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Supporting Information

Molecular Insights into the Electrowetting Behavior of Aqueous Ionic Liquid

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Figure S1: (a) Average hydrogen bond profile along the z-axis. (b) Dipole orientation of water at the solid-fluid interface on the application of the electric field.



Figure S2: Snapshot of aqueous [EMIM][BF₄] drop at the various electric fields.



Figure S3: Droplet deformation of 10 wt.-% aqueous [EMIM][BF₄] under application of an external electric field E = 0.08 V/Å.



Figure S4: Snapshot of aqueous 50 wt-% [EMIM][BF₄] drop at the various positive electric fields.



Figure S5: Configurational snapshot of 30 wt.-% of $[EMIM][BF_4]$ at -0.09 V/Å a) Aqueous droplet b) Cations c) Anions

a. 10 wt.%	b. 30 wt.%	c. 50 wt.%
0.03V/Å	0.03V/Å	0.03V/Å
0.05V/Å	0.05V/Å	0.05V/Å
0.07V/Å	0.07V/Å	0.07V/Å
0.08V/Å	0.08V/Å	0.08V/Å
\mathbf{A}		

Figure S6: Snapshot of aqueous IL [EMIM][NTF₂] droplet at the various electric fields (can be moved to SI)



Figure S7. Snapshot of the first layer (near the surface) of aqueous 30 wt.% [EMIM][NTF₂] ILs droplet with a) E=0 V/Å and b) 0.07 V/Å, representing the distribution and orientation of molecules near graphene surface at the three-phase contact line.



Figure S8: Snapshot of 30 wt.-% [EMIM][NTF2] at -0.08 V/Å and -0.09 V/Å



igure S9. Density profiles of 30wt% a.) [EMIM][NTF₂]. b) [EMIM][BF₄] at ± 0.05 V/Å



Figure S10: Hydrogen bond (HB) geometric criteria (a) for water-water (b) for BF₄-water (c) for NTF₂-water



Figure S11: Hydrogen bond profile of 50 wt. % aqueous IL droplet at different electric fields



Figure S12: Interaction energy between a) [EMIM] –[Water] & [BF₄]-[Water] b) [EMIM] [Water] and [NTF₂]-[Water]