

Supporting Information for

Sterically Crowded 1,4-Diiodobenzene as Precursor to Difunctional Hypervalent Iodine Compounds

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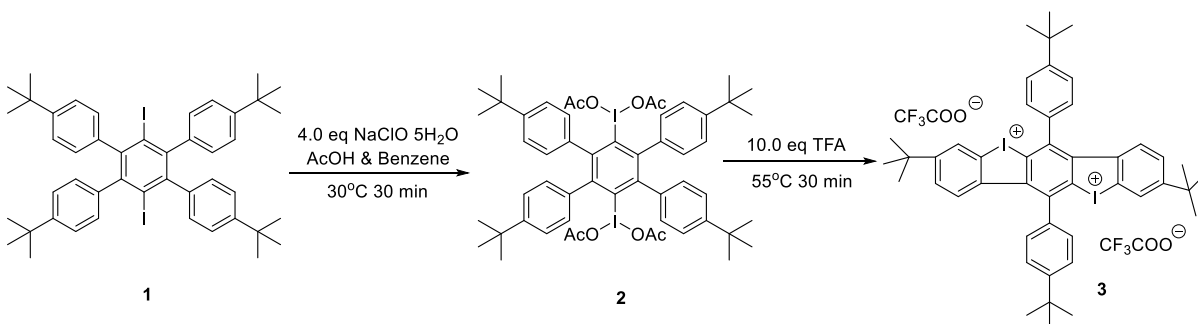
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Experiment Procedures

General Information. All starting materials, reagents, and solvents were purchased from commercial sources and used without further purification. ^1H NMR and ^{13}C $\{^1\text{H}\}$ NMR spectra were obtained on Bruker Avance III HD 500 MHz NMR Spectrometer. Chemical shifts (δ) are reported in parts per million (ppm). Melting points were determined with melting points apparatus and are uncorrected.

Method: As shown in **Scheme 1**, the unusual 1,4-diiodobenzene (**1**) bearing four *para*- t Bu- C_6H_4 groups was prepared according to our previous paper.¹ **1** was then oxidized into (diacetoxyiodo)iodoarenes (**2**) through sodium hyperchlorite pentahydrate.² **2** was expected to react with trifluoroacetic acid (TFA) to synthesize [bis(trifluoroacetoxy)iodoarenes], but it was cyclized to form iodonium salts to give compound **3**.



Scheme S1. Synthesis of hypervalent iodine with di-iodine(III) centers compounds **2** and **3**.

2,3,5,6-tetra-*tert*butylphenyl-1,4-(diacetoxyiodo)benzene (2): Sodium hyperchlorate pentahydrate (1.32 g, 4.0 mmol) and compound **1** (0.858 g, 1.0 mmol) were added into a 50 mL flask with a stir bar. Glacial acetic acid (10 mL) and benzene (10 mL) were added and stirred at 30°C for 30 min. Concentrated to remove the solvent and extracted with DCM. Dried over Na₂SO₄ and concentrated to obtain white solid. Crystals were obtained from slow evaporation of DCM. Yield: 0.81 g, 73.9%. Melting points: greater than 350 °C. ^1H NMR (500 MHz, Chloroform-*d*) δ = 7.12 (s, 2H), 7.01 (d, J = 8.0 Hz, 2H), 1.98 (s, 3H), 1.20 (s, 9H). ^{13}C $\{^1\text{H}\}$ NMR (126 MHz, Chloroform-*d*) δ = 176.30, 151.47, 146.84, 139.33, 137.05, 129.29, 124.24, 34.66, 31.28, 20.69.

2,6,7,10-tetra-*tert*butyl-bisdibenziodolium trifluoroacetate (3): Compound **2** (1.09 g, 1.0 mmol) and TFA (10.0 eq, 0.77 mL) were added into a 50 mL flask with a stir bar. The mixture was stirred at 50°C for 30 min and cooled down to room temperature. DI water (30 mL) was added and the mixture was extracted with DCM. Dried over Na₂SO₄ and concentrated to obtain the solid. Crystals were obtained from vapor diffusion (hexanes into a solution of **3** in DCM). Yield: 0.42 g, 38.8%. Melting points: greater than 350 °C. ^1H NMR (500 MHz, Chloroform-*d*) δ = 8.39 (s, 1H), 7.81 (d, J = 8.4 Hz, 2H), 7.43 (d, J = 8.3 Hz, 2H), 7.34 (d, J = 8.6 Hz, 1H), 6.74 (d, J = 8.6 Hz, 1H), 1.51 (s, 9H), 1.30 (s, 9H). ^{13}C $\{^1\text{H}\}$ NMR (126 MHz, Chloroform-*d*) δ = 162.24, 161.95, 161.66, 161.37, 156.85, 155.70, 141.35, 138.06, 137.53, 137.13, 135.53, 130.33, 128.76, 128.37, 127.46, 123.30, 117.16, 114.84, 36.03, 35.51, 31.42, 30.90. ^{19}F $\{^1\text{H}\}$ NMR (471 MHz, Chloroform-*d*) δ = -75.61.

NMR Spectra

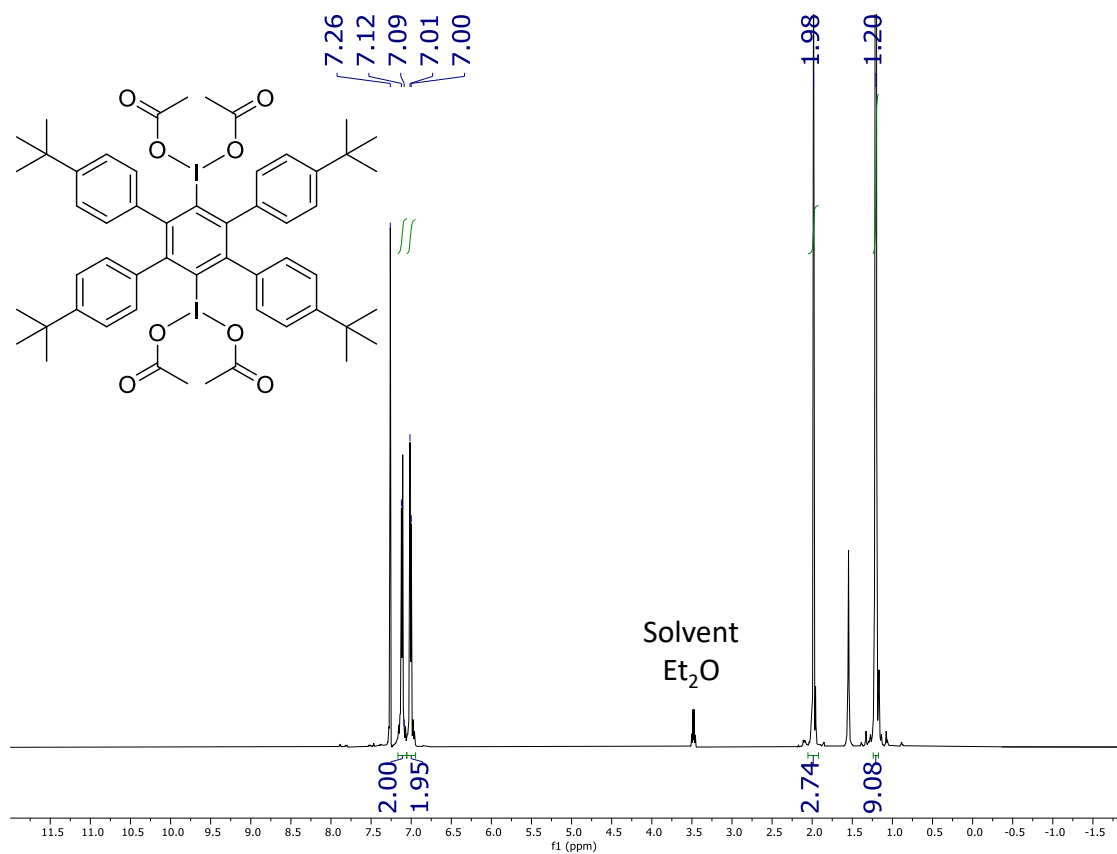


Figure S1. ¹H NMR spectrum of 2 (CDCl₃, 500 MHz, 298K)

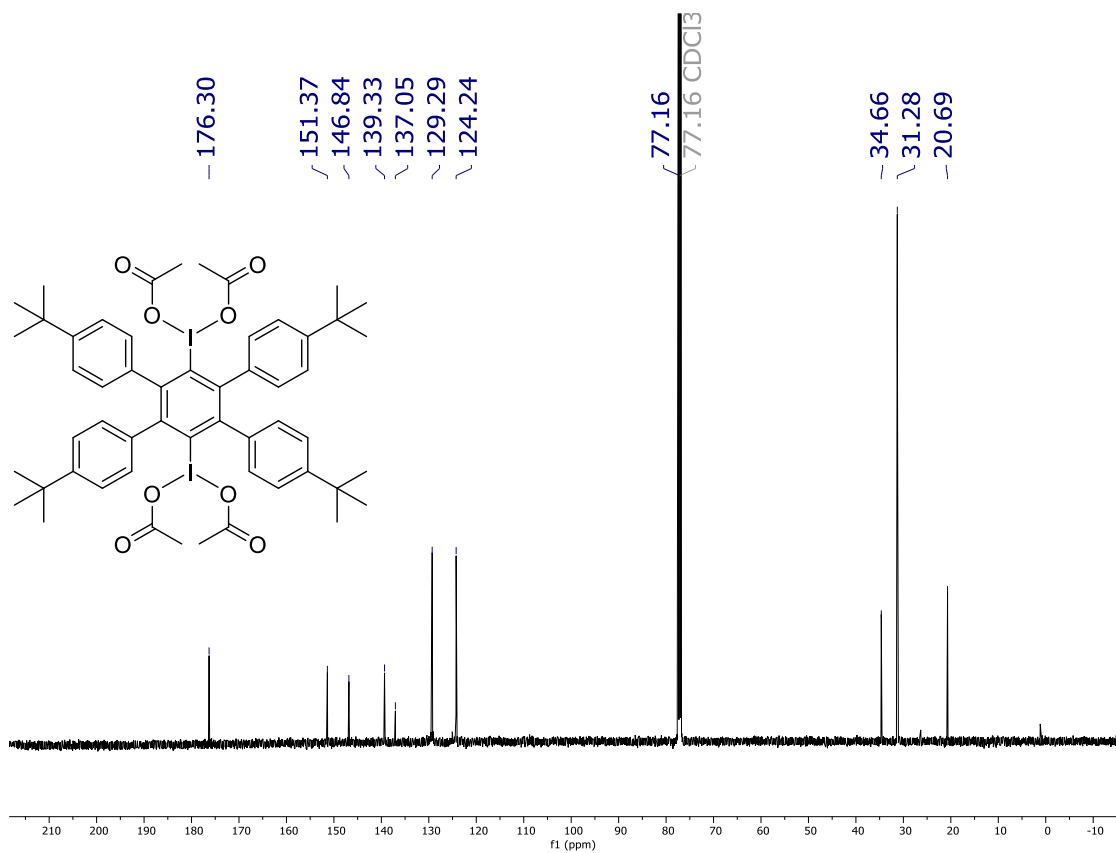


Figure S2. ^{13}C $\{^1\text{H}\}$ NMR spectrum of 2 (CDCl_3 , 126 MHz, 298K)

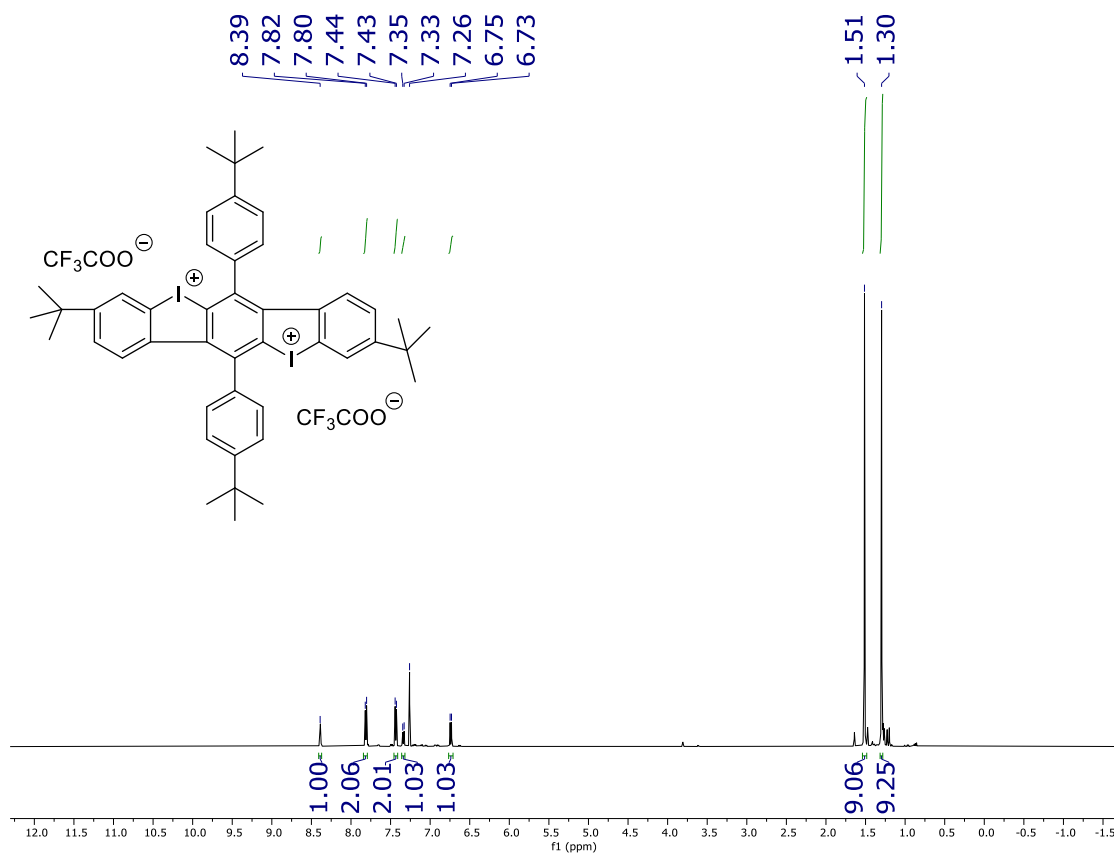


Figure S3. ^1H NMR spectrum of **3** (CDCl_3 , 500 MHz, 298K)

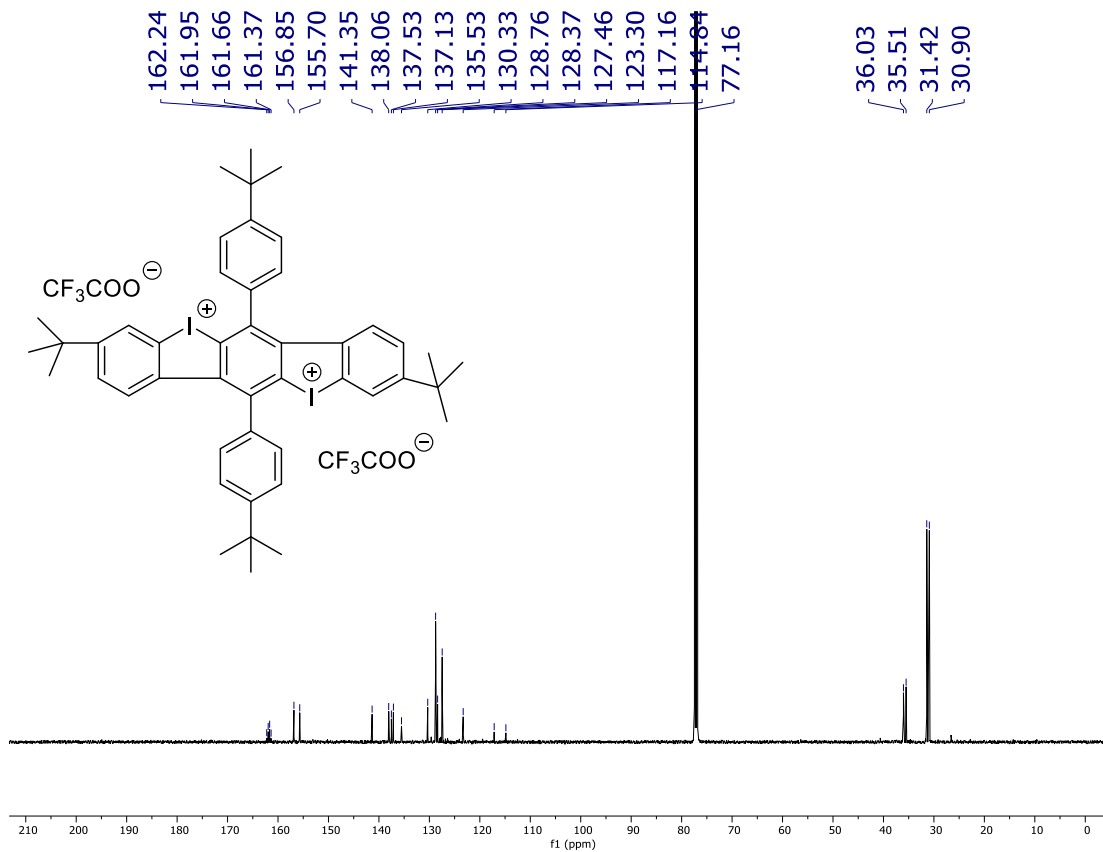


Figure S4. ^{13}C $\{^1\text{H}\}$ NMR spectrum of **3** (CDCl_3 , 126 MHz, 298K)

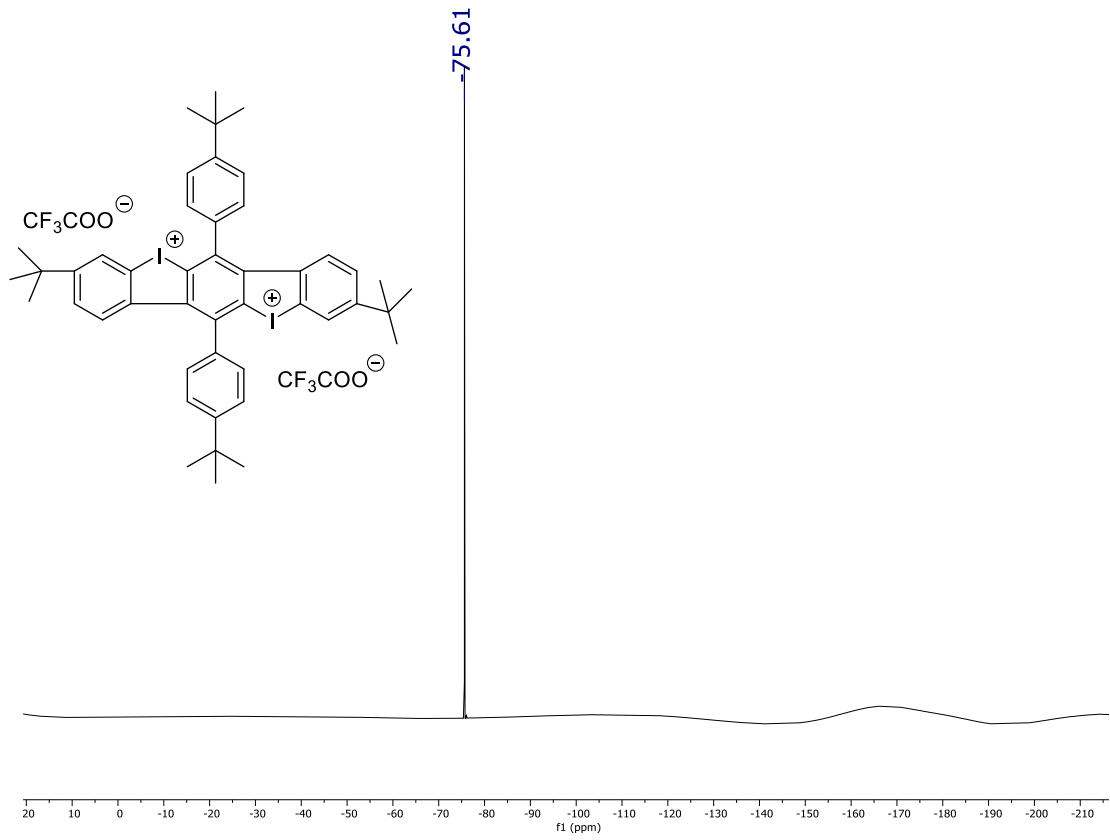


Figure S5. ^{19}F $\{^1\text{H}\}$ NMR spectrum of **3** (CDCl_3 , 471 MHz, 298K)

Crystallographic Analysis

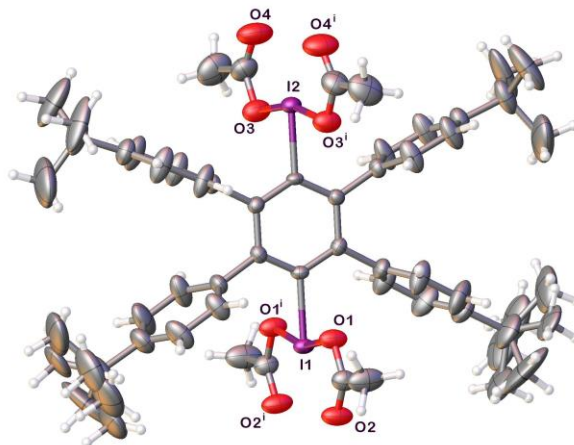


Figure S5. Structural diagram of compound **2**.

Table S1. Crystal data and structure refinement for compound **2_200K_sq**.

Identification code	compound 2	
Empirical formula	C ₅₄ H ₆₄ I ₂ O ₈	
Formula weight	1094.85	
Temperature	100.0 K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	Pnna	
Unit cell dimensions	a = 21.9600(4) Å	∠ = 90°.
	b = 15.2357(3) Å	∠ = 90°.
	c = 18.1317(4) Å	∠ = 90°.
Volume	6066.4(2) Å ³	
Z	4	
Density (calculated)	1.199 Mg/m ³	
Absorption coefficient	1.080 mm ⁻¹	
F(000)	2232	
Crystal size	0.31 x 0.13 x 0.08 mm ³	
Theta range for data collection	1.456 to 26.370°.	
Index ranges	-27<=h<=27, -19<=k<=11, -22<=l<=22	
Reflections collected	38417	
Independent reflections	6222 [R(int) = 0.0618]	
Completeness to theta = 25.242°	100.0 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.5626 and 0.5095	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	6222 / 36 / 330	
Goodness-of-fit on F ²	1.005	
Final R indices [I>2sigma(I)]	R1 = 0.0383, wR2 = 0.0908	
R indices (all data)	R1 = 0.0760, wR2 = 0.1184	
Extinction coefficient	n/a	
Largest diff. peak and hole	1.423 and -0.757 e.Å ⁻³	
SQUEEZE	Found 78e/uc (likely two DCM)	

Table S2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for compound **2_200K_sq**. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)
I(1)	1844(1)	2500	2500	29(1)
I(2)	4999(1)	2500	2500	34(1)
O(1)	2015(1)	3850(2)	2779(2)	46(1)
O(3)	4851(2)	1178(2)	2133(2)	55(1)
O(2)	1011(2)	3878(2)	2773(2)	55(1)
C(4)	4034(2)	2500	2500	23(1)
C(1)	2808(2)	2500	2500	24(1)
C(3)	3742(2)	2344(3)	3166(2)	26(1)
C(2)	3096(2)	2339(3)	3164(2)	28(1)
C(19)	4070(2)	2175(3)	3874(2)	32(1)
O(4)	5856(2)	1157(3)	2173(3)	82(1)
C(9)	2757(2)	2149(3)	3856(2)	35(1)
C(10)	2531(2)	1329(3)	4010(3)	52(1)
C(22)	4592(2)	1873(3)	5270(2)	48(1)
C(5)	1496(2)	4254(3)	2861(3)	42(1)
C(20)	4180(3)	1346(3)	4124(3)	57(2)
C(12)	2119(3)	1799(4)	5173(3)	59(2)
C(11)	2213(3)	1166(4)	4664(3)	64(2)
C(7)	5378(3)	797(4)	2027(3)	59(2)
C(24)	4242(3)	2856(4)	4329(3)	60(2)
C(23)	4502(3)	2696(3)	5006(3)	65(2)
C(21)	4439(3)	1195(3)	4806(3)	66(2)
C(25)	4840(3)	1708(4)	6048(3)	69(2)
C(14)	2654(3)	2799(4)	4368(3)	64(2)
C(6)	1553(3)	5198(3)	3076(4)	72(2)
C(13)	2341(3)	2614(4)	5015(3)	77(2)
C(15)	1779(4)	1622(5)	5900(4)	91(2)
C(27)	5264(5)	2392(5)	6298(5)	134(4)
C(8)	5333(3)	-95(4)	1678(4)	90(2)
C(26)	5079(5)	815(5)	6153(4)	134(4)
C(16)	1506(10)	732(12)	5923(10)	131(7)
C(28)	4273(5)	1798(8)	6589(4)	165(5)
C(17)	1303(10)	2374(14)	6009(13)	191(12)
C(18)	2295(10)	1694(16)	6537(6)	152(8)
C(17A)	1895(15)	2150(20)	6488(14)	98(9)
C(18A)	1870(20)	710(30)	6180(20)	129(12)
C(16A)	1094(12)	1570(30)	5743(13)	116(12)

Table S3. Bond lengths [Å] and angles [°] for compound **2_200K_sq**.

I(1)-O(1)#1	2.152(3)
I(1)-O(1)	2.152(3)
I(1)-C(1)	2.115(5)
I(2)-O(3)#1	2.146(3)
I(2)-O(3)	2.146(3)
I(2)-C(4)	2.119(4)
O(1)-C(5)	1.304(5)
O(3)-C(7)	1.307(6)
O(2)-C(5)	1.219(5)
C(4)-C(3)	1.388(4)
C(4)-C(3)#1	1.388(4)
C(1)-C(2)#1	1.383(4)
C(1)-C(2)	1.383(4)
C(3)-C(2)	1.417(5)
C(3)-C(19)	1.494(5)
C(2)-C(9)	1.487(5)
C(19)-C(20)	1.365(6)
C(19)-C(24)	1.378(6)
O(4)-C(7)	1.214(7)
C(9)-C(10)	1.374(6)
C(9)-C(14)	1.377(7)
C(10)-H(10)	0.9500
C(10)-C(11)	1.398(6)
C(22)-C(23)	1.357(7)
C(22)-C(21)	1.373(7)
C(22)-C(25)	1.533(6)
C(5)-C(6)	1.496(7)
C(20)-H(20)	0.9500
C(20)-C(21)	1.381(6)
C(12)-C(11)	1.352(7)
C(12)-C(13)	1.364(8)
C(12)-C(15)	1.538(7)
C(11)-H(11)	0.9500
C(7)-C(8)	1.503(8)
C(24)-H(24)	0.9500
C(24)-C(23)	1.375(7)
C(23)-H(23)	0.9500
C(21)-H(21)	0.9500
C(25)-C(27)	1.469(9)
C(25)-C(26)	1.472(8)
C(25)-C(28)	1.592(11)
C(14)-H(14)	0.9500
C(14)-C(13)	1.388(7)
C(6)-H(6A)	0.9800
C(6)-H(6B)	0.9800
C(6)-H(6C)	0.9800
C(13)-H(13)	0.9500
C(15)-C(16)	1.485(17)
C(15)-C(17)	1.564(17)
C(15)-C(18)	1.621(18)
C(15)-C(17A)	1.36(3)

C(15)-C(18A)	1.50(3)
C(15)-C(16A)	1.53(3)
C(27)-H(27A)	0.9800
C(27)-H(27B)	0.9800
C(27)-H(27C)	0.9800
C(8)-H(8A)	0.9800
C(8)-H(8B)	0.9800
C(8)-H(8C)	0.9800
C(26)-H(26A)	0.9800
C(26)-H(26B)	0.9800
C(26)-H(26C)	0.9800
C(16)-H(16A)	0.9800
C(16)-H(16B)	0.9800
C(16)-H(16C)	0.9800
C(28)-H(28A)	0.9800
C(28)-H(28B)	0.9800
C(28)-H(28C)	0.9800
C(17)-H(17A)	0.9800
C(17)-H(17B)	0.9800
C(17)-H(17C)	0.9800
C(18)-H(18A)	0.9800
C(18)-H(18B)	0.9800
C(18)-H(18C)	0.9800
C(17A)-H(17D)	0.9800
C(17A)-H(17E)	0.9800
C(17A)-H(17F)	0.9800
C(18A)-H(18D)	0.9800
C(18A)-H(18E)	0.9800
C(18A)-H(18F)	0.9800
C(16A)-H(16D)	0.9800
C(16A)-H(16E)	0.9800
C(16A)-H(16F)	0.9800
O(1)-I(1)-O(1)#1	159.89(17)
C(1)-I(1)-O(1)#1	79.95(8)
C(1)-I(1)-O(1)	79.95(8)
O(3)-I(2)-O(3)#1	162.65(18)
C(4)-I(2)-O(3)#1	81.32(9)
C(4)-I(2)-O(3)	81.32(9)
C(5)-O(1)-I(1)	109.0(3)
C(7)-O(3)-I(2)	109.2(3)
C(3)#1-C(4)-I(2)	117.5(2)
C(3)-C(4)-I(2)	117.5(2)
C(3)-C(4)-C(3)#1	125.0(5)
C(2)-C(1)-I(1)	117.3(2)
C(2)#1-C(1)-I(1)	117.3(2)
C(2)#1-C(1)-C(2)	125.4(5)
C(4)-C(3)-C(2)	117.5(3)
C(4)-C(3)-C(19)	123.7(4)
C(2)-C(3)-C(19)	118.9(3)
C(1)-C(2)-C(3)	117.3(4)
C(1)-C(2)-C(9)	122.6(4)
C(3)-C(2)-C(9)	120.0(4)

C(20)-C(19)-C(3)	122.0(4)
C(20)-C(19)-C(24)	116.7(4)
C(24)-C(19)-C(3)	121.2(4)
C(10)-C(9)-C(2)	122.0(4)
C(10)-C(9)-C(14)	117.2(4)
C(14)-C(9)-C(2)	120.8(4)
C(9)-C(10)-H(10)	119.5
C(9)-C(10)-C(11)	121.0(5)
C(11)-C(10)-H(10)	119.5
C(23)-C(22)-C(21)	116.4(4)
C(23)-C(22)-C(25)	121.8(4)
C(21)-C(22)-C(25)	121.8(5)
O(1)-C(5)-C(6)	114.2(4)
O(2)-C(5)-O(1)	121.9(4)
O(2)-C(5)-C(6)	123.9(4)
C(19)-C(20)-H(20)	119.1
C(19)-C(20)-C(21)	121.7(4)
C(21)-C(20)-H(20)	119.1
C(11)-C(12)-C(13)	116.9(5)
C(11)-C(12)-C(15)	122.3(6)
C(13)-C(12)-C(15)	120.8(5)
C(10)-C(11)-H(11)	119.0
C(12)-C(11)-C(10)	121.9(5)
C(12)-C(11)-H(11)	119.0
O(3)-C(7)-C(8)	113.9(5)
O(4)-C(7)-O(3)	122.1(5)
O(4)-C(7)-C(8)	123.9(5)
C(19)-C(24)-H(24)	119.5
C(23)-C(24)-C(19)	121.0(5)
C(23)-C(24)-H(24)	119.5
C(22)-C(23)-C(24)	122.6(5)
C(22)-C(23)-H(23)	118.7
C(24)-C(23)-H(23)	118.7
C(22)-C(21)-C(20)	121.5(5)
C(22)-C(21)-H(21)	119.2
C(20)-C(21)-H(21)	119.2
C(22)-C(25)-C(28)	105.9(5)
C(27)-C(25)-C(22)	113.2(5)
C(27)-C(25)-C(26)	113.0(6)
C(27)-C(25)-C(28)	104.2(7)
C(26)-C(25)-C(22)	113.5(5)
C(26)-C(25)-C(28)	106.2(7)
C(9)-C(14)-H(14)	119.8
C(9)-C(14)-C(13)	120.4(5)
C(13)-C(14)-H(14)	119.8
C(5)-C(6)-H(6A)	109.5
C(5)-C(6)-H(6B)	109.5
C(5)-C(6)-H(6C)	109.5
H(6A)-C(6)-H(6B)	109.5
H(6A)-C(6)-H(6C)	109.5
H(6B)-C(6)-H(6C)	109.5
C(12)-C(13)-C(14)	122.6(5)
C(12)-C(13)-H(13)	118.7

C(14)-C(13)-H(13)	118.7
C(12)-C(15)-C(17)	107.8(8)
C(12)-C(15)-C(18)	105.1(7)
C(16)-C(15)-C(12)	112.4(9)
C(16)-C(15)-C(17)	113.3(12)
C(16)-C(15)-C(18)	108.9(11)
C(17)-C(15)-C(18)	109.1(13)
C(17A)-C(15)-C(12)	118.4(12)
C(17A)-C(15)-C(18A)	105(2)
C(17A)-C(15)-C(16A)	111.1(16)
C(18A)-C(15)-C(12)	112.9(17)
C(18A)-C(15)-C(16A)	98(2)
C(16A)-C(15)-C(12)	109.1(10)
C(25)-C(27)-H(27A)	109.5
C(25)-C(27)-H(27B)	109.5
C(25)-C(27)-H(27C)	109.5
H(27A)-C(27)-H(27B)	109.5
H(27A)-C(27)-H(27C)	109.5
H(27B)-C(27)-H(27C)	109.5
C(7)-C(8)-H(8A)	109.5
C(7)-C(8)-H(8B)	109.5
C(7)-C(8)-H(8C)	109.5
H(8A)-C(8)-H(8B)	109.5
H(8A)-C(8)-H(8C)	109.5
H(8B)-C(8)-H(8C)	109.5
C(25)-C(26)-H(26A)	109.5
C(25)-C(26)-H(26B)	109.5
C(25)-C(26)-H(26C)	109.5
H(26A)-C(26)-H(26B)	109.5
H(26A)-C(26)-H(26C)	109.5
H(26B)-C(26)-H(26C)	109.5
C(15)-C(16)-H(16A)	109.5
C(15)-C(16)-H(16B)	109.5
C(15)-C(16)-H(16C)	109.5
H(16A)-C(16)-H(16B)	109.5
H(16A)-C(16)-H(16C)	109.5
H(16B)-C(16)-H(16C)	109.5
C(25)-C(28)-H(28A)	109.5
C(25)-C(28)-H(28B)	109.5
C(25)-C(28)-H(28C)	109.5
H(28A)-C(28)-H(28B)	109.5
H(28A)-C(28)-H(28C)	109.5
H(28B)-C(28)-H(28C)	109.5
C(15)-C(17)-H(17A)	109.5
C(15)-C(17)-H(17B)	109.5
C(15)-C(17)-H(17C)	109.5
H(17A)-C(17)-H(17B)	109.5
H(17A)-C(17)-H(17C)	109.5
H(17B)-C(17)-H(17C)	109.5
C(15)-C(18)-H(18A)	109.5
C(15)-C(18)-H(18B)	109.5
C(15)-C(18)-H(18C)	109.5
H(18A)-C(18)-H(18B)	109.5

H(18A)-C(18)-H(18C)	109.5
H(18B)-C(18)-H(18C)	109.5
C(15)-C(17A)-H(17D)	109.5
C(15)-C(17A)-H(17E)	109.5
C(15)-C(17A)-H(17F)	109.5
H(17D)-C(17A)-H(17E)	109.5
H(17D)-C(17A)-H(17F)	109.5
H(17E)-C(17A)-H(17F)	109.5
C(15)-C(18A)-H(18D)	109.5
C(15)-C(18A)-H(18E)	109.5
C(15)-C(18A)-H(18F)	109.5
H(18D)-C(18A)-H(18E)	109.5
H(18D)-C(18A)-H(18F)	109.5
H(18E)-C(18A)-H(18F)	109.5
C(15)-C(16A)-H(16D)	109.5
C(15)-C(16A)-H(16E)	109.5
C(15)-C(16A)-H(16F)	109.5
H(16D)-C(16A)-H(16E)	109.5
H(16D)-C(16A)-H(16F)	109.5
H(16E)-C(16A)-H(16F)	109.5

Symmetry transformations used to generate equivalent atoms:

#1 x,-y+1/2,-z+1/2

Table S4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for compound **2_200K_sq**. The anisotropic displacement factor exponent takes the form: $-2\sigma^2 [h^2 a^*2U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
I(1)	20(1)	27(1)	40(1)	-3(1)	0	0
I(2)	23(1)	42(1)	39(1)	0(1)	0	0
O(1)	33(2)	31(2)	74(2)	-10(2)	-2(2)	-1(1)
O(3)	43(2)	46(2)	76(2)	-12(2)	4(2)	-2(2)
O(2)	33(2)	40(2)	92(3)	-10(2)	-5(2)	1(2)
C(4)	14(2)	25(3)	29(2)	-2(3)	0	0
C(1)	17(2)	24(3)	30(2)	2(3)	0	0
C(3)	25(2)	27(3)	27(2)	-1(2)	-2(2)	2(2)
C(2)	29(2)	28(3)	26(2)	-1(2)	4(2)	0(2)
C(19)	34(2)	34(2)	26(2)	3(2)	-2(2)	0(2)
O(4)	43(2)	78(3)	125(4)	-18(3)	2(2)	7(2)
C(9)	35(2)	42(3)	28(2)	2(2)	7(2)	-2(2)
C(10)	68(4)	35(3)	51(3)	3(2)	24(3)	3(3)
C(22)	60(3)	48(3)	36(2)	-2(2)	-14(2)	3(3)
C(5)	38(3)	33(3)	55(3)	-2(2)	-2(2)	2(2)
C(20)	93(4)	34(3)	44(3)	-10(2)	-32(3)	8(3)
C(12)	59(3)	68(4)	50(3)	3(3)	23(3)	-5(3)
C(11)	78(4)	47(3)	66(4)	12(3)	34(3)	-4(3)
C(7)	54(4)	55(4)	68(4)	-3(3)	6(3)	13(3)
C(24)	97(5)	33(3)	49(3)	2(2)	-28(3)	-4(3)
C(23)	101(5)	43(4)	50(3)	-9(2)	-36(3)	1(3)
C(21)	111(5)	33(3)	55(3)	-4(3)	-39(3)	7(3)
C(25)	109(5)	53(4)	45(3)	-8(3)	-36(3)	4(4)
C(14)	98(5)	46(3)	46(3)	-11(2)	29(3)	-19(3)
C(6)	50(3)	38(3)	129(6)	-19(3)	4(4)	4(3)
C(13)	99(5)	74(5)	58(3)	-28(3)	39(3)	-16(4)
C(15)	99(5)	109(5)	64(4)	4(4)	49(3)	-12(4)
C(27)	202(10)	97(6)	103(6)	11(5)	-110(7)	-17(6)
C(8)	94(6)	58(4)	119(6)	-20(4)	22(5)	7(4)
C(26)	244(11)	89(6)	70(5)	-16(4)	-97(6)	64(6)
C(16)	169(19)	138(8)	86(13)	3(9)	67(11)	-60(10)
C(28)	216(12)	238(13)	40(4)	0(6)	-7(6)	55(10)
C(17)	186(17)	175(13)	210(20)	43(16)	172(16)	55(14)
C(18)	181(13)	240(20)	41(5)	-1(9)	34(6)	-65(12)
C(17A)	96(18)	140(16)	57(9)	-3(10)	55(10)	-24(14)
C(18A)	190(30)	126(9)	70(20)	27(11)	60(20)	8(14)
C(16A)	100(7)	190(30)	60(12)	9(16)	46(8)	-31(12)

Table S5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for compound **2_200K_sq**.

	x	y	z	U(eq)
H(10)	2591	865	3667	62
H(20)	4076	860	3821	68
H(11)	2058	593	4752	76
H(24)	4180	3444	4173	72
H(23)	4624	3181	5300	78
H(21)	4514	608	4959	80
H(14)	2797	3378	4279	76
H(6A)	1604	5243	3611	108
H(6B)	1185	5515	2927	108
H(6C)	1908	5456	2830	108
H(13)	2279	3073	5362	92
H(27A)	5634	2376	5995	201
H(27B)	5372	2288	6815	201
H(27C)	5070	2969	6250	201
H(8A)	5002	-427	1911	135
H(8B)	5718	-410	1745	135
H(8C)	5249	-31	1150	135
H(26A)	4768	386	6009	201
H(26B)	5185	729	6673	201
H(26C)	5443	734	5848	201
H(16A)	1221	663	5511	196
H(16B)	1287	654	6390	196
H(16C)	1828	290	5884	196
H(28A)	4088	2380	6529	247
H(28B)	4409	1726	7100	247
H(28C)	3972	1345	6471	247
H(17A)	1516	2922	6129	286
H(17B)	1027	2220	6413	286
H(17C)	1069	2453	5554	286
H(18A)	2610	1251	6450	228
H(18B)	2107	1595	7020	228
H(18C)	2478	2280	6525	228
H(17D)	2330	2130	6607	146
H(17E)	1658	1953	6914	146
H(17F)	1781	2758	6366	146
H(18D)	1760	288	5789	193
H(18E)	1605	609	6608	193
H(18F)	2294	623	6319	193
H(16D)	949	2138	5562	173
H(16E)	877	1417	6198	173
H(16F)	1017	1117	5370	173

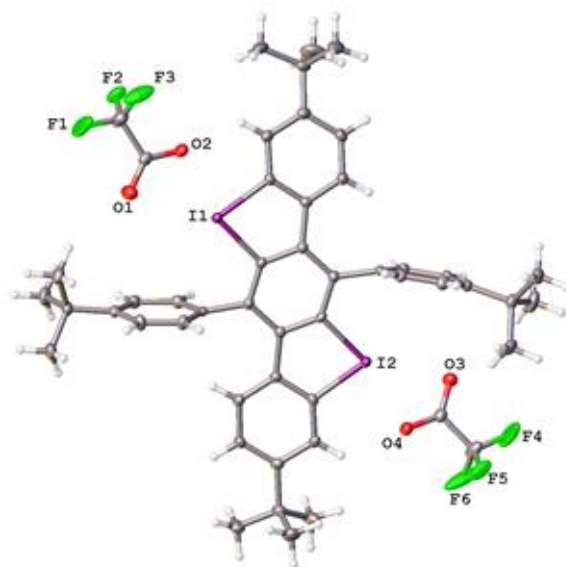


Figure S6. Structural diagram of compound **3**.

Table S6. Crystal data and structure refinement for compound **3_0m_a**.

Identification code	compound 3	
Empirical formula	C ₅₀ H ₅₀ F ₆ I ₂ O ₄	
Formula weight	1082.70	
Temperature	100.0 K	
Wavelength	0.71073 Å	
Crystal system	Triclinic	
Space group	P-1	
Unit cell dimensions	a = 12.7903(7) Å	α = 67.9580(10)°.
	b = 14.1970(6) Å	β = 74.535(2)°.
	c = 14.4995(8) Å	γ = 80.5310(10)°.
Volume	2345.9(2) Å ³	
Z	2	
Density (calculated)	1.533 Mg/m ³	
Absorption coefficient	1.407 mm ⁻¹	
F(000)	1084	
Crystal size	0.26 x 0.12 x 0.08 mm ³	
Theta range for data collection	2.186 to 26.392°.	
Index ranges	-15 ≤ h ≤ 15, -17 ≤ k ≤ 17, -18 ≤ l ≤ 18	
Reflections collected	40984	
Independent reflections	9576 [R(int) = 0.0308]	
Completeness to theta = 25.242°	99.7 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.7454 and 0.6671	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	9576 / 0 / 571	
Goodness-of-fit on F ²	1.031	
Final R indices [I > 2σ(I)]	R1 = 0.0233, wR2 = 0.0600	
R indices (all data)	R1 = 0.0284, wR2 = 0.0634	
Extinction coefficient	n/a	
Largest diff. peak and hole	1.065 and -0.787 e.Å ⁻³	

TableS7. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for compound **3_0m_a**. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)
I(1)	4758(1)	5668(1)	3527(1)	16(1)
I(2)	10004(1)	4265(1)	1581(1)	17(1)
F(2)	880(1)	7246(1)	4277(1)	34(1)
F(5)	14010(1)	3134(1)	861(1)	42(1)
O(4)	11794(1)	3307(1)	952(1)	28(1)
O(2)	3072(1)	6679(1)	4154(1)	28(1)
F(1)	520(1)	6035(2)	5719(2)	73(1)
O(3)	12145(1)	4916(1)	-58(1)	29(1)
F(3)	1381(2)	7267(2)	5552(2)	62(1)
O(1)	2451(2)	5146(1)	5132(2)	38(1)
C(4)	8401(2)	4715(2)	2179(2)	16(1)
C(1)	6350(2)	5207(2)	2903(2)	16(1)
C(6)	6595(2)	4215(2)	2898(2)	17(1)
C(3)	8163(2)	5699(2)	2194(2)	17(1)
C(37)	3994(2)	3152(2)	3078(2)	20(1)
F(6)	13726(2)	2984(2)	-436(2)	85(1)
C(27)	9293(2)	2895(2)	2000(2)	17(1)
C(35)	4384(2)	2519(2)	4733(2)	23(1)
C(5)	7681(2)	3959(2)	2493(2)	16(1)
C(7)	5450(2)	7078(2)	2934(2)	17(1)
C(2)	7071(2)	5973(2)	2550(2)	16(1)
C(36)	3668(2)	2608(2)	4119(2)	19(1)
C(20)	11283(2)	6894(2)	1329(2)	19(1)
C(33)	5655(2)	3554(2)	3301(2)	16(1)
C(34)	5356(2)	2983(2)	4341(2)	21(1)
C(17)	9131(2)	6313(2)	1883(2)	16(1)
C(19)	10701(2)	6577(2)	2350(2)	22(1)
C(38)	4971(2)	3621(2)	2673(2)	20(1)
C(22)	9665(2)	6706(2)	855(2)	19(1)
C(32)	8165(2)	2977(2)	2382(2)	16(1)
C(47)	2369(2)	6070(2)	4770(2)	22(1)

C(21)	10726(2)	6990(2)	587(2)	20(1)
C(12)	4834(2)	7984(2)	2894(2)	19(1)
C(49)	12392(2)	4009(2)	333(2)	20(1)
C(30)	8207(2)	1236(2)	2458(2)	20(1)
C(18)	9648(2)	6288(2)	2628(2)	20(1)
F(4)	14206(2)	4373(2)	-542(2)	122(1)
C(23)	12509(2)	7009(2)	1072(2)	23(1)
C(8)	6568(2)	6988(2)	2541(2)	17(1)
C(48)	1272(2)	6640(2)	5083(2)	30(1)
C(46)	1915(2)	2663(2)	5370(2)	30(1)
C(28)	9883(2)	2034(2)	1860(2)	18(1)
C(31)	7633(2)	2104(2)	2612(2)	19(1)
C(29)	9330(2)	1176(2)	2080(2)	19(1)
C(10)	6482(2)	8821(2)	2043(2)	24(1)
C(43)	2564(2)	2151(2)	4599(2)	22(1)
C(9)	7081(2)	7903(2)	2097(2)	21(1)
C(11)	5356(2)	8888(2)	2425(2)	22(1)
C(50)	13591(2)	3629(2)	55(2)	32(1)
C(39)	9982(2)	217(2)	1921(2)	23(1)
C(45)	1895(2)	2329(2)	3806(2)	31(1)
C(24)	12762(2)	7740(2)	1528(2)	29(1)
C(42)	9246(2)	-526(2)	1913(2)	33(1)
C(44)	2741(2)	999(2)	5157(2)	32(1)
C(26)	12989(2)	7375(2)	-74(2)	37(1)
C(15)	5218(2)	10561(2)	2687(3)	42(1)
C(25)	13054(2)	5940(2)	1559(2)	31(1)
C(13)	4719(2)	9926(2)	2285(2)	29(1)
C(41)	10812(2)	510(2)	892(2)	35(1)
C(40)	10570(2)	-311(2)	2794(2)	33(1)
C(16)	3525(2)	9823(2)	2831(3)	48(1)
C(14)	4781(3)	10481(2)	1130(3)	55(1)

Table S8. Bond lengths [Å] and angles [°] for compound **3_0m_a**.

I(1)-C(1)	2.101(2)
I(1)-C(7)	2.101(2)
I(2)-C(4)	2.098(2)
I(2)-C(27)	2.093(2)
F(2)-C(48)	1.332(3)
F(5)-C(50)	1.319(3)
O(4)-C(49)	1.264(3)
O(2)-C(47)	1.266(3)
F(1)-C(48)	1.310(3)
O(3)-C(49)	1.222(3)
F(3)-C(48)	1.349(3)
O(1)-C(47)	1.214(3)
C(4)-C(3)	1.388(3)
C(4)-C(5)	1.394(3)
C(1)-C(6)	1.393(3)
C(1)-C(2)	1.399(3)
C(6)-C(5)	1.402(3)
C(6)-C(33)	1.501(3)
C(3)-C(2)	1.399(3)
C(3)-C(17)	1.502(3)
C(37)-H(37)	0.9500
C(37)-C(36)	1.395(3)
C(37)-C(38)	1.392(3)
F(6)-C(50)	1.322(3)
C(27)-C(32)	1.403(3)
C(27)-C(28)	1.381(3)
C(35)-H(35)	0.9500
C(35)-C(36)	1.400(3)
C(35)-C(34)	1.383(3)
C(5)-C(32)	1.477(3)
C(7)-C(12)	1.383(3)
C(7)-C(8)	1.394(3)
C(2)-C(8)	1.475(3)
C(36)-C(43)	1.529(3)

C(20)-C(19)	1.405(3)
C(20)-C(21)	1.396(3)
C(20)-C(23)	1.533(3)
C(33)-C(34)	1.397(3)
C(33)-C(38)	1.392(3)
C(34)-H(34)	0.9500
C(17)-C(22)	1.399(3)
C(17)-C(18)	1.397(3)
C(19)-H(19)	0.9500
C(19)-C(18)	1.383(3)
C(38)-H(38)	0.9500
C(22)-H(22)	0.9500
C(22)-C(21)	1.391(3)
C(32)-C(31)	1.398(3)
C(47)-C(48)	1.547(3)
C(21)-H(21)	0.9500
C(12)-H(12)	0.9500
C(12)-C(11)	1.393(3)
C(49)-C(50)	1.537(3)
C(30)-H(30)	0.9500
C(30)-C(31)	1.389(3)
C(30)-C(29)	1.395(3)
C(18)-H(18)	0.9500
F(4)-C(50)	1.308(3)
C(23)-C(24)	1.534(3)
C(23)-C(26)	1.524(3)
C(23)-C(25)	1.549(3)
C(8)-C(9)	1.400(3)
C(46)-H(46A)	0.9800
C(46)-H(46B)	0.9800
C(46)-H(46C)	0.9800
C(46)-C(43)	1.543(3)
C(28)-H(28)	0.9500
C(28)-C(29)	1.399(3)
C(31)-H(31)	0.9500
C(29)-C(39)	1.537(3)

C(10)-H(10)	0.9500
C(10)-C(9)	1.385(3)
C(10)-C(11)	1.400(3)
C(43)-C(45)	1.534(3)
C(43)-C(44)	1.537(3)
C(9)-H(9)	0.9500
C(11)-C(13)	1.531(3)
C(39)-C(42)	1.531(3)
C(39)-C(41)	1.536(3)
C(39)-C(40)	1.529(3)
C(45)-H(45A)	0.9800
C(45)-H(45B)	0.9800
C(45)-H(45C)	0.9800
C(24)-H(24A)	0.9800
C(24)-H(24B)	0.9800
C(24)-H(24C)	0.9800
C(42)-H(42A)	0.9800
C(42)-H(42B)	0.9800
C(42)-H(42C)	0.9800
C(44)-H(44A)	0.9800
C(44)-H(44B)	0.9800
C(44)-H(44C)	0.9800
C(26)-H(26A)	0.9800
C(26)-H(26B)	0.9800
C(26)-H(26C)	0.9800
C(15)-H(15A)	0.9800
C(15)-H(15B)	0.9800
C(15)-H(15C)	0.9800
C(15)-C(13)	1.530(3)
C(25)-H(25A)	0.9800
C(25)-H(25B)	0.9800
C(25)-H(25C)	0.9800
C(13)-C(16)	1.524(4)
C(13)-C(14)	1.545(4)
C(41)-H(41A)	0.9800
C(41)-H(41B)	0.9800

C(41)-H(41C)	0.9800
C(40)-H(40A)	0.9800
C(40)-H(40B)	0.9800
C(40)-H(40C)	0.9800
C(16)-H(16A)	0.9800
C(16)-H(16B)	0.9800
C(16)-H(16C)	0.9800
C(14)-H(14A)	0.9800
C(14)-H(14B)	0.9800
C(14)-H(14C)	0.9800
C(1)-I(1)-C(7)	80.94(8)
C(27)-I(2)-C(4)	81.08(8)
C(3)-C(4)-I(2)	118.96(15)
C(3)-C(4)-C(5)	127.62(19)
C(5)-C(4)-I(2)	113.39(15)
C(6)-C(1)-I(1)	119.93(15)
C(6)-C(1)-C(2)	127.04(19)
C(2)-C(1)-I(1)	113.01(15)
C(1)-C(6)-C(5)	116.48(19)
C(1)-C(6)-C(33)	115.96(18)
C(5)-C(6)-C(33)	127.53(19)
C(4)-C(3)-C(2)	116.39(19)
C(4)-C(3)-C(17)	115.16(18)
C(2)-C(3)-C(17)	128.34(19)
C(36)-C(37)-H(37)	119.3
C(38)-C(37)-H(37)	119.3
C(38)-C(37)-C(36)	121.4(2)
C(32)-C(27)-I(2)	112.43(15)
C(28)-C(27)-I(2)	122.45(16)
C(28)-C(27)-C(32)	125.05(19)
C(36)-C(35)-H(35)	119.0
C(34)-C(35)-H(35)	119.0
C(34)-C(35)-C(36)	121.9(2)
C(4)-C(5)-C(6)	116.05(19)
C(4)-C(5)-C(32)	115.78(18)

C(6)-C(5)-C(32)	128.14(19)
C(12)-C(7)-I(1)	122.55(15)
C(12)-C(7)-C(8)	125.23(19)
C(8)-C(7)-I(1)	112.11(15)
C(1)-C(2)-C(3)	116.29(19)
C(1)-C(2)-C(8)	115.47(18)
C(3)-C(2)-C(8)	128.11(19)
C(37)-C(36)-C(35)	117.18(19)
C(37)-C(36)-C(43)	122.64(19)
C(35)-C(36)-C(43)	120.16(19)
C(19)-C(20)-C(23)	120.14(19)
C(21)-C(20)-C(19)	117.2(2)
C(21)-C(20)-C(23)	122.3(2)
C(34)-C(33)-C(6)	120.67(19)
C(38)-C(33)-C(6)	119.96(19)
C(38)-C(33)-C(34)	118.67(19)
C(35)-C(34)-C(33)	120.2(2)
C(35)-C(34)-H(34)	119.9
C(33)-C(34)-H(34)	119.9
C(22)-C(17)-C(3)	120.30(19)
C(18)-C(17)-C(3)	119.20(19)
C(18)-C(17)-C(22)	118.81(19)
C(20)-C(19)-H(19)	119.0
C(18)-C(19)-C(20)	121.9(2)
C(18)-C(19)-H(19)	119.0
C(37)-C(38)-H(38)	119.7
C(33)-C(38)-C(37)	120.6(2)
C(33)-C(38)-H(38)	119.7
C(17)-C(22)-H(22)	119.8
C(21)-C(22)-C(17)	120.3(2)
C(21)-C(22)-H(22)	119.8
C(27)-C(32)-C(5)	117.14(18)
C(31)-C(32)-C(27)	115.06(19)
C(31)-C(32)-C(5)	127.80(19)
O(2)-C(47)-C(48)	111.8(2)
O(1)-C(47)-O(2)	129.4(2)

O(1)-C(47)-C(48)	118.7(2)
C(20)-C(21)-H(21)	119.3
C(22)-C(21)-C(20)	121.3(2)
C(22)-C(21)-H(21)	119.3
C(7)-C(12)-H(12)	120.8
C(7)-C(12)-C(11)	118.4(2)
C(11)-C(12)-H(12)	120.8
O(4)-C(49)-C(50)	113.1(2)
O(3)-C(49)-O(4)	129.1(2)
O(3)-C(49)-C(50)	117.8(2)
C(31)-C(30)-H(30)	118.6
C(31)-C(30)-C(29)	122.7(2)
C(29)-C(30)-H(30)	118.6
C(17)-C(18)-H(18)	120.0
C(19)-C(18)-C(17)	120.0(2)
C(19)-C(18)-H(18)	120.0
C(20)-C(23)-C(24)	111.13(19)
C(20)-C(23)-C(25)	106.94(18)
C(24)-C(23)-C(25)	108.68(19)
C(26)-C(23)-C(20)	112.29(19)
C(26)-C(23)-C(24)	109.5(2)
C(26)-C(23)-C(25)	108.2(2)
C(7)-C(8)-C(2)	117.76(18)
C(7)-C(8)-C(9)	115.45(19)
C(9)-C(8)-C(2)	126.55(19)
F(2)-C(48)-F(3)	104.8(2)
F(2)-C(48)-C(47)	112.04(19)
F(1)-C(48)-F(2)	108.3(2)
F(1)-C(48)-F(3)	105.9(2)
F(1)-C(48)-C(47)	113.7(2)
F(3)-C(48)-C(47)	111.6(2)
H(46A)-C(46)-H(46B)	109.5
H(46A)-C(46)-H(46C)	109.5
H(46B)-C(46)-H(46C)	109.5
C(43)-C(46)-H(46A)	109.5
C(43)-C(46)-H(46B)	109.5

C(43)-C(46)-H(46C)	109.5
C(27)-C(28)-H(28)	120.6
C(27)-C(28)-C(29)	118.8(2)
C(29)-C(28)-H(28)	120.6
C(32)-C(31)-H(31)	119.5
C(30)-C(31)-C(32)	120.9(2)
C(30)-C(31)-H(31)	119.5
C(30)-C(29)-C(28)	117.43(19)
C(30)-C(29)-C(39)	123.50(19)
C(28)-C(29)-C(39)	119.06(19)
C(9)-C(10)-H(10)	118.6
C(9)-C(10)-C(11)	122.7(2)
C(11)-C(10)-H(10)	118.6
C(36)-C(43)-C(46)	108.30(18)
C(36)-C(43)-C(45)	112.47(19)
C(36)-C(43)-C(44)	109.23(18)
C(45)-C(43)-C(46)	108.13(19)
C(45)-C(43)-C(44)	108.99(19)
C(44)-C(43)-C(46)	109.7(2)
C(8)-C(9)-H(9)	119.8
C(10)-C(9)-C(8)	120.5(2)
C(10)-C(9)-H(9)	119.8
C(12)-C(11)-C(10)	117.7(2)
C(12)-C(11)-C(13)	121.5(2)
C(10)-C(11)-C(13)	120.8(2)
F(5)-C(50)-F(6)	105.2(2)
F(5)-C(50)-C(49)	113.08(19)
F(6)-C(50)-C(49)	112.3(2)
F(4)-C(50)-F(5)	107.4(3)
F(4)-C(50)-F(6)	106.0(3)
F(4)-C(50)-C(49)	112.2(2)
C(42)-C(39)-C(29)	111.78(19)
C(42)-C(39)-C(41)	107.6(2)
C(41)-C(39)-C(29)	110.10(18)
C(40)-C(39)-C(29)	108.23(19)
C(40)-C(39)-C(42)	109.1(2)

C(40)-C(39)-C(41)	110.1(2)
C(43)-C(45)-H(45A)	109.5
C(43)-C(45)-H(45B)	109.5
C(43)-C(45)-H(45C)	109.5
H(45A)-C(45)-H(45B)	109.5
H(45A)-C(45)-H(45C)	109.5
H(45B)-C(45)-H(45C)	109.5
C(23)-C(24)-H(24A)	109.5
C(23)-C(24)-H(24B)	109.5
C(23)-C(24)-H(24C)	109.5
H(24A)-C(24)-H(24B)	109.5
H(24A)-C(24)-H(24C)	109.5
H(24B)-C(24)-H(24C)	109.5
C(39)-C(42)-H(42A)	109.5
C(39)-C(42)-H(42B)	109.5
C(39)-C(42)-H(42C)	109.5
H(42A)-C(42)-H(42B)	109.5
H(42A)-C(42)-H(42C)	109.5
H(42B)-C(42)-H(42C)	109.5
C(43)-C(44)-H(44A)	109.5
C(43)-C(44)-H(44B)	109.5
C(43)-C(44)-H(44C)	109.5
H(44A)-C(44)-H(44B)	109.5
H(44A)-C(44)-H(44C)	109.5
H(44B)-C(44)-H(44C)	109.5
C(23)-C(26)-H(26A)	109.5
C(23)-C(26)-H(26B)	109.5
C(23)-C(26)-H(26C)	109.5
H(26A)-C(26)-H(26B)	109.5
H(26A)-C(26)-H(26C)	109.5
H(26B)-C(26)-H(26C)	109.5
H(15A)-C(15)-H(15B)	109.5
H(15A)-C(15)-H(15C)	109.5
H(15B)-C(15)-H(15C)	109.5
C(13)-C(15)-H(15A)	109.5
C(13)-C(15)-H(15B)	109.5

C(13)-C(15)-H(15C)	109.5
C(23)-C(25)-H(25A)	109.5
C(23)-C(25)-H(25B)	109.5
C(23)-C(25)-H(25C)	109.5
H(25A)-C(25)-H(25B)	109.5
H(25A)-C(25)-H(25C)	109.5
H(25B)-C(25)-H(25C)	109.5
C(11)-C(13)-C(14)	107.8(2)
C(15)-C(13)-C(11)	110.3(2)
C(15)-C(13)-C(14)	109.4(2)
C(16)-C(13)-C(11)	112.0(2)
C(16)-C(13)-C(15)	108.9(2)
C(16)-C(13)-C(14)	108.3(2)
C(39)-C(41)-H(41A)	109.5
C(39)-C(41)-H(41B)	109.5
C(39)-C(41)-H(41C)	109.5
H(41A)-C(41)-H(41B)	109.5
H(41A)-C(41)-H(41C)	109.5
H(41B)-C(41)-H(41C)	109.5
C(39)-C(40)-H(40A)	109.5
C(39)-C(40)-H(40B)	109.5
C(39)-C(40)-H(40C)	109.5
H(40A)-C(40)-H(40B)	109.5
H(40A)-C(40)-H(40C)	109.5
H(40B)-C(40)-H(40C)	109.5
C(13)-C(16)-H(16A)	109.5
C(13)-C(16)-H(16B)	109.5
C(13)-C(16)-H(16C)	109.5
H(16A)-C(16)-H(16B)	109.5
H(16A)-C(16)-H(16C)	109.5
H(16B)-C(16)-H(16C)	109.5
C(13)-C(14)-H(14A)	109.5
C(13)-C(14)-H(14B)	109.5
C(13)-C(14)-H(14C)	109.5
H(14A)-C(14)-H(14B)	109.5
H(14A)-C(14)-H(14C)	109.5

H(14B)-C(14)-H(14C)

109.5

Symmetry transformations used to generate equivalent atoms:

Table S9. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for compound **3_0m_a**. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U^{11}	U^{22}	U^{33}	U^{23}	U^{13}	U^{12}
I(1)	12(1)	15(1)	19(1)	-5(1)	-2(1)	0(1)
I(2)	12(1)	15(1)	22(1)	-6(1)	0(1)	-1(1)
F(2)	25(1)	45(1)	26(1)	-10(1)	-10(1)	12(1)
F(5)	27(1)	66(1)	33(1)	-20(1)	-15(1)	14(1)
O(4)	18(1)	22(1)	37(1)	-8(1)	2(1)	-2(1)
O(2)	19(1)	23(1)	35(1)	-9(1)	2(1)	1(1)
F(1)	28(1)	58(1)	76(1)	15(1)	20(1)	7(1)
O(3)	25(1)	21(1)	32(1)	-4(1)	-3(1)	0(1)
F(3)	58(1)	89(1)	60(1)	-58(1)	-31(1)	41(1)
O(1)	32(1)	23(1)	48(1)	-8(1)	3(1)	-2(1)
C(4)	11(1)	18(1)	18(1)	-7(1)	-1(1)	2(1)
C(1)	12(1)	17(1)	16(1)	-5(1)	-2(1)	1(1)
C(6)	14(1)	17(1)	16(1)	-4(1)	-3(1)	-1(1)
C(3)	16(1)	18(1)	16(1)	-5(1)	-3(1)	-1(1)
C(37)	18(1)	22(1)	23(1)	-8(1)	-6(1)	-1(1)
F(6)	68(1)	145(2)	67(1)	-82(2)	-42(1)	73(1)
C(27)	18(1)	14(1)	16(1)	-3(1)	-3(1)	-3(1)
C(35)	22(1)	25(1)	18(1)	-2(1)	-2(1)	-8(1)
C(5)	15(1)	16(1)	14(1)	-4(1)	-3(1)	-1(1)
C(7)	17(1)	15(1)	18(1)	-5(1)	-4(1)	-4(1)
C(2)	16(1)	16(1)	15(1)	-5(1)	-3(1)	0(1)
C(36)	16(1)	18(1)	23(1)	-9(1)	-2(1)	-2(1)
C(20)	17(1)	17(1)	22(1)	-7(1)	-3(1)	-2(1)
C(33)	12(1)	14(1)	21(1)	-7(1)	-1(1)	0(1)
C(34)	17(1)	25(1)	20(1)	-5(1)	-6(1)	-3(1)
C(17)	13(1)	13(1)	22(1)	-6(1)	-2(1)	0(1)
C(19)	20(1)	28(1)	23(1)	-12(1)	-4(1)	-2(1)
C(38)	21(1)	19(1)	18(1)	-5(1)	-3(1)	-3(1)
C(22)	19(1)	19(1)	19(1)	-3(1)	-7(1)	-3(1)
C(32)	16(1)	15(1)	16(1)	-4(1)	-3(1)	0(1)
C(47)	18(1)	27(1)	21(1)	-9(1)	-4(1)	1(1)

C(21)	18(1)	20(1)	20(1)	-4(1)	-1(1)	-5(1)
C(12)	15(1)	21(1)	22(1)	-9(1)	-6(1)	1(1)
C(49)	18(1)	24(1)	18(1)	-9(1)	-2(1)	-1(1)
C(30)	20(1)	14(1)	25(1)	-4(1)	-4(1)	-2(1)
C(18)	20(1)	21(1)	19(1)	-10(1)	0(1)	-2(1)
F(4)	23(1)	68(2)	163(3)	50(2)	26(1)	3(1)
C(23)	16(1)	28(1)	25(1)	-10(1)	-3(1)	-4(1)
C(8)	18(1)	16(1)	17(1)	-7(1)	-6(1)	2(1)
C(48)	23(1)	35(1)	21(1)	-4(1)	-1(1)	5(1)
C(46)	18(1)	39(2)	33(1)	-18(1)	3(1)	-6(1)
C(28)	15(1)	18(1)	18(1)	-3(1)	-3(1)	1(1)
C(31)	15(1)	19(1)	21(1)	-6(1)	-1(1)	-1(1)
C(29)	21(1)	17(1)	16(1)	-4(1)	-3(1)	2(1)
C(10)	23(1)	17(1)	31(1)	-8(1)	-5(1)	-4(1)
C(43)	16(1)	24(1)	26(1)	-9(1)	-1(1)	-5(1)
C(9)	17(1)	19(1)	28(1)	-9(1)	-2(1)	-3(1)
C(11)	21(1)	20(1)	28(1)	-11(1)	-9(1)	1(1)
C(50)	21(1)	38(2)	22(1)	-1(1)	2(1)	4(1)
C(39)	23(1)	16(1)	27(1)	-7(1)	-2(1)	1(1)
C(45)	21(1)	40(2)	36(1)	-15(1)	-5(1)	-10(1)
C(24)	21(1)	29(1)	39(2)	-12(1)	-7(1)	-6(1)
C(42)	30(1)	24(1)	46(2)	-18(1)	-2(1)	0(1)
C(44)	23(1)	26(1)	43(2)	-8(1)	0(1)	-9(1)
C(26)	18(1)	56(2)	31(1)	-10(1)	0(1)	-10(1)
C(15)	31(2)	27(1)	78(2)	-32(2)	-14(1)	3(1)
C(25)	19(1)	36(1)	40(2)	-16(1)	-10(1)	4(1)
C(13)	26(1)	18(1)	47(2)	-14(1)	-11(1)	2(1)
C(41)	36(2)	20(1)	39(2)	-11(1)	9(1)	2(1)
C(40)	31(1)	25(1)	40(2)	-9(1)	-12(1)	8(1)
C(16)	24(1)	26(1)	97(3)	-31(2)	-11(2)	6(1)
C(14)	73(2)	28(2)	59(2)	-7(1)	-30(2)	16(2)

Table S10. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^{-3}$) for compound **3_0m_a**.

	x	y	z	U(eq)
H(37)	3540	3204	2636	24
H(35)	4198	2128	5441	27
H(34)	5821	2913	4779	25
H(19)	11040	6561	2865	27
H(38)	5173	3990	1961	24
H(22)	9302	6778	337	23
H(21)	11077	7256	-114	24
H(12)	4074	7989	3179	22
H(30)	7819	659	2617	24
H(18)	9276	6073	3326	24
H(46A)	2321	2544	5894	45
H(46B)	1208	2372	5693	45
H(46C)	1804	3397	5012	45
H(28)	10651	2024	1619	22
H(31)	6869	2105	2876	23
H(10)	6851	9431	1733	28
H(9)	7845	7894	1830	25
H(45A)	1201	2016	4147	47
H(45B)	2300	2021	3300	47
H(45C)	1757	3062	3463	47
H(24A)	13550	7792	1361	43
H(24B)	12497	7479	2271	43
H(24C)	12402	8414	1241	43
H(42A)	8710	-732	2564	50
H(42B)	9688	-1130	1816	50
H(42C)	8869	-193	1353	50
H(44A)	3119	881	5700	49
H(44B)	3181	681	4671	49
H(44C)	2035	701	5454	49
H(26A)	12637	8043	-400	55

H(26B)	12867	6886	-358	55
H(26C)	13772	7433	-202	55
H(15A)	5194	10202	3415	62
H(15B)	4804	11223	2590	62
H(15C)	5975	10664	2313	62
H(25A)	13845	5977	1394	46
H(25B)	12880	5461	1290	46
H(25C)	12784	5706	2302	46
H(41A)	10435	905	344	53
H(41B)	11180	-109	771	53
H(41C)	11350	920	905	53
H(40A)	11049	158	2798	50
H(40B)	11003	-918	2698	50
H(40C)	10033	-512	3445	50
H(16A)	3189	9445	2556	71
H(16B)	3148	10502	2726	71
H(16C)	3470	9455	3563	71
H(14A)	5544	10567	767	83
H(14B)	4382	11151	1019	83
H(14C)	4458	10078	873	83

Computational Studies

Single point energy calculations on compounds **1-3** were conducted using the Gaussian09 program³ using the geometries and methods and basis sets as recently reported (B3LYP 6-31+g(d,p) and SDD pseudopotential for iodine).⁴ Bond critical points were analyzed using Multiwfn⁵ and visualized using AIMALL.⁶ Visualization of the non-covalent interactions (NCI plots) were conducted using VMD 1.94.⁷

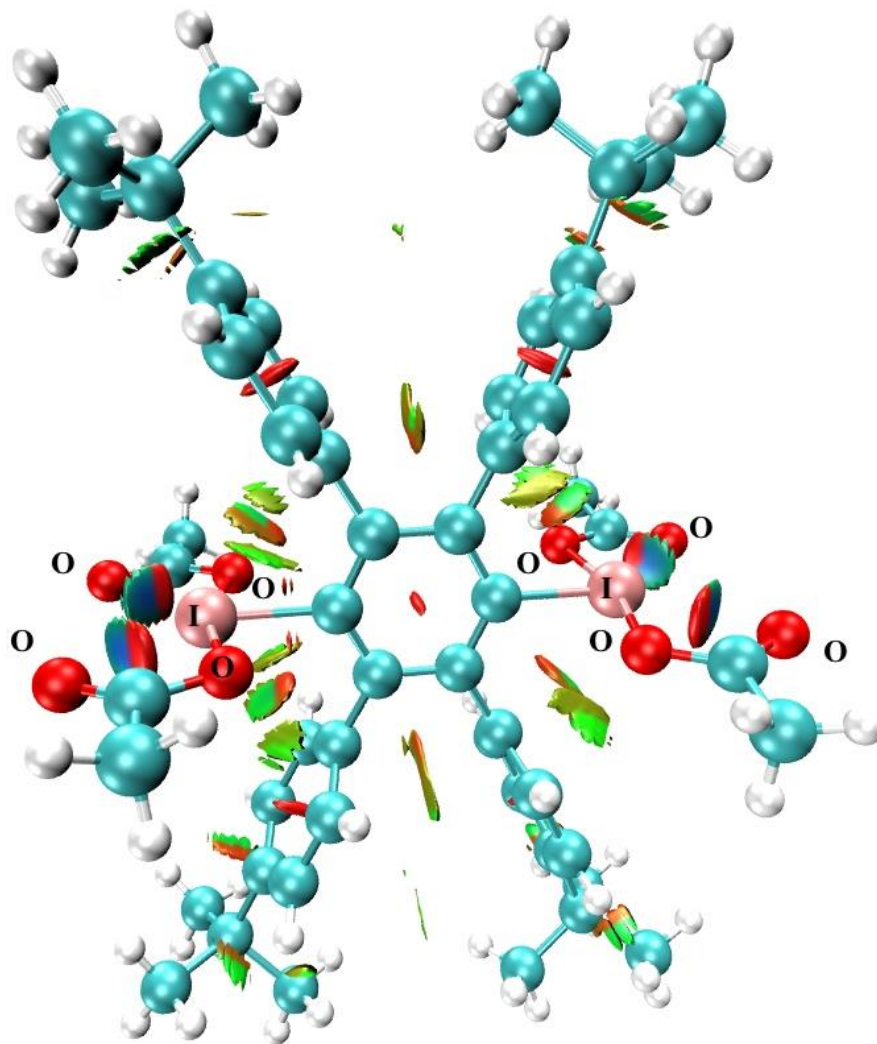


Figure S7. Computed NCI plots of compound **2**. The surface color red indicates repulsions, green van der Waals attractions and/or weak interactions, and blue stronger attractive interactions.

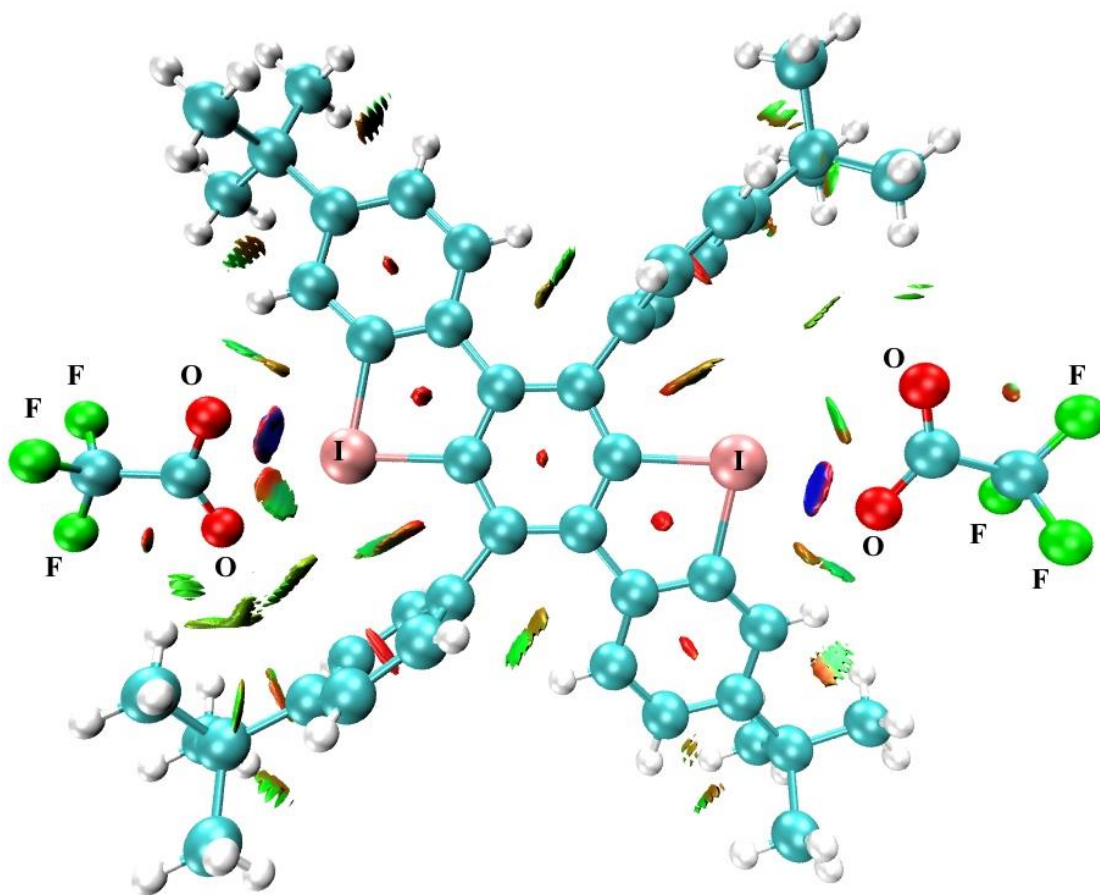


Figure S8. Computed NCI plots of compound **3**. The surface color red indicates repulsions, green van der Waals attractions and/or weak interactions, and blue stronger attractive interactions.

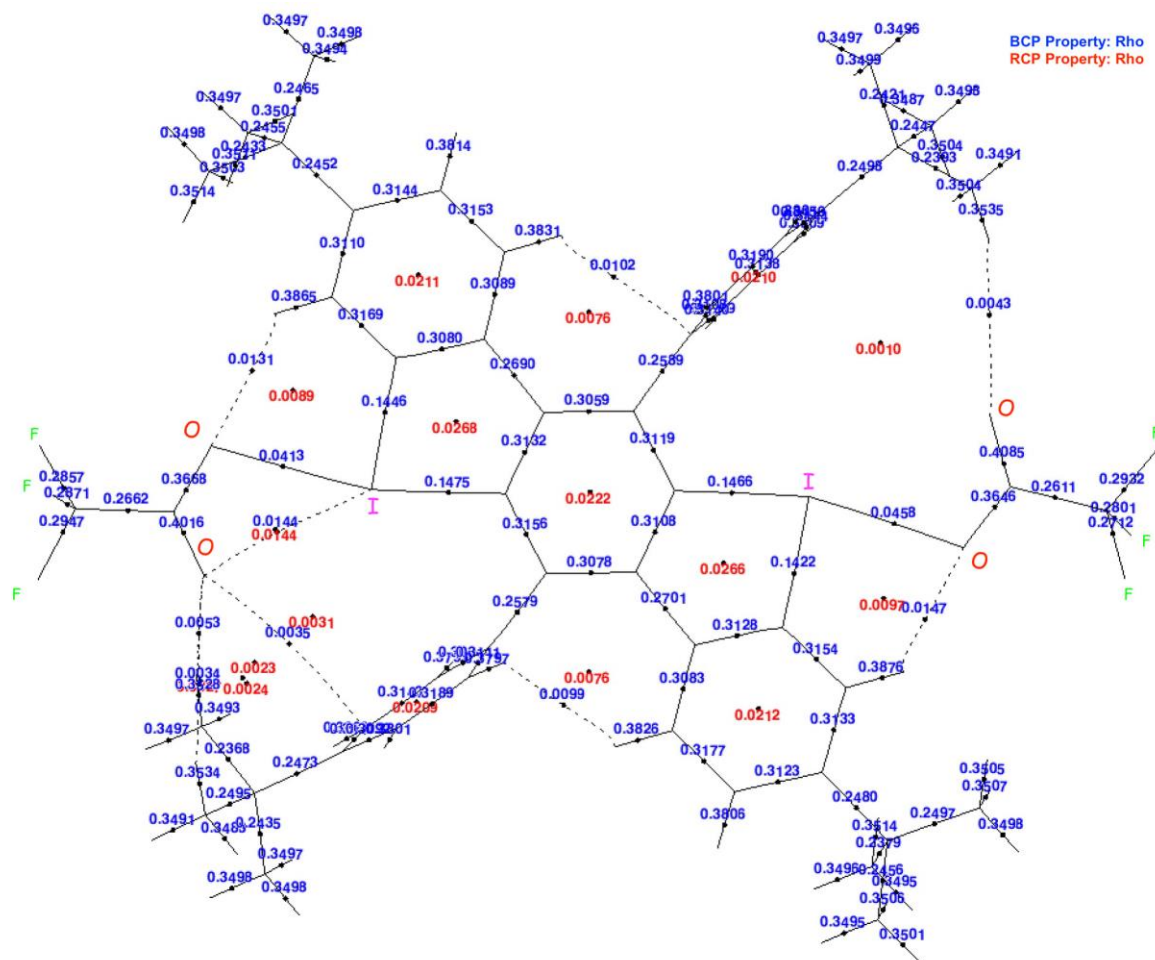


Figure S9. Computed NCI plots of compound 3.

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