## Supporting Information

## A quinoline-benzothiazole based chemosensor coupled with smart phone for rapid detection of In<sup>3+</sup>

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Fig. 1S. FT-IR spectrum of L



Fig. 2S. <sup>1</sup>H NMR spectrum of L







Fig. 4S. HR-MS spectrum of L



Fig. 5S. ESI-MS spectrum of L+In<sup>3+</sup>



Fig. 6S. Bar diagrams depicting the effect of various interfering anions analytes



Fig. 7S. Stability Test of L performed by UV-vis spectral response for seven days

Structure of the probe	Method	LOD	Medium	Additional species detected	Application
[41]	Ratiometric fluorescent	2.68 µM	Near aqueous	-	Real sample analysis and live cell imaging
	fluorimetric	29 nM	DMF/H <sub>2</sub> O (9:1)	Zn <sup>2+</sup> & Ppi in ACN Fluorimetric	Real sample analysis
	fluorimetric	9.62 μM	aqueous	-	Real sample analysis
	fluorimetric	1.02 nM	DMF/H <sub>2</sub> O (9:1)	Zn <sup>2+</sup> (EtOH:H <sub>2</sub> O) Fluorimetric	Real sample analysis
	fluorimetric	5.89 µM	CH₃CN	Fe <sup>3+</sup> (colorimetric)	NA
	fluorimetric	0.05 μΜ	MeOH/H <sub>2</sub> O (1:1)	Zn <sup>2+</sup> , ClO <sup>-</sup> (fluorimetric) (colorimetric)	Real sample analysis
	fluorimetric	0.53 μΜ	DMF	Fe <sup>3+</sup> (colorimetric)	NA
	fluorimetric	0.04 µM	DMF/H <sub>2</sub> O (9:1)	Ppi (secondary sensor)	Real sample analysis
	fluorimetric	7.92 μM	МеОН	Ga <sup>3+</sup> , Al <sup>3+</sup> (fluorimetric)	NA
Present Work	colorimetric	0.2 μΜ	DMSO	Nil	Test strip

**Table 1S.** Comparison of present probe with previously reported probes for In<sup>3+</sup> ions recognition

Code	Abs (nm)	Mol. Abs (L m <sup>-1</sup> cm <sup>-1</sup> )	Wavelength max(nm)	Oscillator strength (f)	Transition	Orbital Contribution
L	360	13500	372.28	0.3957	$S_0 \rightarrow S_2$	H-1→L 86%, H→L 10%
L-In <sup>3+</sup>	450	9200	480.26	0.239	$S_0 \rightarrow S_3$	H-1→L 50%, H→L+1 48%

 Table 2S. Comparison of experimental and theoretical excitation spectral studies.

\*H-HOMO, L-LUMO orbitals

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