

## Supporting Information

### **Pressure-stabilized polymerization of nitrogen in manganese nitrides at ambient and high pressures**

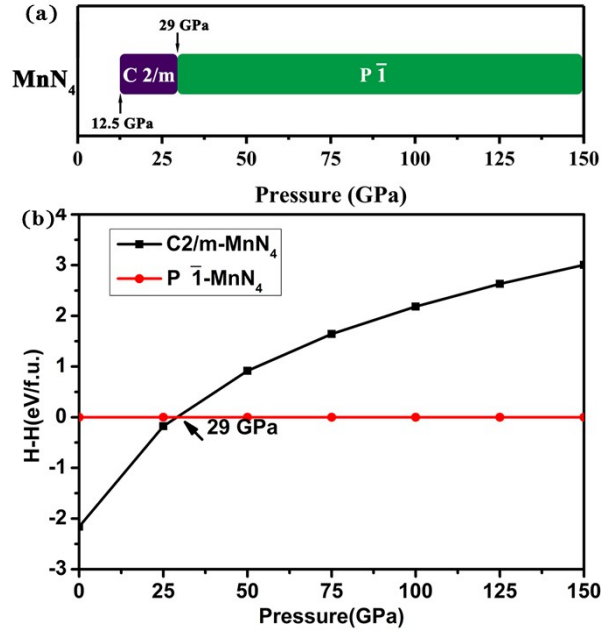
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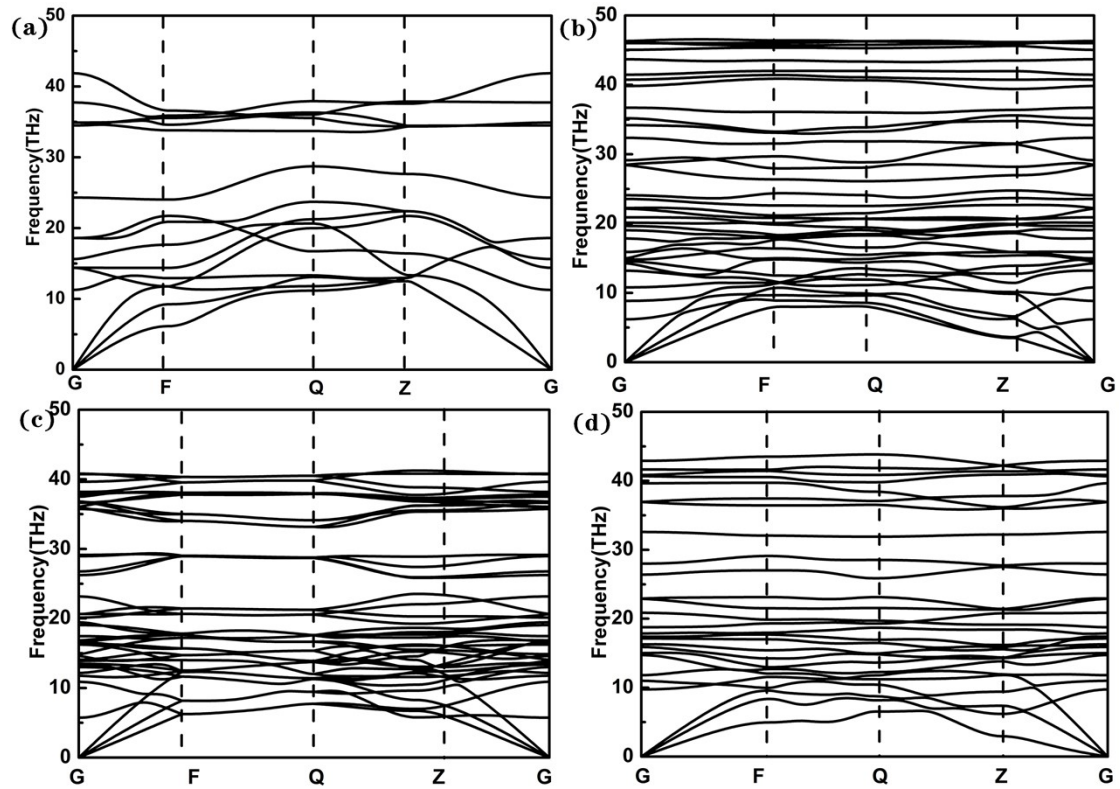
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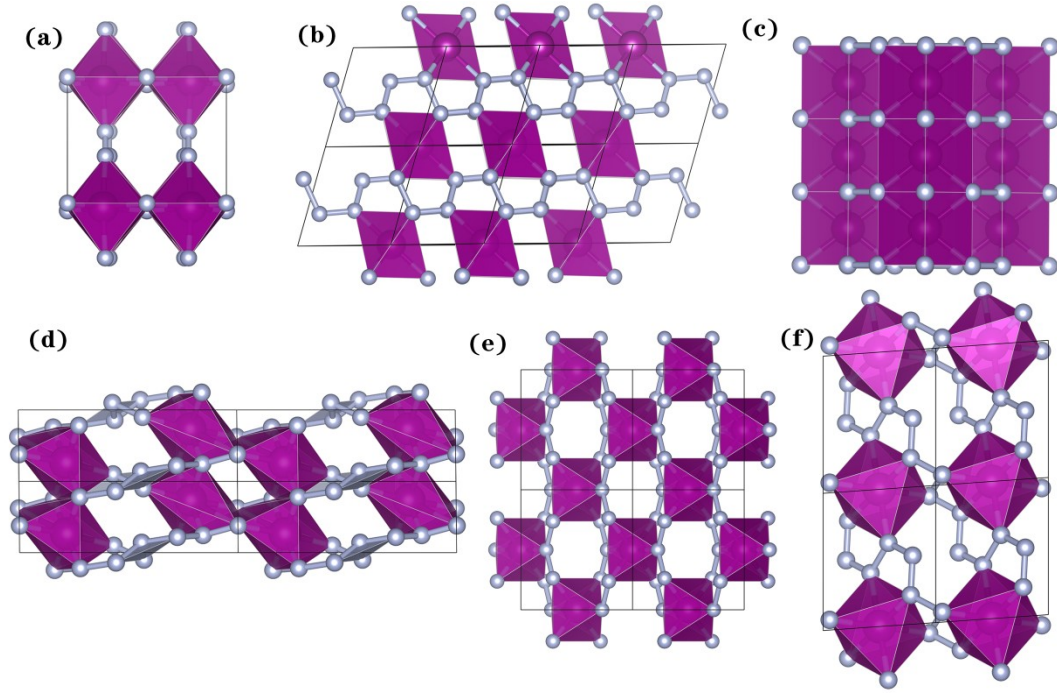
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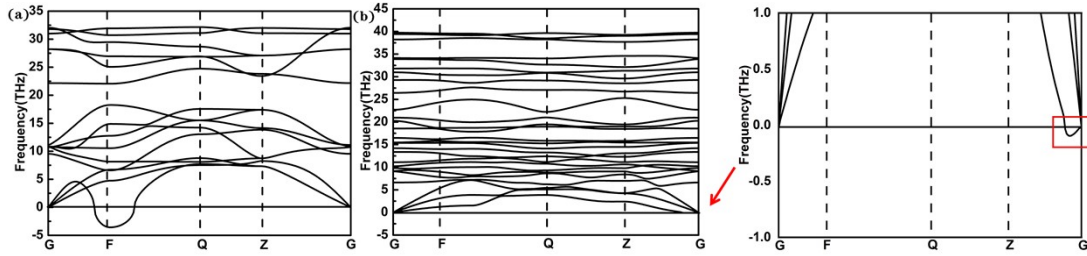
**Fig. S1.** (a) The stable pressure range of  $C2/m$ - $\text{MnN}_4$  and  $P\bar{1}$ - $\text{MnN}_4$ , and (b) enthalpy of the  $C2/m$ - $\text{MnN}_4$  phase relative to the  $P\bar{1}$ - $\text{MnN}_4$  phase with pressure.



**Fig. S2.** The phonon dispersion curves of (a)  $P4/mmm$ - $\text{MnN}_4$  at 100 GPa, (b)  $P\bar{1}$ - $\text{MnN}_5$  at 150 GPa, (c)  $C2/m$ - $\text{MnN}_6$  at 50 GPa, and (d)  $P\bar{1}$ - $\text{MnN}_8$  at 50 GPa.



**Fig. S3.** The polyhedral supercell structures of the predicted phases: (a)  $C2/m$ - $MnN_4$ , (b)  $P \bar{1}$ - $MnN_4$ , (c)  $P4/mmm$ - $MnN_4$ , (d)  $P \bar{1}$ - $MnN_5$ , (e)  $C2/m$ - $MnN_6$ , and (f)  $P \bar{1}$ - $MnN_8$ .



**Fig. S4.** The phonon dispersion curves of (a)  $P4/mmm$ - $MnN_4$  at 0 GPa and (b)  $P \bar{1}$ - $MnN_8$  at 0 GPa.

**Table S1.** Structural parameters of the Mn-N compounds at corresponding pressure.

Structure	Pressure (GPa)	Lattice parameters ( $\text{\AA}$ , $^\circ$ )	Atomic Positions
$P6_3/mmc$ - $Mn_2N$	50	$a = 2.5102$ $b = 2.5102$ $c = 8.6143$ $\alpha = 90.00^\circ$ $\beta = 90.00^\circ$ $\gamma = 120.00^\circ$	$Mn_1(4f)$ (0.6667, 0.3333, 0.6074) $N_1(2d)$ (0.6667, 0.3333, 0.2500)

C2/m-MnN <sub>4</sub>	20	$a = 7.6064$ $b = 5.5432$ $c = 10.3809$ $\alpha = 90.00^\circ$ $\beta = 154.99^\circ$ $\gamma = 90.00^\circ$	Mn <sub>1</sub> (4g) (0.0000, 0.2238, 0.0000) N <sub>1</sub> (8j) (-1.5704, 0.2380, -1.5968) N <sub>2</sub> (4i) (-2.2758, 0.5000, -1.9694) N <sub>3</sub> (4i) (0.3055, 0.0000, -0.0290)
P $\bar{1}$ -MnN <sub>4</sub>	50	$a = 3.7071$ $b = 3.8140$ $c = 5.1901$ $\alpha = 98.21^\circ$ $\beta = 110.51^\circ$ $\gamma = 73.47^\circ$	Mn <sub>1</sub> (2i) (0.7632, 0.0026, 0.7613) N <sub>1</sub> (2i) (0.1861, 0.6554, 0.6873) N <sub>2</sub> (2i) (0.6775, 0.6498, 0.1855) N <sub>3</sub> (2i) (0.0992, 0.3231, 0.9615) N <sub>4</sub> (2i) (0.5984, 0.3191, 0.4640)
P4/mmm-MnN <sub>4</sub>	100	$a = 3.5131$ $b = 3.5131$ $c = 2.3547$ $\alpha = 90.00^\circ$ $\beta = 90.00^\circ$ $\gamma = 90.00^\circ$	Mn <sub>1</sub> (1c) (0.5000, 0.5000, 0.0000) N <sub>1</sub> (4k) (0.8050, 0.1940, 0.5000)
P $\bar{1}$ -MnN <sub>5</sub>	150	$a = 2.3722$ $b = 3.7598$ $c = 7.2847$ $\alpha = 88.00^\circ$ $\beta = 90.62^\circ$ $\gamma = 91.50^\circ$	Mn <sub>1</sub> (2i) (0.2644, 0.6189, 0.2150) N <sub>1</sub> (2i) (0.2188, 0.0321, 0.7226) N <sub>2</sub> (2i) (0.8591, 0.1085, 0.4408) N <sub>3</sub> (2i) (0.3789, 0.8451, 0.0020) N <sub>4</sub> (2i) (0.7555, 0.2581, 0.1595) N <sub>5</sub> (2i) (0.0997, 0.4199, 0.4312)
C2/m-MnN <sub>6</sub>	50	$a = 5.1820$ $b = 4.8642$ $c = 4.3284$ $\alpha = 90.00^\circ$ $\beta = 61.10^\circ$ $\gamma = 90.00^\circ$	Mn <sub>1</sub> (2d) (0.0000, 0.5000, 0.5000) N <sub>1</sub> (8j) (0.7870, 0.2375, 0.8349) N <sub>2</sub> (4i) (0.7265, 0.0000, 0.7307)

P $\bar{1}$ -MnN <sub>8</sub>	50	a = 3.7604	Mn <sub>1</sub> (1e) (0.5000, 0.5000, 0.0000)
		b = 3.7855	N <sub>1</sub> (2i) (0.4204, 0.5923, 0.6062)
		c = 4.4830	N <sub>2</sub> (2i) (0.7774, 0.2247, 0.7866)
		α = 97.15°	N <sub>3</sub> (2i) (0.9421, 0.8681, 0.0742)
		β = 92.31°	N <sub>4</sub> (2i) (0.2046, 0.7742, 0.4997)
		γ = 102.71°	

**Table S2.** The elastic constants  $C_{ij}$  (GPa) of the Mn-N compounds at 0 GPa.

MnN <sub>x</sub>	C <sub>11</sub>	C <sub>22</sub>	C <sub>33</sub>	C <sub>44</sub>	C <sub>55</sub>	C <sub>66</sub>	C <sub>12</sub>	C <sub>13</sub>	C <sub>15</sub>	C <sub>23</sub>	C <sub>25</sub>	C <sub>35</sub>	C <sub>46</sub>
C2/m-MnN <sub>4</sub>	404	214	320	103	15	26	8	4	10	82	-4	-4	0.02
P $\bar{1}$ -MnN <sub>4</sub>	439	629	295	172	185	214	80	120	-137	68	-35	-88	-74
P4/mmm-MnN <sub>4</sub>	720	720	396	109	109	205	100	4	-	-	-	-	-
P $\bar{1}$ -MnN <sub>5</sub>	408	292	584	122	238	164	117	141	-36	70	-33	-102	-28
C2/m-MnN <sub>6</sub>	211	520	441	176	136	134	101	87	-8	107	25	-6	-20
P $\bar{1}$ -MnN <sub>8</sub>	108	423	280	105	55	60	37	49	18	111	-8	-60	-23

Monoclinic structure:

The mechanical stability criteria of monoclinic structure are shown as follows:

$$C_{11} > 0, C_{22} > 0, C_{33} > 0, C_{44} > 0, C_{55} > 0, C_{66} > 0, [C_{11} + C_{22} + C_{33} + 2(C_{12} + C_{13} + C_{23})] > 0$$

$$C_{33}C_{55} - C_{35}^2 > 0, C_{44}C_{66} - C_{46}^2 > 0, C_{22} + C_{33} - 2C_{23} > 0,$$

$$C_{22}(C_{33}C_{55} - C_{35}^2) + 2C_{23}C_{25}C_{35} - (C_{23}^2)C_{55} - (C_{25}^2)C_{33} > 0$$

$$2[C_{15}C_{25}(C_{33}C_{12} - C_{13}C_{23}) + C_{15}C_{35}(C_{22}C_{13} - C_{12}C_{23}) + C_{25}C_{35}(C_{11}C_{23} - C_{12}C_{13})] - [C_{15}^2(C_{22}C_{33} - C_{23}^2) + C_{25}C_{25}(C_{11}C_{33} - C_{13}^2) + C_{35}C_{35}(C_{11}C_{22} - C_{12}^2)] + C_{55}g > 0$$

At ambient pressure, the monoclinic C2/m-MnN<sub>4</sub> and C2/m-MnN<sub>6</sub> are mechanically unstable due to that their elastic tensors  $C_{ij}$  do not satisfy to the criteria.

Triclinic structure:

The mechanical stability criteria of triclinic structure are shown as follows:

$$C_{11} > 0, C_{22} > 0, C_{33} > 0, C_{44} > 0, C_{55} > 0, C_{66} > 0$$

$$[C_{11} + C_{22} + C_{33} + 2(C_{12} + C_{13} + C_{23})] > 0$$

$$(C_{33}C_{55} - C_{35}^2) > 0, (C_{44}C_{66} - C_{46}^2) > 0, (C_{22} + C_{33} - 2C_{23}) > 0$$

$$[C_{22} (C_{33}C_{55} - C_{35}^2) + 2C_{23}C_{25}C_{35} - C_{23}^2C_{55} - C_{23}^2C_{33}] > 0$$

At ambient pressure, the triclinic  $P \bar{1}$ -MnN<sub>4</sub>,  $P \bar{1}$ -MnN<sub>5</sub> and  $P \bar{1}$ -MnN<sub>8</sub> are mechanically stable due to that their elastic tensors C<sub>ij</sub> satisfy to the criteria.

Tetragonal structure:

The mechanical stability criteria of tetragonal structure are shown as follows:

$$C_{11} > |C_{12}|$$

$$2C_{13}^2 < C_{33} (C_{11} + C_{12})$$

$$C_{44} > 0$$

At ambient pressure, the tetragonal P4/mmm-MnN<sub>4</sub> is mechanically stable due to that its elastic tensor C<sub>ij</sub> satisfies to the criteria.

**Table S3.** The Bader charges transfer of the Mn-N compounds, the "N" is the number of each equivalent atom (Mn and N atoms) in one cell and the "sigma (e)" is the corresponding number of electrons lost and gained by the Mn and N atoms, respectively.

Structure	Atom	N	Charge value(e)	σ(e)
<b>C2/m-MnN<sub>4</sub></b>	Mn <sub>1</sub>	4	11.67	-1.30
	N <sub>1</sub>	8	5.20	0.20
	N <sub>2</sub>	4	5.49	0.49
	N <sub>3</sub>	4	5.41	0.41
<b>P <math>\bar{1}</math>-MnN<sub>4</sub></b>	Mn <sub>1</sub>	2	11.81	-1.19
	N <sub>1</sub>	2	5.20	0.20
	N <sub>2</sub>	2	5.22	0.22
	N <sub>3</sub>	2	5.37	0.37
	N <sub>4</sub>	2	5.39	0.39
<b>P4/mmm-MnN<sub>4</sub></b>	Mn <sub>1</sub>	1	11.76	-1.24
	N <sub>1</sub>	4	5.31	0.31
<b>P <math>\bar{1}</math>-MnN<sub>5</sub></b>	Mn <sub>1</sub>	2	11.90	-1.10
	N <sub>1</sub>	2	5.34	0.34

	N <sub>2</sub>	2	5.34	0.34
	N <sub>3</sub>	2	5.25	0.25
	N <sub>4</sub>	2	5.19	0.19
	N <sub>5</sub>	2	4.99	-0.01
<b>C2/m-MnN<sub>6</sub></b>	Mn <sub>1</sub>	2	11.70	-1.30
	N <sub>1</sub>	8	5.23	0.23
	N <sub>2</sub>	4	5.19	0.19
<b>P <math>\bar{1}</math>-MnN<sub>8</sub></b>	Mn <sub>1</sub>	1	11.72	-1.28
	N <sub>1</sub>	2	5.22	0.22
	N <sub>2</sub>	2	5.04	0.04
	N <sub>3</sub>	2	5.16	0.16
	N <sub>4</sub>	2	5.18	0.18