

Electronic Supplementary Information

Spectroscopic Characterization and Photochemistry of the Vinylsulfinyl Radical

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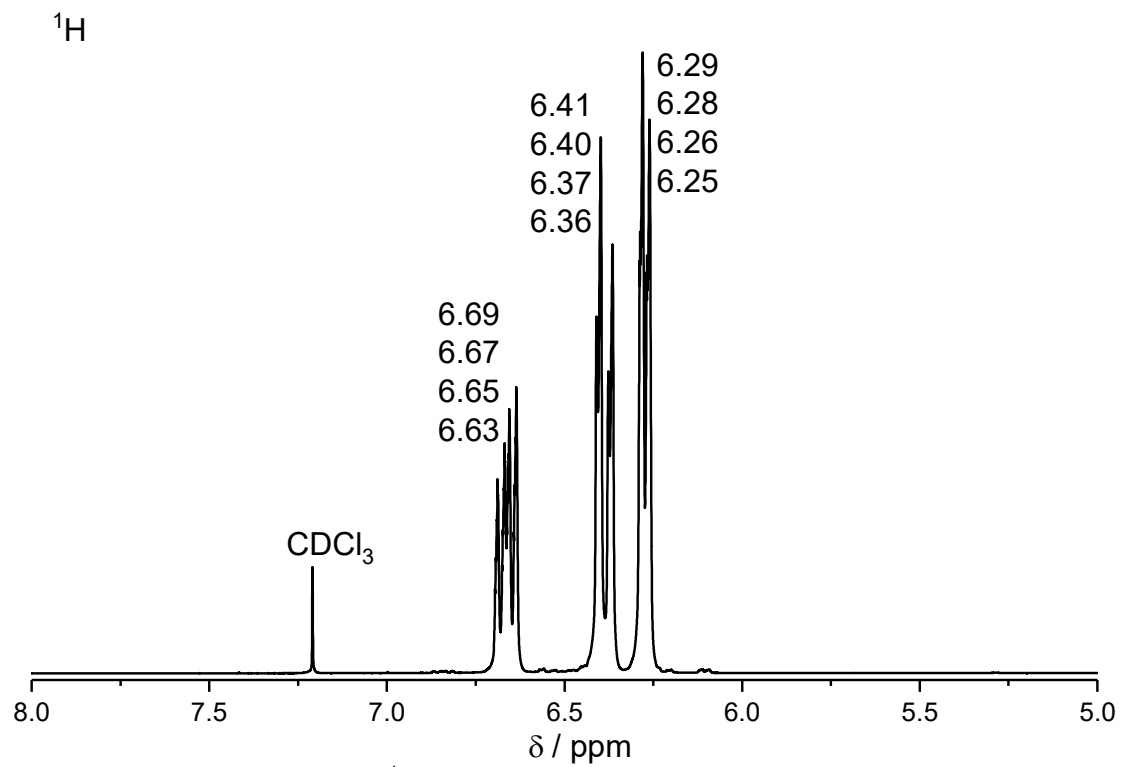


Figure S1. ^1H NMR of $\text{CH}_2=\text{C}(\text{H})\text{S}(\text{O})\text{CF}_3$ in CDCl_3 .

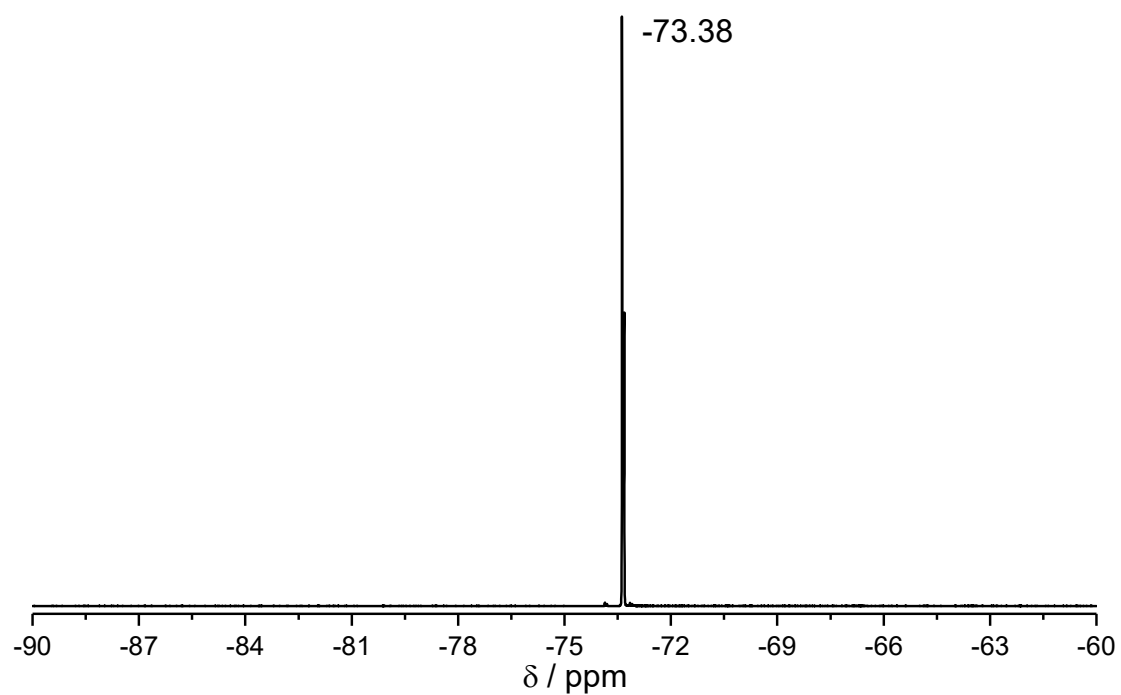


Figure S2. ^{19}F NMR of $\text{CH}_2=\text{C}(\text{H})\text{S}(\text{O})\text{CF}_3$ in CDCl_3 .

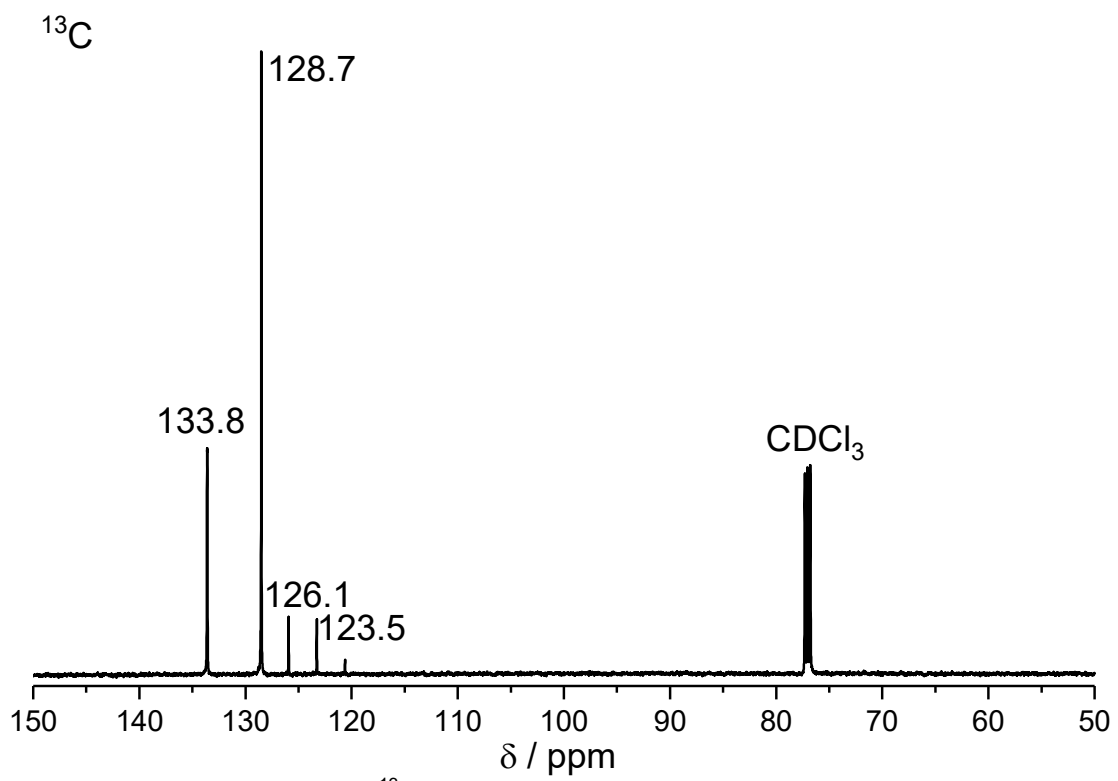


Figure S3. ¹³C NMR of CH₂=C(H)S(O)CF₃ in CDCl₃.

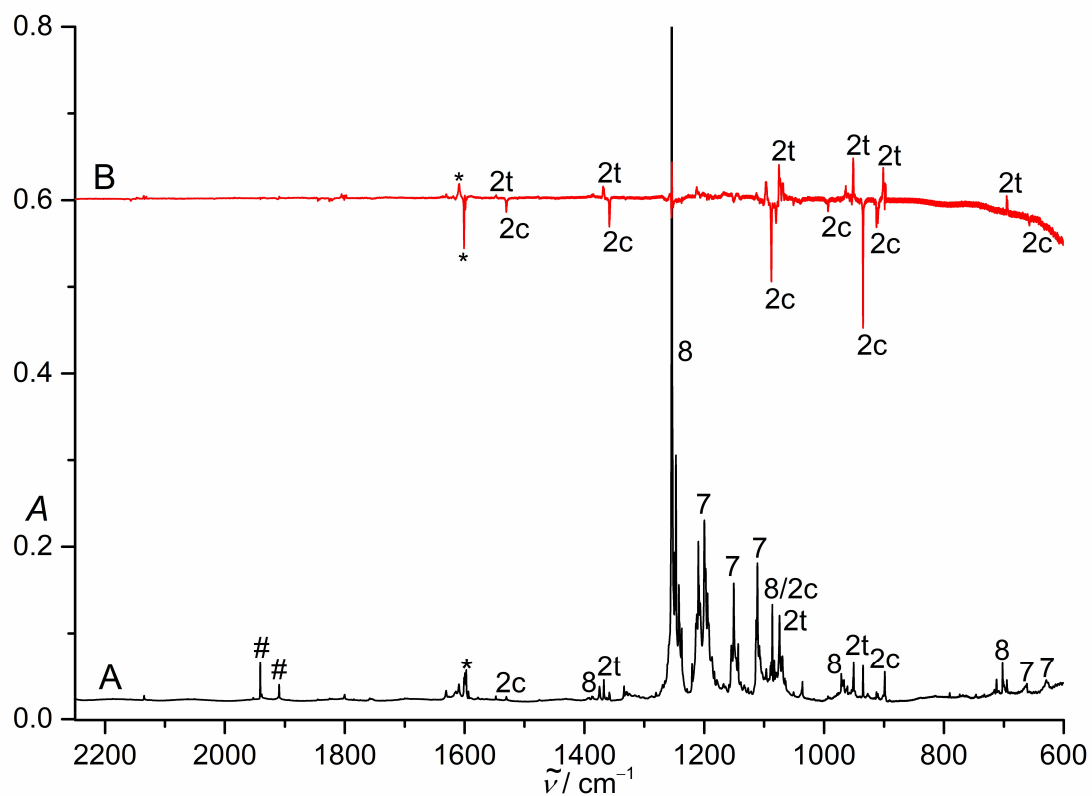


Figure S4. (A) IR spectrum of the high-vacuum flash pyrolysis (HVFP, 800 K) products of $\text{CH}_2=\text{C}(\text{H})\text{S}(\text{O})\text{CF}_3$ (**7**) in solid Ne (**7** : Ne = 1:1000, 150 min deposition) at 2.8 K; (B) IR difference spectrum reflecting the conversion of *cis*- $\text{CH}_2=\text{C}(\text{H})\text{SO}\cdot$ (**2c**) to *trans*- $\text{CH}_2=\text{C}(\text{H})\text{SO}\cdot$ (**2t**) upon irradiation with an UV-light LED (365 nm, 10 min). The IR bands of $\cdot\text{CF}_3$ (**8**), CF_2O (**#**), and H_2O (*) are also labeled.

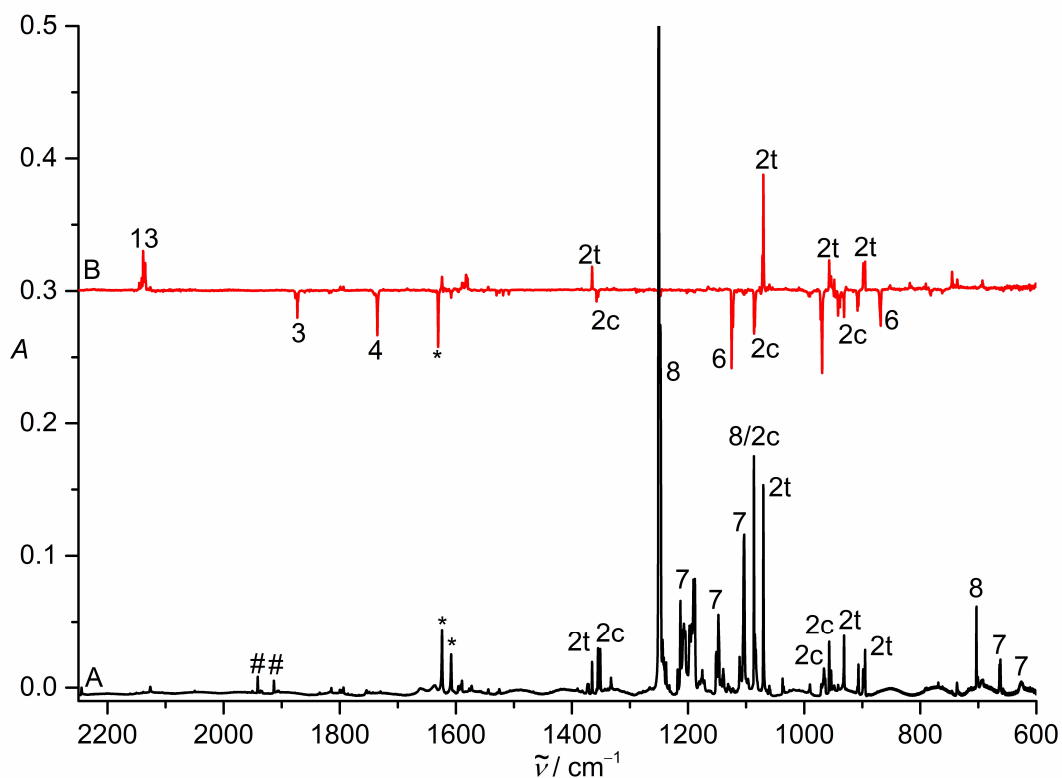


Figure S5. (A) IR spectrum of the high-vacuum flash pyrolysis (HVFP, 800 K) products of $\text{CH}_2=\text{C}(\text{H})\text{S}(\text{O})\text{CF}_3$ (**7**) in solid Ar (**7** : Ar = 1:1000, 150 min deposition) at 10.0 K; (B) IR difference spectrum reflecting the conversion of *cis*- $\text{CH}_2=\text{C}(\text{H})\text{SO}\cdot$ (**2c**) to *trans*- $\text{CH}_2=\text{C}(\text{H})\text{SO}\cdot$ (**2t**) upon irradiation with a UV-light LED (365 nm, 10 min). The IR bands of $\text{CH}_3\text{SCO}\cdot$ (**3**), $\cdot\text{CH}_2\text{SC}(\text{O})\text{H}$ (**4**), $\text{CH}_3\text{C}(\text{O})\text{S}\cdot$ (**6**), $\text{CH}_2=\text{C}(\text{H})\text{S}(\text{O})\text{CF}_3$ (**7**), $\cdot\text{CF}_3$ (**8**), $\text{CH}_3\text{S}\cdot\cdots\text{CO}$ (**9**), CO (**13**), CF_2O (#), and H_2O (*) are also labeled.

Table S1. Calculated anharmonic frequencies of CH₂=C(H)SO• isomers at B3LYP/6 311++G(3df,3pd) level.

<i>trans</i> -CH ₃ SCO•	<i>cis</i> -CH ₃ SCO•	<i>trans</i> -•CH ₂ SC(O)H	<i>cis</i> -•CH ₂ SC(O)H
3016 (3)	3008 (3)	3136 (0.1)	3138 (2)
3006 (3)	2995 (3)	3026 (2)	3026 (0.8)
2956 (16)	2941 (9)	2775 (49)	2812 (51)
1844 (401)	1822 (308)	1744 (431)	1717 (328)
1438 (13)	1437 (4)	1378 (0.2)	1356 (4)
1433 (6)	1414 (7)	1319 (12)	1327 (6)
1328 (2)	1308 (5)	934 (15)	931 (7)
972 (6)	942 (2)	904 (2)	881 (3)
965 (3)	941 (13)	812 (5)	807 (2)
688 (1)	642 (2)	732 (105)	709 (116)
562 (66)	557 (54)	510 (150)	513 (49)
<i>trans</i> -CH ₃ CSO•	<i>cis</i> -CH ₃ CSO•	<i>trans</i> -•CH ₂ C(O)SH	<i>cis</i> -•CH ₂ C(O)SH
2911 (2)	2923 (2)	3118 (1)	3120 (1)
2895 (5)	2906 (3)	3022 (1)	3023 (0.3)
2864 (5)	2864 (0.2)	2552 (1)	2685 (2)
1416 (9)	1407 (9)	1597 (150)	1611 (163)
1400 (10)	1396 (11)	1424 (8)	1422 (12)
1359 (1)	1350 (6)	1154 (115)	1144 (100)
1330 (4)	1339 (0.5)	1015 (12)	1002 (10)
1102 (176)	1046 (135)	878 (4)	870 (8)
991 (1)	992 (0)	745 (27)	746 (27)
971 (1)	966 (2)	611 (20)	605 (66)
722 (7)	703 (8)	534 (1)	534 (1)
<i>trans</i> -CH ₃ OCS•	<i>cis</i> -CH ₃ OCS•	CH ₃ C(O)S•	CH ₃ S•••CO
3014 (8)	3033 (8)	3133 (2)	2913 (106)
2996 (12)	3007 (10)	3105 (1)	2886 (12)
2960 (43)	2967 (16)	3039 (0.2)	2886 (12)
1462 (425)	1464 (156)	1682 (119)	2198 (81)
1454 (1159)	1443 (4)	1466 (13)	1478 (97)
1447 (219)	1440 (12)	1458 (20)	1301 (38)
1339 (458)	1331 (321)	1388 (24)	1297 (20)
1169 (5)	1147 (0.2)	1153 (102)	
1140 (1)	1116 (1)	1027 (1)	
996 (68)	903 (0.6)	933 (66)	
878 (6)	788 (26)	632 (33)	
465 (19)	602 (3)	500 (1)	

Table S2. TD-DFT B3LYP/6-311++G(3df,3pd) calculated vertical transitions (> 190 nm) of CH₂=C(H)SO• isomers.

<i>trans</i> -CH ₂ =C(H)SO•		<i>cis</i> -CH ₂ =C(H)SO•		<i>trans</i> -•CH ₂ C(O)SH		<i>cis</i> -•CH ₂ C(O)SH	
Energy (nm)	oscillator strength	Energy (nm)	oscillator strength	Energy (nm)	oscillator strength	Energy (nm)	oscillator strength
540	0.0007	520	0.0006	710	0.0001	694	0.0001
389	0.0112	368	0.0065	515	0.0154	522	0.0139
269	0.1082	282	0.1622	237	0.0001	265	0.0006
265	0.0003	275	0.0009	235	0.0032	237	0.0001
235	0.0006	232	0.0002	226	0.0002	235	0.0024
228	0.0001	229	0.0148	221	0.001	220	0.0017
225	0.0538	228	0.0002	218	0.0005	215	0.0194
217	0.004	214	0.0019	211	0.0356	211	0.0178
213	0.0173	213	0.0013			209	0.0007

<i>trans</i> -CH ₃ SCO•		<i>cis</i> -CH ₃ SCO•		<i>trans</i> -•CH ₂ SC(O)H		<i>cis</i> -•CH ₂ SC(O)H	
Energy (nm)	oscillator strength	Energy (nm)	oscillator strength	Energy (nm)	oscillator strength	Energy (nm)	oscillator strength
445	0.0005	424	0.001	378	0.0273	412	0.0211
387	0.0007	364	0.0005	361	0.0001	305	0.0026
279	0.0177	284	0.0003	307	0.0041	283	0.0001
265	0.0007	259	0.0012	270	0.0803	275	0.0025
249	0.0044	240	0.0045	234	0.0048	260	0.1035
229	0.0284	235	0.0066	231	0.0001	241	0.0005
227	0.0011	218	0.002	226	0.0045	233	0.0022
214	0.0262	207	0.0112	218	0.008	222	0.0041
212	0.0146						

CH ₃ C(O)S•		CH ₃ S•···CO	
Energy (nm)	oscillator strength	Energy (nm)	oscillator strength
870	0.0060	326	0.002
284	0.0011	275	0.0006
268	0.0001	257	0.0007
258	0.0033	244	0.0003
219	0.0691	228	0.0043
198	0.0012	216	0.0022
		213	0.0013

Calculated atomic coordinates (in Angstroms) and energies (in Hartree) for the isomers of CH₂=C(H)SO• and the transition states with the B3LYP/6-311++G(3df,3pd) method.

***trans*-CH₂=C(H)SO• (2t)**

C	1.32715100	-1.60645400	0.00000000
H	1.39001200	-2.68529900	0.00000000
H	2.26133300	-1.05910100	0.00000000
C	0.14354200	-0.98574700	0.00000000
H	-0.80494600	-1.51046700	0.00000000
S	0.00000000	0.75453600	0.00000000
O	-1.45881900	1.09193700	0.00000000
Zero-point correction=		0.045252 (Hartree/Particle)	
Thermal correction to Energy=		0.050049	
Thermal correction to Enthalpy=		0.050993	
Thermal correction to Gibbs Free Energy=		0.016934	
Sum of electronic and zero-point Energies=		-551.417021	
Sum of electronic and thermal Energies=		-551.412224	
Sum of electronic and thermal Enthalpies=		-551.411280	
Sum of electronic and thermal Free Energies=		-551.445340	

***cis*-CH₂=C(H)SO• (2c)**

C	-1.76847300	0.50693800	-0.00000600
H	-2.84342400	0.39576900	-0.00001400
H	-1.35729500	1.50718200	0.00001600
C	-0.96494100	-0.56135900	-0.00002200
H	-1.33467100	-1.58102200	0.00018800
S	0.77519200	-0.47795300	-0.00001100
O	1.19160100	0.95648000	0.00001800
Zero-point correction=		0.045544 (Hartree/Particle)	
Thermal correction to Energy=		0.050132	
Thermal correction to Enthalpy=		0.051077	
Thermal correction to Gibbs Free Energy=		0.017510	
Sum of electronic and zero-point Energies=		-551.418327	
Sum of electronic and thermal Energies=		-551.413739	
Sum of electronic and thermal Enthalpies=		-551.412794	
Sum of electronic and thermal Free Energies=		-551.446361	

***trans*-•CH₂C(O)SH (5t)**

C	1.30438400	-1.05095900	-0.00004200
H	0.85025600	-2.03056900	0.00056300
H	2.37984400	-0.95329500	-0.00017300
S	-1.27022300	-0.17998200	-0.00003400
H	-1.61103900	1.11823700	0.00034800
C	0.51323400	0.14774000	-0.00002200
O	0.97485000	1.27058100	0.00002400
Zero-point correction=		0.042431 (Hartree/Particle)	
Thermal correction to Energy=		0.047256	
Thermal correction to Enthalpy=		0.048200	
Thermal correction to Gibbs Free Energy=		0.014297	
Sum of electronic and zero-point Energies=		-551.427692	
Sum of electronic and thermal Energies=		-551.422866	
Sum of electronic and thermal Enthalpies=		-551.421922	
Sum of electronic and thermal Free Energies=		-551.455826	

***cis*-•CH₂C(O)SH (5c)**

C	-1.28545200	1.07419200	-0.00001200
H	-0.83482300	2.05388000	-0.00003800
H	-2.36173500	0.97843500	-0.00007300
S	1.29235700	-0.00112000	-0.00000700
H	1.38089200	1.33808500	-0.00010500

C	-0.51286400	-0.13747000	0.00012800
O	-1.00901800	-1.24660200	-0.00004700
Zero-point correction=		0.042343	(Hartree/Particle)
Thermal correction to Energy=		0.047224	
Thermal correction to Enthalpy=		0.048168	
Thermal correction to Gibbs Free Energy=		0.014147	
Sum of electronic and zero-point Energies=		-551.425312	
Sum of electronic and thermal Energies=		-551.420432	
Sum of electronic and thermal Enthalpies=		-551.419487	
Sum of electronic and thermal Free Energies=		-551.453508	

***trans*-CH₃SCO• (3t)**

C	-1.74260400	0.59095900	-0.00000600
H	-2.67036000	0.02345300	-0.00113900
H	-1.69410100	1.20508900	0.89382500
H	-1.69271300	1.20638600	-0.89286400
S	-0.40719500	-0.64344000	0.00000600
C	0.98913500	0.44810700	0.00006300
O	2.13663800	0.20321400	-0.00003400
Zero-point correction=		0.045054	(Hartree/Particle)
Thermal correction to Energy=		0.050338	
Thermal correction to Enthalpy=		0.051282	
Thermal correction to Gibbs Free Energy=		0.015928	
Sum of electronic and zero-point Energies=		-551.431308	
Sum of electronic and thermal Energies=		-551.426024	
Sum of electronic and thermal Enthalpies=		-551.425080	
Sum of electronic and thermal Free Energies=		-551.460433	

***cis*-CH₃SCO• (3c)**

C	1.60374200	0.03570200	0.00000000
H	2.37274700	0.80495000	0.00000000
H	1.69209300	-0.57667700	0.89233600
H	1.69209300	-0.57667700	-0.89233600
S	0.00000000	0.92018100	0.00000000
C	-1.17002600	-0.37246500	0.00000000
O	-1.04490400	-1.54424000	0.00000000
Zero-point correction=		0.044905	(Hartree/Particle)
Thermal correction to Energy=		0.050165	
Thermal correction to Enthalpy=		0.051109	
Thermal correction to Gibbs Free Energy=		0.015661	
Sum of electronic and zero-point Energies=		-551.429378	
Sum of electronic and thermal Energies=		-551.424118	
Sum of electronic and thermal Enthalpies=		-551.423173	
Sum of electronic and thermal Free Energies=		-551.458621	

***trans*-•CH₂SC(O)H (4t)**

C	1.76790400	0.54455000	0.00003100
H	1.57174000	1.60582600	-0.00168800
H	2.77612500	0.16737500	0.00140100
S	0.49783500	-0.59247400	-0.00007700
H	-0.69925600	1.52922400	0.00092900
C	-0.94972000	0.45319700	0.00008300
O	-2.06538400	0.02383400	-0.00001100
Zero-point correction=		0.042633	(Hartree/Particle)
Thermal correction to Energy=		0.048095	
Thermal correction to Enthalpy=		0.049039	
Thermal correction to Gibbs Free Energy=		0.013759	
Sum of electronic and zero-point Energies=		-551.416344	
Sum of electronic and thermal Energies=		-551.410882	
Sum of electronic and thermal Enthalpies=		-551.409937	

Sum of electronic and thermal Free Energies= -551.445218

cis-•CH₂SC(O)H (4c)

C	-1.28545200	1.07419200	-0.00001200
H	-0.83482300	2.05388000	-0.00003800
H	-2.36173500	0.97843500	-0.00007300
S	1.29235700	-0.00112000	-0.00000700
H	1.38089200	1.33808500	-0.00010500
C	-0.51286400	-0.13747000	0.00012800
O	-1.00901800	-1.24660200	-0.00004700
Zero-point correction=	0.042343 (Hartree/Particle)		
Thermal correction to Energy=	0.047224		
Thermal correction to Enthalpy=	0.048168		
Thermal correction to Gibbs Free Energy=	0.014147		
Sum of electronic and zero-point Energies=	-551.425312		
Sum of electronic and thermal Energies=	-551.420432		
Sum of electronic and thermal Enthalpies=	-551.419487		
Sum of electronic and thermal Free Energies=	-551.453508		

CH₃C(O)S• (6)

C	1.45278000	-0.85177400	-0.00001100
H	1.34268600	-1.48590500	-0.88030500
H	2.43450100	-0.38271000	-0.00145400
H	1.34441900	-1.48381200	0.88201900
C	0.39010600	0.20522300	-0.00007000
O	0.59746300	1.39844600	0.00001700
S	-1.30991400	-0.24724000	0.00000500
Zero-point correction=	0.046025 (Hartree/Particle)		
Thermal correction to Energy=	0.051026		
Thermal correction to Enthalpy=	0.051970		
Thermal correction to Gibbs Free Energy=	0.016010		
Sum of electronic and zero-point Energies=	-551.441214		
Sum of electronic and thermal Energies=	-551.436212		
Sum of electronic and thermal Enthalpies=	-551.435268		
Sum of electronic and thermal Free Energies=	-551.471228		

CH₃S•••CO (9)

C	-1.38627300	1.07440900	0.00006100
H	-1.67537800	1.61846000	0.89702000
H	-0.29646300	0.97687400	-0.00035800
H	-1.67604600	1.61880700	-0.89647200
C	2.59086400	0.15666400	-0.00032600
O	3.63753400	-0.25261000	0.00019000
S	-2.04249600	-0.59873200	-0.00000800
Zero-point correction=	0.041102 (Hartree/Particle)		
Thermal correction to Energy=	0.048521		
Thermal correction to Enthalpy=	0.049466		
Thermal correction to Gibbs Free Energy=	0.003939		
Sum of electronic and zero-point Energies=	-551.424776		
Sum of electronic and thermal Energies=	-551.417357		
Sum of electronic and thermal Enthalpies=	-551.416412		
Sum of electronic and thermal Free Energies=	-551.461939		

CH₃S•••OC (9)

C	1.36402200	1.10594800	0.02728900
H	1.58826700	1.40701400	1.05499000
H	2.09662700	1.57917600	-0.62367500
H	0.35631900	1.43916100	-0.21113000
S	1.50476000	-0.68553200	-0.02034700
O	-2.45763900	0.45078200	-0.07447700

C	-2.77339900	-0.61646300	0.08957400
Zero-point correction=		0.041000 (Hartree/Particle)	
Thermal correction to Energy=		0.048477	
Thermal correction to Enthalpy=		0.049421	
Thermal correction to Gibbs Free Energy=		0.003921	
Sum of electronic and zero-point Energies=		-551.424577	
Sum of electronic and thermal Energies=		-551.417100	
Sum of electronic and thermal Enthalpies=		-551.416156	
Sum of electronic and thermal Free Energies=		-551.461656	

trans-CH₃CSO• (10t)

C	-2.18913400	-0.07056700	0.00003200
H	-2.66464500	-0.50445800	-0.88248000
H	-2.66493300	-0.50260300	0.88317200
H	-2.39378700	1.00578300	-0.00147000
C	-0.75464500	-0.34620900	0.00009600
S	0.65110900	0.38176700	0.00003100
O	1.87103600	-0.45079300	-0.00005900
Zero-point correction=		0.044568 (Hartree/Particle)	
Thermal correction to Energy=		0.049767	
Thermal correction to Enthalpy=		0.050711	
Thermal correction to Gibbs Free Energy=		0.015811	
Sum of electronic and zero-point Energies=		-551.362344	
Sum of electronic and thermal Energies=		-551.357145	
Sum of electronic and thermal Enthalpies=		-551.356201	
Sum of electronic and thermal Free Energies=		-551.391101	

cis-CH₃CSO• (10c)

C	-2.04561800	0.28805500	0.00001500
H	-2.64254600	0.05414200	-0.88418700
H	-2.64242500	0.05421800	0.88431800
H	-1.84177200	1.36466600	-0.00007000
C	-0.79372200	-0.45571400	-0.00004800
S	0.79107500	-0.46713800	0.00001200
O	1.43819800	0.87589200	-0.00000700
Zero-point correction=		0.044529 (Hartree/Particle)	
Thermal correction to Energy=		0.049752	
Thermal correction to Enthalpy=		0.050696	
Thermal correction to Gibbs Free Energy=		0.015143	
Sum of electronic and zero-point Energies=		-551.363280	
Sum of electronic and thermal Energies=		-551.358057	
Sum of electronic and thermal Enthalpies=		-551.357112	
Sum of electronic and thermal Free Energies=		-551.392666	