Electronic Supplementary Information for

Interconnected porous Au₃Pt film on Ni foam: an efficient electrocatalyst for

alkaline hydrogen evolution reaction

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Fig. S1 SEM images of the (a) pAu₃Pt/NF (b) and NF.



Fig. S2 EDS spectrum of the pAu₃Pt/NF.



Fig. S3 SEM images of samples prepared by replacing PS-*b*-PEO with (a) F127 and (b) PVP under the typical condition used for the typical synthesis.



Fig. S4 SEM images of samples prepared with different amounts of metallic precursors under the typical synthesis. The added metallic precursor amounts of $HAuCl_4$ and K_2PtCl_4 were (a) 2.0 mL and 0 mL (pAu/NF), (b) 1.5 mL and 0.5 mL (pAu_3Pt/NF), (c) 1.0 mL and 1.0 mL (pAu_2Pt_2/NF), (d) 0.5 mL and 1.5 mL (pAuPt_3/NF).



Fig. S5 TEM images of samples prepared with different reaction times under the typical conditions for (a) 20 s, (b) 1 min, (c) 10 min, and (d) 30 min.



Fig. S6 Digital photograph of experimental set-up for HER.



Fig. S7 Comparison of overpotentials for samples prepared with different PS-*b*-PEO amounts at current density of 100 mA cm⁻².



Fig. S8 CV curves for (a) pAu₃Pt/NF-20, (b) pAu₃Pt/NF-10, (c) pAu₃Pt/NF-5, (d) pAu₃Pt/NF-2 and (e) pAu₃Pt/NF-0. (f) Capacitive current densities at 0.45 V derived from CV curves against scan rate for different samples.



Fig. S9 (a) LSV curves for samples prepared with different amounts of metallic precursors under the typical conditions, and (b) their corresponding Tafel plots.



Fig. S10 Multi-current process of the pAu₃Pt/NF. The current density without *iR* correction started at 20 mA cm⁻² and ended at 200 mA cm⁻² with an increment of 20 mA cm⁻² per 500 s.

Samples	Au:Pt (atomic ratio)
pAu/NF	100:0
pAu ₃ Pt/NF	76.20:23.80
pAu ₂ Pt ₂ /NF	56.08:43.92
pAuPt ₃ /NF	31.16:68.84

Table S1. Atomic ratios of different Au-Pt alloys obtained from EDS.