

Supporting Information

Identification of intermediates of a molecular ruthenium catalyst for water oxidation using *in-situ* electrochemical X-ray absorption spectroscopy

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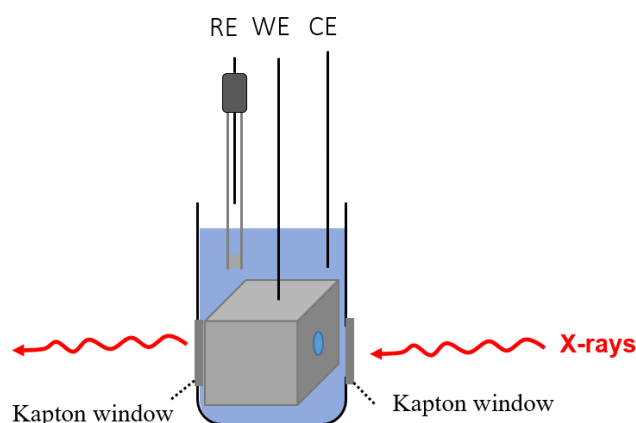


Figure S1. Illustration of in situ electrochemical cell. The cell contains a Kapton film as a window, which allows X-ray pass through. The working electrode (WE), reference electrode (RE), and counter electrode (CE) are glassy carbon foam, Ag/AgCl, and Pt respectively.

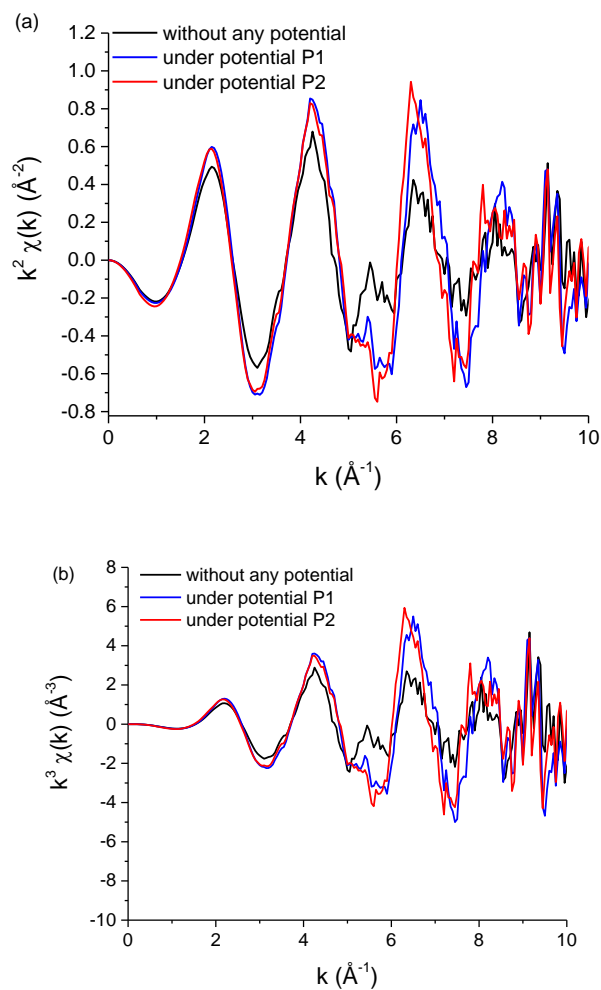


Figure S2. (a) k^2 -weighted and (b) k^3 -weighted EXAFS data $k^2\chi(k)$ of complex 1 in 0.1 M HNO_3 under no controlled potential (black line), under potential P1 (blue line) and potential P2 (red line).

Table S1. Best-fit parameters obtained from the EXAFS Ru K-edge of complex 1 in 0.1 M HNO_3 under no controlled potential and under potentials P1 and P2

Fit	Shell, N^a	R^b , \AA	$\sigma^2(10^{-3})^c$	R-factor	Reduced χ -square
No potential	Ru—N, 4	2.17	4	0.01	165
	Ru—O, 2	2.42	2		
P1	Ru—N, 4	2.12	13	0.03	382
	Ru—O, 3	2.17			
P2	Ru—O3, 1	2.02	10	0.07	233
	Ru—O2, 2	2.10			
	Ru—N3, 4	2.15			

^acoordination number per Ru center, ^bRu-backscatter distance, ^cDebye-Waller factor. R-factor and reduced χ -square are the goodness of the fit parameters. Amplitude reduction factor $S_0^2 = 1$ was used in all the fits.

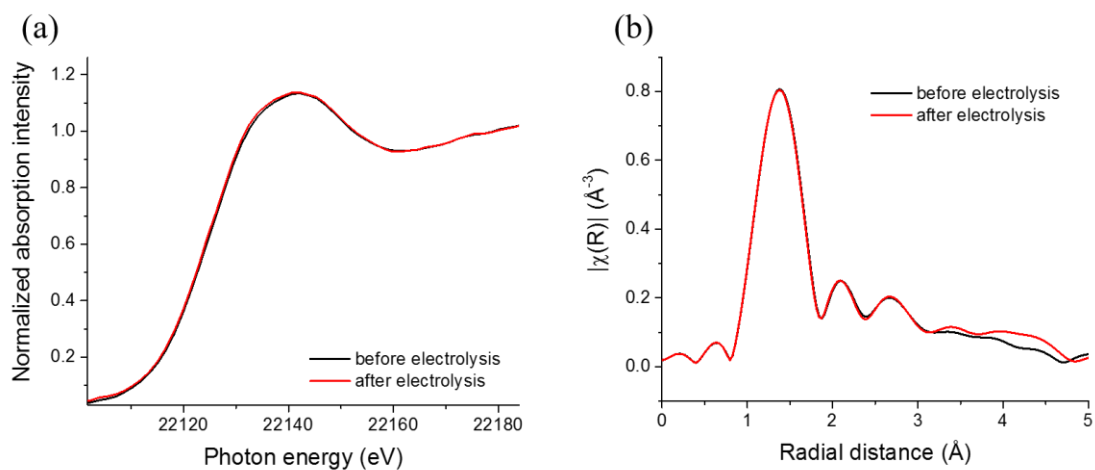


Figure S3 (a). Normalized XANES at Ru K-edge and (b) Fourier transformed EXAFS spectra of complex 1 in 0.1 M HNO₃ before and after electrolysis.

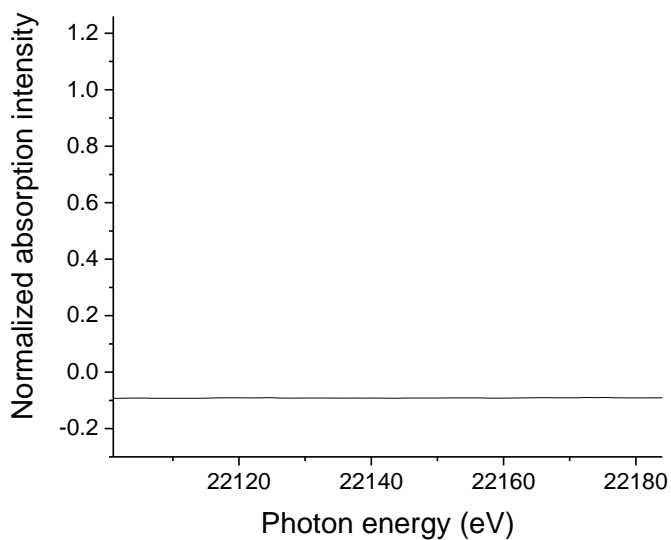


Figure S4. Normalized XANES at Ru K-edge and working electrode 0.1 M HNO₃ after experiment.

Table S2. Fit parameters obtained from EXAFS Ru K-edge of complex 1 in 0.1 M HNO₃ under no controlled potential, under potential P1 and potential P2.

Fit	Shell, N	R, Å	$\sigma^2(10^{-3})$	ΔE_0	ΔR	R _{eff}	R-factor	Reduced χ -square
under no controlled potential The crystal structure used for EXAFS fitting is from reference 1 ¹ 1-7.3 k-space, 1-2.4 R-space								
1	Ru—N, 6	2.19	25	2.36	0.10	2.09	0.08	609
2	Ru—N1, 1	2.07	3	7.47	0.16	1.91	0.14	
	Ru—N2, 1	2.11			0.16	1.94		
	Ru—N4, 2	2.24			0.17	2.07		

	Ru—O2, 1	2.39			0.18	2.17		
	Ru—O3, 1	2.35			0.18	2.21		
3	Ru—N, 4	2.17	5	7.47	0.23	1.95	0.02	581
	Ru—O, 2	2.42			0.25	2.17		
4	Ru—N, 4	2.17	5	7.85	0.23	1.95	0.02	381
	Ru—O, 2	2.42	3		0.25	2.17		
0.8-8 k-space, 1-2.4 R-space								
5	Ru—N, 4	2.17	4	7.67	0.23	1.95	0.01	165
	Ru—O, 2	2.42	2		0.25	2.17		
under no controlled potential The crystal structure used for EXAFS fitting is from reference 2 ² 1-7.3 k-space, 1-2 R-space								
6	Ru—N, 6	2.18	25	2.57	0.08	2.10	0.06	1337
7	Ru—N1, 2	2.06	3	7.47	0.14	1.92	0.12	2400
	Ru—N2, 2	2.22				2.08		
	Ru—O1, 2	2.33				2.32		
8	Ru—N2, 4	2.17	15	6.94	0.08	2.08	0.08	1876
	Ru—O, 2	2.27				2.18		
9	Ru—N1, 4	2.18	3	8.57	0.25	1.92	0.02	375
	Ru—O, 2	2.44		8.57		2.18		
under potential P1 1-7.3 k-space, 1-2 R-space								
10	Ru—N, 6	2.11	12				0.071	5631
under potential P1 The crystal structure used for EXAFS fitting is from reference 2 ² 1-10 k-space, 1-1.86 R-space								
11	Ru—N2, 2	2.09	9	4.61	0.09	1.99	0.03	408
	Ru—O2, 1	2.14				2.04		
	Ru—O3, 1	2.14				2.04		
	Ru—N3, 2	2.17				2.07		
12	Ru—N2, 2	2.09	13	5.03	0.10	1.99	0.03	385
	Ru—O2, 2	2.14				2.04		
	Ru—O3, 1	2.15				2.04		
	Ru—N3, 2	2.18				2.07		
13	Ru—N2, 1	2.08	10	4.61	0.08	1.99	0.03	417
	Ru—O2, 2	2.12				2.04		
	Ru—O3, 1	2.13				2.04		
	Ru—N3, 2	2.16				2.07		
14	Ru—N2, 2	2.09	10	4.61	0.09	1.99	0.03	417
	Ru—O2, 2	2.14				2.04		
	Ru—O3, 1	2.14				2.04		
	Ru—N3, 1	2.17				2.07		
15	Ru—N2, 4	2.12	13	5.03	0.12	1.99	0.03	382
	Ru—O2, 3	2.17				2.04		

16	Ru—N2, 4 Ru—O2, 2	2.12 2.17	9	4.89	0.12	1.99 2.04	0.03	420
17	Ru—N2, 3 Ru—O2, 3	2.11 2.15	10	4.89	0.11	1.99 2.04	0.03	429
18	Ru—O2, 3 Ru—N3, 4	2.13 2.16	14	5.34	0.08	2.04 2.07	0.03	392
19	Ru—O2, 2 Ru—N3, 4	2.11 2.15	10	4.61	0.07	2.04 2.07	0.03	406
20	Ru—O2, 3 Ru—N3, 3	2.11 2.15	10	4.61	0.07	2.04 2.07	0.03	414
21	Ru—N2, 4 Ru—O3, 3	2.12 2.17	13	5.03	0.12	1.99 2.04	0.03	382
22	Ru—N2, 4 Ru—O3, 2	2.12 2.17	8	4.89	0.12	1.99 2.04	0.03	419
23	Ru—N2, 3 Ru—O3, 3	2.11 2.16	9	4.89	0.11	1.99 2.04	0.03	429
under potential P1 The crystal structure used for EXAFS fitting is from reference 2 ² 1-10 k-space, 1-1.92 R-space								
24	Ru—N2,4 Ru—O2,2	2.08 2.19	5	3.51 3.69	0.005	2.08 2.19	0.064	597
25	Ru—N2,4 Ru—O2,3	2.07 2.18	8	3.44 3.61	- 0.005	2.08 2.19	0.064	556
under potential P1 The crystal structure used for EXAFS fitting is from reference 3 ³ 1-10 k-space, 1-1.92 R-space								
26	Ru—N1, 2 Ru—O1, 2 Ru—N2, 2	2.06 2.12 2.14	7	4.07 4.20 4.24	0.07	1.99 2.05 2.07	0.062	597
27	Ru—N1, 2 Ru—O1, 3 Ru—N2, 2	2.06 2.13 2.15	11	4.20 4.33 4.37	0.07	1.99 2.05 2.07	0.061	552
28	Ru—N1, 4 Ru—O1, 2	2.1 2.12	9	4.41 4.45	0.05	2.05 2.07	0.064	605
29	Ru—N1, 4 Ru—O1, 3	2.09 2.15	10	4.32 4.45	0.09	1.99 2.05	0.062	554

under potential P2								
The crystal structure (Ru ^{IV} dimer) used for EXAFS fitting is from reference 1								
1-7.6 k-space, 1-1.92 R-space								
30	Ru—N, 6	2.10	9	1.14	- 0.002	2.10	0.016	1981
31	Ru—O3, 1 Ru—O2, 2 Ru—N3, 4	2.01 2.09 2.13	9	7.47	0.050 0.052 0.053	1.96 2.03 2.08	0.025	3418
32	Ru—O3, 1 Ru—O2, 1 Ru—N3, 4	2.00 2.08 2.13	6	7.07	0.042 0.044 0.045	1.96 2.03 2.08	0.018	3414
under potential P2								
The crystal structure (Ru ^{IV} dimer) used for EXAFS fitting is from reference 1								
1-10 k-space, 1-1.92 R-space								
33	Ru—O3, 1 Ru—O2, 2 Ru—N3, 4	2.02 2.10 2.15	10	8.01	0.063 0.065 0.067	1.96 2.03 2.08	0.079	233

N, coordination number for per Ru center. R, Ru-backscatter distance. σ^2 , Debye-Waller factor. ΔE_0 , the energy shift parameter. ΔR , change in half-path length. R_{eff} , initial half path length. R-factor and Reduced χ -square are the goodness of the fit parameters. Amplitude reduction factor $S_0^2 = 1$ was used in all fits.

References:

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2. Duan, L.; Wang, L.; Inge, A. K.; Fischer, A.; Zou, X.; Sun, L., *Inorg. Chem.* **2013**, *52*, 7844-7852.
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