

Supplementary Information

Photo-sensitive Hybrids Constructed from Diphenyliodonium and Metal-thiocyanates: Photo-induced Structure and Property Transformations

Kai-Yue Song,^a Zhen-Cong Yang,^a Pan-Ke Zhou,^a Lu-Lu Zong,^a Hao-Hong Li,^{* a,b} Rong
Jiang,^{*a} Zhi-Rong Chen^{a,b}

^aCollege of Chemistry, Fuzhou University, Fuzhou, Fujian, 350116, P. R. China

^bFujian Science & Technology Innovation Laboratory for Optoelectronic Information of China,
Fuzhou, Fujian 350108, P. R. China.

Computation Methods

The geometry of DPI radical was simulated using the Gaussian 09 program package.¹ The calculations were performed by using spin restricted DFT wave functions B3LYP. The 6-31g(d) basis set was used for C, H atoms, while CEP-4g basis set was employed for the I atom.

Table S1 Selected bond lengths (Å) and angles (°) for α -1 and β -1

α -1					
Mn(1)-N(1)	2.158(4)	Mn(1)-N(2)	2.184(4)	Mn(1)-N(3)	2.155(3)
Mn(1)-N(4)	2.167(4)	Mn(1)-N(5)	2.196(3)	Mn(1)-O(2)	2.382(4)
N(3)-Mn(1)-N(4)	94.48(14)	N(1)-Mn(1)-N(4)	95.73(16)	N(4)-Mn(1)-N(2)	95.33(15)
N(4)-Mn(1)-N(5)	90.24(14)	N(3)-Mn(1)-O(2)	87.56(14)	N(1)-Mn(1)-O(2)	82.68(16)
N(2)-Mn(1)-O(2)	90.84(15)	N(5)-Mn(1)-O(2)	83.52(15)	N(4)-Mn(1)-O(2)	173.58(14)

β -1					
Mn(1)-N(1)	2.160(6)	Mn(1)-N(2)	2.174(6)	Mn(1)-N(3)	2.152(7)
Mn(1)-N(4)	2.172(6)	Mn(1)-N(5)	2.133(7)	Mn(1)-O(1)	2.478(9)
N(5)-Mn(1)-N(3)	96.7(3)	N(5)-Mn(1)-N(1)	95.3(3)	N(5)-Mn(1)-N(4)	96.9(3)
N(5)-Mn(1)-N(2)	91.5(3)	N(3)-Mn(1)-O(1)	81.1(3)	N(1)-Mn(1)-O(1)	87.3(3)
N(4)-Mn(1)-O(1)	88.4(3)	N(2)-Mn(1)-O(1)	83.2(3)	N(5)-Mn(1)-O(1)	174.3(3)

Table S2 Selected bond lengths (Å) and angles (°) for α -2 and β -2

α -2					
Ni(1)-S(1)	2.5864(14)	Ni(1)-S(1)#1	2.5864(14)	Ni(1)-N(1)#2	2.057(5)
Ni(1)-N(1)#3	2.057(5)	Ni(1)-N(2)	2.024(5)	Ni(1)-N(2)#1	2.024(5)
N(2)#1-Ni(1)-S(1)#1	87.88(13)	N(2)-Ni(1)-S(1)#1	92.12(13)	N(1)#2-Ni(1)-S(1)#1	88.26(13)
N(1)#3-Ni(1)-S(1)#1	91.74(13)	N(2)#1-Ni(1)-S(1)	92.12(13)	N(2)-Ni(1)-S(1)	87.88(13)

N(1)#2-Ni(1)-S(1)	91.74(13)	N(1)#3-Ni(1)-S(1)	88.26(13)	S(1)#1-Ni(1)-S(1)	180.000(1)
Symmetry code: #1 -x,-y+2,-z+2; #2 -x+1,-y+2,-z+2; #3 x-1,y,z					

β -2

Ni(1)-S(1)	2.6106(16)	Ni(1)-S(1)#1	2.6106(16)	Ni(1)-N(1)	2.025(5)
Ni(1)-N(1)#1	2.025(5)	Ni(1)-N(2)	2.058(5)	Ni(1)-N(2)#1	2.058(5)
N(1)#1-Ni(1)-S(1)#1	88.02(16)	N(1)-Ni(1)-S(1)#1	91.98(16)	N(2)-Ni(1)-S(1)#1	88.31(15)
N(2)#1-Ni(1)-S(1)#1	91.69(15)	N(1)#1-Ni(1)-S(1)	91.98(16)	N(1)-Ni(1)-S(1)	88.02(16)
N(2)-Ni(1)-S(1)	91.69(15)	N(2)#1-Ni(1)-S(1)	88.31(15)	S(1)#1-Ni(1)-S(1)	180.000(1)
Symmetry code: -x+1,-y+2,-z+1					

Table S3 Selected bond lengths (Å) and angles (°) for α -3 and β -3

α -3

Cd(1)-S(1)	2.768(3)	Cd(1)-S(4)	2.796(3)	Cd(1)-N(1)	2.393(10)
Cd(1)-N(2)	2.283(8)	Cd(1)-N(3)	2.270(7)	Cd(1)-N(4)	2.308(9)
Cd(2)-S(6)	2.773(3)	Cd(2)-S(6)#1	2.773(3)	Cd(2)-N(5)	2.310(8)
Cd(2)-N(5)#1	2.310(8)	Cd(2)-N(6)#1	2.272(7)	Cd(2)-N(6)	2.272(7)
N(3)-Cd(1)-N(1)	89.8(3)	N(2)-Cd(1)-N(1)	89.0(3)	N(4)-Cd(1)-N(1)	93.8(3)
N(1)-Cd(1)-S(1)	94.0(3)	N(3)-Cd(1)-S(4)	89.8(3)	N(2)-Cd(1)-S(4)	91.5(3)
N(4)-Cd(1)-S(4)	90.8(2)	N(1)-Cd(1)-S(4)	175.4(3)	S(6)-Cd(2)-S(6)#1	180.0
N(6)#1-Cd(2)-S(6)	90.1(3)	N(6)-Cd(2)-S(6)	89.9(3)	N(5)#1-Cd(2)-S(6)	85.9(2)
N(5)-Cd(2)-S(6)	94.1(2)	N(6)#1-Cd(2)-S(6)#1	89.9(3)	N(6)-Cd(2)-S(6)#1	90.1(3)
N(5)#1-Cd(2)-S(6)#1	94.1(2)	N(5)-Cd(2)-S(6)#1	85.9(2)		
Symmetry code: #1 -x,-y+2,-z+1					

β -3

Cd(1)-N(1)	2.166(16)	Cd(1)-N(1)#1	2.166(16)	Cd(1)-N(2)	2.259(9)
Cd(1)-N(2)#1	2.259(9)	Cd(1)-S(1)#2	2.909(5)	Cd(1)-S(1)#3	2.909(5)
N(1)-Cd(1)-S(1)#2	87.2(5)	N(1)#1-Cd(1)-S(1)#2	92.8(5)	N(2)-Cd(1)-S(1)#2	92.5(3)
N(2)#1-Cd(1)-S(1)#2	87.5(3)	N(1)-Cd(1)-S(1)#3	92.8(5)	N(1)#1-Cd(1)-S(1)#3	87.2(5)
N(2)-Cd(1)-S(1)#3	87.5(3)	N(2)#1-Cd(1)-S(1)#3	92.5(3)	S(1)#2-Cd(1)-S(1)#3	180.000(1)
Symmetry code: #1 -x+2,-y+2,-z+1; #2 x+1,y,z; #3 -x+1,-y+2,-z+1					

Table S4 Hydrogen bond details

	D-H	d(H...A)	d(D...A)	\angle DHA	Symmetry Code
α -1	O(2)-H(2B)···O(1)	1.86	2.751(4)	145	
β -1	O(1)-H(1B)···O(2)	1.81	2.66(5)	151	
α -2	C(4)-H(4)···S(2)	2.87	3.799(6)	176	1+x,-1+y,z
β -2	C(4)-H(4)···S(2)	2.90	3.827	174	1+x,-1+y,z
α -3	C(4)-H(4)···S(3)	2.84	3.764(15)	175	-x,-1/2+y,1/2-z
β -3	C(2)-H(2)···S(1)	2.87	3.388(13)	117	

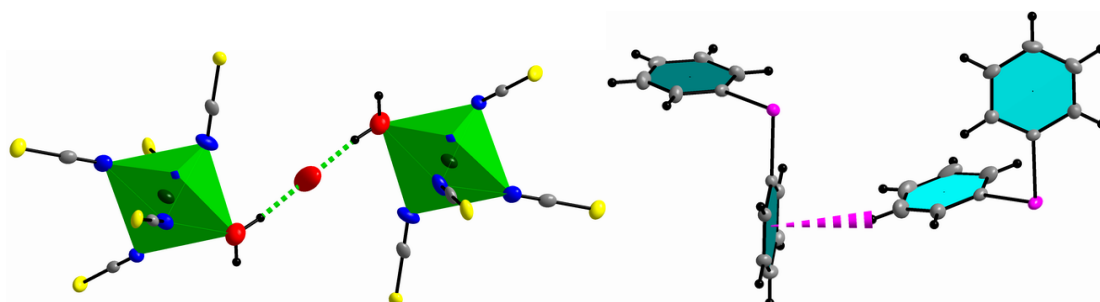
Table S5 C-H... π interaction parameters

	C-H... π	C...C _g /Å	H...C _g /Å	\angle (C-H...C _g) ^o	Symmetry transformation
α -1	C(3)-H(3)···C _g (1)	3.805(6)	2.93	157	1-x,1-y,1-z
	C _g (1): C(13)→C(14)→C(15)→C(16)→C(17)→C(18)→				
β -1	C(17)-H(17)···C _g (2)	3.886(15)	2.99	163	2-x,1-y,1-z

	C _g (1): C(7)→C(8)→C(9)→C(10)→C(11)→C(12)→				
	C(10)-H(10)⋯C _g (1)	3.710(7)	2.92	144	3/2-x,1/2+y,1/2-z
α-2	C(12)-H(12)⋯C _g (1)	3.714(7)	2.80	168	-1+x,y,z
	C _g (1): C(1)→C(2)→C(3)→C(4)→C(5)→C(6)→				
β-2	C(8)-H(8)⋯C _g (1)	3.766(8)	2.84	171	-1+x,y,z
	C _g (1): C(1)→C(2)→C(3)→C(4)→C(5)→C(6)→				
	C(12)-H(12)⋯C _g (1)	3.869(13)	2.95	172	
	C _g (1): C(13)→C(14)→C(15)→C(16)→C(17)→C(18)→				
α-3	C(28)-H(28)⋯C _g (2)	3.774(16)	2.97	145	-1-x,-1/2+y,1/2-z
	C _g (2): C(19)→C(20)→C(21)→C(22)→C(23)→C(24)→				
	C(30)-H(30)⋯C _g (3)	3.790(15)	2.86	175	-1+x,y,z
	C _g (3): C(1)→C(2)→C(3)→C(4)→C(5)→C(6)→				
β-3	C(8)-H(8)⋯C _g (1)	3.907(14)	2.99	170	1+x,y,z
	C _g (1): C(1)→C(2)→C(3)→C(4)→C(5)→C(6)→				

Table S6 Secondary hypervalent I(III)⋯S interactions (lengths in Å)

α-1		β-1	
I(1)⋯S(1)	3.249	I(1)⋯S(2)#1	3.192
I(1)⋯S(1)#1	3.340	I(1)⋯S(5)#1	3.289
I(2)⋯S(4)	3.267	I(2)⋯S(3)	3.355
I(2)⋯S(5)	3.173	I(2)⋯S(3)#1	3.299
I(3)⋯S(2)	3.297	I(3)⋯S(1)	3.269
I(3)⋯S(3)	3.227	I(3)⋯S(4)	3.347
Symmetry Code: #1: 1-x,1-y,1-z			
α-2		β-2	
I(1)⋯S(1)#1	3.385	I(1)⋯S(1)	3.410
I(1)⋯S(2)#2	3.159	I(1)⋯S(2)#3	3.176
Symmetry Code: #1: 1/2-x,-1/2+y,3/2-z; #2: -1/2+x,3/2-y,-1/2+z; #3: -x,2-y,1-z			
α-3		β-3	
I(1)⋯S(4)#1	3.411	I(1)⋯S(1)	3.416
I(1)⋯S(5)#1	3.200	I(1)⋯S(2)	3.193
I(2)⋯S(2)	3.188		
I(2)⋯S(6)	3.523		
I(3)⋯S(3)#2	3.171		
Symmetry Code: #1: -x,2-y,1-z; #2: -1-x,2-y,1-z			



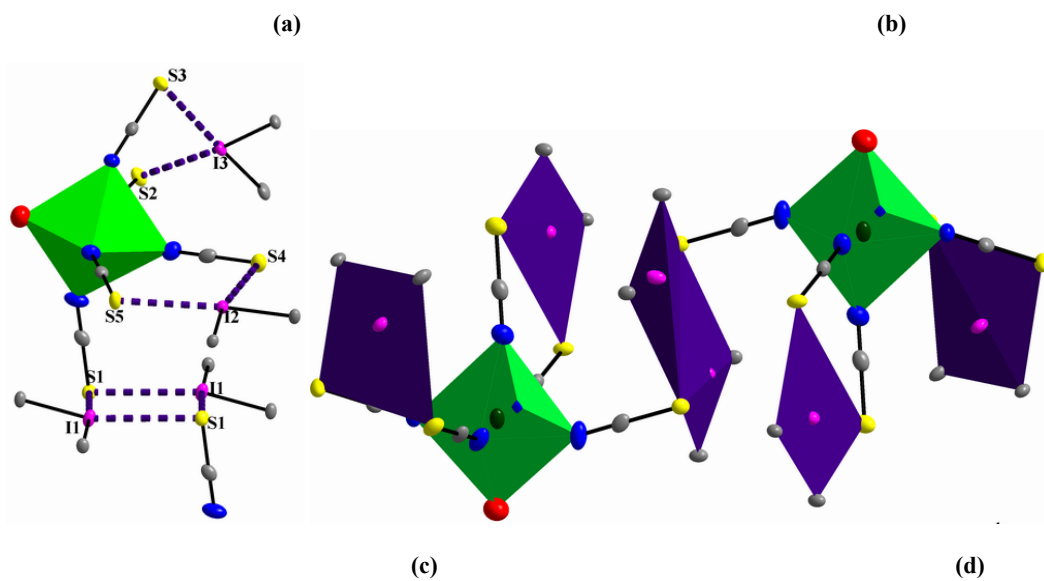
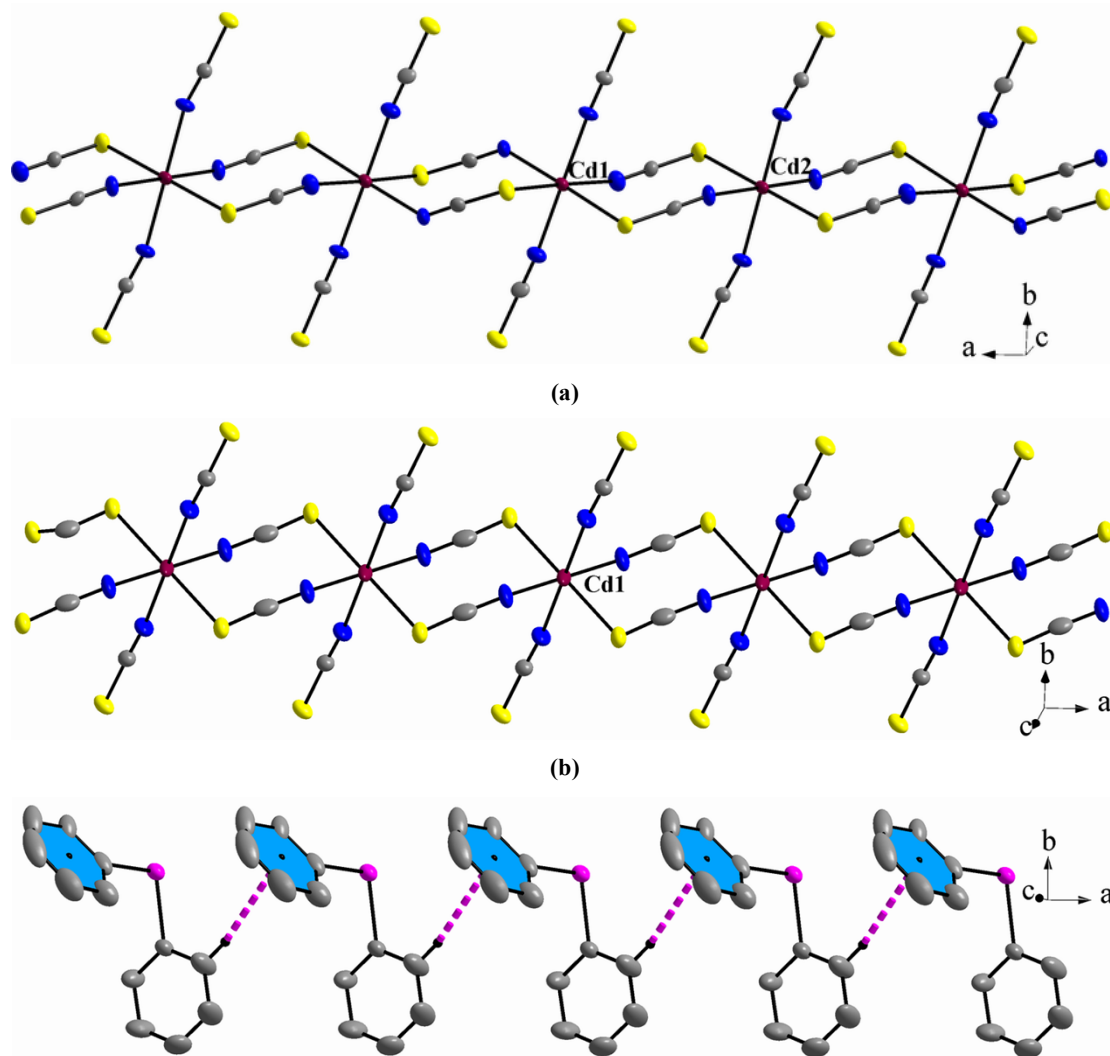
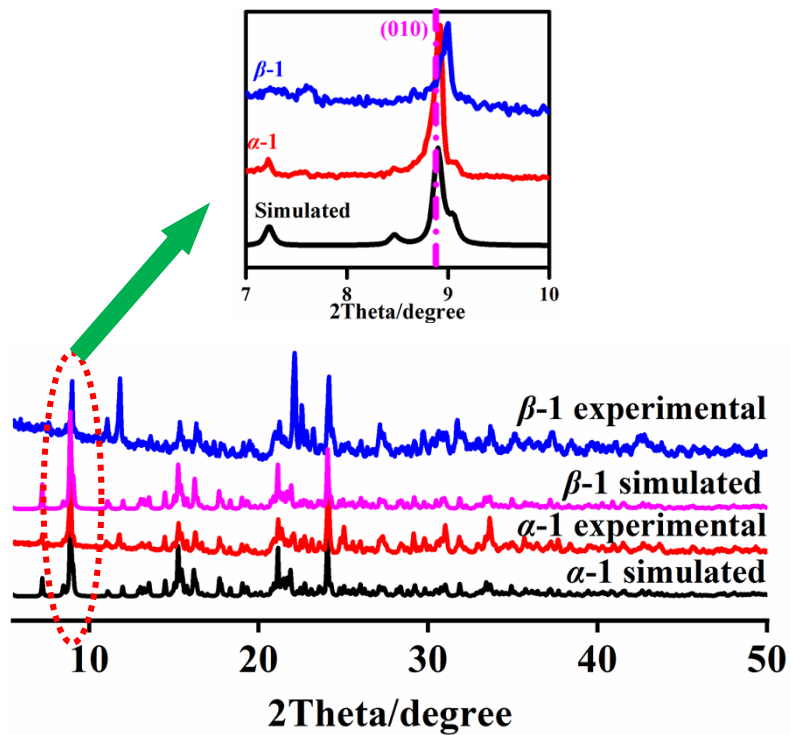


Fig. S1 (a) $\{[\text{Mn}(\text{SCN})_5(\text{H}_2\text{O})]_2\}^{6-} \cdot \text{H}_2\text{O}$ dimer based on O-H \cdots O hydrogen bond; (b) DPI⁺ dimer based on C-H \cdots π interaction of α -1; (c) the structure of $(\text{DPI})_3[\text{Mn}(\text{SCN})_5(\text{H}_2\text{O})]$ based on the secondary I(III) \cdots S contacts in α -1; (d) $(\text{DPI})_3[\text{Mn}(\text{SCN})_5(\text{H}_2\text{O})]$ dimer based on bridged DPI⁺ in α -1 (other C and H atoms on benzene rings were omitted for clarity, green: MnN₅O octahedron, purple: IS₂C₂ planar square)

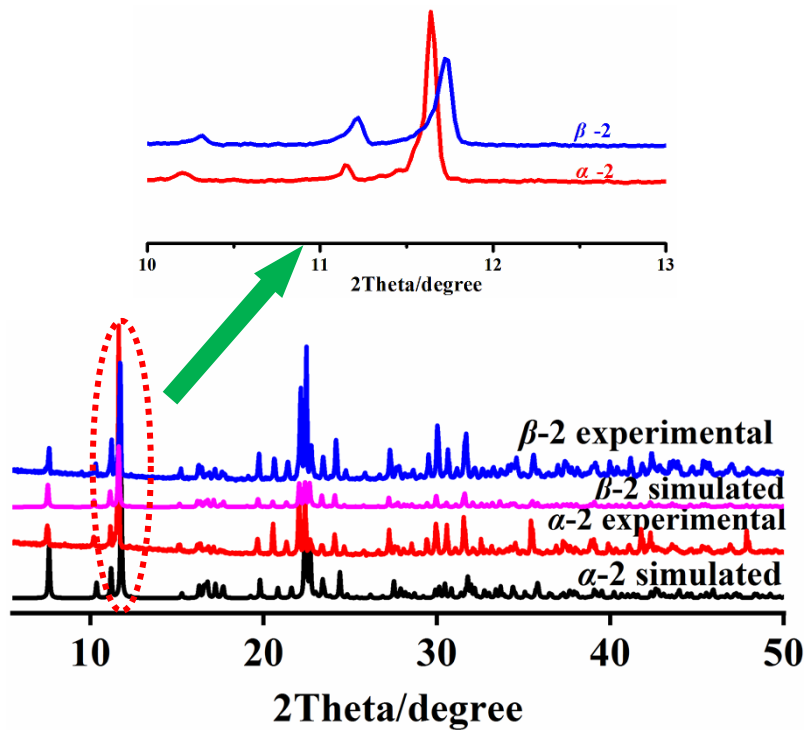


(c)

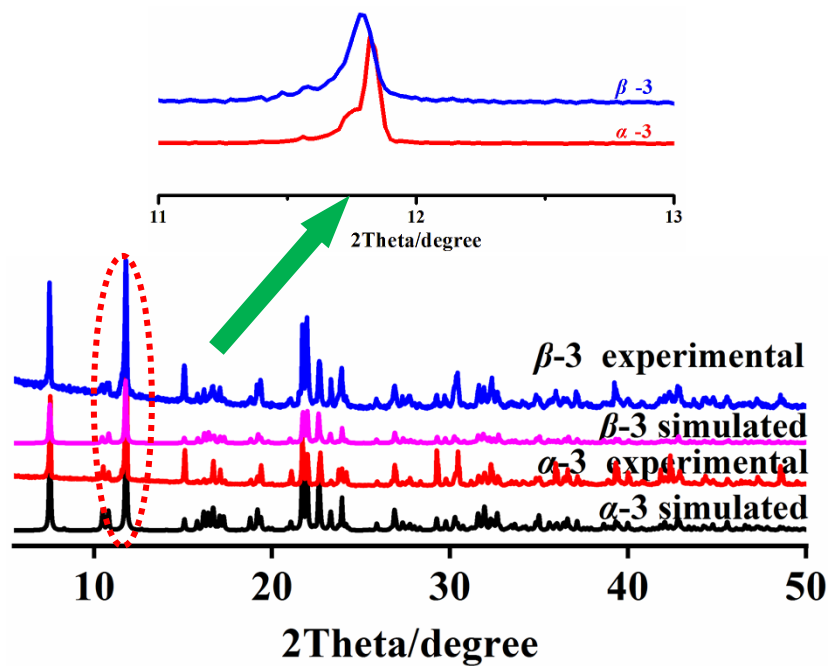
Fig. S2 (a) 1-D zigzag $\{[\text{Cd}(\text{SCN})_4]_3^{6-}\}_n$ anionic chain of α -3; (b) 1-D $\{[\text{Cd}(\text{SCN})_4]^{2-}\}_n$ of β -3; (c) 1-D $(\text{DPI})_n^{n+}$ chain based on $\text{C-H}\cdots\pi$ interaction in β -3



(a)



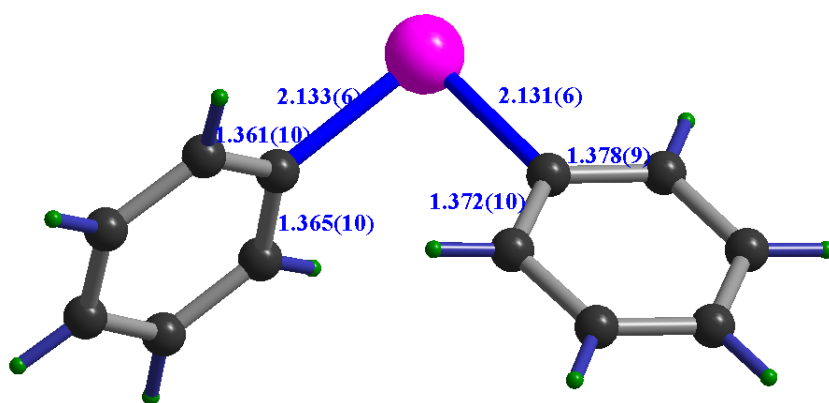
(b)



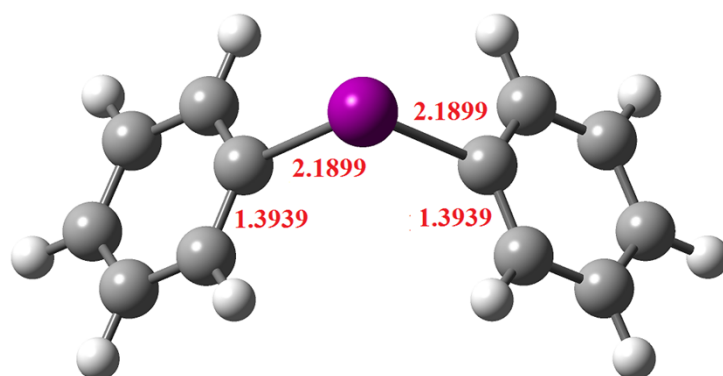
(c)

Fig. S3 Powder X-ray diffraction (PXRD) patterns of three hybrids before and after irradiating: (a) 1; (b) 2;

(c) 3

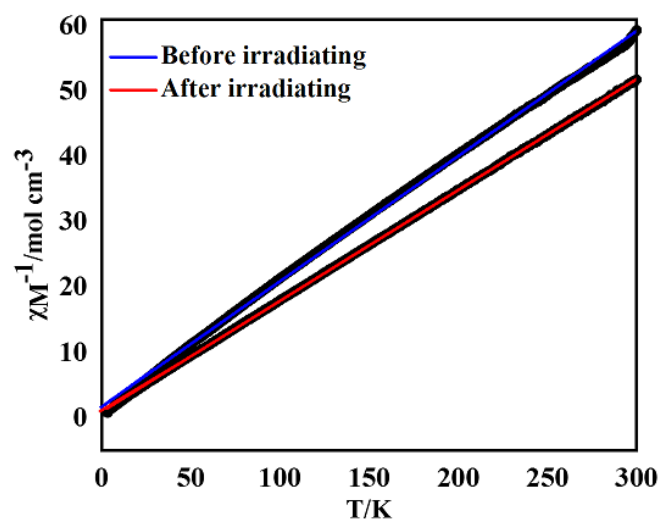


(a)

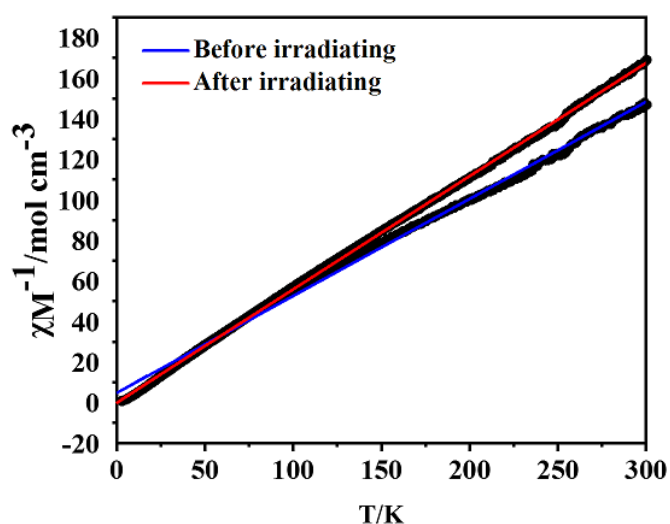


(b)

Fig. S4 The geometry of DPI⁺ cation: (a) in β -2; (b) the calculated model based on DFT calculation



(a)



(b)

Fig. S5 The $1/\chi_M$ vs. T plots: (a) α -1; (b) α -2. The red and blue lines represent the fits

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