Electronic Supplementary Material (ESI) for Analytical Methods. This journal is © The Royal Society of Chemistry 2021

## **Electronic Supplementary Information**

## Voltammetric Sensor Based on Carbon Black and Chitosan-Stabilized Gold Nanoparticles Nanocomposite for Ketoconazole Determination

Laís Pereira Silvaª, Tiago Almeida Silva<sup>b</sup>, Fernando Cruz Moraes<sup>a</sup> and Orlando

Fatibello-Filho<sup>a\*</sup>

<sup>a</sup>Department of Chemistry, Federal University of São Carlos, 13560-970, São Carlos, São Paulo, Brazil.

<sup>b</sup>Department of Chemistry, Federal University of Viçosa, 36570-900, Minas Gerais, Brazil.

\* Corresponding author.

Tel.: +55 16 33518098; Fax: +55 16 33518350

E-mail address: <u>bello@ufscar.br</u> (O. Fatibello-Filho)



Fig. S1. Cyclic voltammograms recorded at different potential scan rates  $(10 - 500 \text{ mV} \text{ s}^{-1})$  for GCE (a), CB–CTS/GCE (b) and CB–CTS–AuNPs/GCE (c) using  $1.0 \times 10^{-3}$  mol  $L^{-1}$  [Fe(CN)<sub>6</sub>]<sup>4–/3–</sup> in 0.1 mol  $L^{-1}$  KCl supporting electrolyte. (d) Plots of  $I_p$  vs.  $v^{1/2}$  for anodic peak currents ( $I_a$ ) for GCE, CB–CTS/GCE and CB–CTS–AuNPs/GCE.



**Fig. S2.** Cyclic voltammograms for  $1.0 \times 10^{-4}$  mol L<sup>-1</sup> KTO in 0.2 mol L<sup>-1</sup> phosphate buffer, pH 9.0 (\_\_\_\_\_), 0.04 mol L<sup>-1</sup> Britton-Robinson, pH 9.0 (\_\_\_\_\_) and 0.1 mol L<sup>-1</sup> sodium chloride, pH 9.0 adjusted with a 0.1 mol L<sup>-1</sup> NaOH solution (\_\_\_\_\_) supporting electrolyte solutions. v = 50 mV s<sup>-1</sup>.



**Fig. S3.** Optimization of KTO anodic response  $(I_p)$  by square-wave voltammetry using CB–CTS–AuNPs/GCE for different parameters: frequency (*f*), plot of  $I_p$  vs. *f* (a), amplitude (*a*), plot of  $I_p$  vs. *a* (b), and potential increment ( $\Delta E_s$ ), plot of  $I_p$  vs.  $\Delta E_s$  (c).