

**Deep-ultraviolet transparent alkali metal-rare earth metal
sulfate $\text{NaY}(\text{SO}_4)_2 \cdot \text{H}_2\text{O}$ as a nonlinear optical crystal: synthesis
and characterization**

Chao Wu, Lin Lin, Tianhui Wu, Zhipeng Huang, Chi Zhang*

China-Australia Joint Research Center for Functional Molecular Materials, School of Chemical
Science and Engineering, Tongji University, Shanghai 200092, P. R. China

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Table S1. Selected bond distances (\AA) and angles (deg) for $\text{NaY}(\text{SO}_4)_2\text{H}_2\text{O}$.^a

Y(1)-O(2)#1	2.355(5)	Y(1)-O(2)#2	2.355(5)
Y(1)-O(1W)	2.386(11)	Y(1)-O(4)#3	2.393(6)
Y(1)-O(4)#4	2.393(6)	Y(1)-O(1)#5	2.441(5)
Y(1)-O(1)	2.441(5)	Y(1)-O(3)#5	2.474(5)
Y(1)-O(3)	2.474(5)	S(1)-O(4)	1.461(5)
S(1)-O(2)	1.471(5)	S(1)-O(3)	1.473(5)
S(1)-O(1)	1.482(5)	Na(1)-O(3)#9	2.356(6)
Na(1)-O(3)	2.356(6)	Na(1)-O(1)#10	2.415(6)
Na(1)-O(1)#4	2.415(6)	Na(1)-O(4)#11	2.533(5)
Na(1)-O(4)#2	2.533(5)	Na(1)-O(2)#2	2.888(5)
Na(1)-O(2)#11	2.888(5)		
O(4)-S(1)-O(2)	107.8(3)	O(4)-S(1)-O(3)	112.0(3)
O(2)-S(1)-O(3)	111.4(3)	O(4)-S(1)-O(1)	111.1(3)
O(2)-S(1)-O(1)	109.7(3)	O(3)-S(1)-O(1)	104.8(3)

^aSymmetry codes: #1 $y, x-1, -z$; #2 $-x+y+1, -x+1, z-1/3$; #3 $-x+y+1, -x, z-1/3$; #4 $y+1, x, -z$; #5 $x-y, -y, -z-1/3$; #6 $-y+1, x-y, z+1/3$; #7 $x, y-1, z$; #8 $-y, x-y-1, z+1/3$; #9 $y, x, -z$; #10 $x, y+1, z$; #11 $-x+1, -x+y+1, -z+1/3$.

Table S2. Atomic coordinates ($\times 10^4$), equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) of $\text{NaY}(\text{SO}_4)_2 \cdot \text{H}_2\text{O}$. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

Atom	Wyck.	x	y	z	$U_{\text{eq}}(\text{\AA}^2)$
Y(1)	6c	5631(1)	0	-1667	8(1)
S(1)	6c	5573(3)	139(3)	741(1)	8(1)
Na(1)	3a	4716(6)	4716(6)	0	17(1)
O(1)	6c	6164(8)	-1250(9)	56(4)	14(1)
O(2)	6c	7573(8)	1742(9)	1350(4)	14(1)
O(3)	6c	4848(10)	1328(9)	9(3)	14(1)
O(4)	6c	3790(9)	-1277(10)	1486(4)	15(1)
O(1W)	6c	9131(16)	0	-1667	39(3)

Table S3. Hydrogen-bonding interactions for NaY(SO₄)₂·H₂O.

D-H···A	<i>d</i> (D-H)	<i>d</i> (H···A)	<i>d</i> (D···A)	\angle (DHA)
O1W-H1···O1	0.7600	2.4700	3.2274	173.00
O1W-H1···O2	0.7600	2.4800	3.0032	128.00

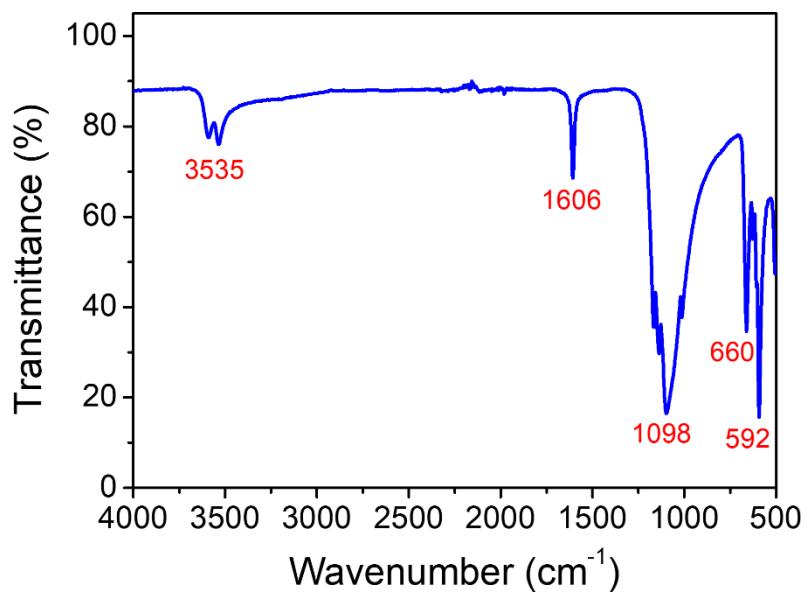


Figure S1. IR spectrum of $\text{NaY}(\text{SO}_4)_2 \cdot \text{H}_2\text{O}$.