

Supplementary Information

Short-range structure, the role of bismuth and property-structure correlations in bismuth borate glasses

C.P.E. Varsamis^{1*}, N. Makris², C. Valvi³ and E.I. Kamitsos^{2*}

¹*Applied Physics Laboratory, Faculty of Engineering, University of West Attica, 250 Thivon, 112 41 Egaleo, Attica, Greece*

²*Theoretical and Physical Chemistry Institute, National Hellenic Research Foundation, 48 Vass. Constantinou Ave., 116 35 Athens, Greece*

³*Electronic Devices and Materials Laboratory, Department of Electrical and Electronics Engineering, University of West Attica, 250 Thivon, 112 41 Egaleo, Attica, Greece*

*Corresponding authors: cvars@uniwa.gr (C.P.E. Varsamis); eikam@eie.gr (E.I. Kamitsos)

S1. Application of infrared spectroscopy for the calculation of the molar fractions of the borate species present in glasses $x=0.50$ and 0.60 of the system $x\text{Bi}_2\text{O}_3-(1-x)\text{B}_2\text{O}_3$.

Based on the discussion in Sections 3.1 and 4.1, the borate speciation in glasses $x=0.50$ and 0.60 involves pyro-borate BO_2^{2-} and ortho-borate BO_3^{3-} trigonal units and tetrahedral units in the form of either meta-borate BO_4^- or ortho-borate $\text{B}\text{O}_2\text{O}_2^{3-}$ species. For the evaluation of the corresponding molar fractions we consider the mass balance equation:

$$X_{4m} + X_{4o} + X_1 + X_0 = 1 \quad (\text{S.1})$$

X_{4m}, X_{4o} denote the molar fraction of the BO_4^- and $\text{B}\text{O}_2\text{O}_2^{3-}$ tetrahedral units, and X_1, X_0 are the molar fractions of BO_2^{2-} and BO_3^{3-} triangular units, respectively. Besides Eq. (S1), we apply a pair of equations similar to Eqs. (15) in the manuscript for the two component bands in Table 2 with frequency ν_2 and ν_3 in the ranges $1290\text{-}1315 \text{ cm}^{-1}$ and $1185\text{-}1199 \text{ cm}^{-1}$, respectively:

$$A_2 = (X_1 + X_0) a_{B-O^-} \quad (\text{S.2a})$$

$$A_3 = X_1 a_{B-\emptyset} + 0.14 X_0 a_{B-O^-} \quad (\text{S.2b})$$

where A_2 and A_3 are the normalized integrated intensities of the bands with frequency ν_2 and ν_3 , respectively. Using the previously calculated values for $a_{B-O^-} = 0.785 \text{ cm}^{-2}$ and $a_{B-\emptyset} = 0.4624 \text{ cm}^{-2}$, the molar fractions of pyroborate species $B\emptyset O_2^{2-}$, X_1 , and orthoborate species BO_3^{3-} , X_0 , can be evaluated directly from Eqs. (S.2a) and (S.2b) as follows:

$$X_0 = \frac{A_3 - A_2 \frac{a_{B-\emptyset}}{a_{B-O^-}}}{(0.14 a_{B-O^-}) - a_{B-\emptyset}} \quad (\text{S.3})$$

$$X_1 = \frac{A_2}{a_{B-O^-}} - \frac{A_3 - A_2 \frac{a_{B-\emptyset}}{a_{B-O^-}}}{(0.14 a_{B-O^-}) - a_{B-\emptyset}} = \frac{A_3 - 0.14 A_2}{a_{B-\emptyset} - 0.14 a_{B-O^-}} \quad (\text{S.4})$$

The calculated molar fractions through Eqs. (S.1), (S.3) and (S.4) for the glass $x=0.50$ are $X_1 = 0.145$, $X_0 = 0.447$ and $X_4 = X_{4m} + X_{4o} = 0.408$, whereas for the glass $x=0.60$ the corresponding values are $X_1 = 0.089$, $X_0 = 0.576$ and $X_4 = X_{4m} + X_{4o} = 0.335$. The individual fractions X_{4m} and X_{4o} were obtained by employing Eqs. (19) of the manuscript, resulting in $X_{4m} = 0.366$ and $X_{4o} = 0.042$ for $x=0.50$ and $X_{4m} = 0.281$ and $X_{4o} = 0.054$ for $x=0.60$. The obtained relative fractions X_{4m}/X_4 are and X_{4o}/X_4 are shown in Fig. 8b.