

Selective Hydrogen Bonding Controls Temperature Response of Layer-by-Layer Micellar Assemblies

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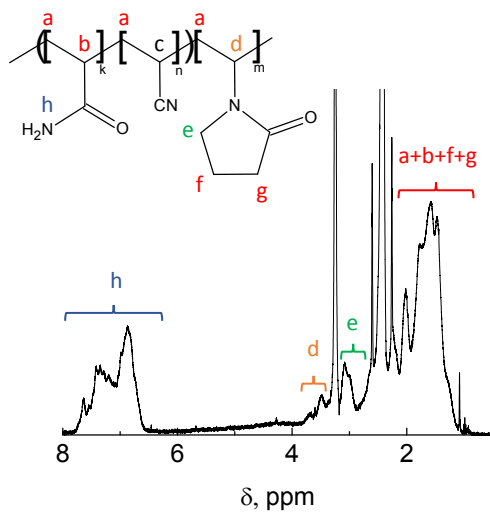


Figure S1. ¹H NMR of P(AAm-co-AN)-b-PVP in d₆-DMSO.

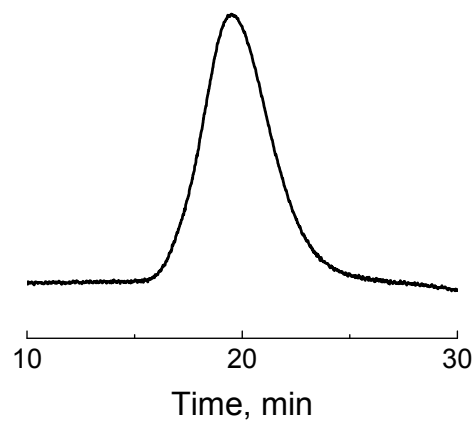


Figure S2. A GPC trace of 2 mg/ml P(AAm-co-AN)-b-PVP in DMSO. Flow rate of the eluent was 0.1 ml/min at 45 °C.

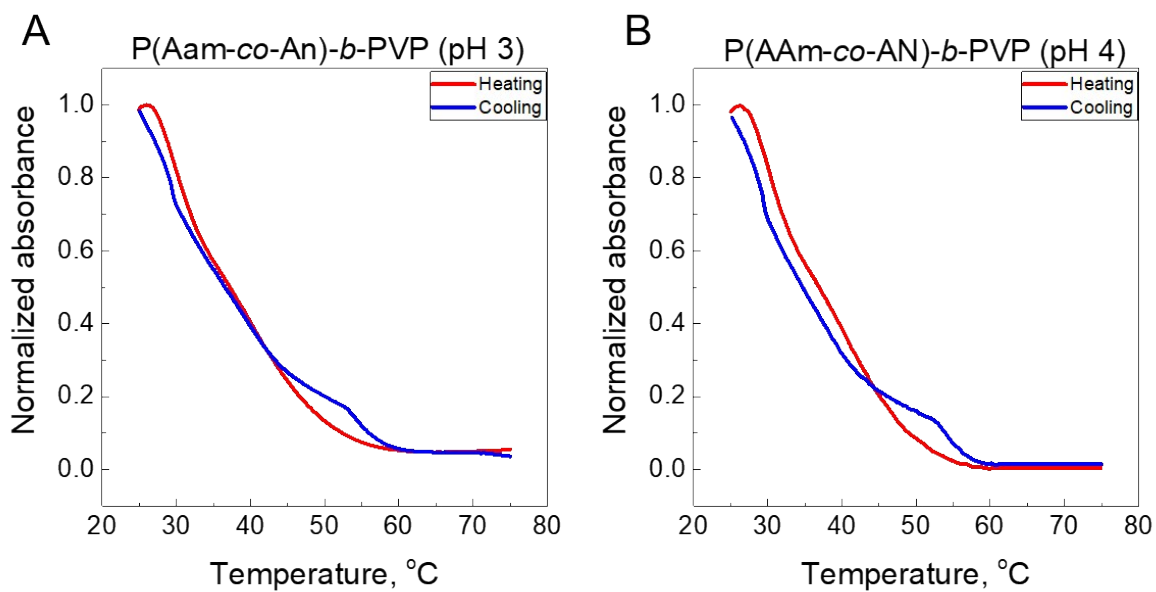


Figure S3. Temperature dependent turbidity in 1 mg/ml P(AAm-co-AN)-b-PVP solutions in PBS at pH 3 (A) and pH 4 (B).

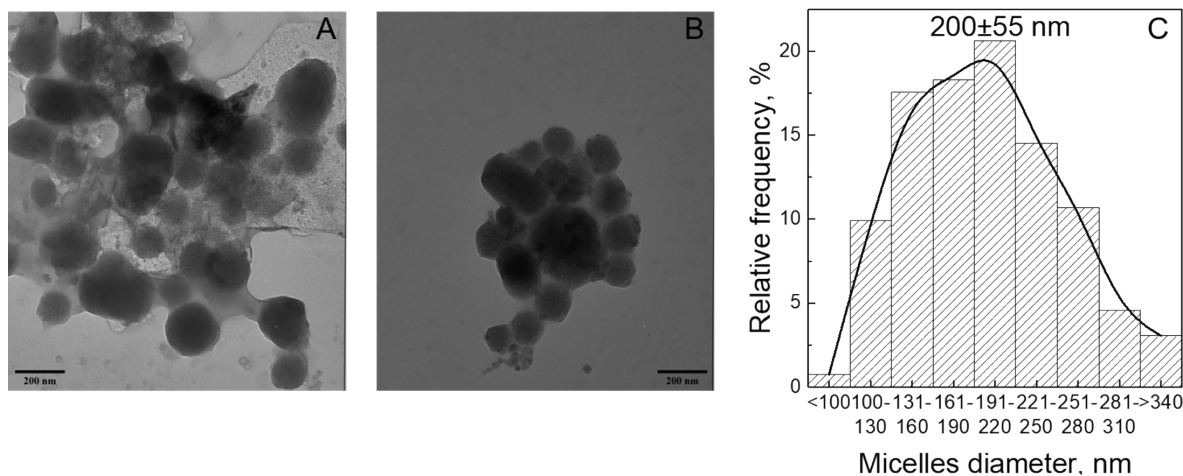


Figure S4. TEM images of UCSTMs dried from solutions at pH 3 (A) and pH 4 (B), as well as the micellar size distribution (C) dried from pH 4 solution.

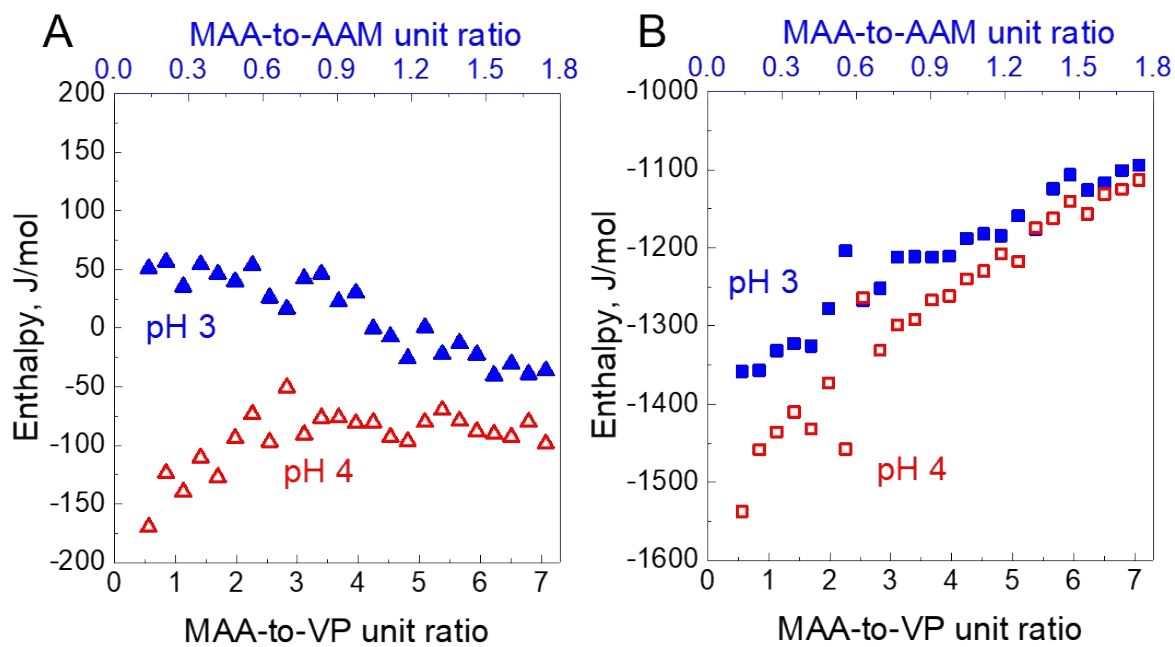


Figure S5. Enthalpy of dilution for PMAA to PBS at pH 3 and pH 4 at 25 °C (A) and 55 °C (B).

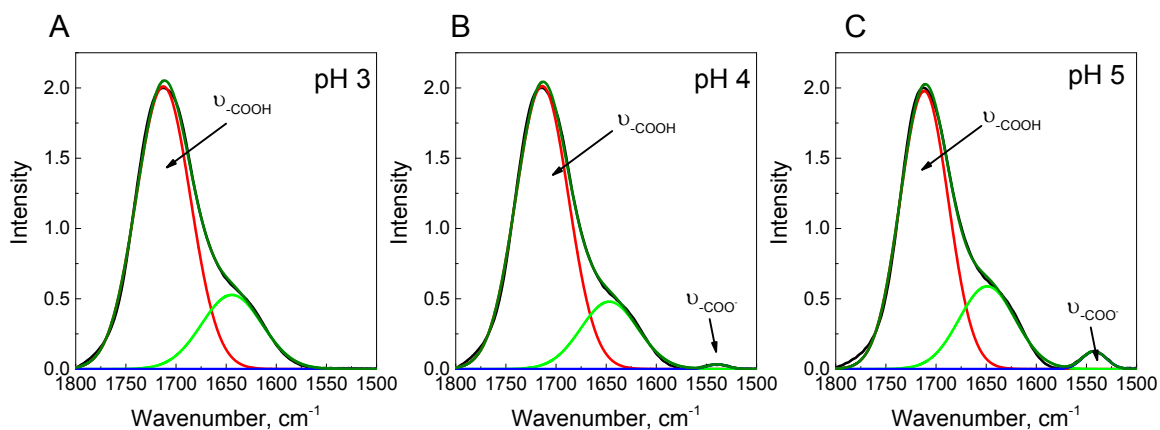


Figure S6. FTIR spectra of PMAA freeze-dried at pH 3 (A), pH 4 (B) and pH 5 (C). Assuming that the extinction coefficients of stretching carbonyl vibrations of ionized and nonionized PMAA are equal, ionization of PMAA was determined as $0.6 \pm 0.1 \%$ at pH 4 and $3.8 \pm 0.2 \%$ at pH 5 from the ratio of 1540 cm^{-1} to the sum of two 1540 and 1720 cm^{-1} peaks in the $1500\text{-}1800 \text{ cm}^{-1}$ region.

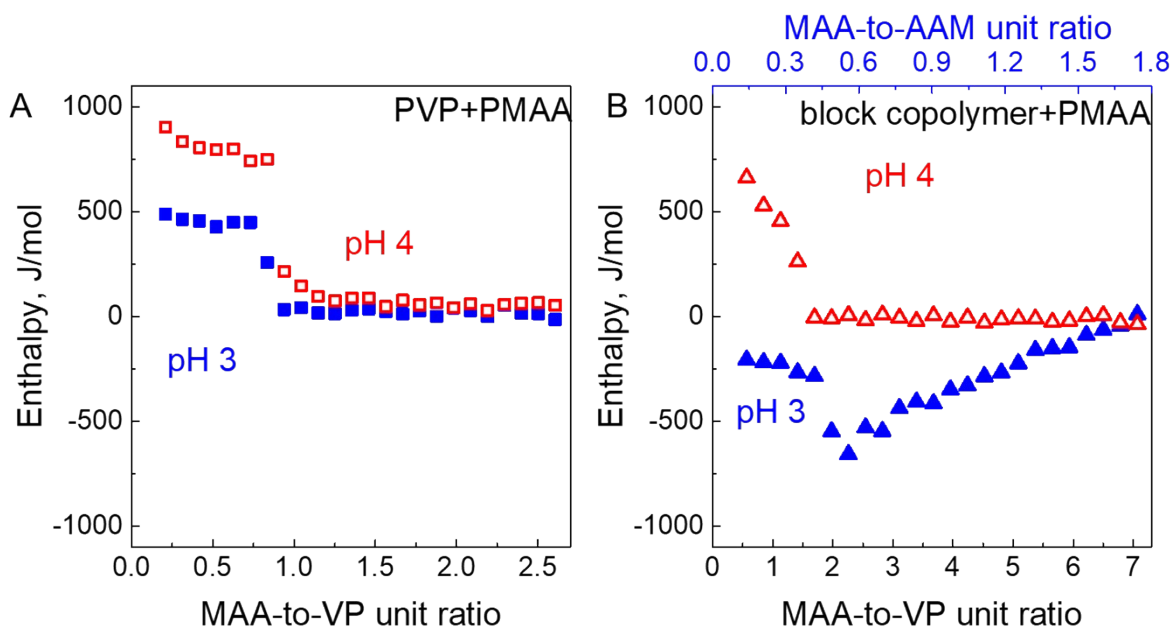


Figure S7. ITC titration of PVP (A), and P(AAm-co-AN)-*b*-PVP (B) with PMAA at pH 3 (closed symbols) and pH 4 (open symbols) at $55 \text{ }^\circ\text{C}$.

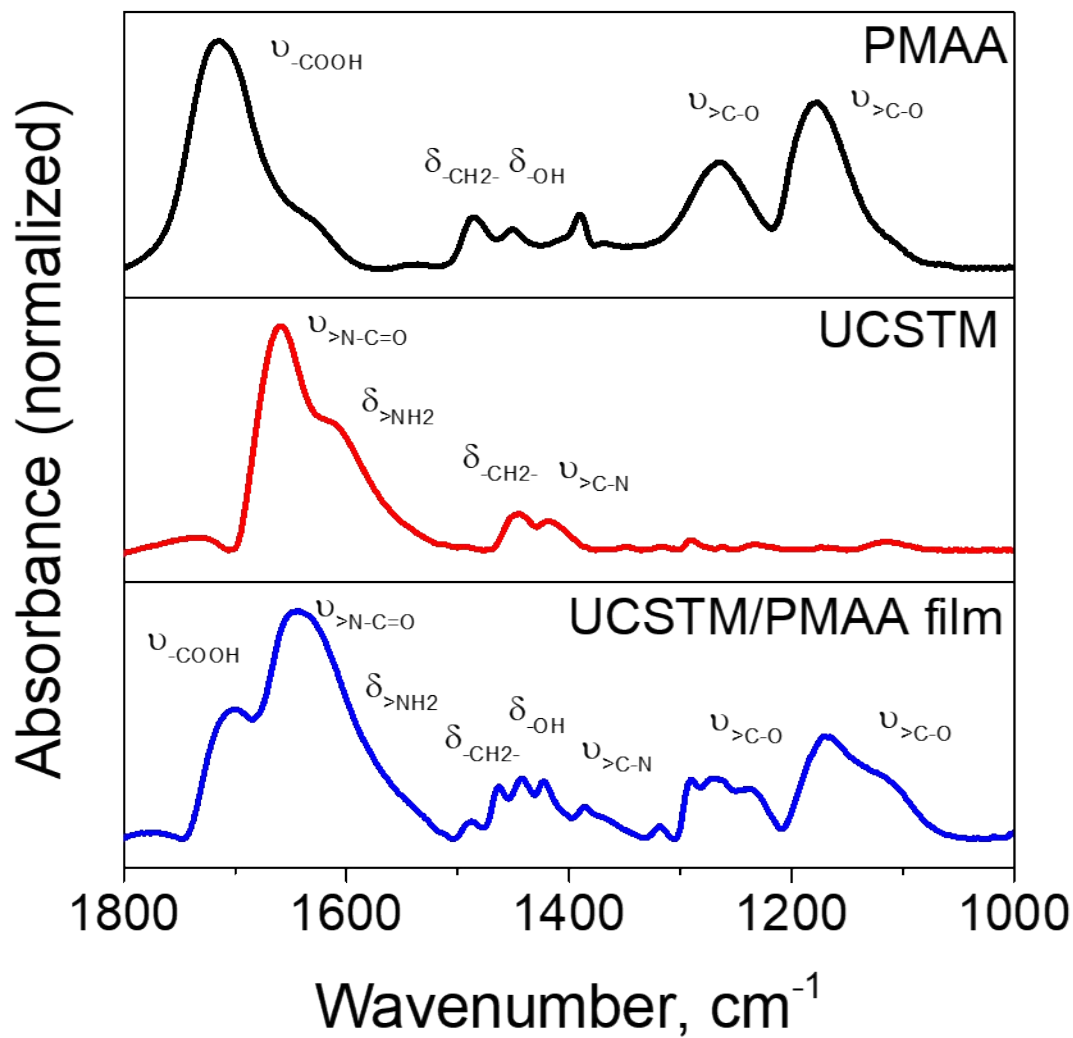


Figure S8. FTIR spectra of PMAA and UCSTM drop casted and dried on pre-cleaned silicon wafers, as well as UCSTMs/PMAA films deposited at pH 3.

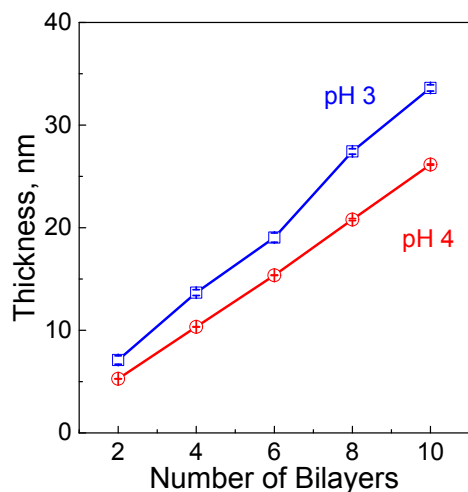


Figure S9. Ellipsometric dry thickness of PVP/PMAA films deposited at pH 3 (open squares, blue) and pH 4 (open circles, red) at room temperature.

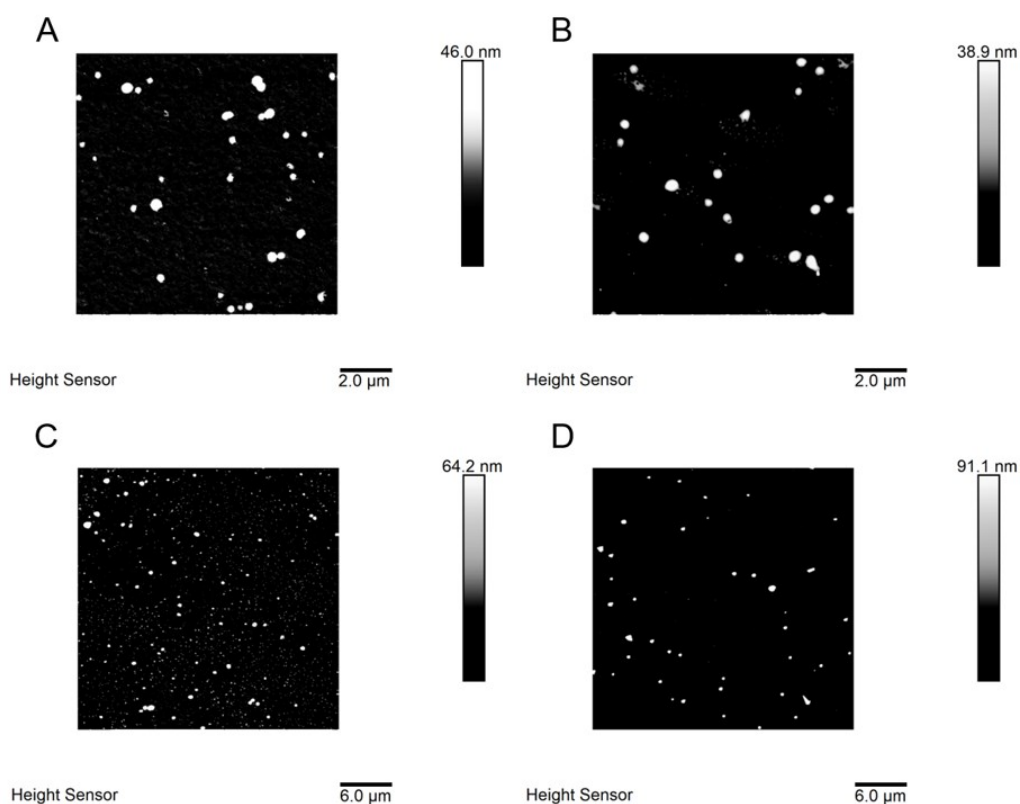


Figure S10. AFM images of monolayer of micelles deposited at pH 3 (A) and 4 (B) and 2 bilayer (UCSTM/PMAA) films deposited at pH 3 (C) and 4 (D) on silicon substrate. All substrates were primed with BPEI/PMAA layer prior to the micelles deposition.

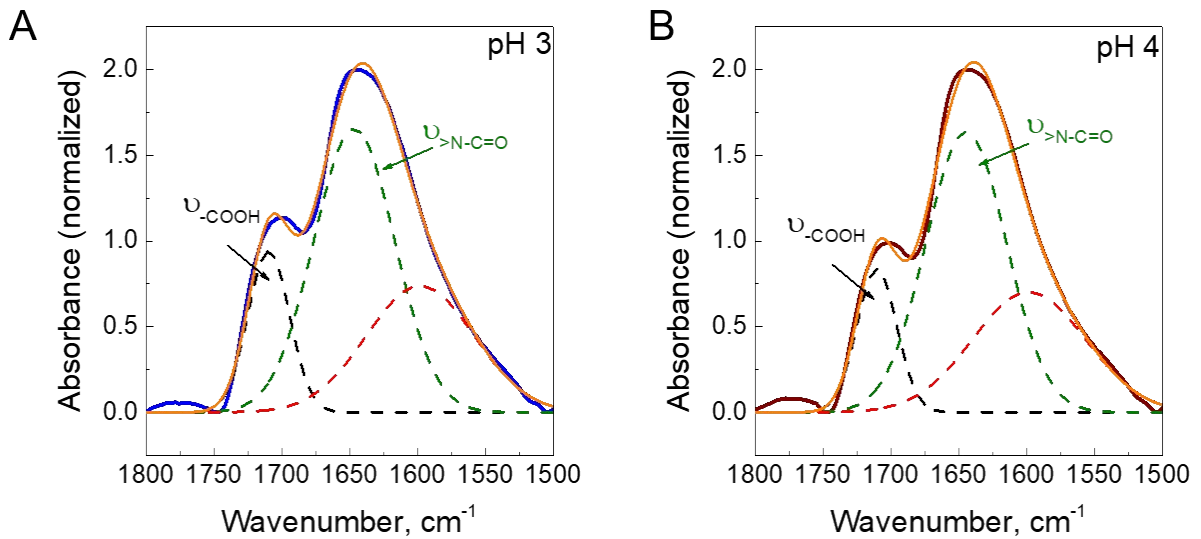


Figure S11. Deconvolution of FTIR spectra of UCSTM/PMAA films deposited at pH 3 (A) and pH 4 (B) in the 1500-1800 cm^{-1} region. The ratio of integrated intensities of stretching vibrations of PMAA ($\nu_{\text{-COOH}}$) to those of stretching vibrations of P(AAm-co-AN) block of UCSTMs ($\nu_{\text{>N-C=O}}$) is 0.3 ± 0.01 and 0.25 ± 0.01 at pH 3 and 4, respectively, suggesting a lower content of PMAA in films deposited at pH 4.

Fitting Model For Neutron Reflectometry Data:

Below is the model used for fitting the data obtained from neutron reflectometry experiments.

The thicknesses of the three stacks are then:

$$d_H^{pure} = \sum_{l=0}^7 d_l$$

$$d_D^{pure} = \sum_{l=8}^9 d_l$$

$$d_{H1}^{pure} = \sum_{l=10}^{17} d_l$$

where the initial bilayer thickness d_l (or d_0) is a fitted parameter.

If the material deposited in sharply defined bilayers, the summation results in the total thickness of the film. The marker layer, d_D , yields information on how much the bilayers intermix with preceding and subsequently deposited material, so we can fit d_D as well and then we can get the thicknesses of the H-blocks, where in addition to d_0 and ϵ we now fit d_D .

$$d_H = \sum_{l=0}^8 d_l - d_D/2,$$

$$d_{H1} = \sum_{l=9}^{17} d_l - d_D/2,$$

To breakdown the SLD data, the varying film density should be taken into account and can be represented as:

$$\Sigma = \rho \frac{N_A \sum_j b_j}{M} \equiv \rho S,$$

where ρ is the mass density, N_A Avogadro's number, $\sum_j b_j$ the sum of the scattering lengths in the formula unit (e.g. monomer), and M the atomic mass of the formula unit. In this way, we can separate the mass density from the stoichiometry S for each of the components of the film (PMAA, dPMAA, UCSTMs, and H₂O) and treat mass density as a fitted parameter.

The non-deuterated strata of the film follow the same hydration and mixing, except with PMAA rather than dPMAA,

$$\Sigma_H = \rho(f_{H_2O}S_{H_2O} + (1 - f_{H_2O})[f_{PMAA}S_{PMAA} + (1 - f_{PMAA})S_{UCSTM}]),$$

where f_{H_2O} is the water volume fraction, f_{PMAA} the fraction of PMAA deposited per bilayer. The SLD of marker layer is the following:

$$\Sigma_D^{pure} = \rho(f_{H_2O}S_{H_2O} + (1 - f_{H_2O})[f_{PMAA}S_{dPMAA} + (1 - f_{PMAA})S_{UCSTM}]),$$

The ratio between the fitted thickness of the marker layer and thickness of an unmixed marker bilayer d_D^{pure} can be used to constrain the SLD of the marker layer. The actual marker layer does not go down as a perfectly sharp, pure layer, but is instead distributed over a greater thickness so that the marker layer has mixed into the material from the adjacent layers,

$$f_d = d_D^{pure}/d_D$$

$$\Sigma_D = f_d \Sigma_D^{pure}$$

Parameters obtained from and used in the fit are presented in Table S1 and S2

Table S1. Model parameters for a (UCSTM/PMAA)₈/(UCSTM/dPMAA)₂/(UCSTM/PMAA)₈ film deposited from buffer solution at pH 3.0.

Layer	Nb , (Å ⁻²)	d , (Å)	σ_{int} , (Å)	d_0 , (Å)	f_{PMAA}	f_{H_2O}	ρ , g/cm ³	f_d
(UCSTM/PMAA) ₈	1.07e-6	285.7	230.8	44.7	0.82	0.10	1.11	0.62
(UCSTM/dPMAA) ₂	2.52e-6	144.1	126.3					
(UCSTM/PMAA) ₈	1.07e-6	285.7	77.6					
BPEI	5.0e-7	27.3	27.3	N/A	N/A	N/A	N/A	N/A
SiO2	3.23e-6	21.6	6.0					
Si	2.05e-6	100.0	10.0					

Table S2. Model parameters for a (UCSTM/PMAA)₈/(UCSTM/dPMAA)₂/(UCSTM/PMAA)₈ film deposited from buffer solution at pH 4.0.

Layer	Nb , (Å ⁻²)	d , (Å)	σ_{int} , (Å)	d_0 , (Å)	f_{PMAA}	f_{H_2O}	ρ , g/cm ³	f_d
(UCSTM/PMAA) ₈	1.11e-6	207.0	173.5	24.2	0.65	0.09	1.10	0.69
(UCSTM/dPMAA) ₂	2.37e-6	70.7	62.9					
(UCSTM/PMAA) ₈	1.11e-6	207.0	44.5					
BPEI	4.95e-7	15.1	15.1	N/A	N/A	N/A	N/A	N/A
SiO2	3.2e-6	24.3	10.0					
Si	2.02e-6	100.0	5.0					