Supporting information

An ultrathin, strong, flexible composite solid electrolyte for high-voltage lithium metal batteries

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Figure S1. Transmission electron microscopy (TEM) image of the BNNFs.

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Figure S2. Cross-sectional SEM images of the PAN-LiClO₄-BNNF composite electrolyte at magnifications of 5000 (left) and 20000 (right).



Figure S3. XRD patterns of the PAN-LiClO₄ polymer electrolyte and the PAN-LiClO₄-BNNF composite electrolyte.



Figure S4. The TGA curve of the PAN-LiClO₄-BNNF composite electrolyte.



Figure S5. Chronoamperometry of the Li/PAN-LiClO₄/Li symmetric cell with an applied voltage of 0.1 V. The inset is the impedance spectra of the symmetric cells before and after polarization.



Figure S6. LSV scans for the PAN-LiClO₄-BNNF composite electrolyte from 0 to 3 V.



Figure S7. Top SEM images of the BNNF layer of PBCEB at magnifications of 30000 (left) and 80000 (right).



Figure S8. Cross sectional SEM image of the BNNF-PAN-LiClO₄-BNNF-BNNF electrolyte.



Figure S9. The EIS results of the Li/PAN-LiClO₄-BNNF/Li and Li/PBCEB/Li symmetric cells.