

Supporting Information for

Understanding the Reaction Mechanism and Performances of 3d Transition Metal Cathodes for All-solid-state Fluoride Ion Batteries

Datong Zhang¹, Kentaro Yamamoto^{1}, Aika Ochi¹, Yanchang Wang¹, Takahiro Yoshinari¹, Koji Nakanishi¹, Hiroyuki Nakano¹, Hidenori Miki^{1,2}, Shinji Nakanishi², Hideki Iba², Tomoki Uchiyama¹, Toshiki Watanabe¹, Koji Amezawa³, and Yoshiharu Uchimoto¹*

¹Graduate School of Human and Environmental Studies, Kyoto University, Yoshida-nihonmatsucho, Sakyo, Kyoto 606-8501, Japan

²Battery Research Div., Toyota Motor Corporation, 1200 Mishuku, Shizuoka 410-1193, Japan

³IMRAM, Tohoku University, 2-1-1 Katahira, Aoba, Sendai, Miyagi 980-8577, Japan

*Corresponding Author (yamamoto.kentaro.4e@kyoto-u.ac.jp)

ORCID: Kentaro Yamamoto (0000-0002-8739-4246)

Table S1. The parameters of RF sputtering on LaF₃ substrate

Element	Application	Thickness / nm	RF Sputtering Power / W
Cu	Working electrode	2.3	50
Co	Working electrode	2.3	120
Ni	Working electrode	2.3	100
Pb	Reference electrode	1000	50
Pt	Current collector	50	50
W	Current collector	50	120

RF Sputtering Temperature: Room temperature (~25 °C)

Atmosphere: Argon

Atmosphere pressure: 1.0 Pa

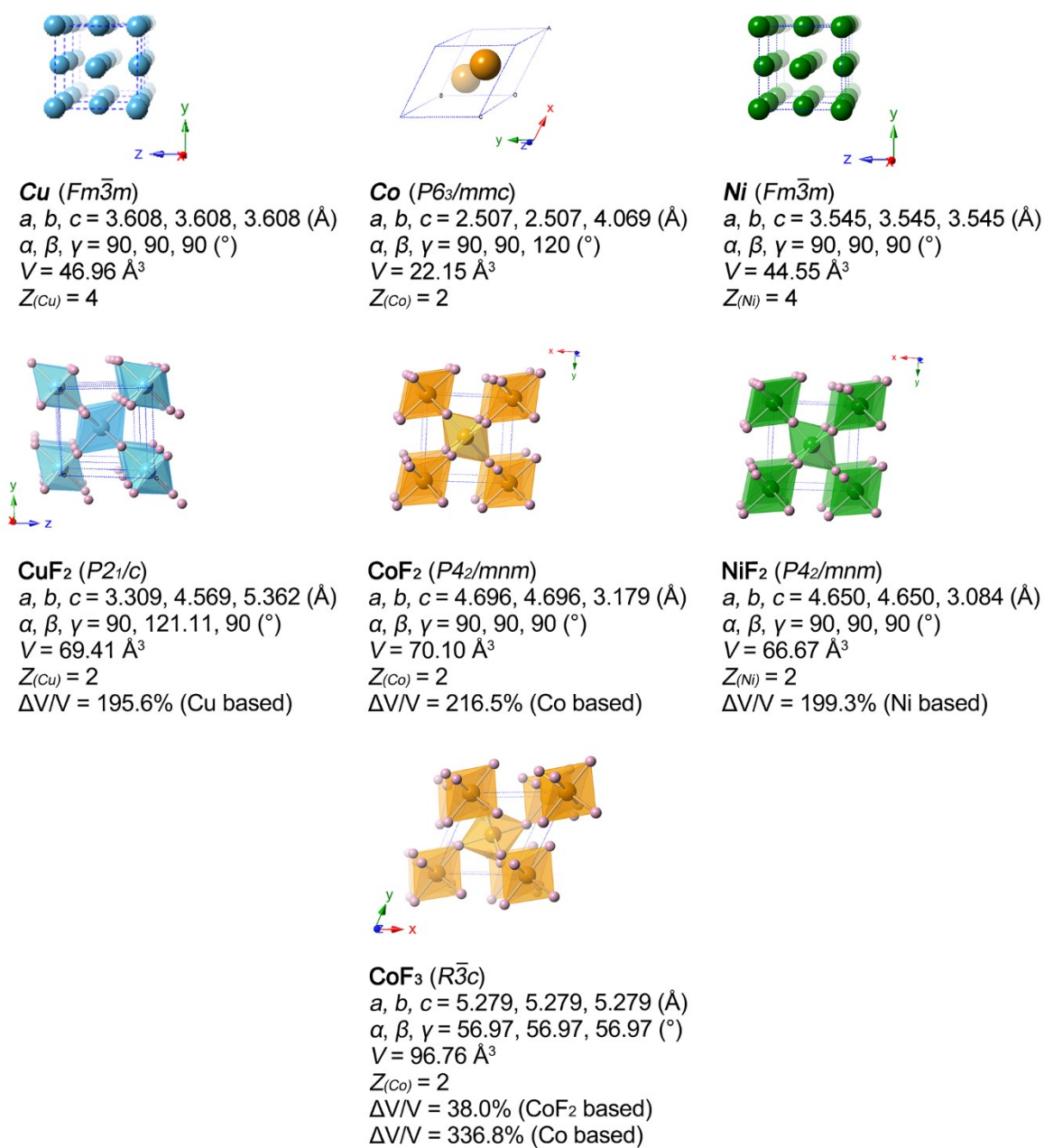


Figure S1. The crystallographic models of Cu, Co, Ni and their fluorides. The volumetric changes are listed under each model. ¹⁻⁶

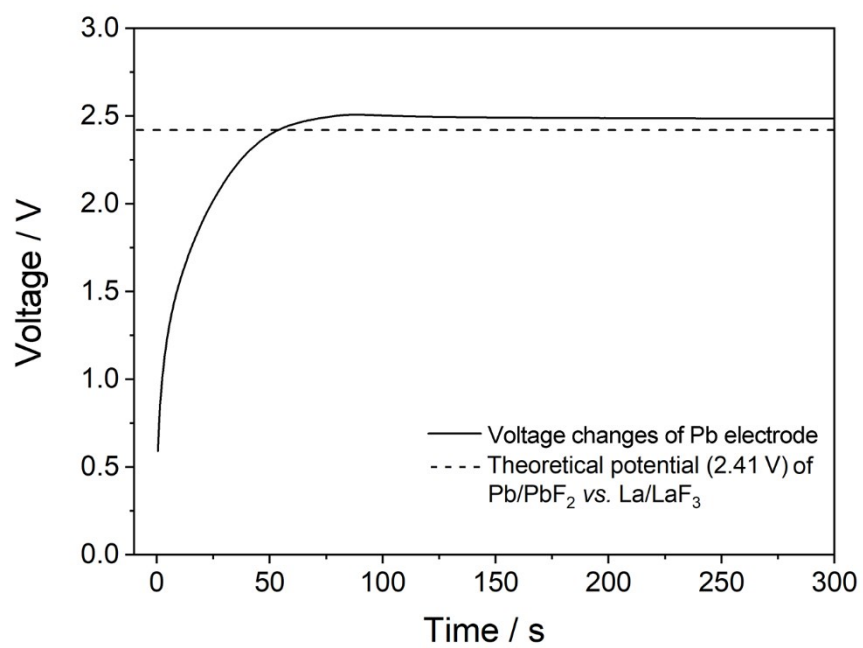


Figure S2. The voltage changes of Pb reference electrode versus counter electrode upon pre-fluorination treatment. The dash line represents the theoretical voltage (2.41 V) of Pb/PbF₂ vs. La/LaF₃.

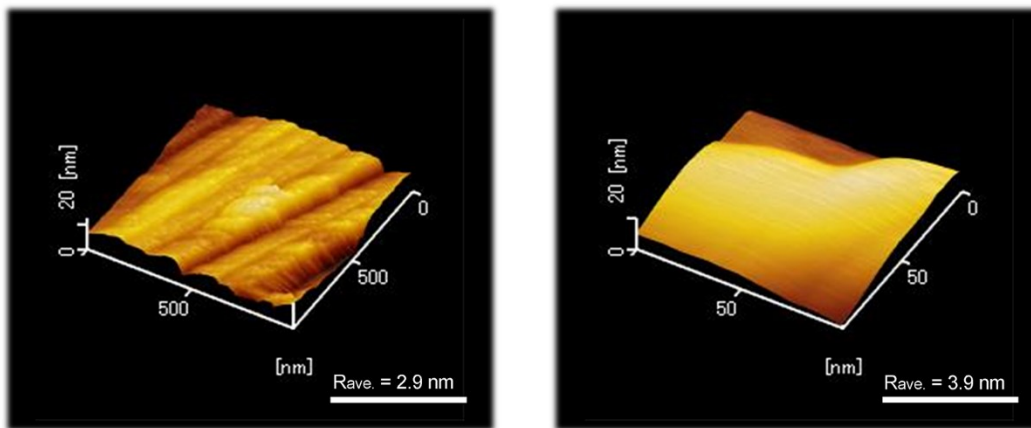


Figure S3. Surface smoothness of the substrate. Atomic force microscopy (AFM) images of (a) mirror polished LaF_3 substrate and (b) Cu deposited LaF_3 substrate. The surface roughnesses of the substrate before and after the Cu deposition are 2.9 and 3.9 nm, respectively.

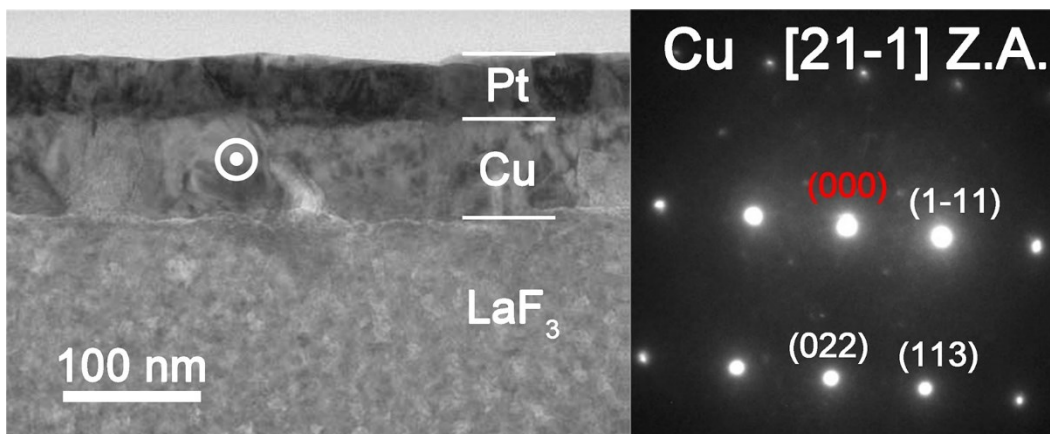


Figure S4. The cross-sectional TEM images of Pt, Cu film and LaF₃ substrates, and the nano-area electron diffraction (NAED) patterns of Cu along [21 $\bar{1}$] zone axis.

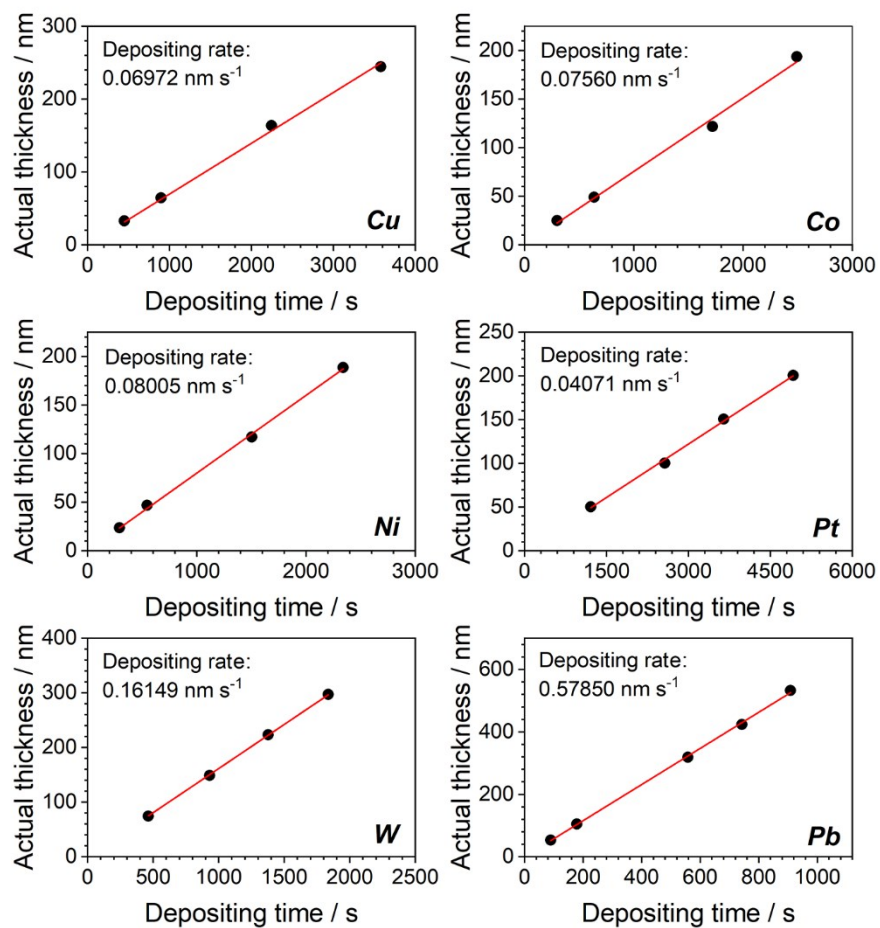


Figure S5. The direct proportional fittings ($y=kx$ model) of depositing rates of all utilized metals.

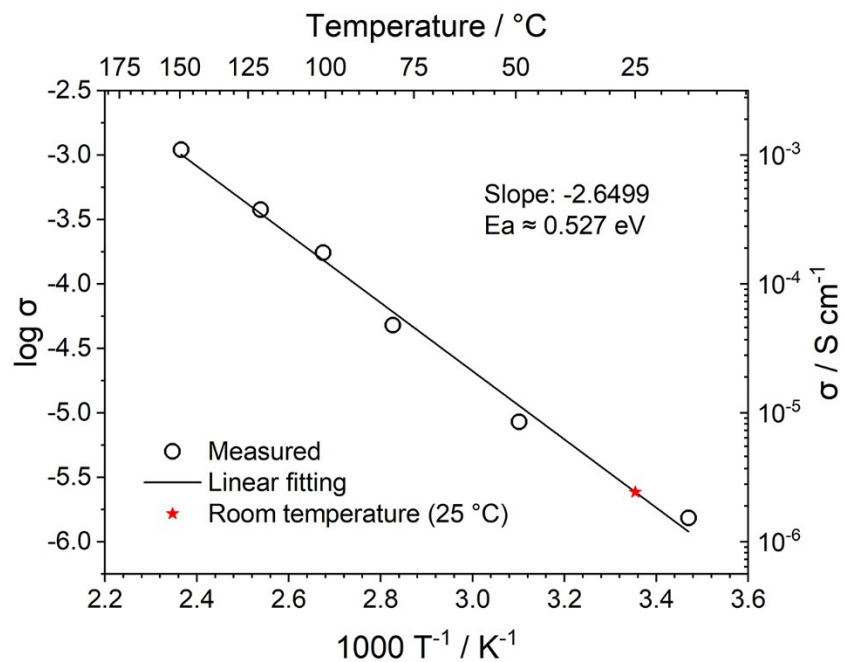


Figure S6. The ionic conductivities of LaF_3 substrates at temperatures of 15, 50, 80, 100, 125, 150 °C, respectively, and corresponding Arrhenius linear fitting.

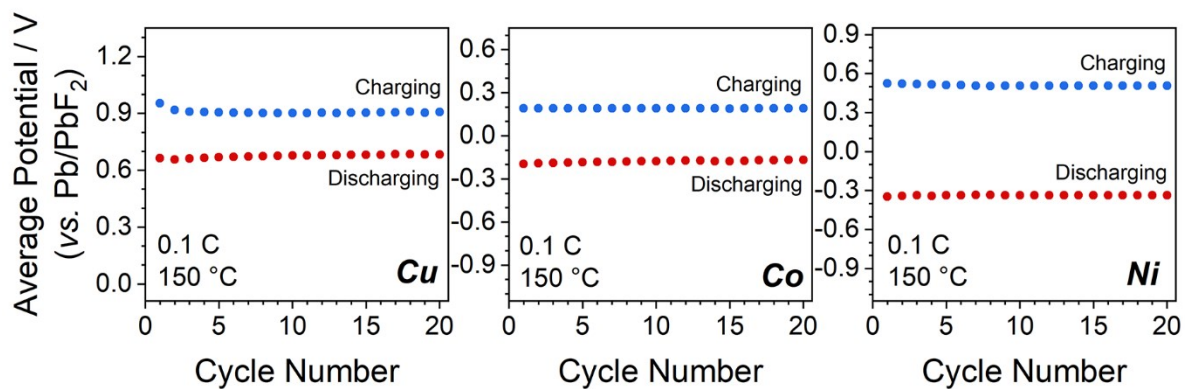


Figure S7. The average working potentials in the first 20 cycles of Cu, Co and Ni at 0.1 C and 150 °C.

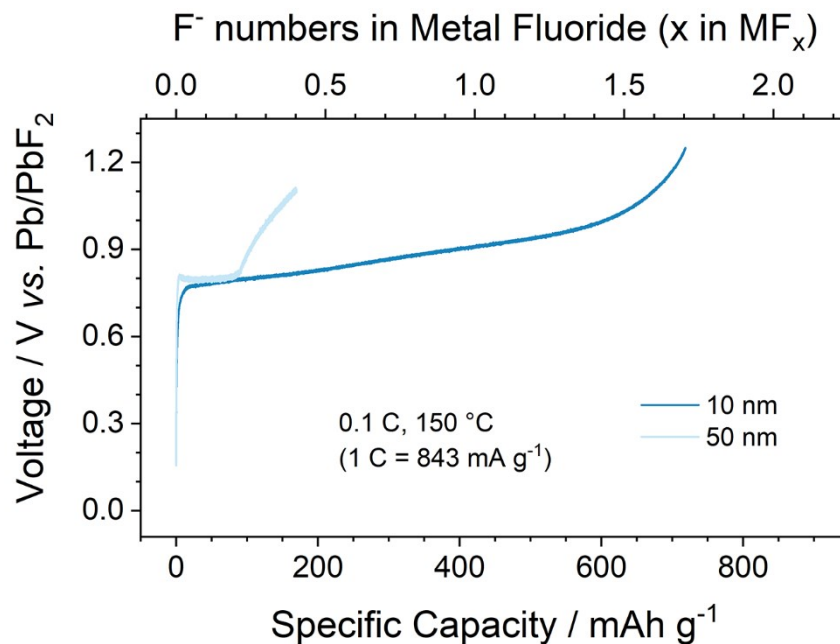


Figure S8. The electrochemical properties of Cu with various thicknesses. The 10 nm sample possessed a similar initial performance with 2.3 nm sample in this study, while the thickest sample only delivered an initial charging capacity of 170 mAh g⁻¹ which corresponded to c.a. 20% of theoretical values, i.e. diffusion depth of 10 nm out of 50 nm.

REFERENCES

1. Owen, E. A.; Yates, E. L. Precision measurements of crystal parameters Locality: synthetic Sample: at T = 18 °C. *Philos. Mag.* **1933**, *15*, 472-488.
2. Fischer, P.; Hälg, W.; Schwarzenbach, D.; Gamsjäger, H. Magnetic and crystal structure of copper (II) fluoride. *J. Phys. Chem. Solids* **1974**, *35* (12), 1683-1689.
3. Zemmann, J. Crystal structures, 2nd edition. Vol. 1 by R. W. G. Wyckoff. *Acta Crystallogr.* **1965**, *18* (1), 139.
4. Jorgensen, J. E.; Smith, R. I. On the compression mechanism of FeF₃. *Acta Crystallogr. Sect. B: Struct. Sci.* **2006**, *62* (Pt 6), 987-992.
5. Costa, M. M. R.; Paixão, J. A.; de Almeida, M. J. M.; Andrade, L. C. R. Charge Densities of Two Rutile Structures: NiF₂ and CoF₂. *Acta Crystallogr. Sect. B: Struct. Sci.* **1993**, *49* (4), 591-599.
6. Hepworth, M. A.; Jack, K. H.; Peacock, R. D.; Westland, G. J. The Crystal Structures of the Trifluorides of Iron, Cobalt, Ruthenium, Rhodium, Palladium and Iridium. *Acta Crystallogr.* **1957**, *10* (1), 63-69.