

Vibrational Spectrum and Photochemistry of Phosphaketene HPCO

Supporting Information

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Computational details

Local minima were confirmed by vibrational frequencies analysis, and transition states were further confirmed by intrinsic reaction coordinate (IRC) calculations.^[1] The natural population analysis (NPA), natural spin densities, and Wiberg bond indices (WBI) were computed at the CCSD(T)/cc-pVTZ level with the Natural Bond Orbital (NBO) method.^[2] The structures and harmonic frequencies were performed with ORCA.^[3] The electrostatic potential maps were performed with Multiwfn^[4] and Gaussian 09 software,^[5] respectively.

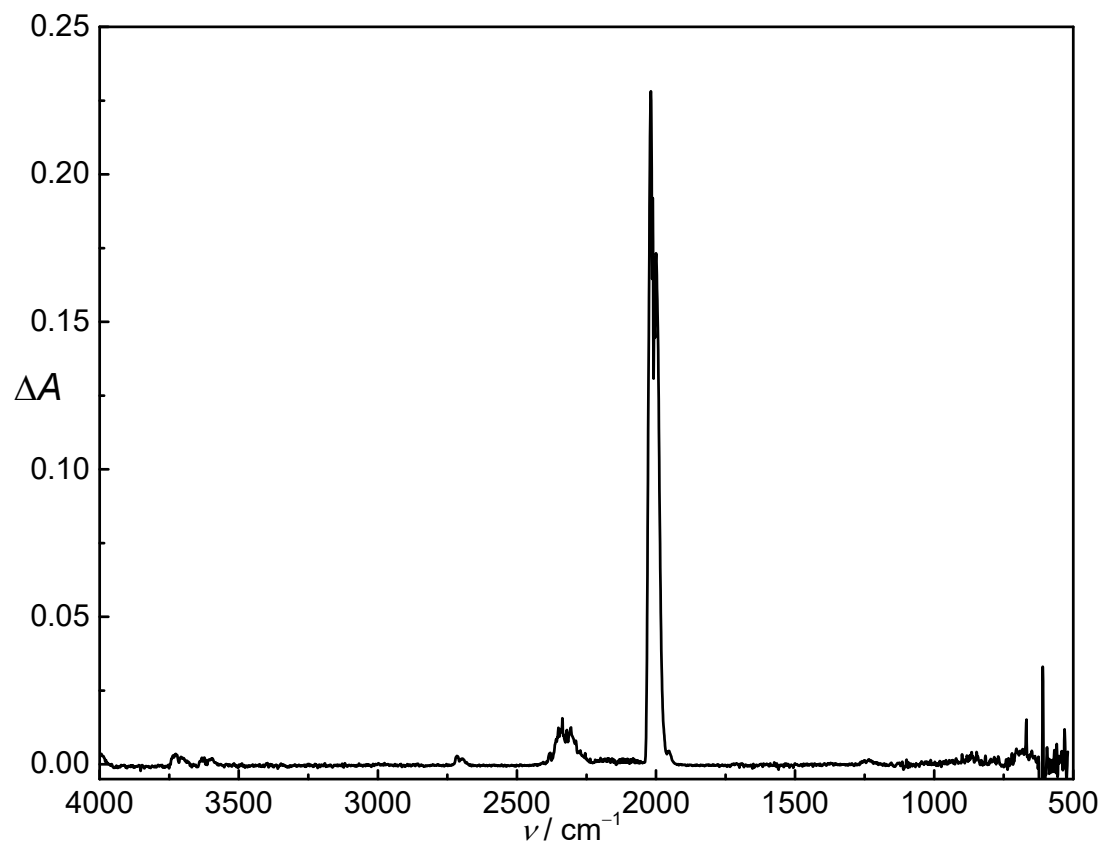


Figure S1. Gas phase IR spectrum of HPCO (130 Pa).

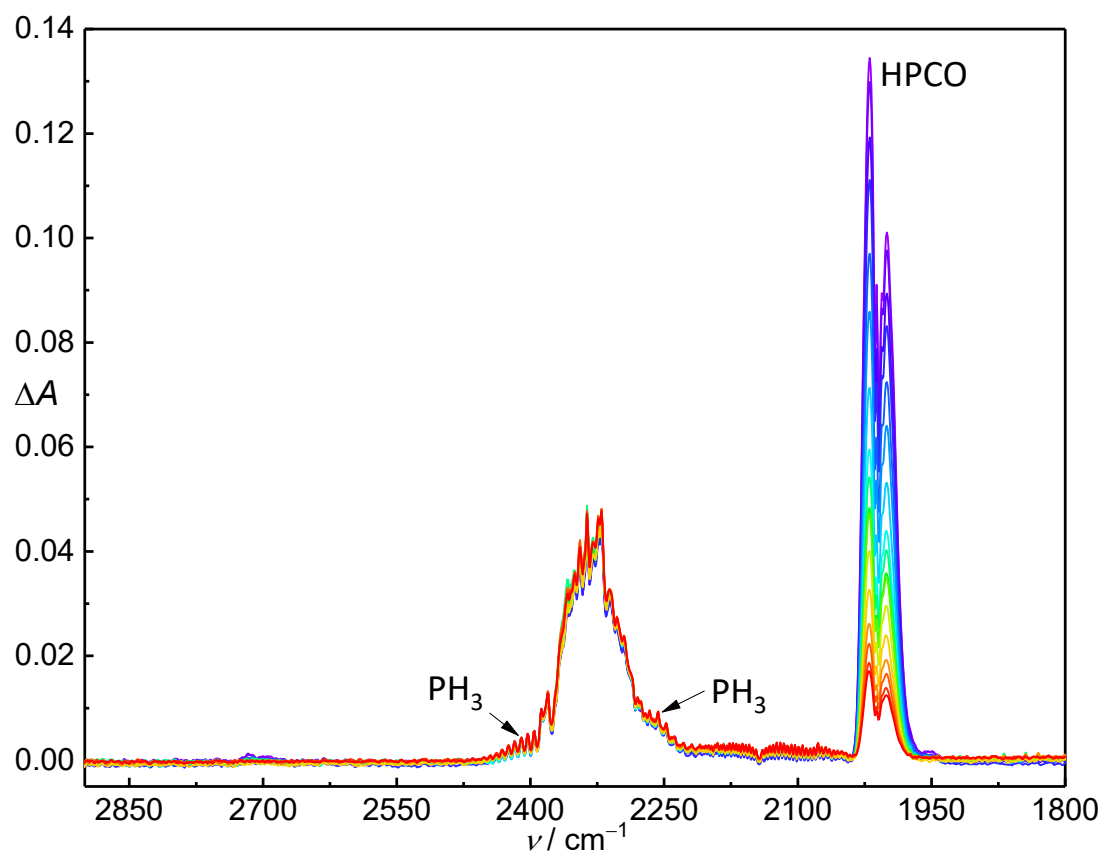


Figure S2. IR spectra reflecting the decomposition of HPCO (300.0 Pa) and formation of PH_3 in the gas phase at 26 °C over a period of 124.5 min.

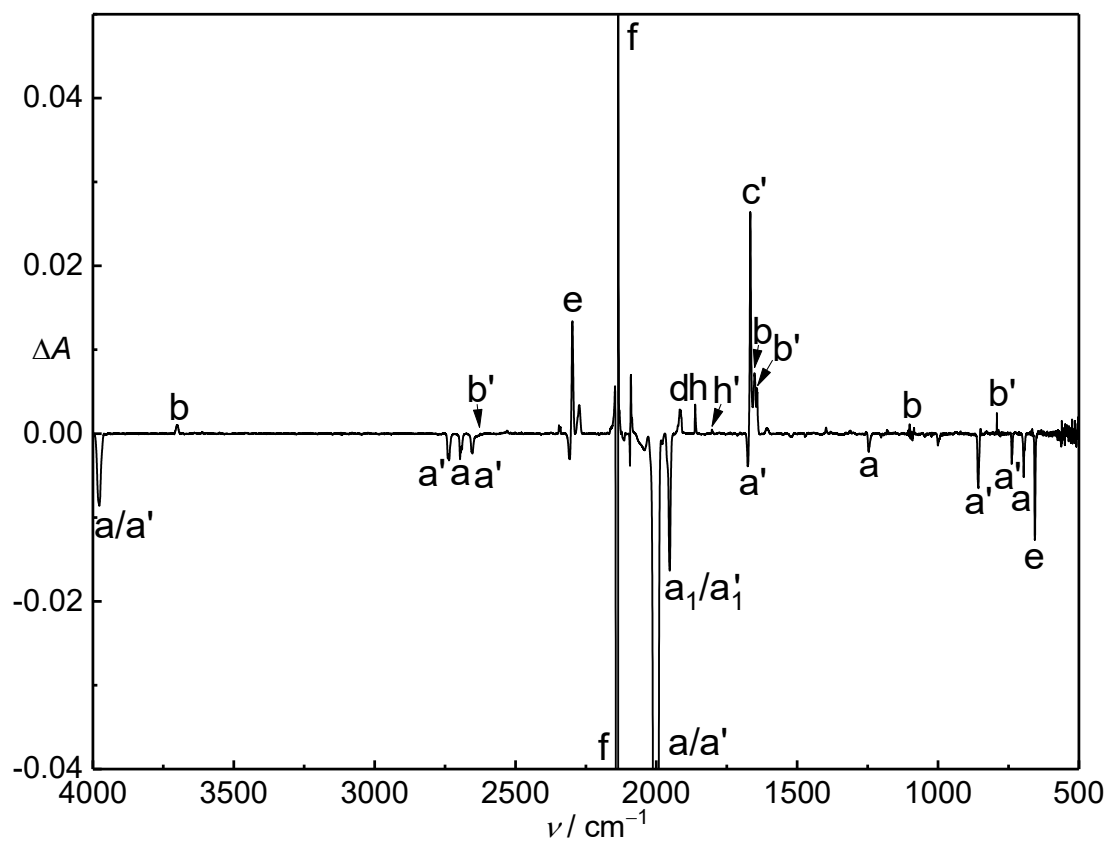


Figure S3. IR difference spectrum showing the changes of DPCO/CO/Ar=1:50:1000 upon 193 nm laser irradiation (5.0 min). The bands of HPCO (a), HP^{13}CO (a_1), DP^{13}CO (a_1'), $\text{DP}(c')$, PCO (d), CO_2 (e), CO (f), HCO (h), and DCO (h') are labeled.

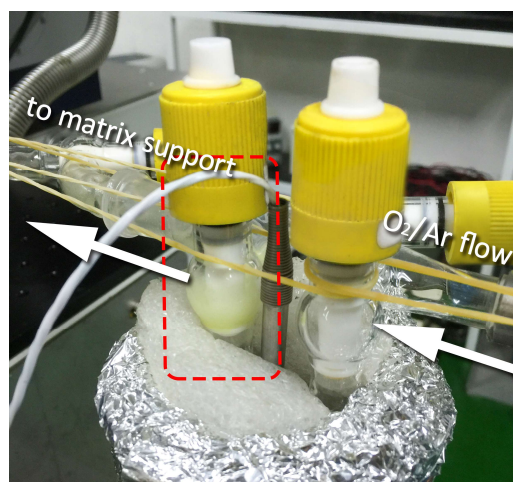
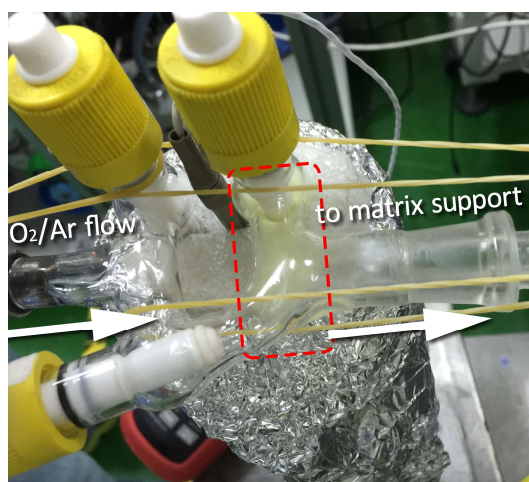


Figure S4. The oxidation of the HPCO vapor by passing the O₂/Ar mixture (3:1000) through the cold trap containing solid HPCO (ca. -139 °C).

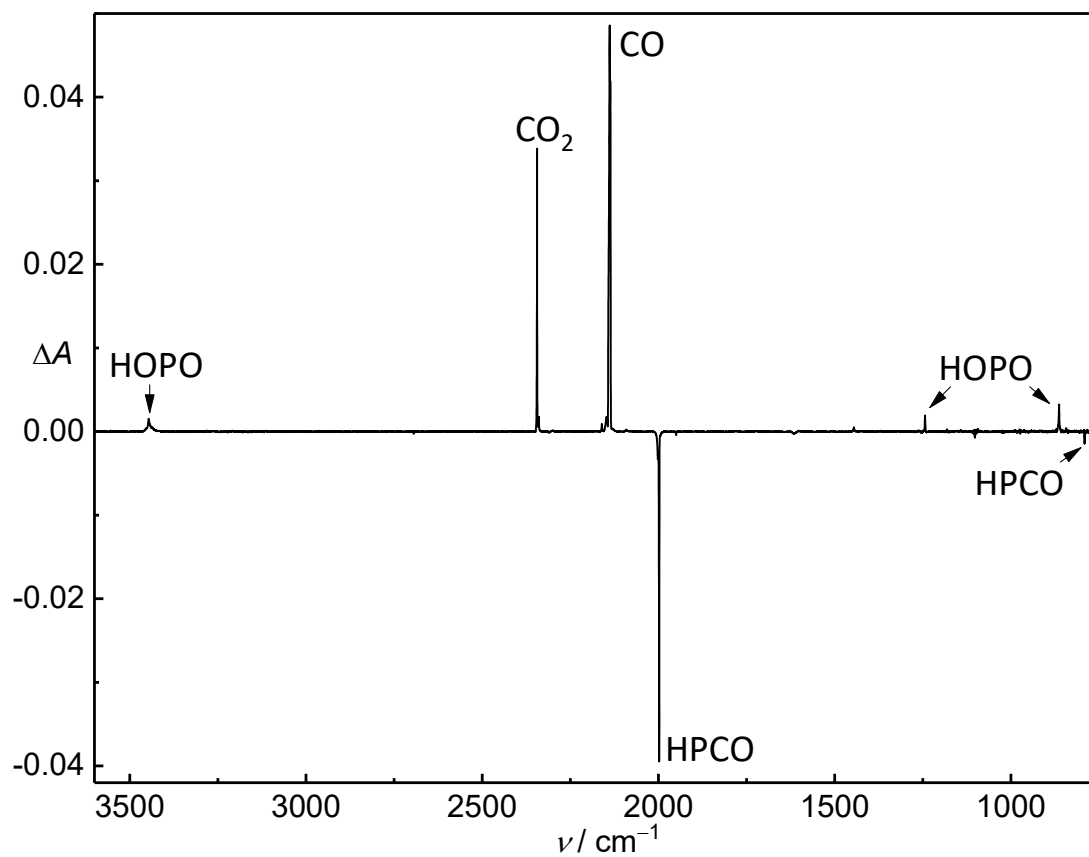


Figure S5. IR difference spectrum showing the changes of HPCO/O₂/Ar=1:3:1000 upon 365 nm LED irradiation (12.0 K, 30.0 min).

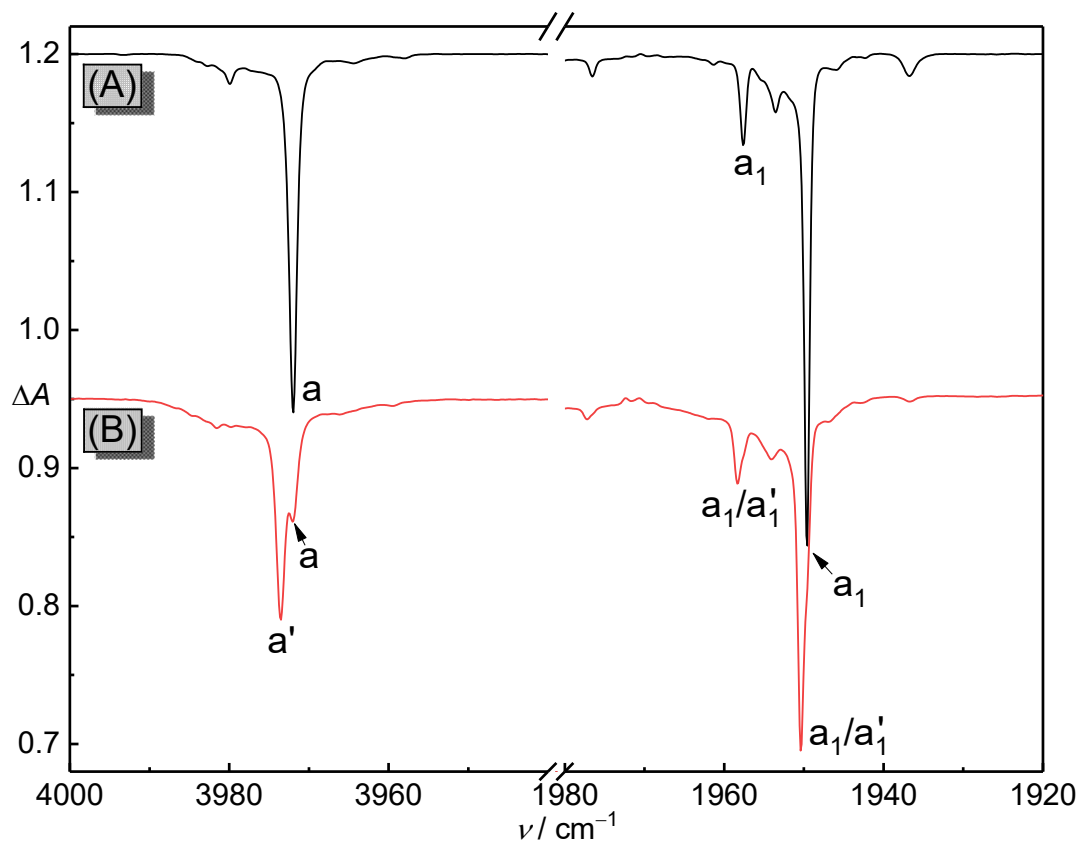


Figure S6. A) IR difference spectrum in the ranges of 4000–3940 and 1980–1920 cm^{-1} showing the depletion of HPCO (a) upon 193 nm laser irradiation (5 mJ, 7.5 min). B) IR difference spectrum in the range of 4000–3940 and 1980–1920 cm^{-1} showing the depletion of DPCO (a') in Ar-matrix upon 193 nm laser irradiation (5 mJ, 5.5 min). The IR bands of HP^{13}CO (a₁), DP^{13}CO (a₁') are also labeled.

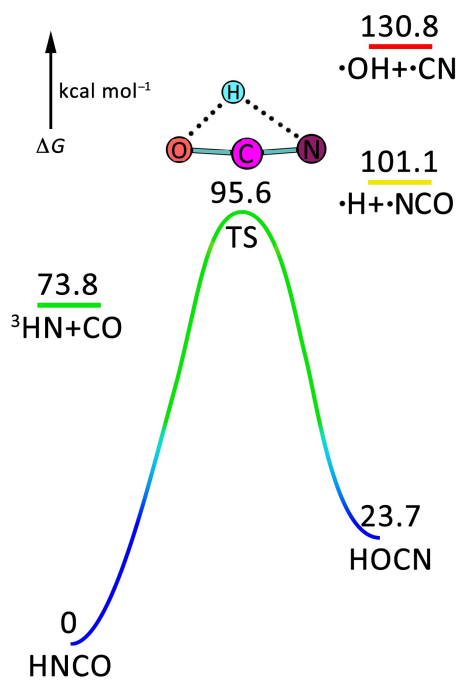


Figure S7. Potential energy profile for HNCO/HOCN, and relative energies with ZPE corrections (in kcal mol⁻¹) calculated at the CCSD(T)/cc-pVTZ level.

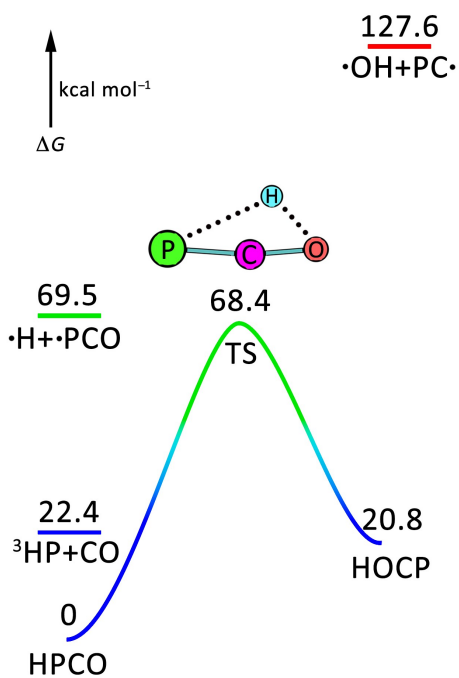


Figure S8. Potential energy profile for HPCO/HOCP, and relative energies with ZPE corrections (in kcal mol⁻¹) calculated at the CCSD(T)/cc-pVTZ level.

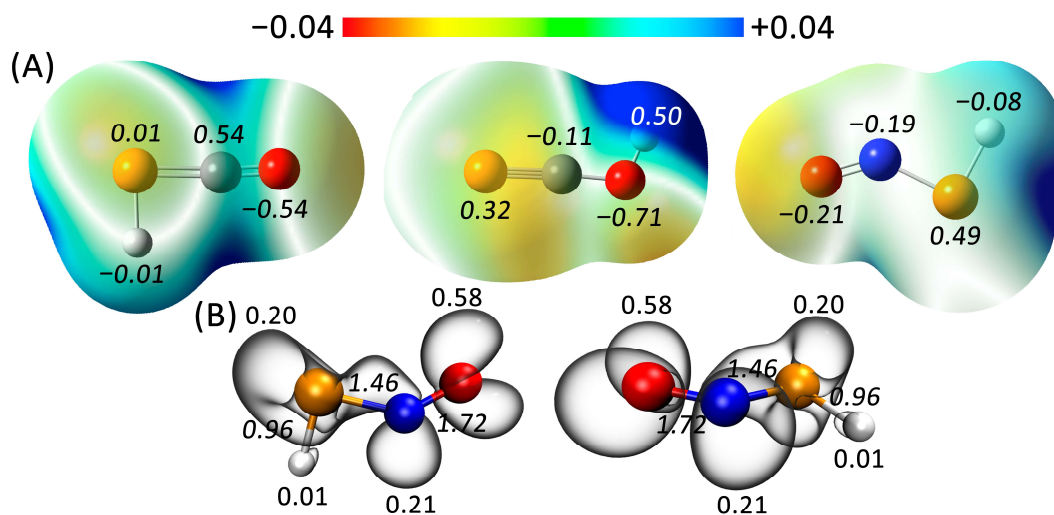


Figure S9. A) CCSD(T)/cc-pVTZ calculated molecular electrostatic potential (ESP) maps (isovalue = 0.001) and natural population analysis (NPA) charges (numbers in italics) for HPCO (a, left), HOCP (b, middle), and *anti*-HPNO• (g, right). B) Calculated spin densities and Wiberg bond indices (WBI, numbers in italics) for *anti*-HPNO•.

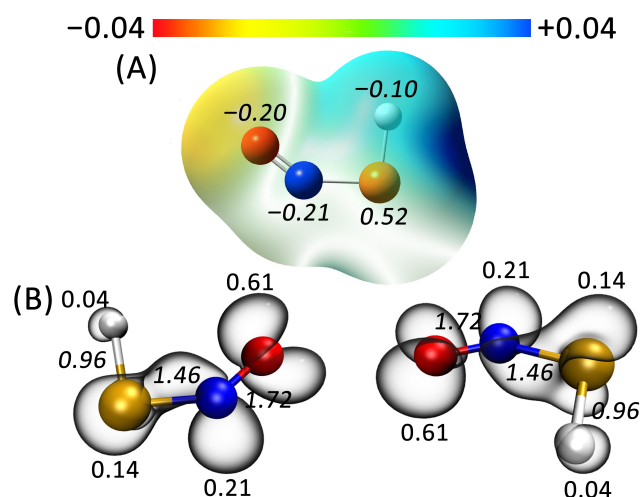


Figure S10. A) CCSD(T)/cc-pVTZ calculated molecular electrostatic potential (ESP) maps (isovalue = 0.001) and natural population analysis (NPA) charges (numbers in italics) for *syn*-HPNO•. B) Calculated spin densities and Wiberg bond indices (WBI, numbers in italics) for *syn*-HPNO•.

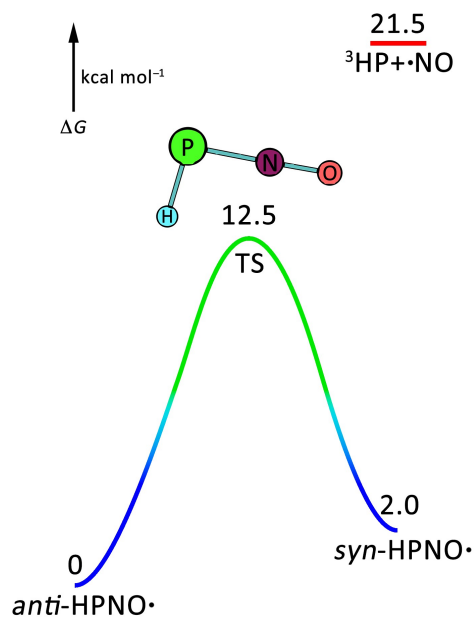


Figure S11. Potential energy profile for *anti*-HPNO•, and relative energies with ZPE corrections (in kcal mol^{-1}) calculated at the UCCSD(T)/cc-pVTZ level.

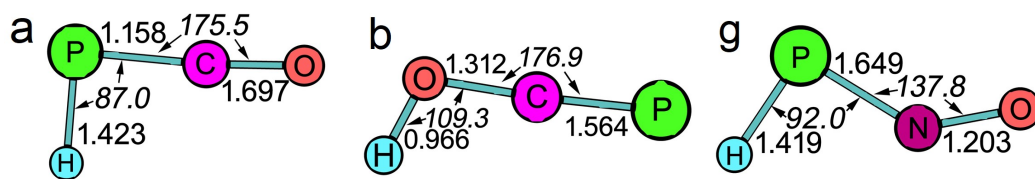


Figure S12. The CCSD(T)/cc-pVTZ computed bond lengths (Å) and angles (°).

Table S1. Observed and Calculated IR Frequencies (cm^{-1}) for HPCO.

$\nu_{\text{calc.}}^{[a]}$		$\nu_{\text{obs.}}^{[b]}$	$\Delta \nu(\text{H/D})$		assignment ^[c]
harmonic	VPT2		obs. ^[d]	calc. ^[e]	
2390.9 (16)	2292.9 (17)	2311.9 (1.22)	634.7	627.3	ν_{1s} , $\nu(\text{PH})$
2042.1 (631)	2011.0 (605)	1998.5 (100)	-1.1	-0.6	ν_{2s} , $\nu(\text{CO})$
867.4 (11)	851.1 (11)	857.9 (1.67)	118.5	122.8	ν_{3s} , $\delta(\text{HPC})$
703.1 (5)	686.2 (4)	698.4 (0.76)	40.7	36.0	ν_{4s} , $\nu(\text{PC})$
467.0 (<1)	470.5 (<1)	473.4 ^[f]	-1.6 ^[f]	5.6	ν_{5s} , $\delta_{\text{o.o.p.}}(\text{PCO})$
381.3 (7)	386.6 (7)	392.6 ^[f]	47.9 ^[f]	43.6	ν_{6s} , $\delta_{\text{i.p.}}(\text{PCO})$

[a] Calculated frequencies and intensities (km mol^{-1} , in parenthesis) at the CCSD(T)/cc-pVTZ level.

[b] Observed frequencies and relative intensities (in parenthesis) in Ar-matrix. [c] Tentative

assignment of the vibrational modes. [d] Observed isotope shifts in the Ar-matrix. [e] VPT2

computed isotope shifts. [f] Derived from overtone bands.

Table S2. Calculated and Observed IR Frequencies (4000–500 cm⁻¹) of HPCO and DPCO.

HPCO		DPCO		assignment ^[c]
$\nu_{\text{calc.}}$ ^[a]	$\nu_{\text{obs.}}$ ^[b]	$\nu_{\text{calc.}}$ ^[a]	$\nu_{\text{obs.}}$ ^[b]	
2292.9 (16.76)	2311.8 (1.22)	1665.6 (8.25)	1677.1 (0.42)	ν_1 , $\nu(\text{PH})$
2011.0 (604.79)	1998.5 (100)	2011.6 (605.72)	1999.6 (100)	ν_2 , $\nu(\text{CO})$
851.1 (10.66)	857.9 (1.67)	728.3 (1.96)	739.4 (0.52)	ν_3 , $\delta(\text{HPC})$
686.2 (4.13)	698.4 (0.76)	650.2 (6.32)	657.7 (1.61)	ν_4 , $\nu(\text{PC})$
470.5 (0.02)	473.4 ^[d]	464.9 (0.07)	471.8 ^[d]	ν_5 , $\delta_{\text{o.o.p.}}(\text{PCO})$
386.6 (6.92)	392.6 ^[d]	343.0(6.04)	344.7 ^[d]	ν_6 , $\delta_{\text{i.p.}}(\text{PCO})$
4302.6 (0.34)	n.o. ^[e]	3286.3 (0.02)	n.o. ^[e]	$2\nu_1$
3998.9 (7.61)	3972.0 (1.36)	4000.1(7.60)	3973.6 (3.83)	$2\nu_2$
1697.6 (0.87)	1709.1 (0.27)	1450.5 (0.31)	1473.6 (0.06)	$2\nu_3$
1364.7 (0.37)	1389.7 (0.05)	1297.0 (0.06)	n.o. ^[e]	$2\nu_4$
941.6 (0.48)	946.8 (0.02)	930.6 (0.42)	943.6 (0.08)	$2\nu_5$
779.7 (0.86)	785.2 (0.10)	689.5 (1.35)	689.4 (0.00)	$2\nu_6$
4302.6 (0.34)	n.o. ^[e]	3676.2(0.19)	3674.8 (0.02)	$\nu_1 + \nu_2$
3124.4 (0.05)	3151.7 (0.02)	2391.2 (0.07)	2414.7 (0.01)	$\nu_1 + \nu_3$
2977.8 (0.17)	3009.2 (0.03)	2309.3 (0.15)	2311.7 (0.33)	$\nu_1 + \nu_4$
2763.3 (0.01)	2774.2 (0.03)	2130.4 (0.00)	n.o. ^[e]	$\nu_1 + \nu_5$
2679.1 (0.12)	2689.9 (0.01) ^[f]	2007.2 (0.32)	2010.9 ^[g]	$\nu_1 + \nu_6$
2859.0 (0.11)	2852.2 (0.03)	2738.2 (2.71)	2736.6 (0.82)	$\nu_2 + \nu_3$
2696.0 (4.58)	2694.8 (0.83)	2658.6 (2.00)	2652.9 (0.36)	$\nu_2 + \nu_4$
2475.3 (0.04)	n.o. ^[e]	2470.3 (0.03)	n.o. ^[e]	$\nu_2 + \nu_5$
2393.0 (0.08)	2383.5 (0.01)	2352.5 (0.00)	n.o. ^[e]	$\nu_2 + \nu_6$
1535.4 (0.26)	1554.2 (0.07)	1376.2 (0.17)	1389.6 (0.07)	$\nu_3 + \nu_4$
1324.2 (0.02)	n.o. ^[e]	1191.5 (0.01)	n.o. ^[e]	$\nu_3 + \nu_5$
1236.4 (3.01)	1246.4 (0.60)	1073.7 (1.08)	1091.2 (0.27)	$\nu_3 + \nu_6$
1152.2 (0.00)	n.o. ^[e]	1115.2 (0.01)	n.o. ^[e]	$\nu_4 + \nu_5$
1062.8 (0.02)	1082.2 (0.01)	984.6 (1.01)	1001.6 (0.30)	$\nu_4 + \nu_6$
870.9 (0.00)	n.o. ^[e]	813.1 (0.00)	n.o. ^[e]	$\nu_5 + \nu_6$

[a] Computed anharmonic IR frequencies (cm⁻¹) and intensities (km mol⁻¹, in parentheses) at the CCSD(T)/cc-pVTZ level using VPT2. [b] Observed band positions (cm⁻¹) and normalized relative intensities (in parentheses) in Ar-matrix at 12 K. [c] Tentative assignment of the vibrational modes according to the computed vibrational displacement vectors. [d] Derived from combination and overtone bands. [e] Not observed due to either out of range (4000–500 cm⁻¹) or low intensity. [f] Overlap with the strong band at 2694.8 cm⁻¹. [g] Overlap with the strong band at 1996.9 cm⁻¹.

Table S3. Experimental and Calculated IR Frequencies (cm^{-1}) for HP.

$\nu_{\text{calc.}}^{[a]}$	$\nu_{\text{obs.}}$		$\Delta \nu(\text{H/D})$		assignment ^[b]
	Ar-matrix	N ₂ -matrix	obs. ^[c]	calc. ^[d]	
2264.4 (75)	2300.9	2291.3	633.1	620.0	$\nu_1, \nu(\text{PH})$

[a] Calculated VPT2 anharmonic frequencies and intensities (km mol^{-1} , in parenthesis) at the UCCSD(T)/cc-pVTZ level. [b] Tentative assignment of the vibrational modes according to the computed vibrational displacement vectors. [c] Observed isotope shift in Ar-matrix. [d] VPT2 computed isotope shift.

Table S4. Observed and Calculated IR Frequencies (cm^{-1}) for PCO•.

$\nu_{\text{calc.}}^{[a]}$	$\nu_{\text{obs.}}$			assignment ^[b]
	ref. ^[c]	Ar-matrix	N ₂ -matrix	
1921.2 (1189)	1912.7	1912.8	1919.3	$\nu_1, \nu(\text{CO})$
718.7 (9)	n.o.	n.o.	n.o.	$\nu_2, \nu(\text{CP})$
407.2 (9)	n.o.	n.o.	n.o.	$\nu_3, \delta(\text{PCO})$

[a] Calculated VCI (utilizing quartic force fields) anharmonic frequencies and intensities (km mol^{-1} , in parenthesis) at the UCCSD(T)-F12/cc-pVTZ-F12 level. [b] Tentative assignment of the vibrational modes according to the computed vibrational displacement vectors. [c] In Ar-matrix, see reference 6.

Table S5. Calculated IR Frequencies (cm^{-1}) for *syn*-HPNO•.

$\nu_{\text{calc.}}^{[a]}$	$\Delta \nu_{\text{calc.}}(\text{H/D})$	$\Delta \nu_{\text{calc.}}(^{14}\text{N}/^{15}\text{N})$	assignment ^[b]
2194.8 (74)	594.8	0.1	$\nu_1, \nu(\text{HP})$
1598.4 (241)	-1.4	29.2	$\nu_2, \nu(\text{NO})$
905.9 (33)	222.9	3.0	$\nu_3, \delta(\text{HP})$
721.4 (2)	-5.4	7.2	$\nu_4, \nu(\text{PN})$
503.9 (<1)	59.9	6.6	$\nu_5, \tau(\text{HPN})$
291.8 (4)	12.9	4.7	$\nu_6, \delta(\text{PNO})$

[a] VPT2 computed anharmonic frequencies and intensities (km mol^{-1} , in parenthesis) at the UCCSD(T)/cc-pVTZ level. [b] Tentative assignment of the vibrational modes according to the computed vibrational displacement vectors.

Table S6. Calculated and Observed IR Frequencies (4000–500 cm⁻¹) for HOCP.

$\nu_{\text{calc.}}$ ^[a]	$\nu_{\text{obs.}}$ ^[b]		$\Delta \nu(\text{H/D})$		assignment ^[c]
	Ar-matrix	N ₂ -matrix	obs. ^[d]	calc. ^[e]	
3599.0 (138)	3549.6 (43)	3514.8 (49)	924.2	940.6	ν_1 , $\nu(\text{OH})$
1647.7 (274)	1647.8 (100)	1653.4 (100)	9.3	10.0	ν_2 , $\nu(\text{CP})$
1188.4 (173)	1174.8 (55)	1186.0 (44)	254.4	261.9	ν_3 , $\delta(\text{COH})$
793.8 (5)	n.o.	n.o.	n.o.	15.8	ν_4 , $\nu(\text{CO})$
423.6 (7)	n.o.	n.o.	n.o.	12.2	ν_5 , $\tau(\text{CP})$
376.8 (22)	n.o.	n.o.	n.o.	28.6	ν_6 , $\delta(\text{OCP})$

[a] VPT2 computed anharmonic frequencies and intensities (km mol⁻¹, in parenthesis) at the CCSD(T)/cc-pVTZ level. [b] Observed frequencies and relative intensities (in parenthesis). [c] Tentative assignment of the vibrational modes according to the computed vibrational displacement vectors. [d] Observed isotope shifts in the Ar-matrix. [e] VPT2 computed isotope shifts.

Computed Atomic Coordinates (in Angstrom) and Energies (in Hartree) for All Optimized Structures

HPCO

CCSD(T)/cc-pVTZ

P	-1.100245047	0.048807065	0.000000000
H	-1.090711152	-1.373954481	0.000000000
C	0.595063156	-0.027036592	0.000000000
O	1.752883159	0.012341826	0.000000000
Electronic energy	...	-454.64156814 Eh	
Total thermal energy		-454.62238243 Eh	
Final Gibbs free enthalpy	...	-454.65076644 Eh	

HOCP

CCSD(T)/cc-pVTZ

O	-1.740823680	-0.061696019	-0.000061935
H	-2.106797437	0.831972640	-0.000085531
C	-0.430008729	0.006372289	0.000166622
P	1.134111968	0.002320421	-0.000029787
Electronic energy	...	-454.61202767 Eh	
Total thermal energy		-454.58962864 Eh	
Final Gibbs free enthalpy	...	-454.61761738 Eh	

HP

UCCSD(T)/cc-pVTZ

P	0.000000000	0.000000000	0.045028872
H	0.000000000	0.000000000	-1.383884561
Electronic energy	...	-341.43243743 Eh	
Total thermal energy		-341.42471969 Eh	
Final Gibbs free enthalpy	...	-341.44603401 Eh	

CO

CCSD(T)/cc-pVTZ

O	0.000000000	0.000000000	0.057140293
C	0.000000000	0.000000000	1.192859706
Electronic energy	...	-113.15557882 Eh	
Total thermal energy		-113.14831307 Eh	
Final Gibbs free enthalpy	...	-113.16980983 Eh	

•OH

UCCSD(T)/cc-pVTZ

O	0.695488760	-1.228722864	0.000000000
H	0.371339909	-0.313354755	0.000000000
Electronic energy	...	-75.63772341 Eh	

Total thermal energy		-75.62683265 Eh
Final Gibbs free enthalpy	...	-75.64611268 Eh

CP•

UCCSD(T)/cc-pVTZ

C	-0.653083910	0.857418100	0.000000000
P	0.916614530	0.857418100	0.000000000
Electronic energy		...	-378.78325596 Eh
Total thermal energy			-378.77800658 Eh
Final Gibbs free enthalpy		...	-378.80160969 Eh

anti-HPNO•

UCCSD(T)/cc-pVTZ

15	-1.038126422	0.111786535	0.000000000
1	-1.484825179	-1.235056968	0.000000000
7	0.544302781	-0.353112824	0.000000000
8	1.627343016	0.170487738	0.000000000
Electronic energy		...	-471.19848650 Eh
Total thermal energy			-471.17969340 Eh
Final Gibbs free enthalpy		...	-471.20880834 Eh

syn-HPNO•

UCCSD(T)/cc-pVTZ

P	-1.063708608	-0.017849466	0.000000000
H	-0.889245425	-1.440578177	0.000000000
N	0.546334228	0.337811611	0.000000000
O	1.637576881	-0.170409484	0.000000000
Electronic energy		...	-471.19466453 Eh
Total thermal energy			-471.17627530 Eh
Final Gibbs free enthalpy		...	-471.20566272 Eh

Transition State between anti-HPNO• and syn-HPNO•

UCCSD(T)/cc-pVTZ

P	1.079466172	-0.092598085	0.000580892
H	1.317499597	1.331149001	0.000008963
N	-0.523003292	-0.015719276	-0.001379523
O	-1.722561477	0.022807360	0.000789667
Electronic energy		...	-471.17680516 Eh
Total thermal energy			-471.15998894 Eh
Final Gibbs free enthalpy		...	-471.18884990 Eh

•NO

UCCSD(T)/cc-pVTZ

N	-0.444719623	0.857418100	0.000000000
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O	0.708250243	0.857418100	0.000000000
Electronic energy		...	-129.71683452 Eh
Total thermal energy			-129.70965925 Eh
Final Gibbs free enthalpy		...	-129.73201664 Eh

HNCO

CCSD(T)/cc-pVTZ

N	-1.171115912	-0.142141094	0.000024010
H	-1.813621773	0.631516391	0.000075577
C	0.039215417	0.023101154	-0.000083948
O	1.208968268	0.021270548	0.000026360
Electronic energy		...	-168.43278174 Eh
Total thermal energy			-168.40832301 Eh
Final Gibbs free enthalpy		...	-168.43450920 Eh

HCPO

CCSD(T)/cc-pVTZ

C	1.518133822	-0.284235130	0.000000000
H	2.445485779	0.267824417	0.000000000
P	0.000913866	0.195934792	0.000000000
O	-1.436662468	-0.182136079	0.000000000
Electronic energy		...	-454.56033210 Eh
Total thermal energy			-454.54067364 Eh
Final Gibbs free enthalpy		...	-454.56915187 Eh

HC(O)P

UCCSD(T)/cc-pVTZ

C	0.000457206	0.781580643	-0.000000010
H	-0.782662110	1.563931260	0.000000003
O	1.178350141	1.074180219	0.000000003
P	-0.579361237	-0.994681123	0.000000002
Electronic energy		...	-454.57934304 Eh
Total thermal energy			-454.55818747 Eh
Final Gibbs free enthalpy		...	-454.58797734 Eh

HNCO

CCSD(T)/cc-pVTZ

N	-1.171115912	-0.142141094	0.000024010
H	-1.813621773	0.631516391	0.000075577
C	0.039215417	0.023101154	-0.000083948
O	1.208968268	0.021270548	0.000026360
Electronic energy		...	-168.43278174 Eh
Total thermal energy			-168.40832301 Eh
Final Gibbs free enthalpy		...	-168.43450920 Eh

HOCN**CCSD(T)/cc-pVTZ**

O	-0.143197172	-1.124264726	-0.000000000
H	0.732552710	-1.530925334	-0.000000000
C	-0.003116049	0.175951619	0.000000000
N	0.057698510	1.339341442	-0.000000000
Electronic energy	...	-168.39491798 Eh	
Total thermal energy		-168.37014483 Eh	
Total Enthalpy	...	-168.36920062 Eh	

HONC**CCSD(T)/cc-pVTZ**

O	-0.025775253	-1.103505582	-0.000000000
H	-0.969430144	-1.317130319	-0.000000000
C	0.189909632	1.401261104	-0.000000000
N	0.008377765	0.234011796	0.000000000
Electronic energy	...	-168.30115974 Eh	
Total thermal energy		-168.27714809 Eh	
Total Enthalpy	...	-168.27620388 Eh	

HCNO**CCSD(T)/cc-pVTZ**

C	-0.049835343	-1.181202165	-0.000000023
H	-0.073360528	-2.241518724	0.000000015
O	0.046590201	1.188114095	0.000000007
N	0.000231671	-0.016294204	-0.000000000
Electronic energy	...	-168.32119851 Eh	
Total thermal energy		-168.29954789 Eh	
Final Gibbs free enthalpy	...	-168.32224866 Eh	

HC(O)N**UCCSD(T)/cc-pVTZ**

C	-0.008730904	0.384329647	-0.000000006
H	0.031804281	1.484125745	0.000000002
O	-1.058648635	-0.244859578	0.000000002
N	1.217379258	-0.261461814	0.000000001
Electronic energy	...	-168.29852456 Eh	
Total thermal energy		-168.27635840 Eh	
Total Enthalpy	...	-168.27541419 Eh	

References

- [1] E. D. Glendening, J. K. Badenhoop, A. E. Reed, J. E. Carpenter, J. A. Bohmann, C. M. Morales, P. Karafiloglou, C. R. Landis and F. Weinhold, Theoretical Chemistry Institute, University of Wisconsin, Madison, WI (2018).
- [2] F. Neese, *WIREs Comput. Mol. Sci.*, 2018, **8**, No. e1327.
- [3] T. Lu and F. W. Chen, *J. Comput. Chem.*, 2012, **33**, 580–592.
- [4] M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, Ö. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski, D. J. Fox, *Gaussian 09*, Revision A.2, Gaussian, Inc., Wallingford CT, 2009.
- [5] a) K. Fukui, *Acc. Chem. Res.*, 1981, **14**, 363368. b) J. A. Montgomery Jr. and M. J. Frisch, *J. Chem. Phys.*, 1999, **110**, 2822.
- [6] Z. Mielke and L. Andrews, *Chem. Phys. Lett.*, 1991, **181**, 355.