

Supporting Information

Electrochemiluminescence aptasensor for diethylstilbestrol detection based on resonance energy transfer between Ag₃PO₄-Cu-MOF(II) and silver nanoparticles and its application

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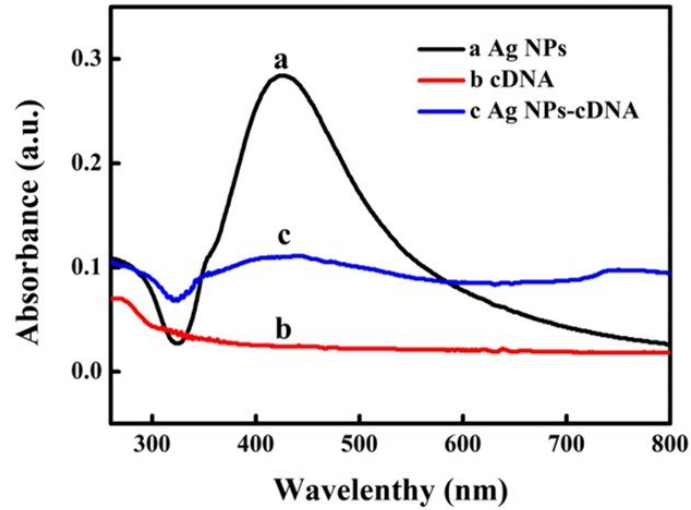


Figure S1. UV-Vis absorption spectra of Ag NPs (a), cDNA (b) and Ag NPs-cDNA (c)

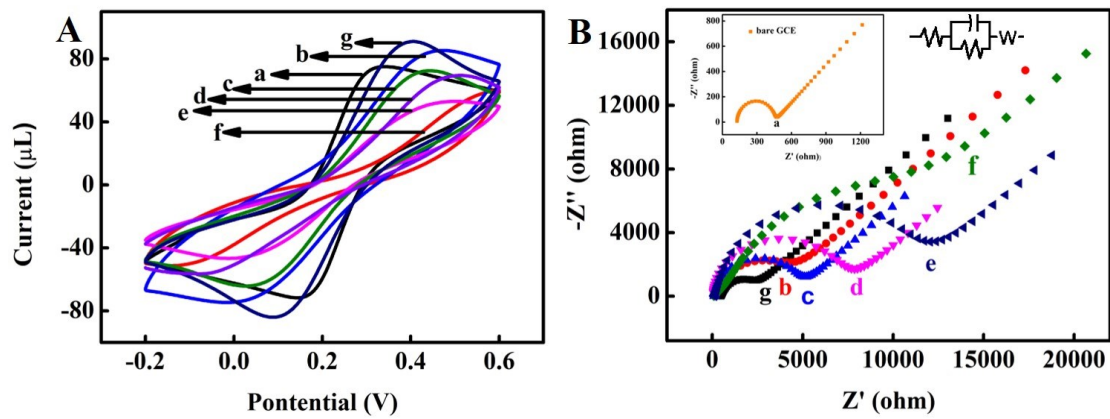


Figure S2. (A) CV curves and (B) EIS of different modified electrodes: (a) bare GCE; (b) Cu-MOF/GCE; (c) Ag_3PO_4 /GCE; (d) Ag_3PO_4 -Cu-MOF/GCE; (e) apt/ Ag_3PO_4 -Cu-MOF/GCE; (f) Ag NPs-cDNA/apt/ Ag_3PO_4 -Cu-MOF/GCE; (g) DES-Ag NPs-cDNA/apt/ Ag_3PO_4 -Cu-MOF/GCE

As can be seen in Figure S3, after modifying Ag_3PO_4 (Figure S3(c)), the impedance of $\text{Ag}_3\text{PO}_4\text{-Cu-MOF}$ did not change much compared with the single Ag_3PO_4 , so the Cu-MOF did not affect the experiment.

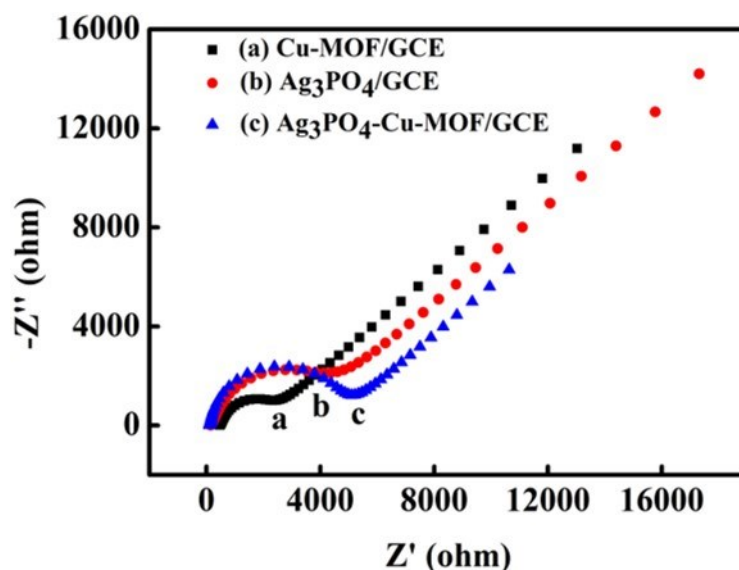


Figure S3. EIS of different modified electrodes: (a) Cu-MOF/GCE; (b) Ag_3PO_4 /GCE; (d) $\text{Ag}_3\text{PO}_4\text{-Cu-MOF}$ /GCE

Table S1. Comparison of this developed aptasensor with those in the literature for
DES detection.

Method	Linear range	LOD	Reference
ECL	$1 \times 10^{-8} \sim 1.2 \times 10^{-6}$ M	4.6×10^{-9} M	1
DPV	$2.0 \times 10^{-9} \sim 2 \times 10^{-7}$ M	9×10^{-11} M	2
CV	$2.0 \times 10^{-5} \sim 1.0 \times 10^{-7}$ M	1.5×10^{-8} M	3
Electrochemical immunosensor	$1.9 \times 10^{-11} \sim 1.9 \times 10^{-9}$ M	6.98×10^{-12} M	4
ECL(MMIPs-QDs-Aptamer)	$0.3 \sim 1.0 \times 10^5$ pg/mL	0.1 pg/mL	5
ECL(CdTe@ZnS/r-GO)	$1.8 \times 10^3 \sim 25.0$ nM	0.25 pM	6
ECL	$1.0 \times 10^{-12} \sim 1.0 \times 10^{-4}$	7.2×10^{-13}	This work

Table S2 DES determination result table in fish pond water

Sample	Added value/fM	Measured value/fM	Recovery rate/%	RSD/%
Fishpond water		50.00	100.00	
	50.00	52.00	102.00	4.16
		48.00	98.00	

References

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