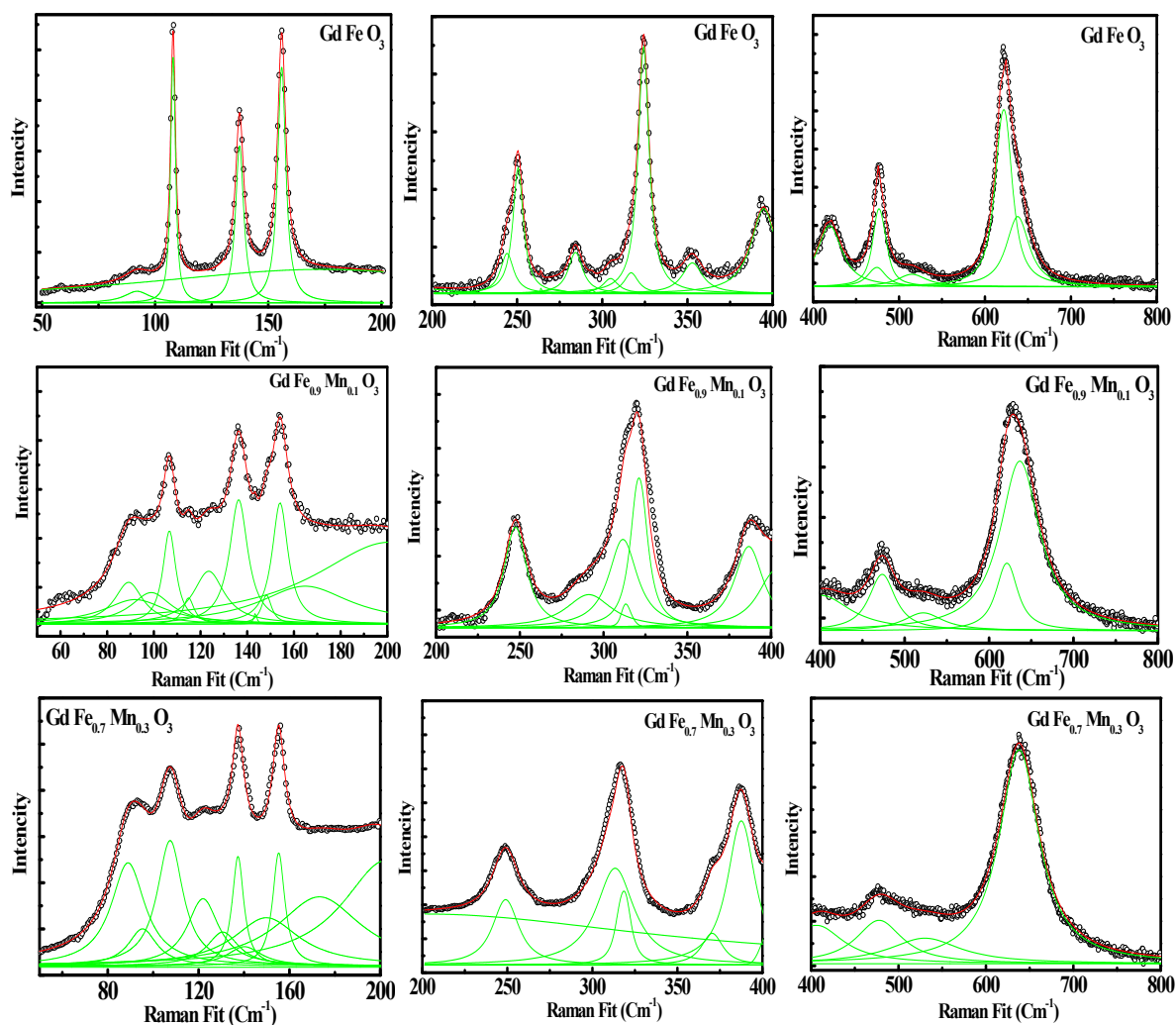
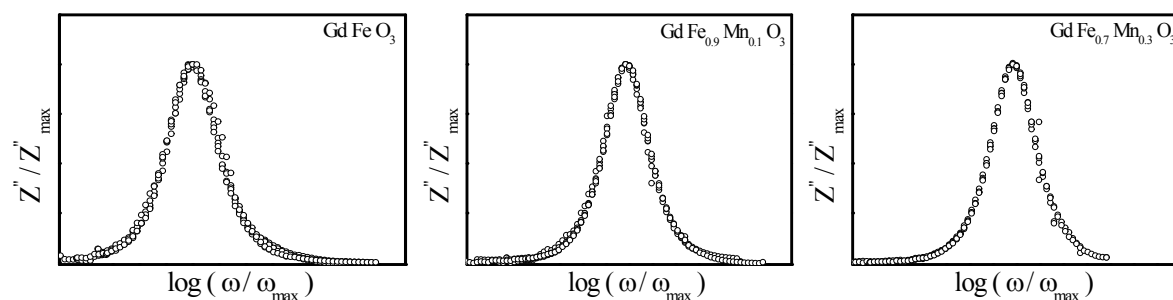


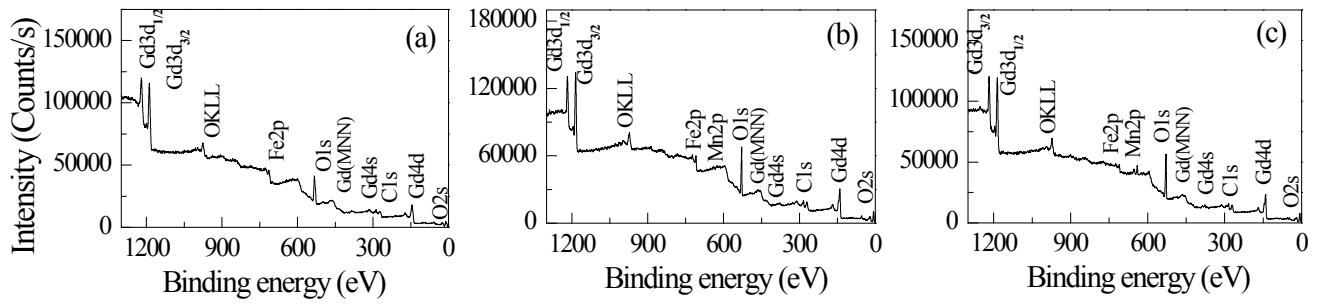
## Supporting Figure



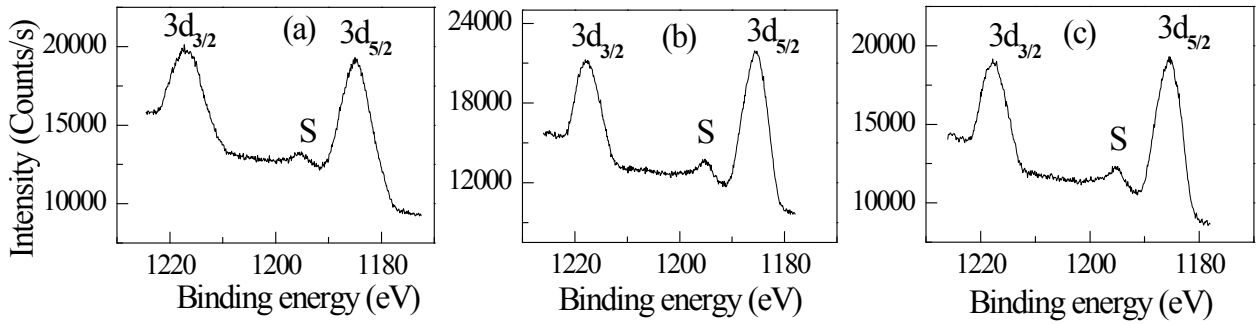
S 1. Raman spectra of  $\text{GdFeO}_3$ ,  $\text{GdFe}_{0.9}\text{Mn}_{0.1}\text{O}_3$  &  $\text{GdFe}_{0.7}\text{Mn}_{0.3}\text{O}_3$  composites.



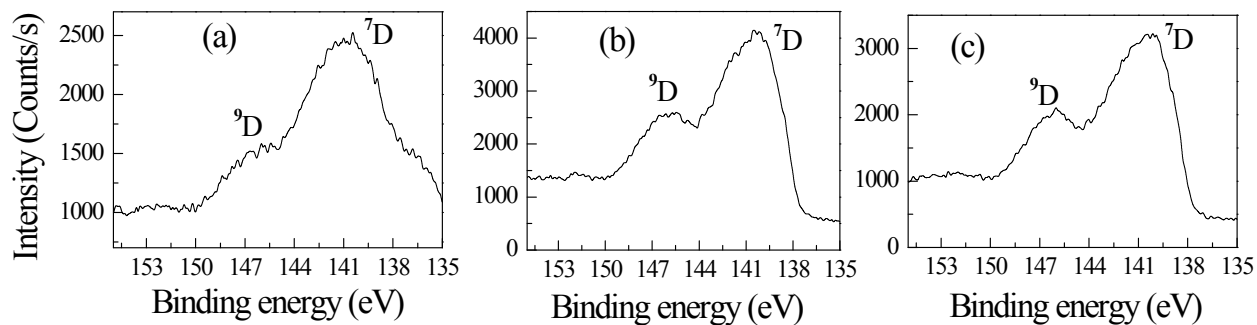
S 2. Scaling behaviour of  $Z''$  for  $\text{GdFeO}_3$ ,  $\text{GdFe}_{0.9}\text{Mn}_{0.1}\text{O}_3$  &  $\text{GdFe}_{0.7}\text{Mn}_{0.3}\text{O}_3$ .



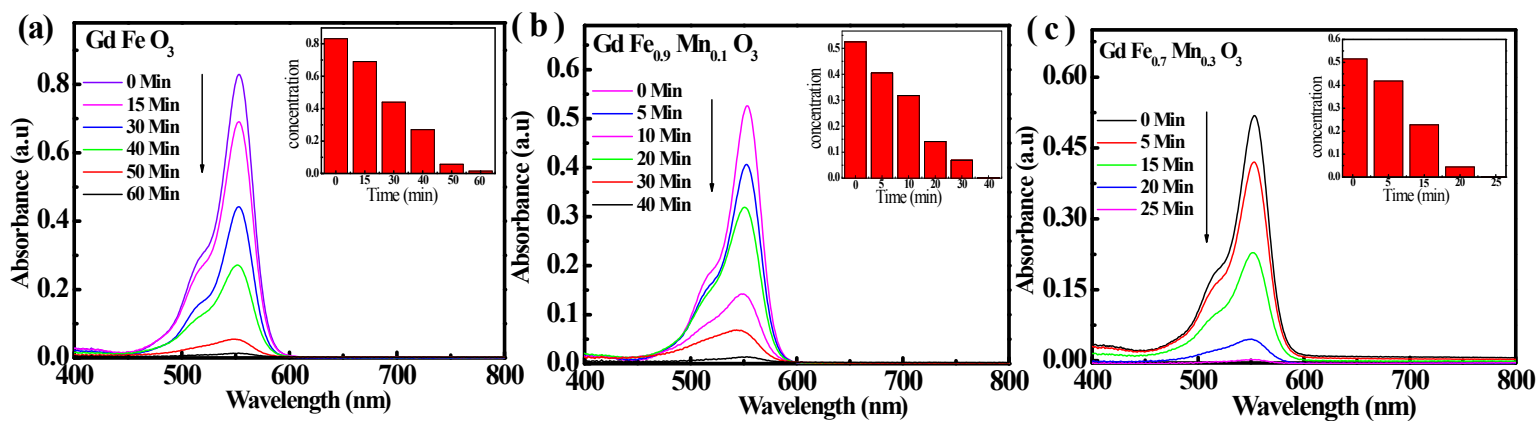
S 3. X-ray photoemission spectra of (a)  $\text{GdFeO}_3$  (b)  $\text{GdFe}_{0.9}\text{Mn}_{0.1}\text{O}_3$  & (c)  $\text{GdFe}_{0.7}\text{Mn}_{0.3}\text{O}_3$ .



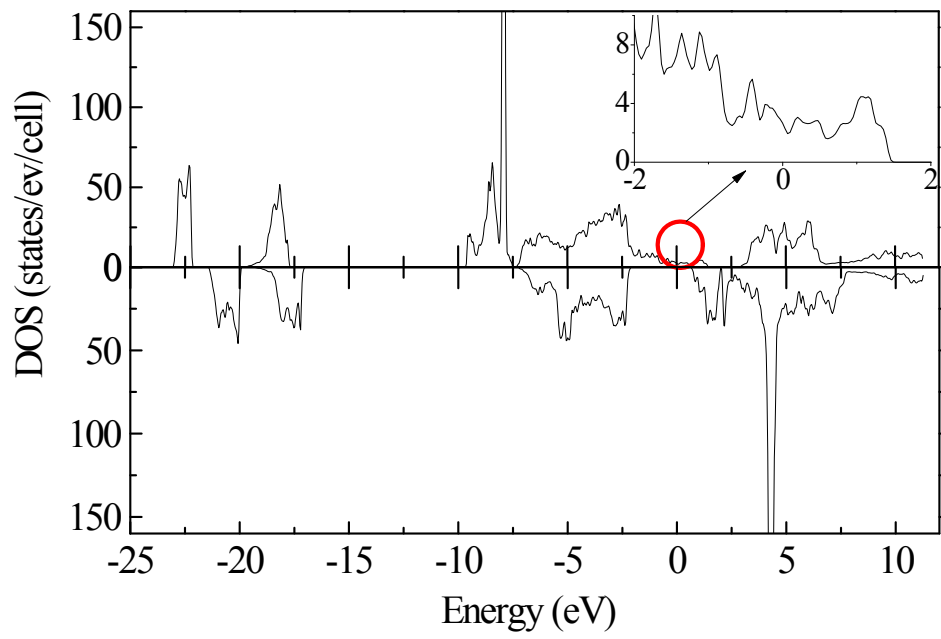
S 4. Core level XPS of Gd-3d (a)  $\text{GdFeO}_3$  (b)  $\text{GdFe}_{0.9}\text{Mn}_{0.1}\text{O}_3$  & (c)  $\text{GdFe}_{0.7}\text{Mn}_{0.3}\text{O}_3$ .



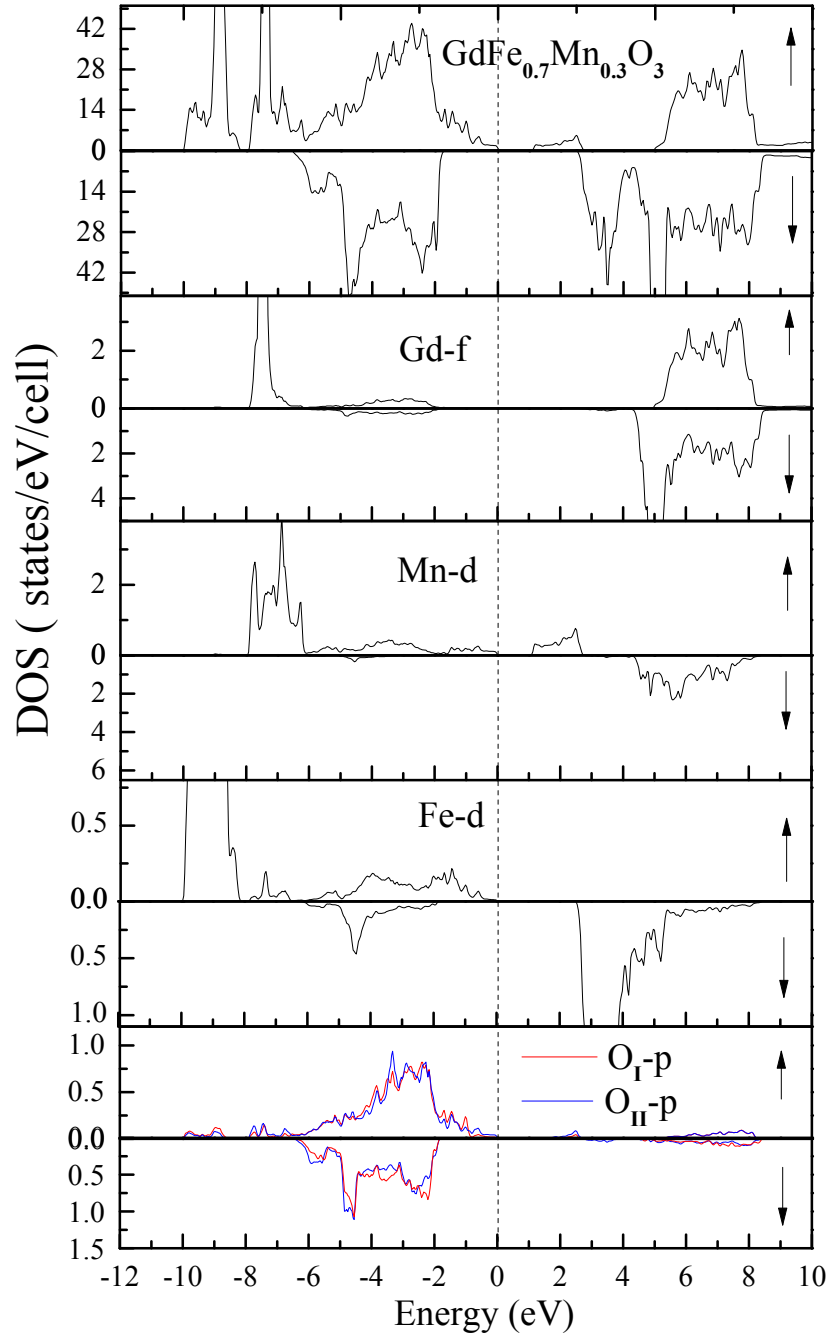
S 5. Core level XPS of Gd-4d (a)  $\text{GdFeO}_3$  (b)  $\text{GdFe}_{0.9}\text{Mn}_{0.1}\text{O}_3$  & (c)  $\text{GdFe}_{0.7}\text{Mn}_{0.3}\text{O}_3$ .



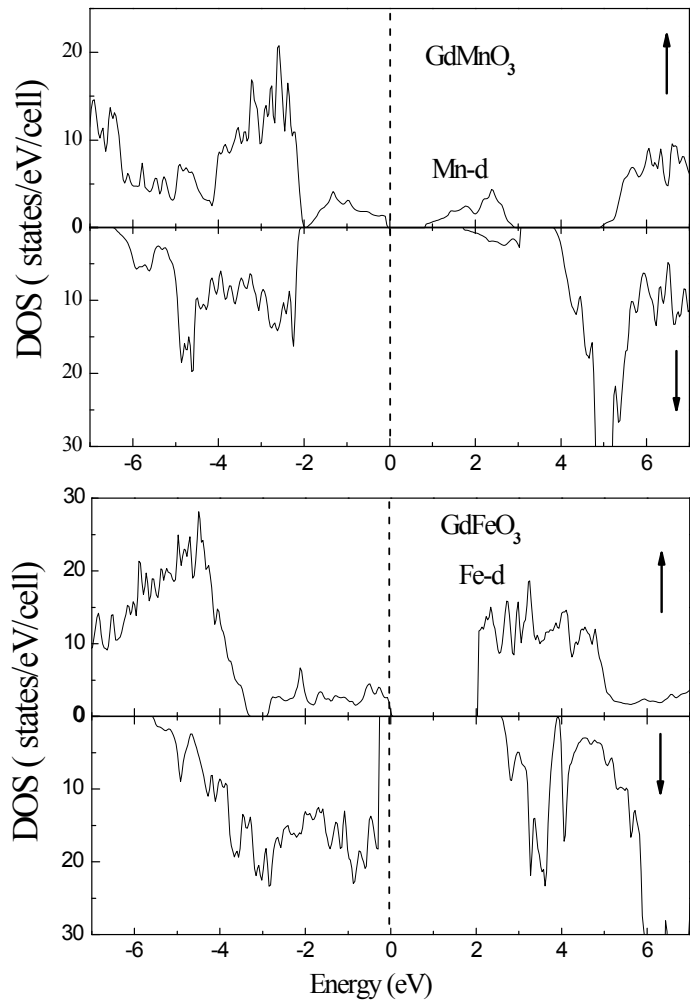
S 6. Absorption spectra of Rh-B solution in visible-light-induced photocatalytic process for (a)  $\text{GdFeO}_3$  (b)  $\text{GdFe}_{0.9}\text{Mn}_{0.1}\text{O}_3$  & (c)  $\text{GdFe}_{0.7}\text{Mn}_{0.3}\text{O}_3$ .



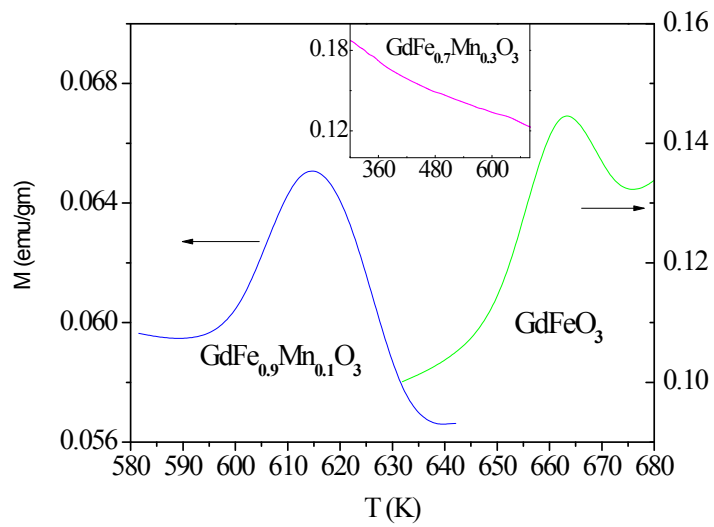
S 7. Spin-polarized DOS spectra of GdFe<sub>0.7</sub>Mn<sub>0.3</sub>O<sub>3</sub> obtained by GGA scheme



S 8. Spin-polarized DOS and PDOS spectra of  $\text{GdFe}_{0.7}\text{Mn}_{0.3}\text{O}_3$  obtained by GGA+U scheme



S 9. Spin-polarized DOS of GdMnO<sub>3</sub> and GdFeO<sub>3</sub> obtained by mBJ+U scheme



**S 10.** Variation of magnetization as a function of temperature.

**Table ST1** Comparison table of GdFeO<sub>3</sub> based catalysts for organic pollutant degradation.

Catalyst	Synthesis Method	Calcination Temp.	Particle Size (nm)	Dye Used	Band gap (eV)	Degradation efficiency
GdFeO <sub>3</sub> [Ref. 100]	Sol Gel	900°C	21 nm	Turquoise Blue KGL, Brilliant Blue KGR, Brilliant Red X-3B, Brilliant Orange K3N	-	74.9% in 120 min. 79.9% in 120 min. 96.5% in 120 min. 98.2 in 120 min.
GdFO <sub>3</sub> /Carbon nanotubes [Ref. 101]	Precipitation	600°C	Below 100 nm	Methyl Orange	1.9	94% in 70 min.
GdFeO <sub>3</sub> (Nano crystalline) [Ref. 102]	microwave synthesis		80 nm	Methyl Orange	2.1	70% in 120 min.
GdFeO <sub>3</sub> nanoparticles [Ref. 103]	glycol-assisted sol-gel technique	800°C	80 nm	Rhodamine B	2.4	100% in 150 min.
GdFeO <sub>3</sub> [Ref. 104]	Microwave-assisted synthesis	NA	44–57 nm	Rhodamine B	2.1	90% in 180 min.
BiFeO <sub>3</sub> /GdFeO <sub>3</sub> nanocomposite [Ref. 105]	Sol-Gel	500°C	14–35 nm	Methylene Blue	1.8	56% in 180 min.
GdFeO <sub>3</sub> microspheres [Ref. 106]	Hydrothermal	800°C	50–150 nm	4-CP	2.02	85% in 300 min.
GdFe <sub>0.7</sub> Mn <sub>0.3</sub> O <sub>3</sub> [Present work]	Sol-gel	800°C	80 nm	Rhodamine B	1.72	99% in 25 min.