

Supporting Information

Fig. S1 FT-IR spectra of the zinc complexes



Fig. S2 UV-Vis spectra of the zinc complexes in methanol







Fig. S4 ¹H NMR spectrum of ZnP11^{N3}



Fig. S5 Molecular structure of the Schiff base complex with 30% ellipsoid probability



Fig. S6 Hirshfeld surface analysis plots of ZnP11^{NCS}.



Fig. S7 Hirshfeld surface analysis plots of $ZnP11^{N3}$.



Fig. S8 Tauc plots of the zinc complexes.



Fig. S9 PXRD plots with de-convoluted peaks of ZnP11^{N3} complex before and after use in device B construction

Bond lengths (Å)								
Zn1–O1	2.422(2)	Zn1–O2	1.	9798(17)				
Zn1–O5	1.9975(17)	Zn1–N1	1.993((3)				
Zn1–N2	21.988(2)							
Bond angles (°)								
O1–Zn–O2	73.33(7)	01–Zn1–O	5	81.57(7)				
O1–Zn–N1	84.18(11)	O1–Zn1–N	2	168.92(10)				
O2–Zn–O5	138.31(7)	O2–Zn1–N	1	103.82(10)				
O2–Zn–N2	102.00(9)	O5 –Zn1–N	J1	106.16(9)				
O5–Zn1–N2	96.35(9)	N1-Zn1 -N	12	106.80(10)				

Table S1. Bond distances and bond angles for the ZnP11^{NCS} complex

 Table S2.
 Bond distances and bond angles for the ZnP11^{N3} complex

Bond lengths (Å))		
Zn1–O1	1.9869(15)	Zn1 –O2	2.0094(16)
Zn1–O3	2.3827(17)	Zn1 –06	2.6041(17)
Zn1–N1	2.0189(18)	Zn1 –N2	2.0440(18)
Zn2 –O1	2.0541(16)	Zn2 –O2	2.0772(16)
Zn2 – O4	2.6605(19)	Zn2 –O5	2.5506(17)
Zn2 –N3	1.962(2)	Zn2 –N6	1.942(3)
Bond angles (°)			
O1 –Zn1 –O2	78.57(7)	O1 –Zn1 –O3	132.49(6)
O1 –Zn1 –O6	88.65(6)	O1 –Zn1 –N1	138.85(7)
O1 –Zn1 –N2	90.15(7)	O2 –Zn1 –O3	90.65(6)
O2 –Zn1 –O6	129.25(6)	O2 –Zn1 –N1	91.27(7)
O2 –Zn1 –N2	143.98(7)	O3 –Zn1–O6	130.47(6)
O3–Zn1 –N1	86.78(6)	O3 –Zn1 –N2	71.97(6)
O6 –Zn1 –N1	67.19(6)	O6 –Zn1 –N2	83.87(6)
N1 –Zn1–N2	118.03(8)	O1 –Zn2 –O2	75.54(6)
O1 –Zn2 –O4	65.32(6)	O1 –Zn2 –O5	141.05(6)
O1 –Zn2 –N3	110.88(9)	O1 –Zn2 –N6	112.36(9)
O2 –Zn2 –O4	139.60(6)	O2 –Zn2 –O5	66.40(6)
O2 –Zn2 –N3	102.39(8)	O2 –Zn2 –N6	118.52(9)
O4 –Zn2 –O5	153.58(6)	O4 –Zn2 –N3	83.29(8)

O4 –Zn2 –N6	86.62(9)	O5 –Zn2 –N3	85.79(8)
O5 –Zn2 –N6	80.47(9)	N3 –Zn2 –N6	125.94(11)

Table S3. Crystallographic data and structure refinement parameters for HL

Parameters	HL
Empirical formula	C ₁₅ H ₁₅ NO ₃
Formula weight	257.28
Temperature (K)	296
Crystal system	Orthorhombic
Space group	Pca2 ₁
a (Å)	23.723(3)
b (Å)	7.7146(9)
c (Å)	7.3608(5)
Volume (Å ³)	1347.1(2)
Z	4
ρ (gcm ⁻³)	1.269
$\mu (mm^{-1})$	0.089
F (000)	544
R _{int}	0.040
θ ranges (°)	2.6-32.7
Number of unique reflections	4469
Total number of reflections	10018
Final R indices	0.0601, 0.2106
Largest peak and hole (eA°-3)	0.17, -0.16

Table S4. Contribution of different energy components to the stabilisation of molecularframeworks for $ZnP11^{NCS}$

rystal		Atoms Surface	Energies						
ntera is th otal e	ctior e dis ener	n Energies (kJ/mol) stance between molecula gies, only reported for tw yy components, scaled ap	r centroid vo benchr propriate	ls (mean atomic posi marked energy mode ly (see the scale fac	tion) in Å Is, are ti tor table	he sum below)	of the		
	N	Symop	R	Electron Density	E_ele	E_pol	E_dis	E_rep	E_tot
	2	x, y, z	10.83	B3LYP/6-31G(d,p)	-9.5	-2.9	-26.1	22.6	-21.0
	1	-x, -y, -z	16.25	B3LYP/6-31G(d,p)	1.3	-0.3	-1.3	0.0	0.0
	1	-x, -y, -z	8.40	B3LYP/6-31G(d,p)	-85.4	-1.0	-130.9	107.9	-138.4
	2	x, -y, z+1/2	9.89	B3LYP/6-31G(d,p)	-23.9	0.0	-25.0	11.7	-39.8
	1	-x+1/2, -y+1/2, -z	8.78	B3LYP/6-31G(d,p)	-82.6	-0.5	-133.1	100.9	-141.3
	2	x, -y, z+1/2	11.79	B3LYP/6-31G(d,p)	-2.4	-6.1	-11.5	11.0	-10.3
	1	-x, y, -z+1/2	11.81	B3LYP/6-31G(d,p)	-20.5	0.0	-91.0	61.0	-63.2
	2	x+1/2, -y+1/2, z+1/2	18.83	B3LYP/6-31G(d,p)	0.8	-0.6	-6.3	2.8	-3.4
	1	-x+1/2, -y+1/2, -z	12.38	B3LYP/6-31G(d,p)	7.5	-1.1	-20.0	5.7	-6.8
	2	-x+1/2, y+1/2, -z+1/2	13.99	B3LYP/6-31G(d,p)	1.2	-2.5	-24.1	10.0	-15.4
			-				2		

Scale factors for benchmarked energy models See Mackenzie et al. IUCrJ (2017)

Energy Model	k_ele	k_pol	k_disp	k_rep
CE-HF HF/3-21G electron densities	1.019	0.651	0.901	0.811
CE-B3LYP B3LYP/6-31G(d,p) electron densities	1.057	0.740	0.871	0.618

Table S5. Contribution of different energy components to the stabilisation of molecular frameworks for ZnP11^{N3}

Information ZnP11N3

Crystal Atoms Surface Energies

Interaction Energies (kJ/mol) R is the distance between molecular centroids (mean atomic position) in Å.

Total energies, only reported for two benchmarked energy models, are the sum of the four energy components, scaled appropriately (see the scale factor table below)

N	Symop	R	Electron Density	E_ele	E_pol	E_dls	E_rep	E_tot
1	x+1/2, y, -z+1/2	7.99	B3LYP/6-31G(d,p)	-63.7	-21.8	-119.9	84.3	-135.8
1	x, -y+1/2, z+1/2	12.10	B3LYP/6-31G(d,p)	-19.4	-4.1	-33.3	23.5	-38.0
0	-x, y+1/2, -z+1/2	8.44	B3LYP/6-31G(d,p)	-36.2	0.0	-71.8	49.0	-70.5
2	-x+1/2, y+1/2, z	11.30	B3LYP/6-31G(d,p)	-3.9	-7.0	-15.2	11.0	-15.7
1	-x, -y, -z	12.13	B3LYP/6-31G(d,p)	3.1	0.0	-33.7	18.9	-14.4
1	x+1/2, -y+1/2, -z	14.33	B3LYP/6-31G(d,p)	2.4	-8.9	-8.5	3.0	-9.6
0	-x+1/2, -y, z+1/2	14.94	B3LYP/6-31G(d,p)	-8.5	0.0	-4.4	1.6	-11.8
2	-x+1/2, -y, z+1/2	14.45	B3LYP/6-31G(d,p)	-1.1	-8.8	-2.9	0.0	-10.2
1	-x, -y, -z	12.54	B3LYP/6-31G(d,p)	0.5	0.0	-15.1	6.2	-8.7

Scale factors for benchmarked energy models See Mackenzie et al. IUCrJ (2017)

Energy Model	k_ele	k_pol	k_cisp	k_rep
CE-HF HF/3-21G electron densities	1.019	0.651	0.901	0.811
CE-B3LYP B3LYP/6-31G(d,p) electron densities	1.057	0.740	0.871	0.618