

ARTICLE

Received 00th January 20xx,
Accepted 00th January 20xx

DOI: 10.1039/x0xx00000x

A Photo-responsive p-Si/TiO₂/Ag Heterostructure with Charge Transfer for Recyclable Surface-Enhanced Raman Scattering Substrate

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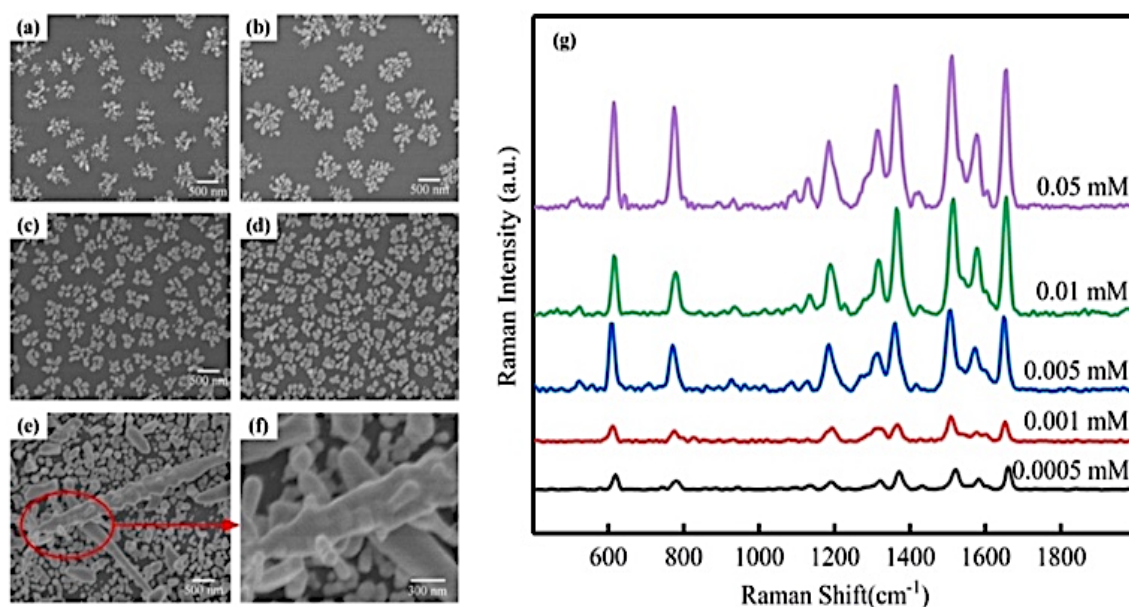


Fig. S1 SEM images of Ag nanoflowers prepared on Si/ TiO₂ substrates with different AgNO₃ concentrations and SERS spectra of R6G absorbed on Si/TiO₂/Ag substrates. (a)~(e) are the SEM images of Ag nanoflowers with the AgNO₃ concentrations of 0.0005 mM, 0.001 mM, 0.005 mM, 0.01 mM, 0.05 mM sequentially. (f) is the magnification image of the selected in (e). (g) is the SERS spectra of 10⁻⁵ M R6G absorbed on the corresponding substrates.

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See DOI: 10.1039/x0xx00000x

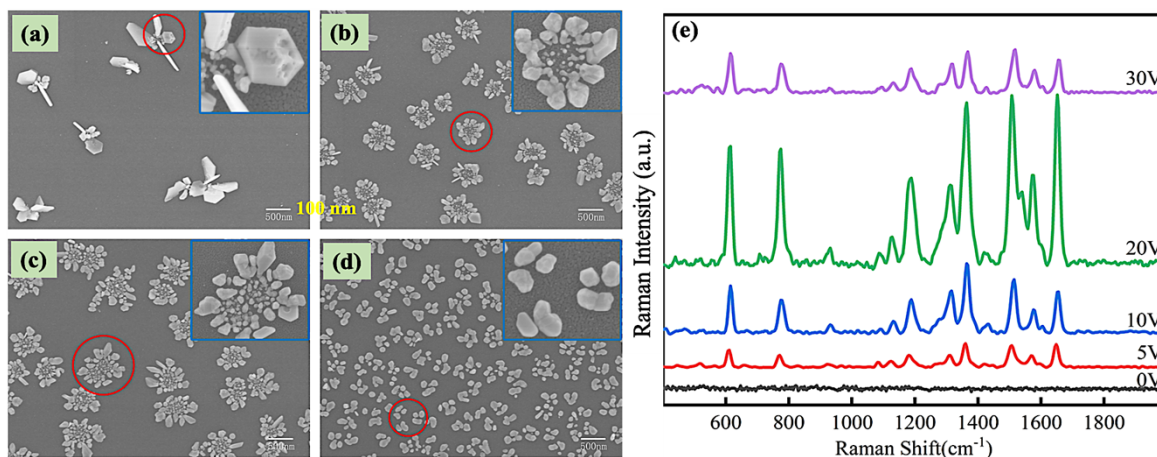


Fig. S2 SEM images of Ag nanoflowers deposited on TiO₂ thin film substrate with different applied voltages (a-d represent 5V, 10V, 20V and 30V, respectively) and the corresponding SERS spectra of R6G absorbed on it (e).

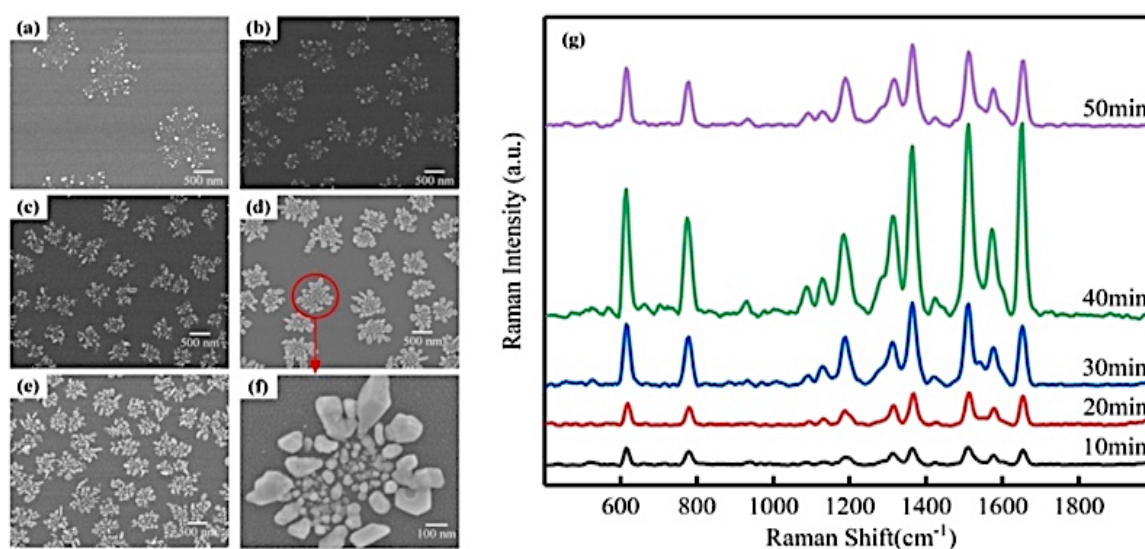


Fig. S3 SEM images of Ag nanoflowers on TiO₂ substrates prepared with different reaction times (a-f) and the corresponding SERS spectra of the 10⁻⁵M R6G (g).

The band gap energy (E_g) of TiO₂ is determined based on UV-Vis Spectra (UV-2450, Shimadzu). According to the formula:

$$\left(\frac{Ah\nu}{K}\right)^{1/2} = h\nu - E_g$$

, the E_g of TiO₂ is about 3.22 eV, as seen from the following Fig. S4.

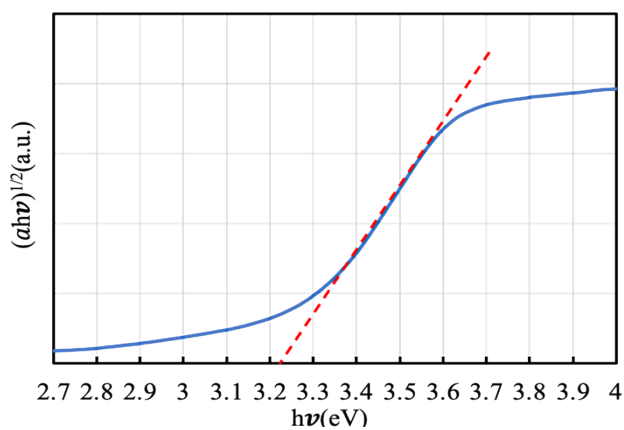


Fig.S4 Energy gap of TiO₂ based on UV -Vis spectra

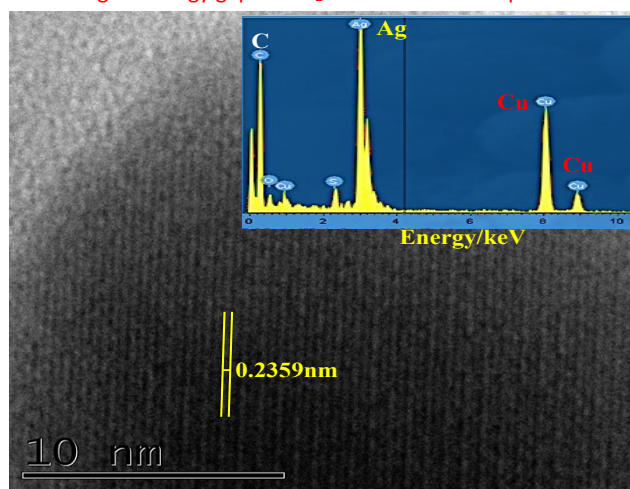


Fig. S5 the HR-TEM image of Ag and the inserted EDS result

The EF calculated formula is as follows:

$$EF = (I_{\text{SERS}}/N_{\text{SERS}}) / (I_{\text{Normal}}/N_{\text{Normal}})$$

Here, I_{SERS} and I_{Normal} is the Raman intensity of SERS and normal Raman spectrum of R6G, respectively. The N_{SERS} and N_{Normal} is the concentration of R6G, respectively. In our experiment, the SERS peak at 1649 cm^{-1} is selected and the EF is estimated to be about 1.23×10^{12} .