Supporting Information

Polycyclic Motifs Engineering in Cynaostilbene Based Donors Towards

Highly Efficient Modulable Emission Properties in Two-Component System

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Figure S1. ¹HNMR of NPN

¹H NMR (400 MHz, Chloroform-*d*) δ 8.18 (d, *J* = 2.0 Hz, 1H), 7.99 – 7.87 (m, 5H), 7.77 (dd, *J* = 8.7, 2.0 Hz, 1H), 7.68 (s, 1H), 7.59 – 7.44 (m, 5H).



Figure S2. ¹HNMR of ATN

¹H NMR (400 MHz, Chloroform-*d*) δ 8.53 (s, 1H), 8.39 (d, *J* = 1.5 Hz, 1H), 8.07 (d, *J* = 9.8 Hz, 3.85H), 7.79 (d, *J* = 8.3 Hz, 2H), 7.53 (t, *J* = 3.3 Hz, 4H), 7.36 (d, *J* = 8.0 Hz, 2H), 2.47 (s, 3H).



Figure S3. ¹HNMR of PTN

¹H NMR (400 MHz, Chloroform-*d*) δ 8.63 (d, *J* = 8.1 Hz, 1H), 8.53 (s, 1H), 8.28 – 8.20 (m, 4H), 8.20 – 8.04 (m, 4H), 7.74 (d, *J* = 8.2 Hz, 2H), 7.33 (d, *J* = 8.0 Hz, 2H), 2.46 (s, 3H).



Figure S4. Twisted angle between two planes of anthracene and phenyl planes are 81.84 for 1b co-crystal.



Figure S5. Twisted angle between two planes of anthracene and phenyl planes are 18.09 for **1c** co-crystal



Figure S6. Comparison of experimental PXRD patterns of the cocrystals and the corresponding starting materials



Figure S7. Fluorescence decay curves for solid-state CS-based products

Table S1. QY and life time values of the physical mixture

Code	Ia (1:1)	Ib (1:2)	Ic (1:2)
PLQY $\Phi_F(\%)$ for physical mixture	6.04	4.04	6.42
Fluorescence lifetime $\tau_F(ns)$	10.05	8.26	10.18