

Supporting Information to

Highly sensitive detection of rutin in pharmaceuticals and human serum using ITO electrodes modified with vertically-ordered mesoporous silica-graphene nanocomposite films

Xinyu Ma,^{‡,a} Wenyan Liao,^{‡,b} Huaxu Zhou,^a Yun Tong,^a Fei Yan,^{a,*} Hongliang Tang,^{c,*} and

Jiyang Liu^{a,*}

^a Department of Chemistry, Zhejiang Sci-Tech University, 928 Second Avenue, Xiasha Higher Education Zone, Hangzhou, 310058, PR China

^b Affiliated International Zhuang Medicine Hospital, Guangxi University of Chinese Medicine, Nanning, 530023, PR China

^c The First Affiliated Hospital of Guangxi University of Chinese Medicine, Nanning, 530023, PR China.

* Corresponding author. E-mail: feifei19881203@126.com; tanghongliang@gxctmu.edu.cn; liujy@zstu.edu.cn

‡ These two authors contributed equally.

Table of Contents

S1. XPS characterizations of GO/ITO and ErGO/ITO

S2. EIS characterizations of VMSF/ErGO/ITO

S3. CVs of rutin at the VMSF/ITO and VMSF/ErGO/ITO

S4. The effect of scan rate on the CV responses

S5. Transmittance spectra of bare ITO and VMSF/ErGO/ITO

S6. Optimized conditions for electrochemical detection

S6.1 The concentration of GO

S6.2 pH of supporting electrolyte

S6.3 Preconcentration time

S7. Anti-interference study of the VMSF/ErGO/ITO

S1 XPS characterization of GO/ITO and ErGO/ITO

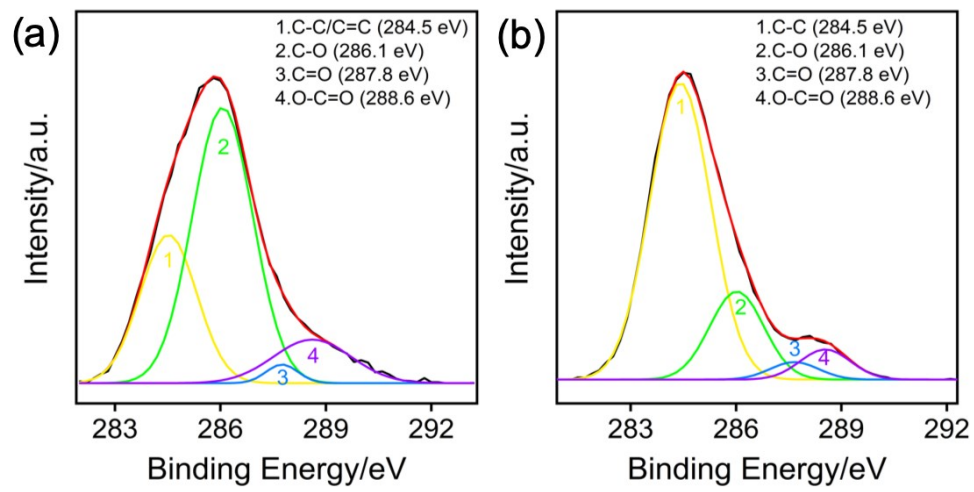


Fig. S1 C 1s XPS profiles of GO/ITO and ErGO/ITO.

S2 EIS characterization of VMSF/ErGO/ITO

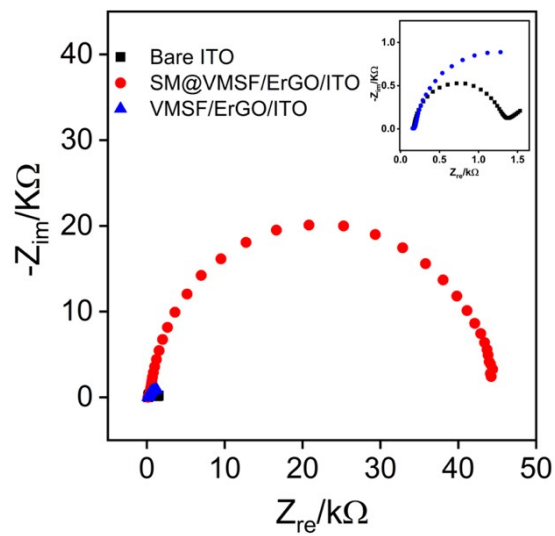


Fig. S2 EIS plots of the Bare ITO, SM@VMSF/ErGO/ITO and VMSF/ErGO/ITO electrodes in 0.1 M KHP solution containing 2.5 mM $\text{Fe}(\text{CN})_6^{3-/4-}$ at a frequency of 0.1 Hz to 100 kHz..

S3. CVs of rutin at the VMSF/ITO and VMSF/ErGO/ITO

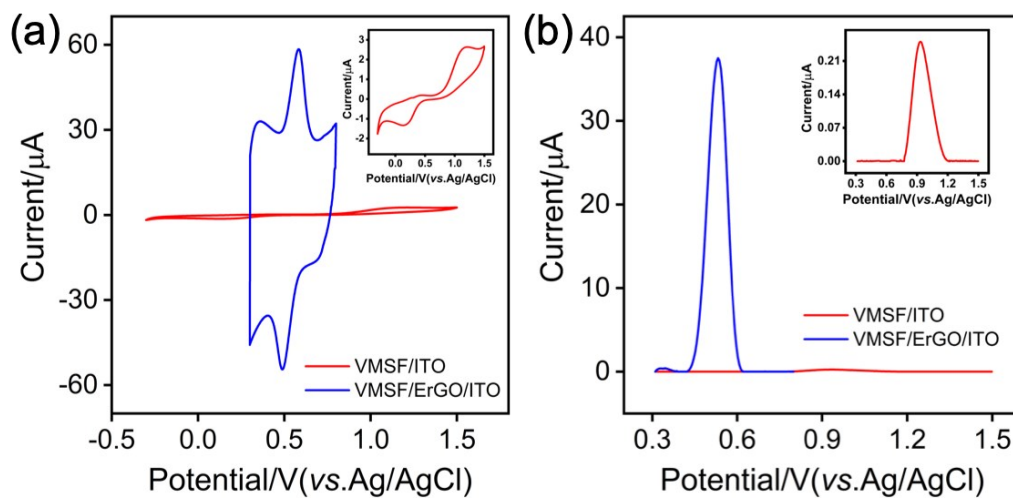


Fig. S3 CVs (a) and DPVs (b) of 30 μM rutin at the VMSF/ITO and VMSF/ErGO/ITO electrodes in 0.1 M PBS (pH = 3). The scan rate for CVs was 50 mV/s and the insets were the corresponding amplified view of the VMSF/ITO electrode.

S4. The effect of scan rate on the CV responses

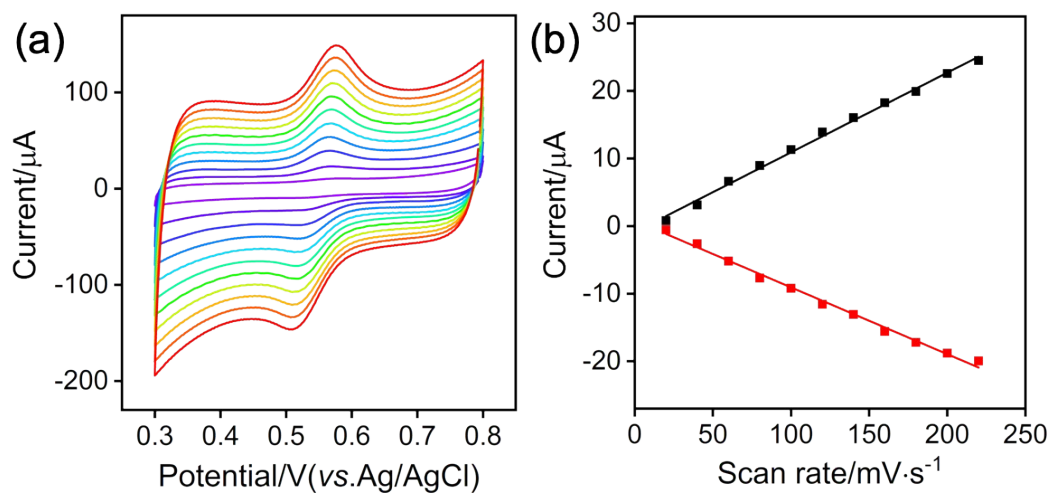


Fig. S4 (a) CV curves obtained from VMSF/ErGO/ITO in PBS (0.1 M, pH 3.0) containing 30 μM rutin at various scan rates. (b) The dependence of anodic and cathodic peak potential on scan rate.

S5. Transmittance spectra of bare ITO and VMSF/ErGO/ITO

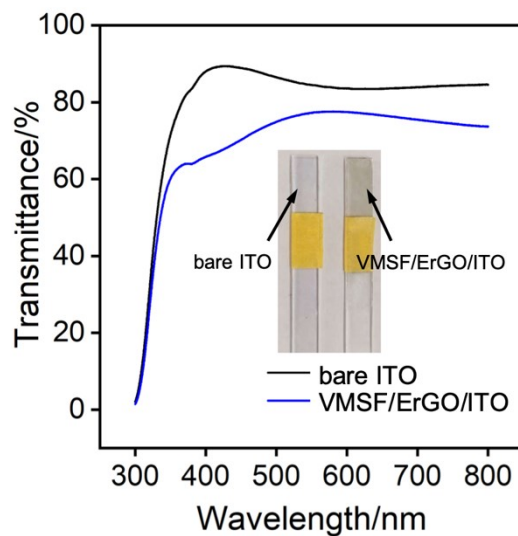


Fig. S5 Transmittance spectra of the bare ITO and VMSF/ErGO/ITO. Inset was the photographs of the bare ITO and VMSF/ErGO/ITO electrode. The concentration of GO used here is 0.1 mg/mL.

S6. Optimized conditions for electrochemical detection

S6.1. The concentration of GO

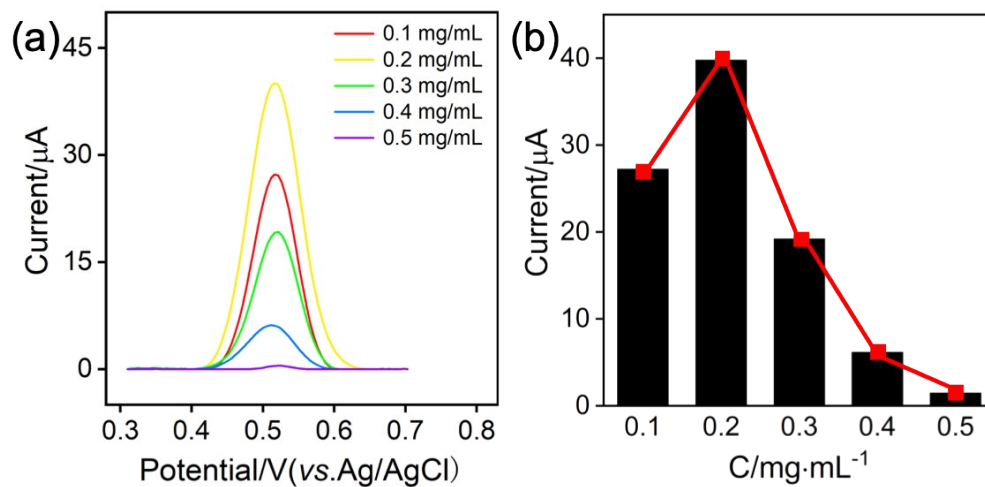


Fig. S6 (a) DPVs of the VMSE/ErGO/ITO electrode prepared by various concentrations of GO in a 0.1 M PBS (pH=3) solution containing 30 μM rutin. (b) The dependence of anodic peak current on the GO concentration.

S6.2 pH of supporting electrolyte

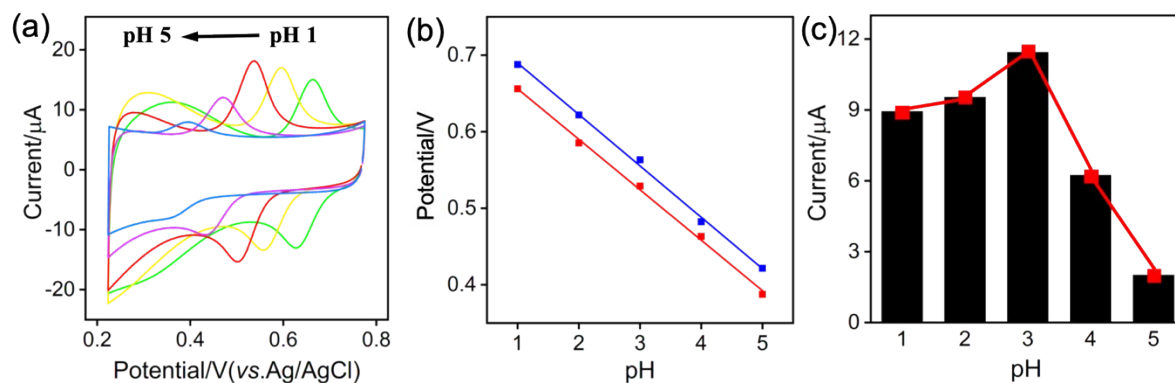


Fig. S7 (a) CVs of the VMSF/ErGO/ITO electrode in a 0.1 M PBS solution containing 30 μM rutin at various pH values. (b) The dependence of cathodic peak potential (E_{pc}) and anodic peak potential (E_{pa}) on the pH value. (c) The dependence of anodic peak current on the pH value.

S6.3 Preconcentration time

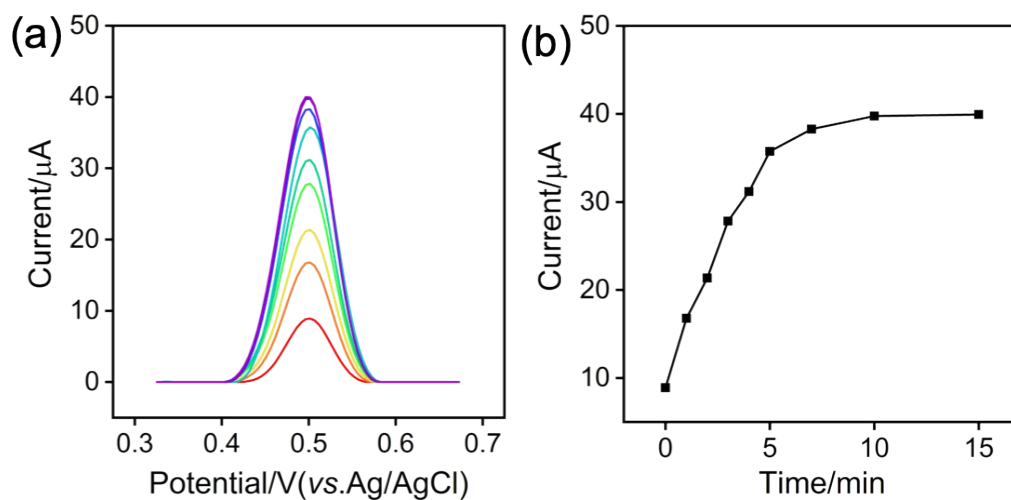


Fig. S8 (a) DPVs of the VMSF/ErGO/ITO electrode in a 0.1 M PBS (pH=3) solution containing 30 μM rutin at different accumulation time. (b) The dependence of anodic peak current on the accumulation time.

S7. Anti-interference study of the VMSF/ErGO/ITO

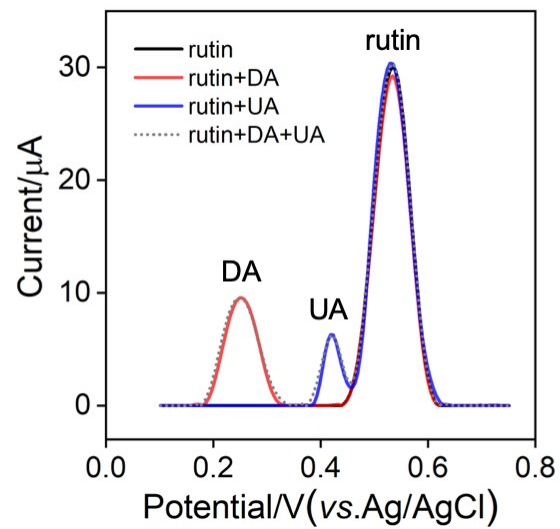


Fig. S9 DPVs of the VMSF/ErGO/ITO electrode in a 0.1 M PBS (pH=3) solution containing 20 μM rutin in the absence and presence of 40 μM DA or/and UA.