*Supporting Information for*

## **Doping engineering of the flexible polyaniline electrochromic material through H2SO4-HClO<sup>4</sup> multiple acids for the radiation regulation in snow environment**

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## <span id="page-2-0"></span>**General methods**

The value of  $\varepsilon$  can be calculated by weighting  $(1 - R(\lambda))$  (namely spectral emittance) with the black body spectrum for a particular wavelength and integrating over the entire measured wavelength range according to the following two equations:

$$
B_{(\lambda)} = \frac{c_1 \lambda^{-5}}{\exp[c_2/(\lambda T)] - 1}
$$
 (1)

$$
\varepsilon = \frac{\int_{\lambda \min}^{\lambda \max} (1 - R(\lambda)) B(\lambda) d\lambda}{\int_{\lambda \min}^{\lambda \max} B(\lambda) d\lambda}
$$
 (2)

where c<sub>1</sub> is the first radiation constant (3.7418 × 108 W  $\mu$ m<sup>4</sup> m<sup>-2</sup>), c<sub>2</sub> is the second radiation constant (1.4388 × 10<sup>4</sup> μm K), λ is the wavelength, and T is the temperature.

**The route for the assembly of the device:** The PANI films doped with  $H_2SO_4$ -HClO<sup>4</sup> on the Au substrate are used as the front and back electrodes, and nanoporous polyethylene (PE) is devoted to a protective layer. The (poly vinylidenefluoridehexafluoro propylene) copolymer containing electrolyte is used as the electrolyte film. After each layer is assembled together, the device is firmly compressed using a roll-toroll mode at 190 °C.

**Cyclic durability testing of polyaniline:** After the initial test of emissivity, the device was scanned for cyclic voltammetry. After 10 cycles of scanning, the emissivity was tested in situ.

**The simulation of the device:** The thermal properties of the device were simulated by finite element method, implemented in the commercial software ANSYS. In all cases, steady-state thermal model was used to record the temperature at which the pipe reaches the steady state of heat transfer. In addition, the transient thermal module was also used to record the temperature of the pipe at different times. For simplicity, the heat dissipation from the device to the air is only mediated by natural convection and radiation. And, the pipe is set to have different heat flow.

**The thermal regulation performance test of the device:** The apparatus is composed of a heater and insulation materials blocking the conductive and convective heat exchange with surroundings. The EC device was placed on the apparatus. Subsequently, a thermal imager was used to record the surface temperature of the EC device.

EC devices	Working materials	Doping acids	Tunable spectral range	Response time	$\Delta \varepsilon$ at 8-14 $\mu$ m	Flexible	UV <sub></sub> reflectivity	Working voltage	Ref.
Device 1	Polyaniline	$H2SO4-HClO4$	$2.5 \mu m - 25 \mu m$	$13-14 s$	0.47	Yes	$>80\%$	$0.5 V/-0.4$ V	<b>This</b> work
Device 2	Polyaniline/ $WO_3$	Camphorsulfonic acid	$5 \mu m-20 \mu m$	9 <sub>s</sub>	0.41	NO	<10%	$0.45V/-$ 0.2V	
Device 3	Polyaniline/poly( diphenyl amine)	Poly(styrene sulfonate)	$2.5 \mu m - 18 \mu m$	2s	0.39	Not reported	<10%	$0 V/-1.1$ V	$\overline{c}$
Device 4	Polyaniline/poly( diphenyl amine)	Poly(styrene sulfonate)	$2.5 \mu m - 25 \mu m$	2s	< 0.4	Yes	$<$ 20%	$0.8 V/-1.1$ V	3
Device 5	Polyaniline	Poly(styrene sulfonate)	$2.5 \mu m - 25 \mu m$	Not reported	0.35	Yes	$< 5\%$	$0.3 V/-0.9$ V	4
Device 6	Ppoly-O- anisidine	H <sub>2</sub> SO <sub>4</sub>	$2.5 \mu m - 25 \mu m$	Not reported	< 0.3	NO	$< 5\%$	$0.2 V/-1.1$ V	5
Device 7	Polyaniline	H <sub>2</sub> SO <sub>4</sub>	$2.5 \mu m - 25 \mu m$	Not reported	< 0.3	NO	Not reported	$1.0V/-$ 0.6V	6
Device 8	Polyaniline	H <sub>2</sub> SO <sub>4</sub>	$2.5 \mu m - 25 \mu m$	$<$ 5 s	< 0.2	NO	<b>Not</b> reported	$0.55$ V/- 0.25V	$\overline{7}$
Device 9	Polyaniline	Dodecylbenzene sulfonate acid	$2.5 \mu m - 25 \mu m$	Not reported	< 0.4	Yes	Not reported	$0.45$ V/- 0.25V	$8\,$
Device 9	Polyaniline	Dodecylbenzene sulfonate acid	$2.5 \mu m - 25 \mu m$	Not reported	< 0.3	NO	$<$ 20%	$0.85$ V/- $0.25~\mathrm{V}$	$\mathbf{9}$
Device 10	Polyaniline	Dodecylbenzene sulfonate acid	$2.5 \mu m - 25 \mu m$	$>15$ s	< 0.4	<b>YES</b>	$<$ 20%	$0.2 V/-$ 1.2V	10
Device 11	Polyaniline	H <sub>2</sub> SO <sub>4</sub>	$2.5 \mu m - 25 \mu m$	Not reported	0.39	<b>YES</b>	Not reported	<b>Not</b> reported	11

<span id="page-3-0"></span>**Table S1.** Comparison for the performances of the EC device with the reported devices.



<span id="page-4-0"></span>**Fig. S1**. EIS of the PANI films at different polymerization current densities. (a) 0.0001 A/cm<sup>2</sup>. (b) 0.00015 A/cm<sup>2</sup>. (c) 0.0002 A/cm<sup>2</sup>. (d) 0.00025 A/cm<sup>2</sup>. (e) 0.0003 A/cm<sup>2</sup>. (f) R<sup>s</sup> of the PANI films at different polymerization current densities.



<span id="page-4-1"></span>**Fig. S2.** SEM images of PANI films at different polymerization current densities. (a) 0.0001 A/cm<sup>2</sup>. (b) 0.00015 A/cm<sup>2</sup>. (c) 0.0002 A/cm<sup>2</sup>. (d) 0.00025 A/cm<sup>2</sup>. (e) 0.0003A/cm<sup>2</sup> . (f) The thickness of the PANI films at different polymerization current densities. Inset: SEM cross-section images of PANI films at different polymerization current densities.



<span id="page-5-0"></span>**Fig. S3.** The ratio of chlorine element to sulfur element at different polymerization current densities.



<span id="page-5-1"></span>**Fig. S4.** (a) Emittance curves of PANI films prepared at different polymerization current densities at potential of 0.4 V. (b) Emittance curves of PANI films prepared at different polymerization current densities at potential of -0.5 V.



<span id="page-6-0"></span>**Fig. S5.** The emissive power curves of PANI films prepared at different polymerization current densities at potential of  $0.4$  V and  $-0.5$  V. (a)  $0.0001$  A/cm<sup>2</sup>. (b)  $0.00015$  A/cm<sup>2</sup>. (c) 0.0002 A/cm<sup>2</sup> . (d) 0.00025 A/cm<sup>2</sup> . (e) 0.0003A/cm<sup>2</sup> .



<span id="page-6-1"></span>**Fig. S6.** (a) The I-t curve of the device under applied voltage. (b) Cyclic durability of the device. (c) IR thermal images of the EC device under bending state on the Al plate.



<span id="page-7-0"></span>**Fig. S7.** Schematic of the heat transfer of the EC device in the finite element simulation.



<span id="page-7-1"></span>**Fig. S8.** The schematic setup for thermal regulation of the device.

## <span id="page-7-2"></span>**References:**

- 1. P. Topart and P. Hourquebie, *Thin Solid Films*, 1999, **352**, 243-248.
- 2. P. Chandrasekhar, G. C. Birur, P. Stevens, S. Rawal, E. A. Pierson, and K. L. Miller, *Synth. Met.*, 2001, **119**, 293-294.
- 3. P. Chandrasekhar, B. J. Zay, G. C. Birur, S. Rawal, E. A. Pierson, L. Kauder and S. Theodore, *Adv. Funct. Mater.* 2002, **12**, 95-103.
- 4. P. Chandrasekhar, B. J. Zay, T. Mcqueeney, G. C. Birur, V. Sitaram, R. Menon, M. Coviello and R.L. Elsenbaumer, *Synth. Met.*, 2005, **155**, 623-627.
- 5. L. Tu, C. Jia, X. Weng and L. Deng, *Synth. Met.*, 2011, **161**, 2045-2048.
- 6. S. Wu, C. Jia, X. Fu, X. Weng, J. Zhang and L. Deng, *Electrochim. Acta*, 2013, **88**, 322-329.
- 7. Y. Tian, S. Dou, X. Zhang, L. Zhang, L. Wang, J. Zhao, X. Zhao and Y. Li,

*Synth. Met.*, 2017, **232**, 111-116.

- 8. Y. Tian, X. Zhang, S. Dou, L. Zhang, H. Zhang, H. Lv, L. Wang, J. Zhao and Y. Li, *Sol. Energy Mater. Sol. Cells*, 2017, **170**, 120-126.
- 9. L. Zhang, D. Li, X. Li, B. Wang, G. Xu, X. Zhao, G. Xia, A. Gavrilyuk, J. Zhao and Y. Li, *Dyes Pigm*., 2019, **170**, 107570.
- 10. X. Li, L. Zhang, B. Wang, G. Xu, S. Yu, M. Pan, S. Dou, Y. Li and J. Zhao, *Electrochim. Acta*, 2020, **332**.
- 11. F. Lu, P. Tan and Y. Han, *J. Appl. Polym. Sci*., 2020, e49622.