

Unusual binary aggregates of perylene bisimide revealed by their electronic transitions in helium nanodroplets and DFT calculations

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Supplementary Information

Table of Contents

Figure S1. Vibrational motions of the important fundamental bands.....	S2
Table S1. Summary of calculations results of all vibronic simulations of the PBI dimers.....	S3
Figure S2. Simulated vibronic spectra of some PBI dimers.....	S4
Completion of references 31 and 32.....	S4

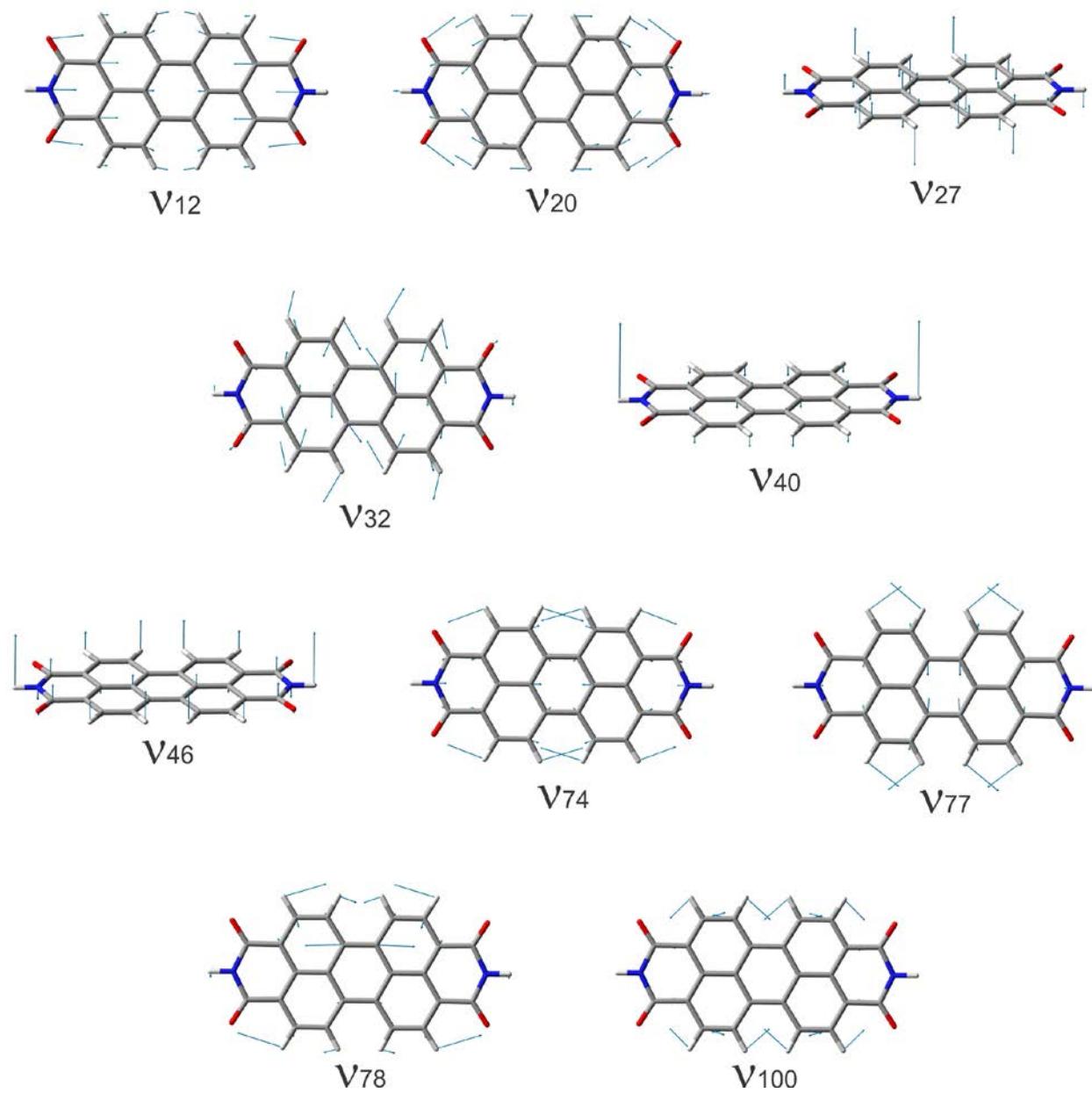


Figure S1. Vibrational motions of the ten fundamental bands which contribute significantly to the $S_0 \rightarrow S_1$ transition calculated at the CAM-B3LYP-D3BJ/6-311++G(2d,p) level.

Table S1. Calculated properties of the PBI monomer and dimers at the three levels of theory.^a

APFD/6-311++G(2d,p)						
PBI monomer gs \rightarrow ex1 AH 0-0 (cm ⁻¹)			18137			
PBI dimers	I $\tau_s(0)$	I $\tau_s(70)$	I $\tau_s(90)$	I $\tau_s(90)$ _shifted	II $\tau_s(90)$	III $\tau_s(90)$
E_b (kcal/mol) ^b	9.04/8.75	9.82/9.44	9.78/9.27	11.29/10.53	10.84	/31.07
ΔE (kcal/mol) ^c	2.25/1.78	1.47/1.09	1.51/1.26	0.00/0.00	0.45	
top \rightarrow top AH 0-0 (cm ⁻¹) ^d	18058/18056	/18035	18024/18019	/17963	18178	
$\Delta(0-0)$ (cm ⁻¹) ^e	-79/-80	/-102	-113/-118	/-174	+41	
bot \rightarrow bot AH 0-0 (cm ⁻¹) ^d			17776/		18199	
$\Delta(0-0)$ (cm ⁻¹) ^e			-361/		+62	
B3LYP+GD3BJ/6-311++G(2d,p)						
PBI monomer gs \rightarrow ex1 AH 0-0 (cm ⁻¹)			17670			
PBI dimers	I $\tau_s(0)$	I $\tau_s(70)$	I $\tau_s(90)$	I $\tau_s(90)$ _shifted	II $\tau_s(90)$	III $\tau_s(90)$
E_b (kcal/mol) ^b	7.03/6.74		7.56/7.17	8.86/		/24.34
ΔE (kcal/mol) ^c						-
top \rightarrow top AH 0-0 (cm ⁻¹) ^d	17591/17593		17573/17566			/17876
$\Delta(0-0)$ (cm ⁻¹) ^e	-79/-77		-97/-104			/+206
bot \rightarrow bot AH 0-0 (cm ⁻¹) ^d						
$\Delta(0-0)$ (cm ⁻¹) ^e						
APFD/6-31G(d)						
PBI monomer gs \rightarrow ex1 AH 0-0 (cm ⁻¹)			18745			
PBI dimers	I $\tau_s(0)$	I $\tau_s(70)$	I $\tau_s(90)$	I $\tau_s(90)$ _shifted	II $\tau_s(90)$	III $\tau_s(90)$
E_b (kcal/mol) ^b	7.37	7.53	7.56	8.67	10.01	29.45
ΔE (kcal/mol) ^c	1.30	1.14	1.11	0.00		-
top \rightarrow top AH 0-0 (cm ⁻¹) ^d	18665	18498	18628	18560		18709
$\Delta(0-0)$ (cm ⁻¹) ^e	-80	-247	-117	-185		-36
bot \rightarrow bot AH 0-0 (cm ⁻¹) ^d						
$\Delta(0-0)$ (cm ⁻¹) ^e						

^a Black colour indicates that the PBI dimers are fully relaxed, the bottom PBI ring is slightly bowed. The red colour indicates that the bottom PBI subunit is constrained as planar.

^b The binding energy of the PBI dimer with the ZPE and BSSE correction.

^c Relative energies compared to the most stable Type I PBI dimer calculated. Positive values mean less stable.

^d See main text for the definition of top \rightarrow top and bot \rightarrow bot.

^e The shift of the 0-0 band origin of the PBI dimer from that of the PBI monomer.

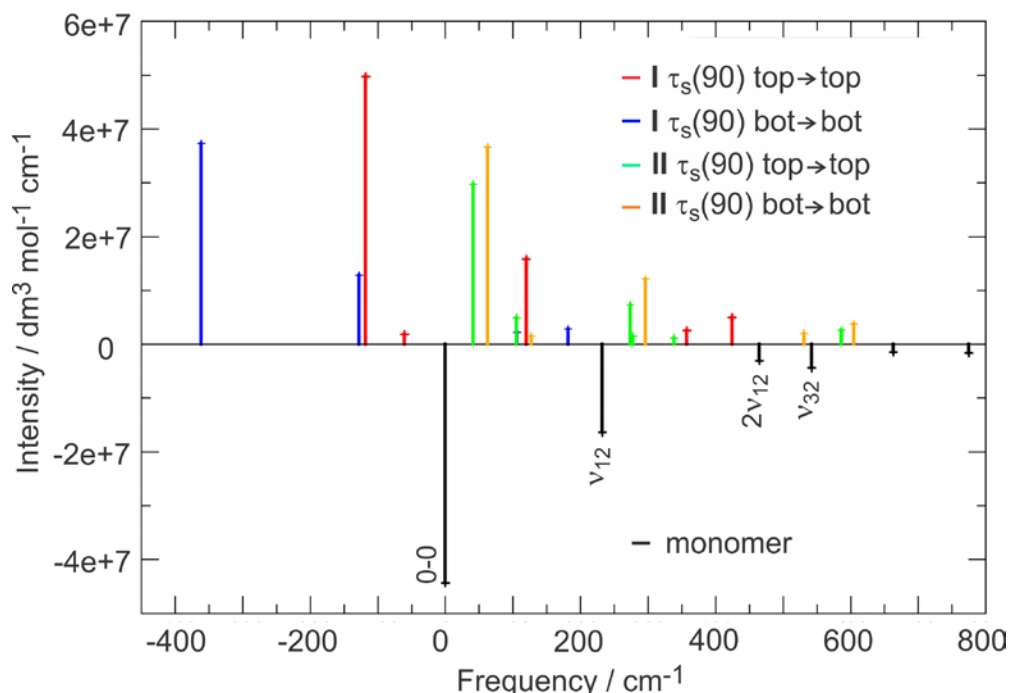


Figure S2. Simulated vibronic spectra of the top → top and bot → bot electronic transitions of **I** $\tau_s(90)$ and **II** $\tau_s(90)$ PBI dimer structures (fully optimized) in comparison to that of the PBI monomer at the APFD/6-311++G(2d,p) level of theory.

Completion of references 31 and 32.

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