Supporting Information

Hydrothermal Synthesis, Crystal Structures, and X-ray Photoelectron Spectroscopy of Lead Tellurium(IV) and Tellurium(VI) Oxycompounds: Ba₃PbTe₆O₁₆ and Na₂Pb₉(μ₆-O)₂(Te₂O₁₀)₂

Han-Yin Li^a and Kwang-Hwa Lii*,a,b

^{*a*}Department of Chemistry, National Central University, Zhongli, Taiwan 320, R.O.C. ^{*b*}Institute of Chemistry, Academia Sinica, Taipei, Taiwan 115, R.O.C.

Figure S1. Energy dispersive X-ray spectroscopy analysis on a crystal of Ba₃PbTe₆O₁₆.

Figure S2. Pawley fit of the powder diffraction data to the structure of Ba₃PbTe₆O₁₆.

Figure S3. Energy dispersive X-ray spectroscopy analysis on a crystal of $Na_2Pb_9(\mu_6-O)_2(Te_2O_{10})_2$.

Figure S4. Pawley fit of the powder diffraction data to the structure of $Na_2Pb_9(\mu_6-O)_2(Te_2O_{10})_2$.

Figure S5. The infrared spectrum of Ba₃PbTe₆O₁₆ (KBr method).

Figure S6. The infrared spectrum of $Na_2Pb_9(\mu_6-O)_2(Te_2O_{10})_2$ (KBr method).

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0 1 Full Scale 82 Spectrum Spectrum 1 Spectrum 2	2 28 cts 0 70.96 69.76	Ba 8.56 8.86	Te Te 17.38 18.23	бо 6 7 Рb 3.1 3.14	8	9	10 keV

Figure S1. Energy dispersive X-ray spectroscopy analysis on a crystal of Ba₃PbTe₆O₁₆.



Figure S2. Pawley fit of the powder diffraction data to the structure of Ba₃PbTe₆O₁₆. The small peaks marked with an arrow correspond to the impurity Ba₃Te₄O₁₁.



Figure S3. Energy dispersive X-ray spectroscopy analysis on a crystal of $Na_2Pb_9(\mu_6-O)_2(Te_2O_{10})_2$



Figure S4. Pawley fit of the powder diffraction data to the structure of $Na_2Pb_9(\mu_6-O)_2(Te_2O_{10})_2$.



Figure S5. The infrared spectrum of Ba₃PbTe₆O₁₆ (KBr method).



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