

Supplementary information

Defect-Rich Boron doped Carbon Nanotube as Electrocatalyst for Hybrid Li-air Battery

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Catalogue:

1. Experimental preparation
2. Calculation of DFT
3. Analysis of experimental results

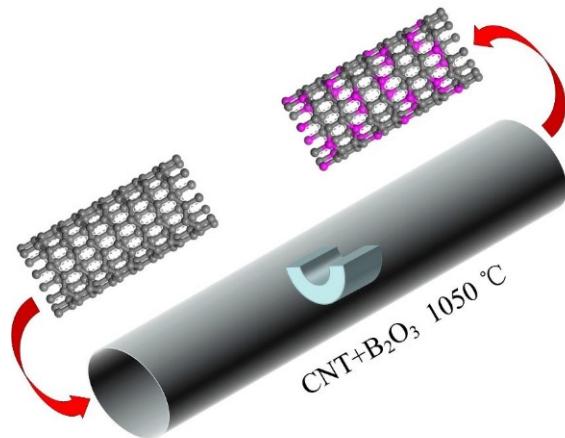
1. Experimental preparation

Fig. S1 Schematic illustration of synthesis process of BC₃NT.

Table S1 List of chemicals

Name	Purity	Manufacturer
LiOH	95%	Sinopharm Group
CNT	99. 9%	Beijing Deke Daojin Science and Technology Co., LTD
PTFE		Sinopharm Group
NMP	AR	Sinopharm Group
B ₂ O ₃	99%	Sinopharm Group
LISICON		Shenzhen Kejing Star Technology Co.
LITFSI-TEGDME		NJ Scientific.

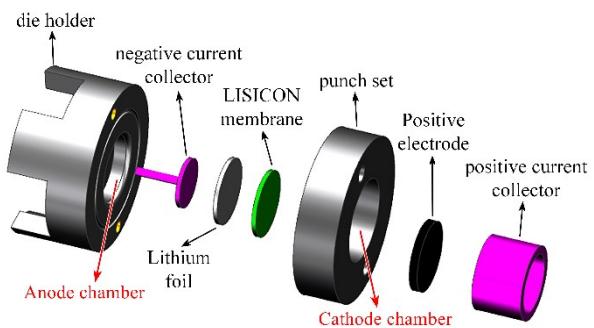


Fig. S2 Illustrate of HLAB mold.

2. Calculation of DFT

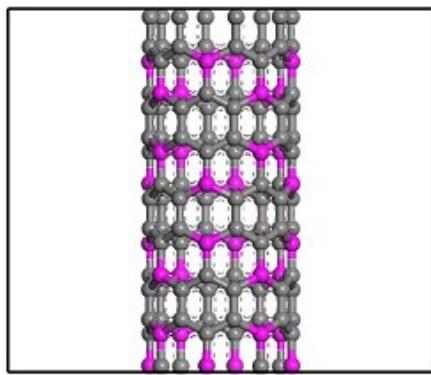


Fig. S3 Model of nanotube.

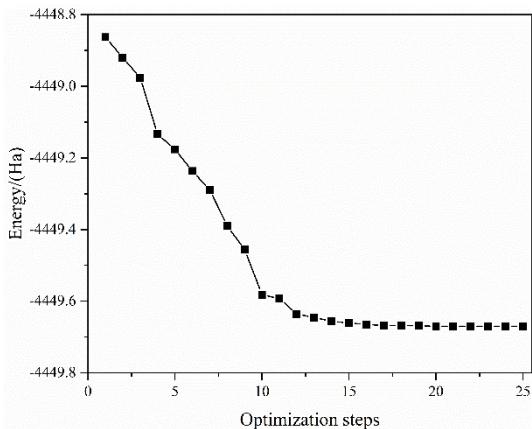
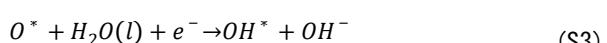
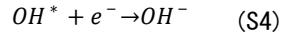


Fig. S4 Energy convergence diagram of optimization process.

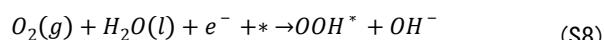
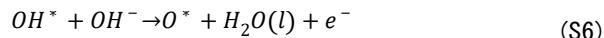
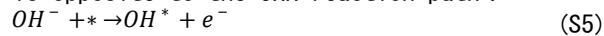
Under alkaline conditions, the ORR and OER four-electron reaction path is in the ESI. †

$$O_2(g) + H_2O(l) + e^- \rightarrow OOH^* + OH^- \quad (\text{S1})$$




* represents the active site on the catalyst surface, and (l) and (g) represent the liquid and gas phases, respectively.

The OER reaction path is opposite to the ORR reaction path :



3. Analysis of experimental results

Table S2 Content of element

	B ₂ O ₃ : CNT=2:1	B ₂ O ₃ : CNT=5:1	B ₂ O ₃ : CNT=8:1
C	92.20%	87.86%	88.58%
B	2.84%	4.68%	5.15%
O	4.10%	5.83%	2.98%

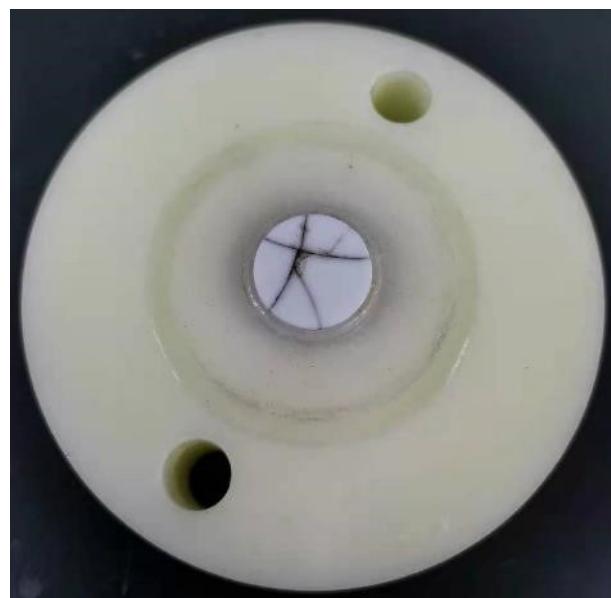


Fig. S5 LISICON membrane after test.

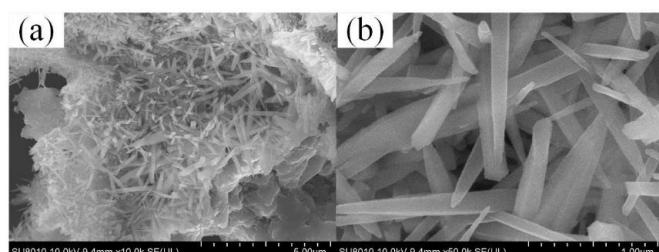


Fig. S6 Lithium dendrites.

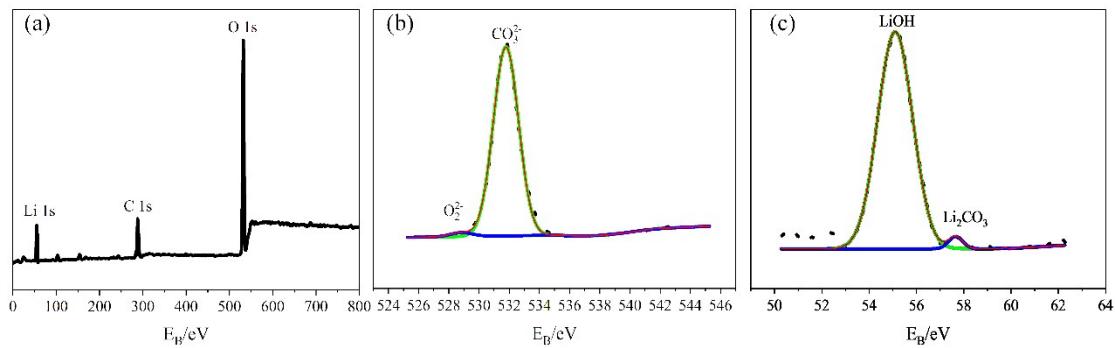


Fig. S7 XPS spectrum: (a)Total, (b)O 1s, (c)Li 1s.

Table S3 Comparison of battery performance

Catalyst	Cycling number	Initial discharge capacity (mAh·g ⁻¹)	Overpotential (V)	Ref
Boron doped CNT	165	8900	0.3	This work
Pt–carbon paper	50		0.75	Sun ¹
MnO–CNT–CNFFs	21		0.15 (low current)	Ji ²
Mn ₃ O ₄	20	221 (0.5 mA·cm ⁻²)	1.3	Li ³
Graphite			0.47	Zhou ⁴

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