Electronic Supplementary Information for Structural Adaptations in Bovine Serum Albumin Protein in Archetypal Deep Eutectic Solvent Reline and Its Aqueous Mixtures

Monika Kumari, Pratibha Kumari, and Hemant K. Kashyap*

Department of Chemistry, Indian Institute of Technology Delhi, Hauz Khas, New Delhi 110016, India

> E-mail: hkashyap@chemistry.iitd.ac.in Phone: +91-(0)11-26591518. Fax: +91-(0)11-26581102

S1 Force-Field Validation

Methodology: Table S1 summarizes the compositions of reline-water mixtures studied. PACKMOL¹ was used to randomly pack the molecules in cubic simulation boxes. The parameters for reline components ([Ch]⁺, Cl⁻, and urea) were taken from CHARMM general force-field² as used in our previous studies^{3,4} while water molecules were considered as TIP3P model.⁵ All the simulation have been carried out for ~60 ns in NPT ensemble using GROMACS 5.1.1.⁶ The temperature was maintained to 303 K using Nosé-Hoover thermostat⁷⁻⁹ and the pressure was kept constant at 1 bar using Parrinello-Rahman barostat.¹⁰ The equations of motion were integrated at a timestep of 1 fs. Periodic boundary conditions were applied in all the directions. A cutoff distance of 1.2 nm was considered for short-range interactions. The long-range electrostatics were evaluated using PME.^{11,12} The last 10 ns trajectory with frame saved at every 100 fs has been utilized to calculate density and X-ray scattering structure function S(q). The X-ray structure functions have been calculated as per the previous literature.^{4,13,14}

System	Reline (wt%)	Water (wt%)	$[Ch]^+/Cl^-$ ion pairs	urea	water
rel-1w	93.5	6.5	1000	2000	1000
rel-2w	87.8	12.2	1000	2000	2000
rel-4w	78.3	21.7	1000	2000	4000
rel-10w	59.0	41.0	1000	2000	10000
rel-20w	41.9	58.1	1000	2000	20000
water	0.0	100	-	-	1000

Table S1: Summary of the system compositions studied.

Results and Discussion: Figure S1 illustrates the comparison of bulk densities of relinewater mixtures at varying compositions in presence of TIP3P water with the previously reported SPC/E model.⁴ A very good agreement is observed at all the studied concentrations. Figure S2 displays the comparative plots of S(q)s of reline-water mixtures at varying concentrations. A good overlap can be noticed from the plots in presence of both the water models. However, a slight difference in the S(q)s of pure SPC/E and TIP3P water can be noted with respect to splitting of the principal peak, which is a known fact for TIP3P water model.⁵ Apart from this, Figure S2 also conveys that in corroboration with the SPC/E water model, TIP3P water also shows the DES structural transition at and above 41 wt% of water (rel-10w). The molecular arrangement of reline is changed from "water-in-reline" to an aqueous solution of reline components i.e. below 41 wt%, all the S(q)s are similar to that of pure reline but at and above 41 wt% of water the S(q)s resemble pure water structure function. Recently, this structural transition at 41 wt% has also been observed experimentally through 2D IR spectroscopy,¹⁵ which further validates the computational findings.



Figure S1: Comparison plots for densities of reline-water mixtures in presence of SPC/E⁴ and TIP3P water models. Density data for SPC/E water model has been reprinted with permission from ref.⁴ Copyright 2018 American Chemical Society.



Figure S2: Comparison of simulated X-ray structure functions for reline-water mixtures using SPC/E^4 and TIP3P water model. X-ray scattering data for SPC/E water model has been reprinted with permission from ref.⁴ Copyright 2018 American Chemical Society.

Table S2:	Summary	of all the	systems	investigated	for re	versibility (of structura
modulatio	$\mathbf{ons.}^*$						

System	Denatured BSA conformation taken from	water
R50	50/50 reline/water	50706
R75	75/25 reline/water	60072
R100	pure reline	36576

*Each system was neutralized using 17 K^+ ions.

Table S3: Average percentage of BSA secondary structures for 350 ns reversibility runs calculated using DSSP algorithm. Last 100 ns trajectories were used for the computation of average values.

System	α -helix	3_{10} -helix	coil	turn	bend
R50	60.1	3.0	18.3	8.2	10.3
R75	46.3	3.2	23.5	14.1	12.4
R100	52.8	8.3	18.9	14.0	10.8



Figure S3: Time-dependent variations in radius of gyration (R_g) on aqueous re-equilibration of denatured protein conformations. Aqueous system data are also plotted for the comparison.



Figure S4: Time evolution of percentage α -helical content for re-equilibration runs of denatured protein systems.

References

- Martínez, L.; Andrade, R.; Birgin, E. G.; Martínez, J. M. Packmol: A Package for Building Initial Configurations for Molecular Dynamics Simulations. J. Comput. Chem. 2009, 30, 2157–2164.
- (2) Vanommeslaeghe, K.; Hatcher, E.; Acharya, C.; Kundu, S.; Zhong, S.; Shim, J.; Darian, E.; Guvench, O.; Lopes, P.; Vorobyov, I. et al. CHARMM general force field: A force field for drug-like molecules compatible with the CHARMM all-atom additive biological force fields. J. Comput. Chem. 2010, 31, 671–690.
- (3) Kaur, S.; Sharma, S.; Kashyap, H. K. Bulk and Interfacial Structures of Reline Deep Eutectic Solvent: A Molecular Dynamics Study. J. Chem. Phys. 2017, 147, 194507.
- (4) Kumari, P.; Shobhna,; Kaur, S.; Kashyap, H. K. Influence of Hydration on the Structure of Reline Deep Eutectic Solvent: A Molecular Dynamics Study. ACS Omega 2018, 3, 15246–15255.
- (5) Jorgensen, W. L.; Chandrasekhar, J.; Madura, J. D.; Impey, R. W.; Klein, M. L. Comparison of Simple Potential Functions for Simulating Liquid Water. J. Chem. Phys. 1983, 79, 926–935.
- (6) van der Spoel, D.; Lindahl, E.; Hess, B.; Groenhof, G.; Mark, A. E.; Berendsen, H. J. C. GROMACS: Fast, Flexible, and Free. J. Comput. Chem. 2005, 26, 1701–1718.
- (7) Nose, S. A Unified Formulation of the Constant Temperature Molecular Dynamics Methods. J. Chem. Phys. 1984, 81, 511–519.
- (8) Nose, S. A Molecular Dynamics Method for Simulations in the Canonical Ensemble. Mol. Phys. 1984, 52, 255 – 268.
- (9) Hoover, W. G. Canonical Dynamics: Equilibrium Phase-Space Distributions. *Phys. Rev. A* 1985, *31*, 1695.

- (10) Parrinello, M.; Rahman, A. Polymorphic Transitions in Single Crystals: A New Molecular Dynamics Method. J. Appl. Phys. 1981, 52, 7182–7190.
- (11) Essmann, U.; Perera, L.; Berkowitz, M. L.; Darden, T.; Lee, H.; Pedersen, L. G. A Smooth Particle Mesh Ewald Method. J. Chem. Phys. 1995, 103, 8577–8593.
- (12) Darden, T.; York, D.; Pedersen, L. Particle Mesh Ewald: An Nlog(n) Method for Ewald Sums in Large Systems. J. Chem. Phys. 1993, 98, 10089–10092.
- (13) Kashyap, H. K.; Hettige, J. J.; Annapureddy, H. V. R.; Margulis, C. J. SAXS Anti-Peaks Reveal the Length-Scales of Dual Positive-Negative and Polar-Apolar Ordering in Room-Temperature Ionic Liquids. *Chem. Commun.* **2012**, *48*, 5103–5105.
- (14) Kaur, S.; Kashyap, H. K. Unusual Temperature Dependence of Nanoscale Structural Organization in Deep Eutectic Solvents. J. Phys. Chem. B 2018, 122, 5242–5250.
- (15) Sakpal, S. S.; Deshmukh, S. H.; Chatterjee, S.; Ghosh, D.; Bagchi, S. Transition of a Deep Eutectic Solution to Aqueous Solution: A Dynamical Perspective of the Dissolved Solute. J. Phys. Chem. Lett. 2021, 12, 8784–8789.