

-Electronic Supplementary Information-

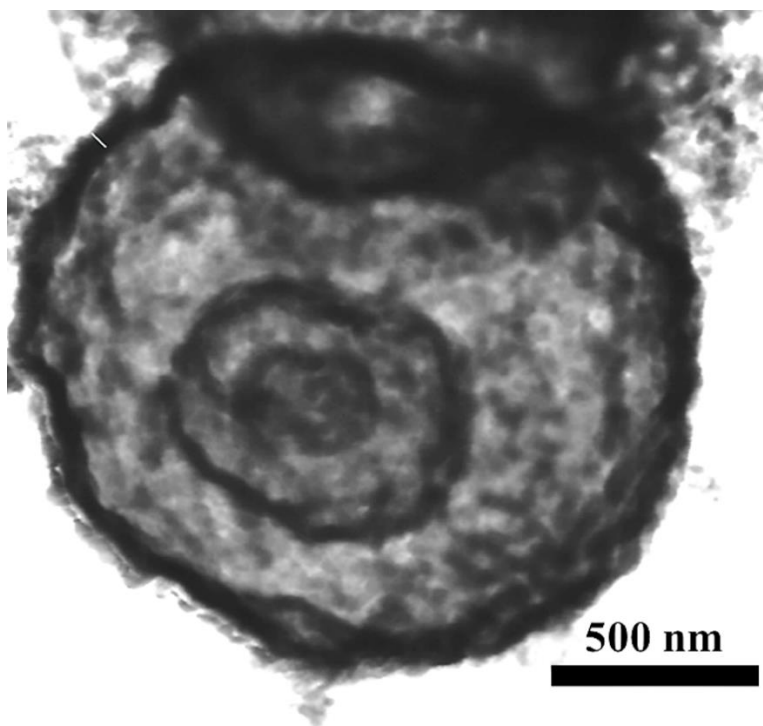


Figure S1. Enlarge TEM images of CeO₂/Co@NCH.

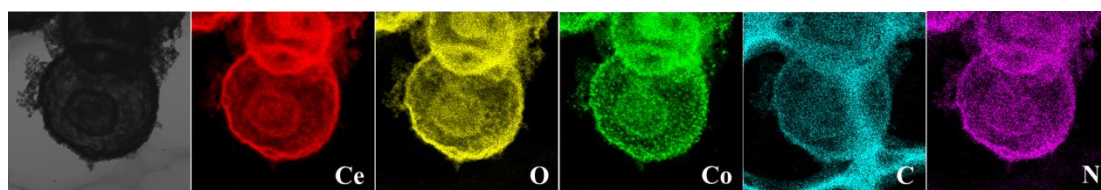


Figure S2. The element mapping of Ce, O, Co, C, N in CeO₂/Co@NCH.

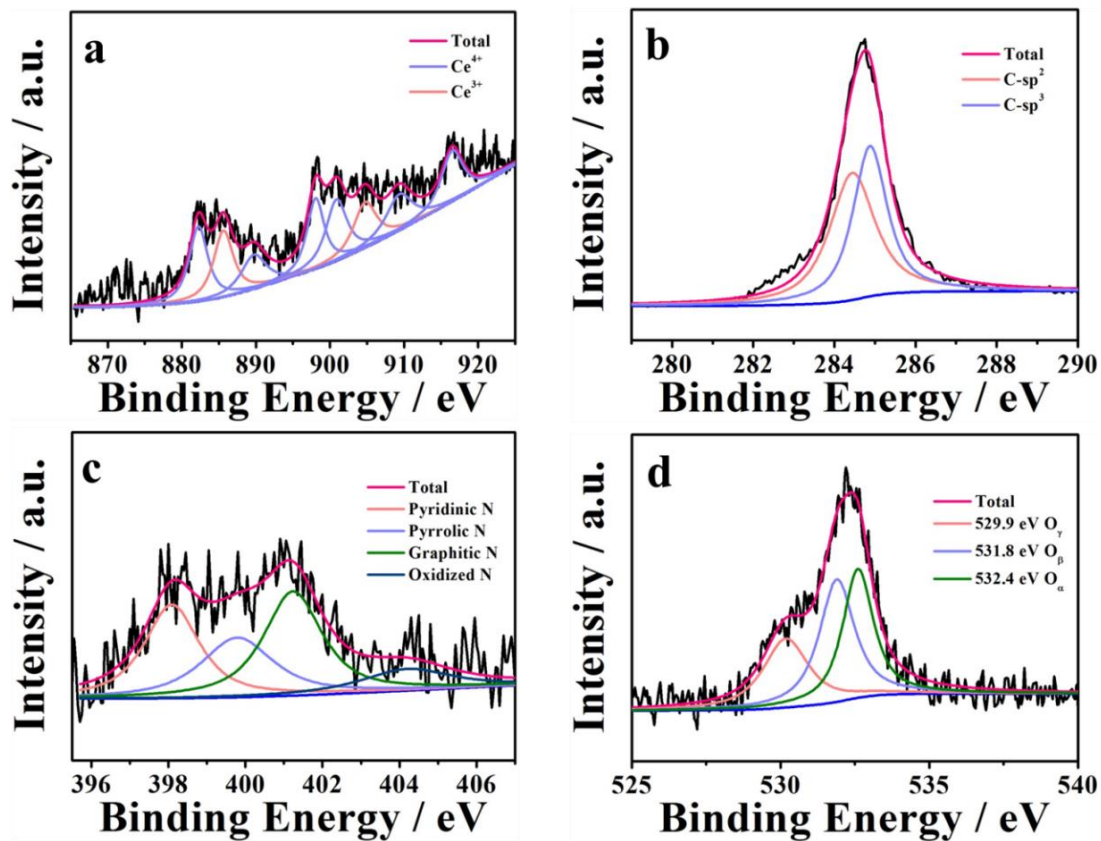


Figure S3. XPS high-resolution Ce 3d (a), C 1s (b), N 1s (c) and O 1s (d) spectra of CeO₂@NCH.

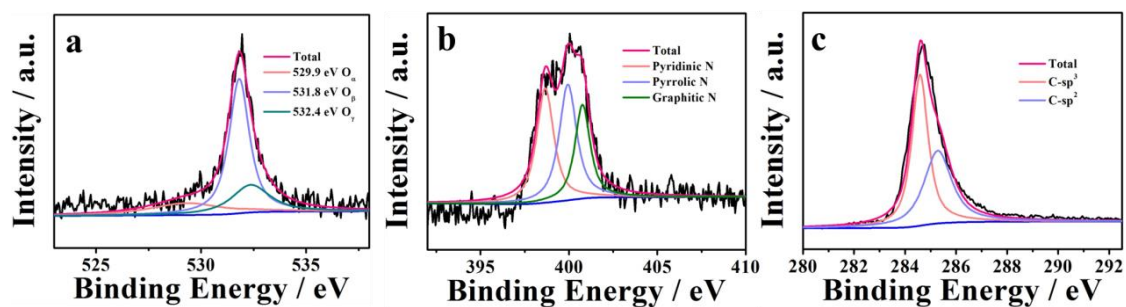


Figure S4. XPS high-resolution O 1s (a), N 1s (b) and C 1s (c) spectra of NCH.

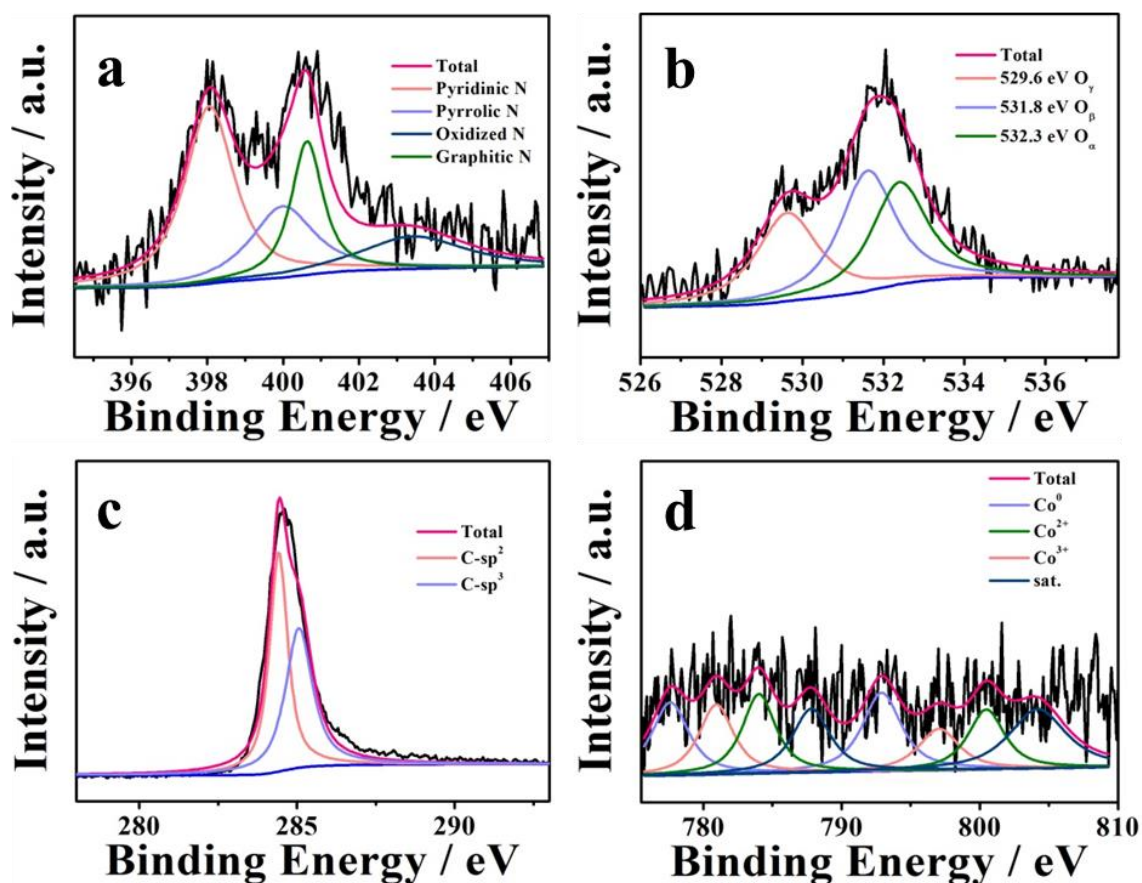


Figure S5. XPS high-resolution N 1s (a), O 1s (b), C 1s (c) and Co (d) spectra of Co@NCH.

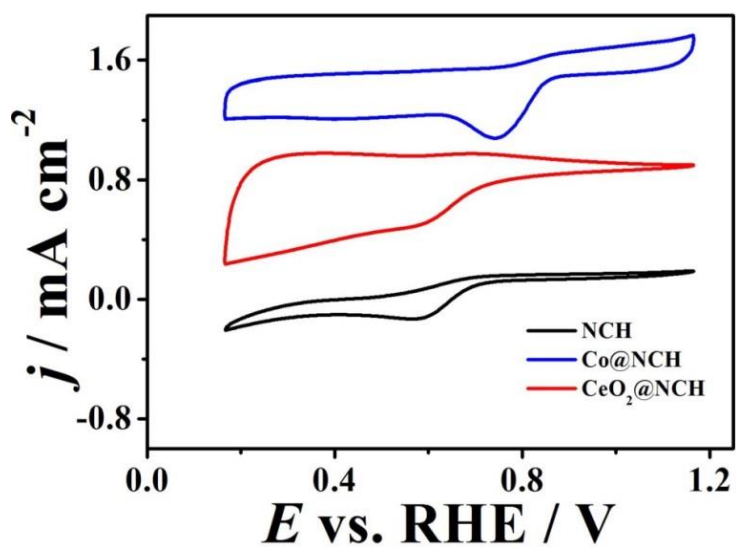


Figure S6. CV curves of NCH, Co@NCH and CeO₂@NCH recorded in O₂ saturated 0.1 M KOH.

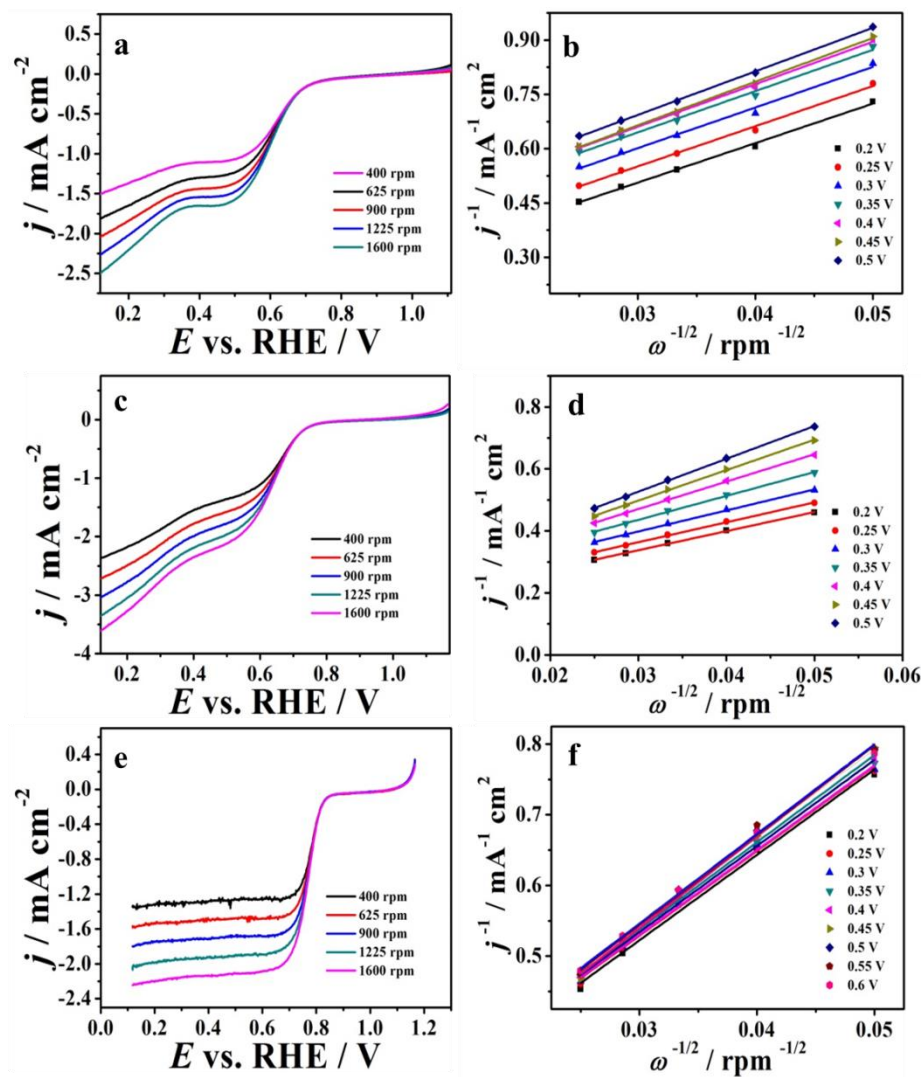


Figure S7. LSV curves of NCH, CeO₂@NCH and Co@NCH measured at different rotating rates and corresponding K-L plots at various potential range.

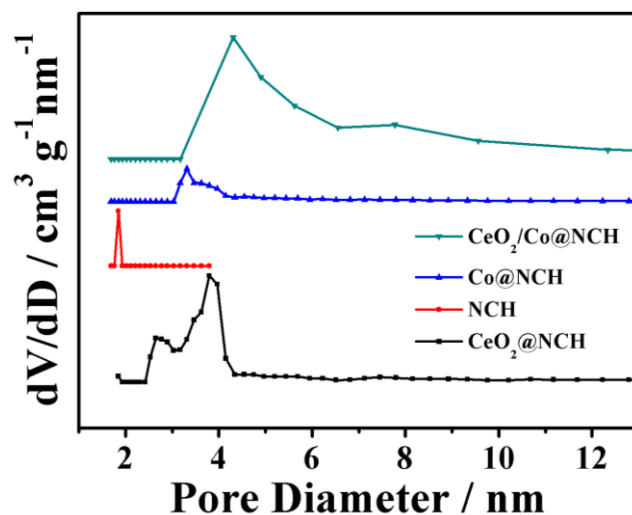


Figure S8. The pore size distribution curves of NCH, Co@NCH, CeO₂@NCH and CeO₂/Co@NCH.

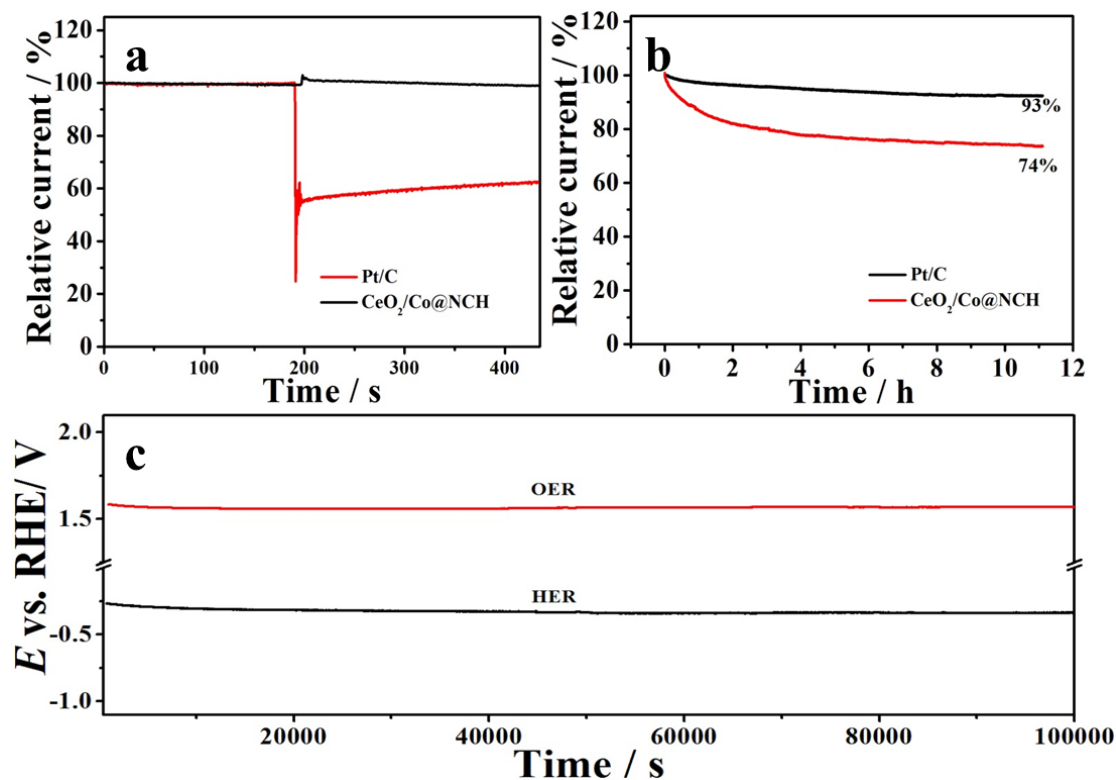


Figure S9. (a) methanol tolerance measurement of CeO₂/Co@NCH and Pt/C samples upon introducing 3M methanol after about 200 s at 0.6 V vs. RHE. (b) i-t chronoamperometric stability measurement of CeO₂/Co@NCH and Pt/C samples at 0.565 V vs. RHE. (c) The OER and HER stability of CeO₂/Co@NCH under a constant current density of 10 mA cm⁻² for 100000 s.

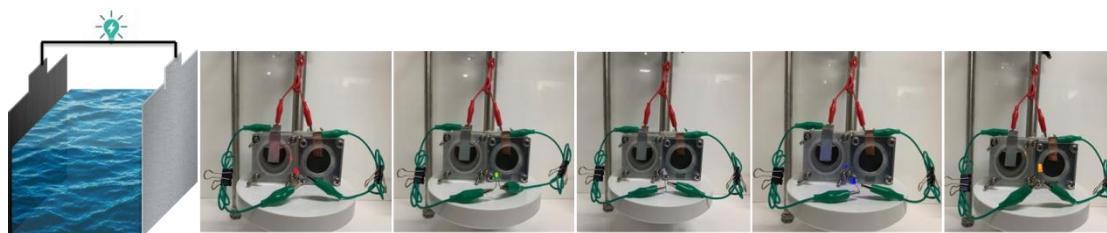


Figure S10. (a) Schematic illustration of the rechargeable zinc-air battery. (b) Photograph showing a series of LED powered by two liquid zinc-air batteries with the CeO₂/Co@NCH cathode connected in series.

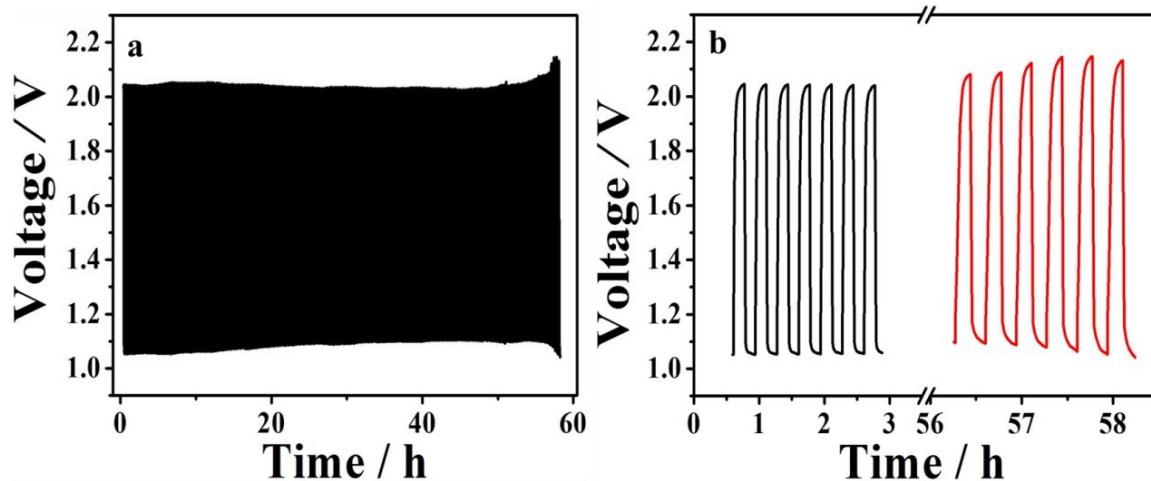


Figure S11. Galvanostatic discharge-charge cycling curves at 5 mA cm^{-2} of recharged Zn-air batteries with the Pt/C catalyst on carbon paper.

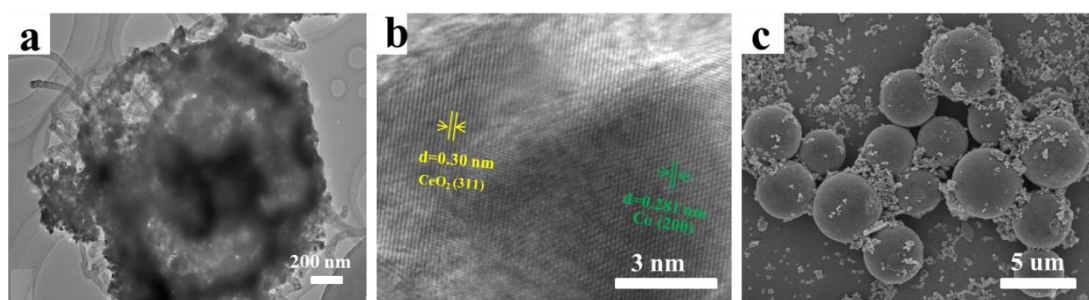


Figure S12. TEM (a) HRTEM (b) and SEM (c) images for $\text{CeO}_2/\text{Co}@NCH$ after 40000 s chronoamperometric test.

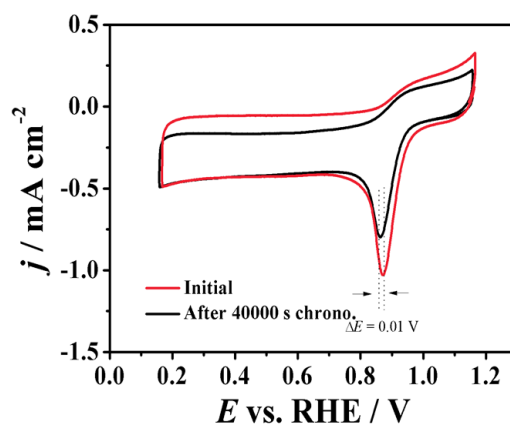


Figure S13. CV curves of $\text{CeO}_2/\text{Co}@NCH$ in O_2 saturated 0.1 M KOH solution

before and after 40000 s chronoamperometric test.

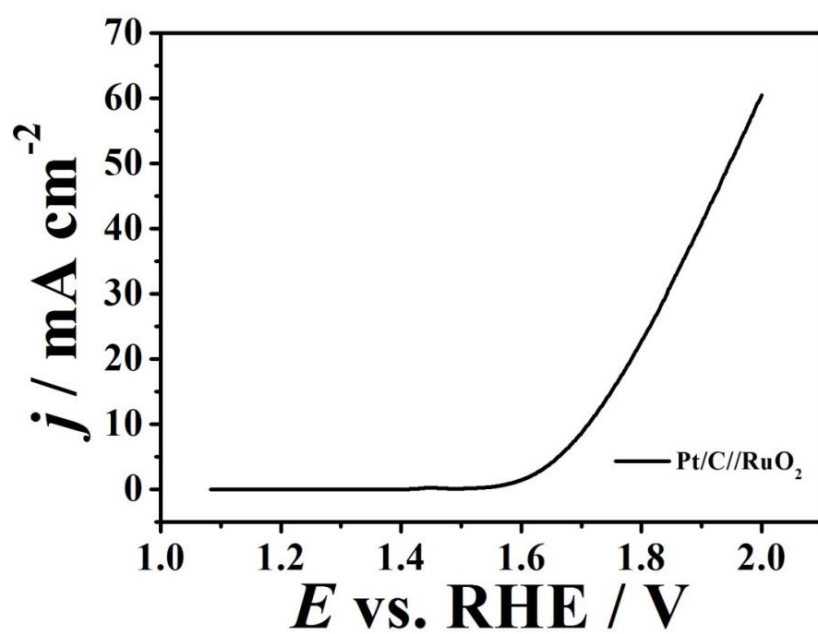


Figure S14. Polarization curves of Pt/C//RuO₂ in N₂-saturated 1M KOH.

Table S1 Comparison of ORR/OER performances of some reported CeO₂-based electrocatalysts.

Electrocatalysts	E_{onset} [V]	$E_{1/2}$ [V]	Medium	η_{OER} , [mV]@ j [mA·cm ⁻²]	η_{HER} , [mV]@ j [mA·cm ⁻²]	Medium	Refs.
CeO ₂ /Co(OH) ₂ HCs				410@10	317@10	1M KOH	[1]
MoS ₂ -G-Ni					>600@10	0.1M KOH	[2]
Ni-Zn/CeO ₂					410@2.74	1M KOH	[3]
g-C ₃ N ₄ /CeO ₂ /Fe ₃ O ₄				400@10	310@10	1M KOH	[4]
3D-rGO-CeO ₂					341@10	1 M KOH	[5]
Co ₃ O ₄ @Z67-NT@CeO ₂	-0.95	0.88	0.1 M	350@10		0.1M KOH	[6]
Ni ₂ P-CeO ₂ /TM					84@20	1 M KOH	[7]
Fe _x Ni _y /CeO ₂ /NC				240@10	240@10	1 M KOH	[8]
Co-CeO ₂ /N-CNR	-0.90	0.82		410@10		0.1M KOH	[9]
Co@NPC-900	0.88	0.76		380@10		0.1M KOH	[10]
Ni/CeO ₂ /CNTs					~220@10	1 M KOH	[11]
CoCe-600N ₂	-0.97	-0.86	0.1 M	274@10		0.1M KOH	[12]
CeO ₂ -Cu ₃ P/NF					~91@10	1 M KOH	[13]
Co-W/CeO ₂					166@10	1 M NaOH	[14]
Co-CeO ₂ -N-C	0.89	0.82	0.1 M	326@10		1 M KOH	[15]
CeO ₂ /rGO	0.946	0.84	0.1 M	500@10		0.1M KOH	[16]
CeO ₂ nanowires	0.756	0.666	0.1 M	700@10		0.1M KOH	[17]

References

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