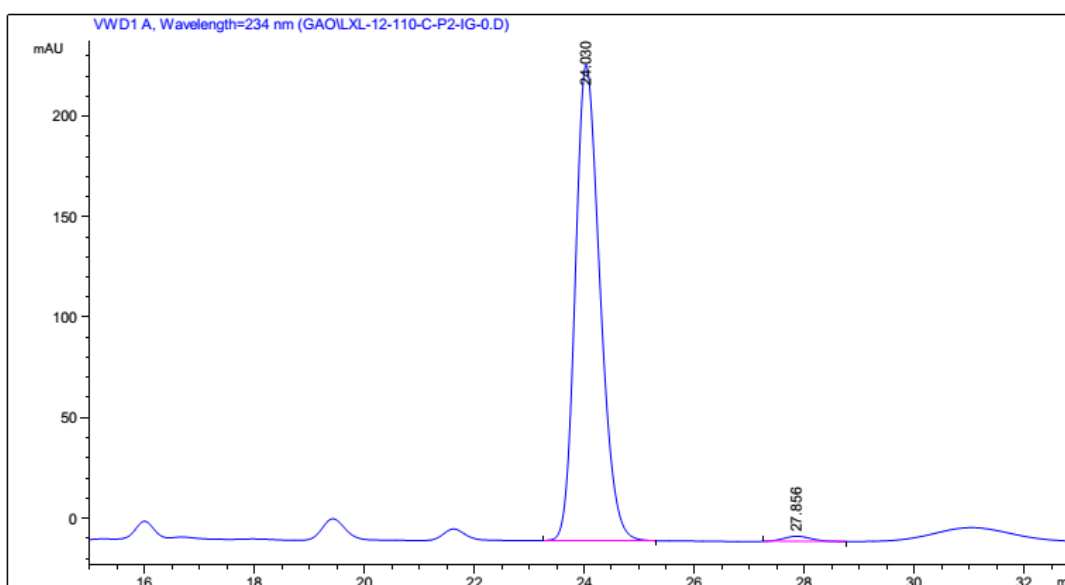


Signal 1: VWD1 A, Wavelength=234 nm

Peak #	RetTime [min]	Type	Width [min]	Area mAU *s	Height [mAU]	Area %
1	26.406	MM	0.5051	456.41907	15.06138	4.3573
2	31.770	BB	0.5988	1.00185e4	257.10541	95.6427

Totals : 1.04749e4 272.16679

7. **Table S1**, entry 11, -97% e.e. for product **14a'**: HPLC analysis, Chiralpak IG-H, *i*-PrOH/hexane = 20/80, 0.8 mL/min, 234 nm; t_{r1} (major) = 24.030 min, t_{r2} (minor) = 27.856 min.

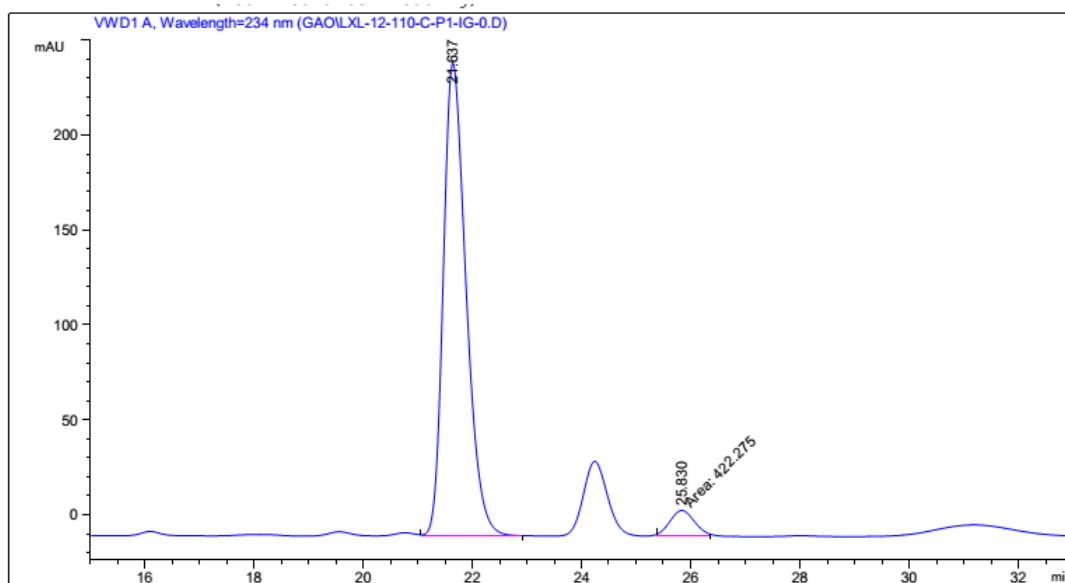


Signal 1: VWD1 A, Wavelength=234 nm

Peak #	RetTime [min]	Type	Width [min]	Area mAU *s	Height [mAU]	Area %
1	24.030	BB	0.4795	7378.74805	236.65709	98.7760
2	27.856	BB	0.4136	91.43444	2.60232	1.2240

Totals : 7470.18249 239.25941

8. **Table S1**, entry 11, -89% e.e. for product **14b'**: HPLC analysis, Chiralpak IG-H, *i*-PrOH/hexane = 20/80, 0.8 mL/min, 234 nm; t_{r1} (major) = 21.637 min, t_{r2} (minor) = 25.830 min.



Signal 1: VWD1 A, Wavelength=234 nm

Peak #	RetTime [min]	Type	Width [min]	Area mAU *s	Height [mAU]	Area %
1	21.637	VB	0.4438	7187.94141	248.97035	94.4512
2	25.830	MM	0.5234	422.27490	13.44546	5.5488

Totals : 7610.21631 262.41581