

## Supporting Information

### **A facile electrochemical strategy for engineering the sulfur deficiencies of CdS Nanosheets to promote catalytic conversion of polysulfides for lithium sulfur batteries**

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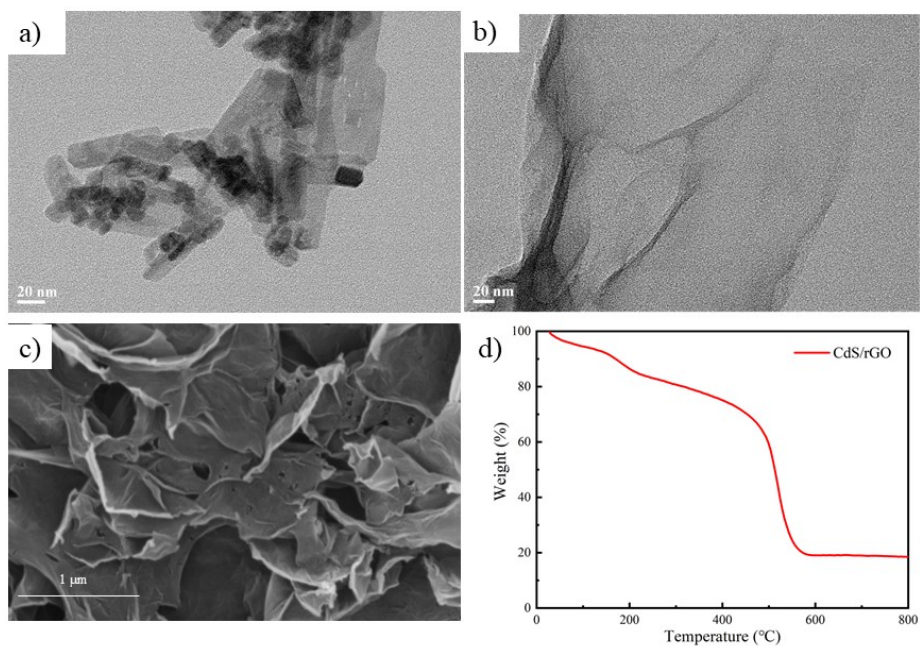


Fig. S1. a) TEM image of CdS NSs. TEM b) and SEM c) images of a thin GO film. d) TGA curve of CdS NSs/rGO composite in air.

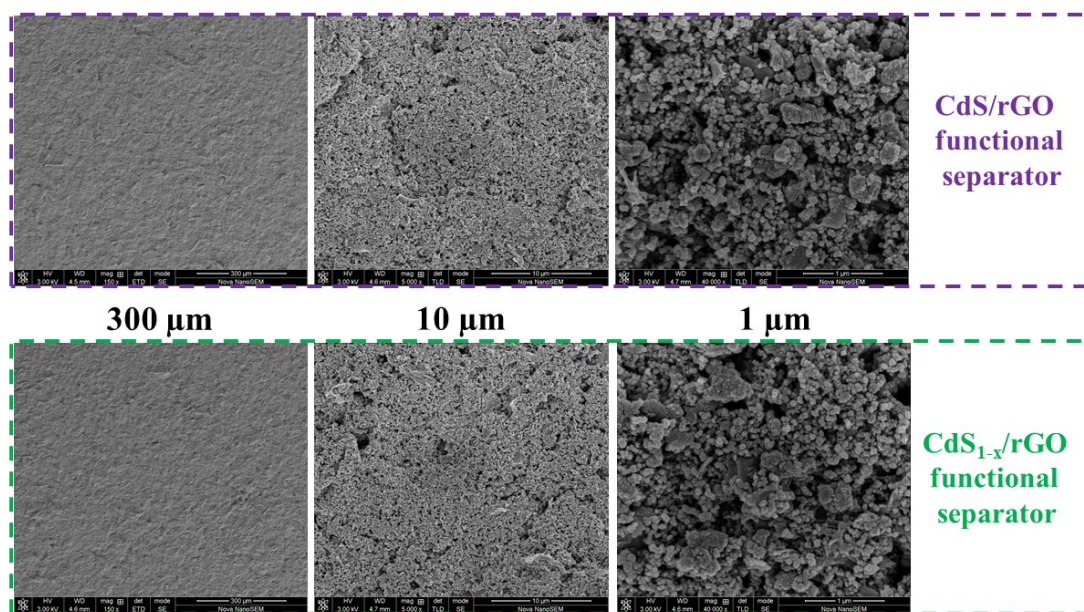


Fig. S2. SEM images of the CdS<sub>1-x</sub> NSs/rGO and CdS NSs/rGO functional separators.

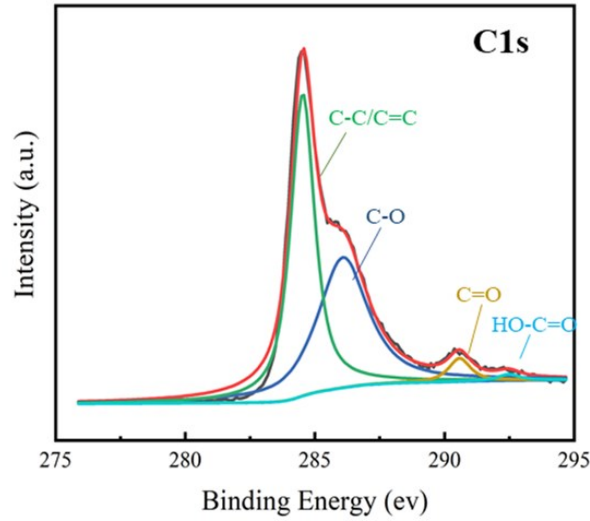


Fig. S3. C 1s XPS spectrum of the  $\text{CdS}_{1-x}$  NSs/rGO functional separator.

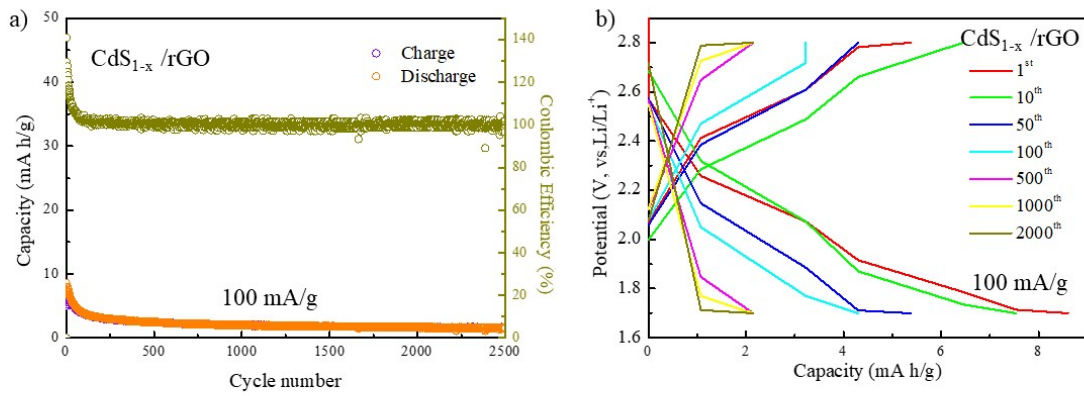


Fig. S4. The intrinsic lithium storage performance of  $\text{CdS}_{1-x}$  NSs/rGO.

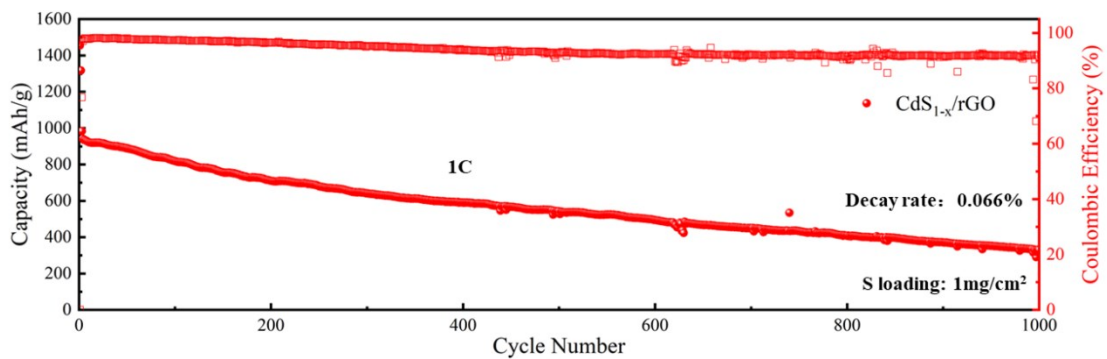


Fig. S5. Cycling performance of the  $\text{CdS}_{1-x}$  NSs/rGO functional separator with  $1 \text{ mg/cm}^2$  S loading at 1C.

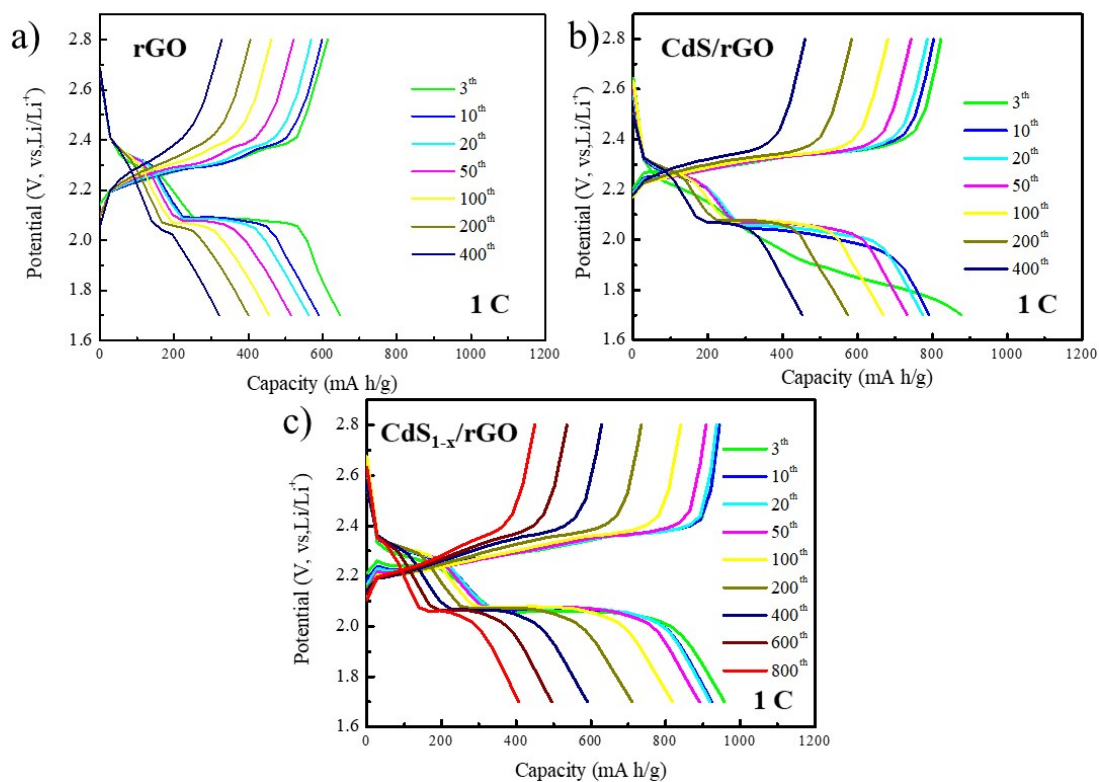


Fig. S6. The discharge/charge curves of the cells with the rGO, CdS NSs/rGO, and CdS<sub>1-x</sub> NSs/rGO functional separators at 1 C.

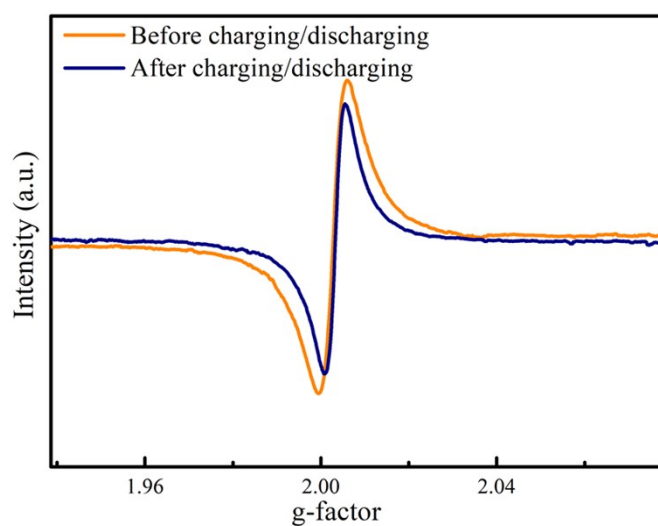


Fig. S7. EPR patterns of the cells with the CdS<sub>1-x</sub> NSs/rGO functional separator before and after repeated charging/discharging.

Table S1 Comparison of catalysts with deficiencies reported to date for Li-S batteries

Catalysts	Rate performance	Initial capacity/Capacity decay (%)	Methods for fabricating deficiencies	References
MoS <sub>2-x</sub> /HMC	730 mA h/g (2 C)	/	Annealing in H <sub>2</sub> /Ar	1
MoS <sub>2-x</sub> /rGO	826 mA h/g (8 C)	1159.9 mA h/g / 0.33 after 100 cycles at 0.2 C	Annealing in H <sub>2</sub> /Ar	2
CNT@TiO <sub>2-x</sub>	597 mA h/g (2 C)	1149 mA h/g / 0.33 after 100 cycles at 1 C	Annealing in H <sub>2</sub> /Ar	3
NiCo <sub>2</sub> O <sub>4-x</sub>	855 mA h/g (5 C)	1221 mA h/g / 0.045 after 800 cycles at 0.2 C	Hydrothermal followed by annealing in air	4
O <sub>2</sub> plasma treated PP separator	/	1028 mA h/g / 0.49 after 105 cycles at 0.2 C	O <sub>2</sub> plasma treatment	5
TiS <sub>2-x</sub>	807 mA h/g (2 C)	956 mA h/g / 0.04 after 1000 cycles at 1 C	Vacuum heat treatment	6
Ti <sub>1-x</sub> O <sub>2</sub>	746 mA h/g (2 C)	656 mA h/g / 0.025 after 500 cycles at 1 C	Solvothermal reaction followed by thermal calcination	7
Fe/Co <sub>3</sub> O <sub>4</sub>	783 mA h/g (2 C)	902 mA h/g / 0.017 after 1000 cycles at 1 C	Hydrothermal reaction followed by thermal calcination	8
SnS <sub>2</sub> /TiO <sub>2</sub>	449 mA h/g (2 C)	841 mA h/g / 0.064 after 500 cycles at 0.5 C	Hydrothermal reaction	9
Co <sub>3</sub> S <sub>4-x</sub>	634mA h/g (2 C)	750 mA h/g / 0.017after 400 cycles at 1 C	Hydrothermal reaction	10
TiO <sub>2</sub> -Ar	571 mA h/g (2 C)	911 mA h/g / 0.082 after 500 cycles at 1 C	Annealing in Ar	11
Co <sub>5.47</sub> N <sub>x</sub> -C	640 mA h/g (2 C)	708 mA h/g / 0.04 after 1000 cycles at 2 C	Thermal calcination in NH <sub>3</sub>	12
CdS <sub>1-x</sub> NSs/rGO	837 mA h/g (2 C)	983 mA h/g / 0.089 after 500 cycles at 1 C	Electroreduction treatment	This Work

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## References

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