

Supplementary information for

Temperature sensitivity modulation through vanadium concentration in La₂MgTiO₆: V⁵⁺, Cr³⁺ double perovskite optical thermometer

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Table S1. The calculated molar of precursors used in the syntheses.

Sample (0.5 g)	La(CH ₃ COO) ₃	Mg(NO ₃) ₂	Ti(C ₃ H ₇ O) ₄	NH ₄ VO ₃	Cr(NO ₃) ₃
LMT	2.24E-03	1.23E-03	1.1212E-03	0.00E+00	0
LMT:0.05V	2.24E-03	1.23E-03	1.1206E-03	5.61E-07	0
LMT:0.1V	2.24E-03	1.23E-03	1.1201E-03	1.12E-06	0
LMT:0.5V	2.24E-03	1.23E-03	1.1156E-03	5.61E-06	0
LMT:1V	2.24E-03	1.23E-03	1.1100E-03	1.12E-05	0
LMT:2V	2.24E-03	1.23E-03	1.0988E-03	2.24E-05	0
LMT:0.1Cr	2.24E-03	1.23E-03	1.1201E-03	0	1.12E-06
LMT:0.1V, 0.1Cr	2.24E-03	1.23E-03	1.1189E-03	1.12E-06	1.12E-06
LMT:2V, 2Cr	2.24E-03	1.23E-03	1.0768E-03	2.24E-05	2.24E-05

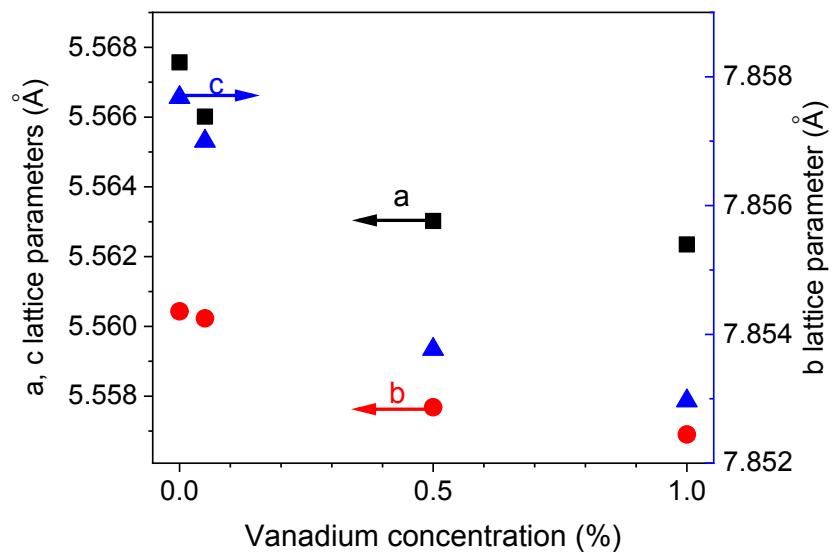


Fig. S1 Unit cell *a*, *b*, *c*, parameters of the $\text{La}_2\text{MgTiO}_6$ in the function of vanadium concentration.

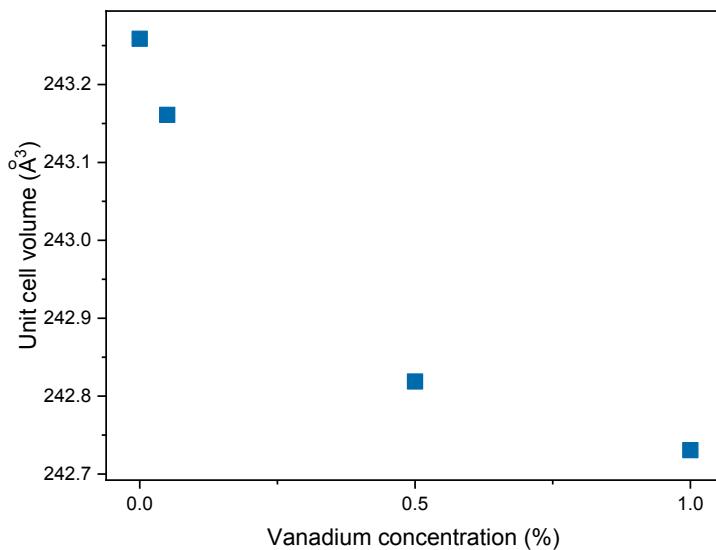


Fig. S2 Unit cell volume of the $\text{La}_2\text{MgTiO}_6$ in the function of vanadium concentration.

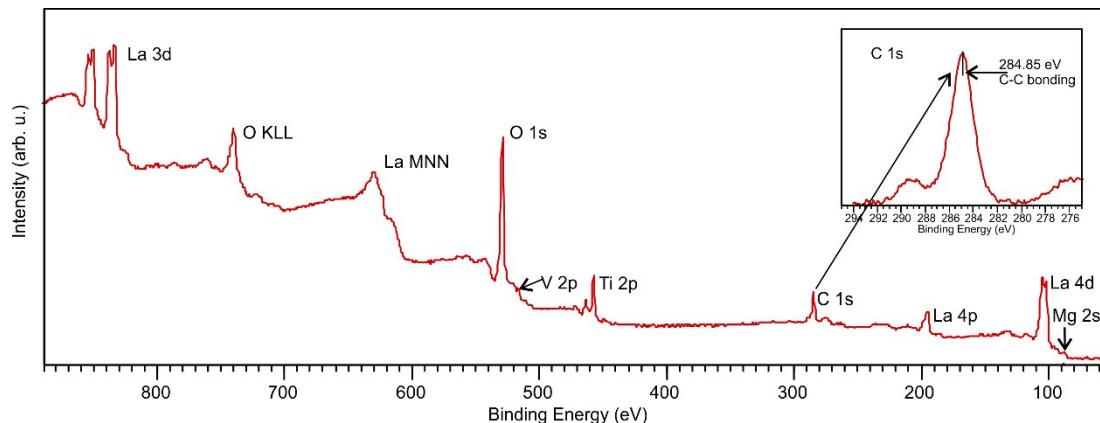


Fig. S3 Broad XPS spectrum for $\text{La}_2\text{MgTiO}_6\cdot\text{V}$ with core level lines of $\text{La } 3\text{d}_{5/2}$, $\text{O } 1\text{s}$, $\text{Ti } 2\text{p}_{3/2}$, $\text{Mg } 2\text{s}$.

Table S2 Positions and full width at half maxima (FWHM) of main core level lines.

State	Position (eV)	FWHM (eV)
$\text{La } 3\text{d}_{5/2}$	833.95	6.6
$\text{O } 1\text{s}$	529.35	2.75
$\text{Ti } 2\text{p}_{3/2}$	458.05	1.85
$\text{Mg } 2\text{s}$	87.65	2.2
$\text{V } 2\text{p}_{3/2} \text{ V(V)}$	516.7	1.7
$\text{V } 2\text{p}_{3/2} \text{ V(III)}$	515.25	2.0

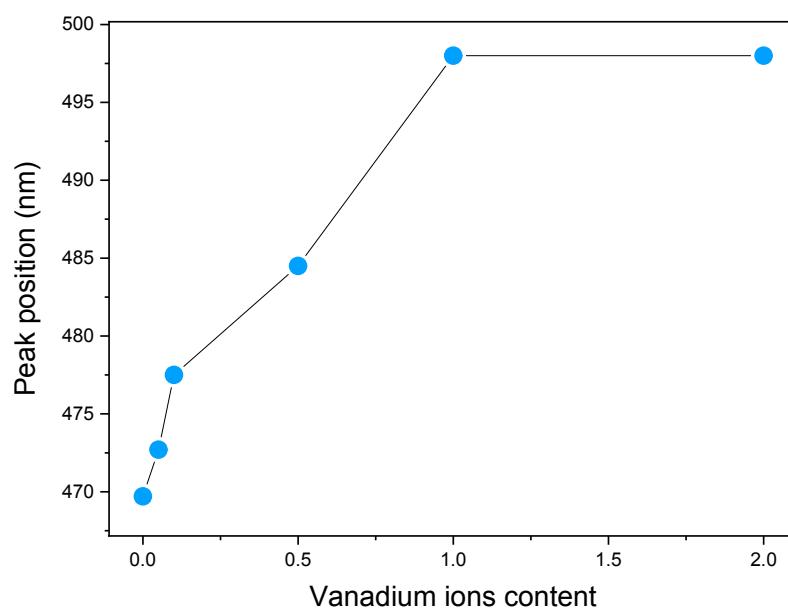


Fig. S4 Shift of maximum emission band with vanadium concentration.

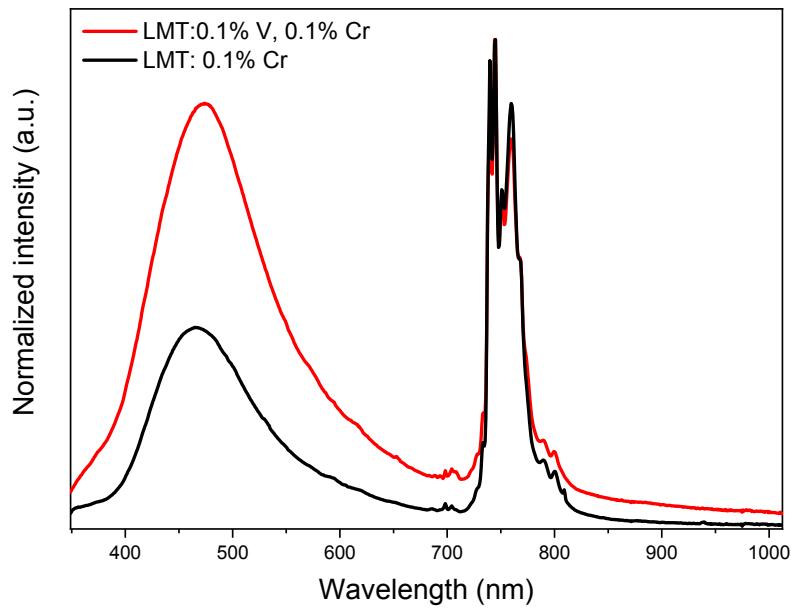


Fig. S5 Normalized emission spectra of $\text{La}_2\text{MgTiO}_6:0.1\%$ V, 0.1% Cr and $\text{La}_2\text{MgTiO}_6:0.1\%$ Cr.

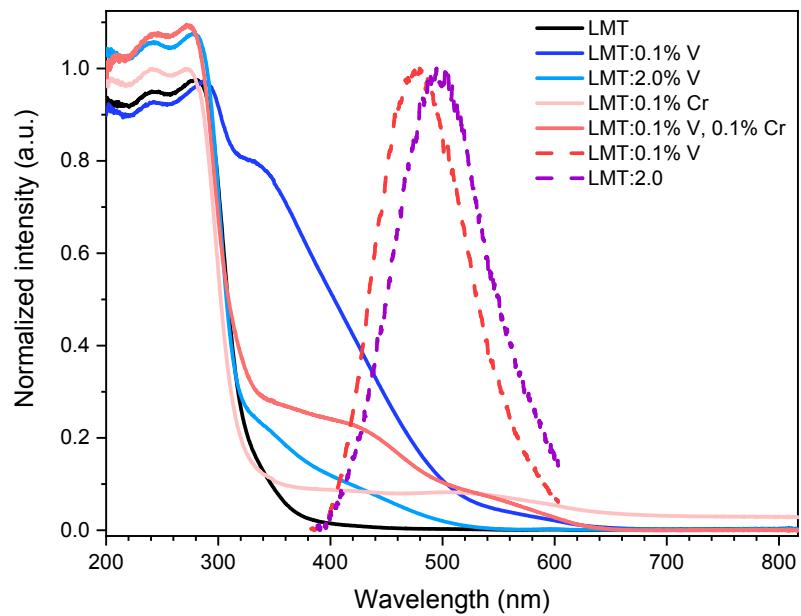


Fig. S6 Absorption (solid line) and emission (dashed line) of investigated samples.

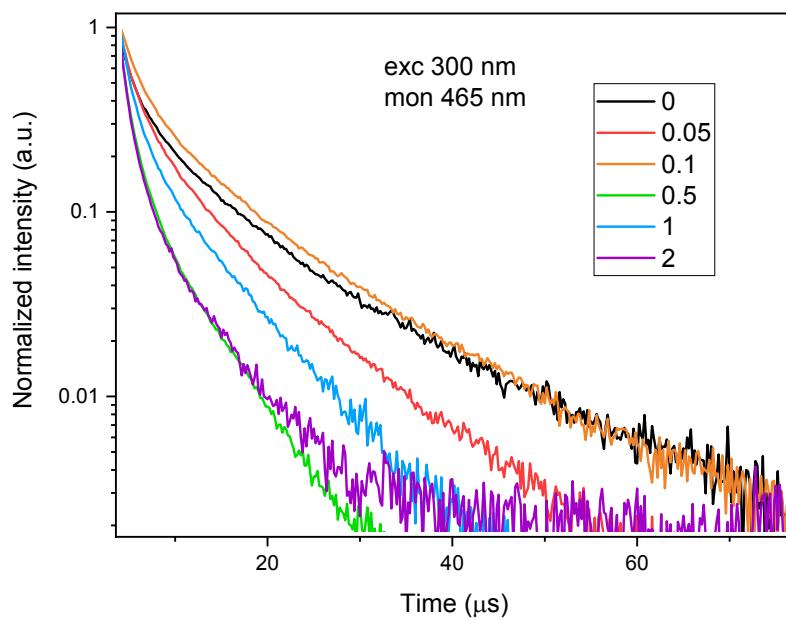


Fig. S7 Emission Decay curve of $\text{La}_2\text{MgTiO}_6$ and samples doped with vanadium ions monitored at 465 nm.

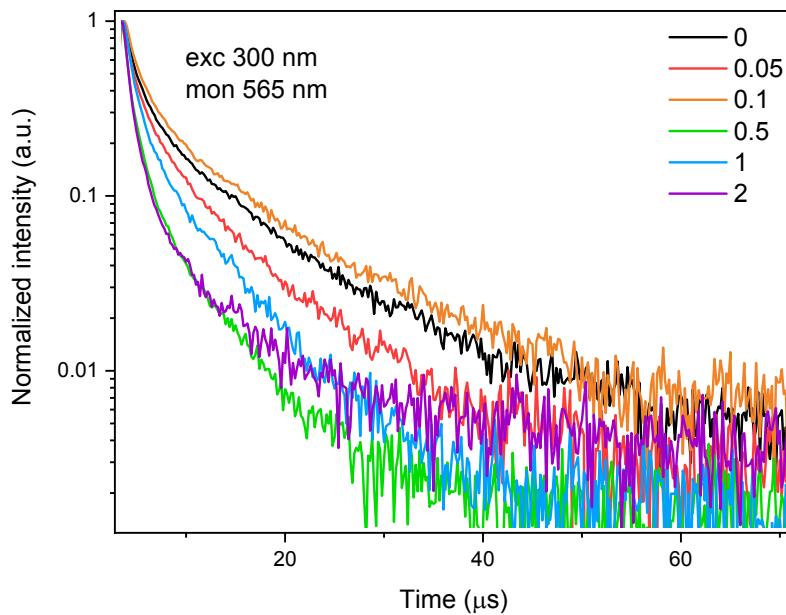


Fig. S8 Emission Decay curves of $\text{La}_2\text{MgTiO}_6$ and samples doped with vanadium ions monitored at 565 nm.

Table S3 Emission decay times of investigated samples.

LMT: xV	mon 465 nm		mon 565 nm	
	τ_1 [μs]	τ_2 [μs]	τ_1 [μs]	τ_2 [μs]
0	1.78	10.93	-	-
0.05	1.57	8.33	0.98	5.95

0.1	2.02	11.13	1.27	8.51
0.5	0.91	4.75	0.66	3.53
1.0	1.13	6.55	0.87	5.37
2.0	0.83	4.23	0.63	3.61

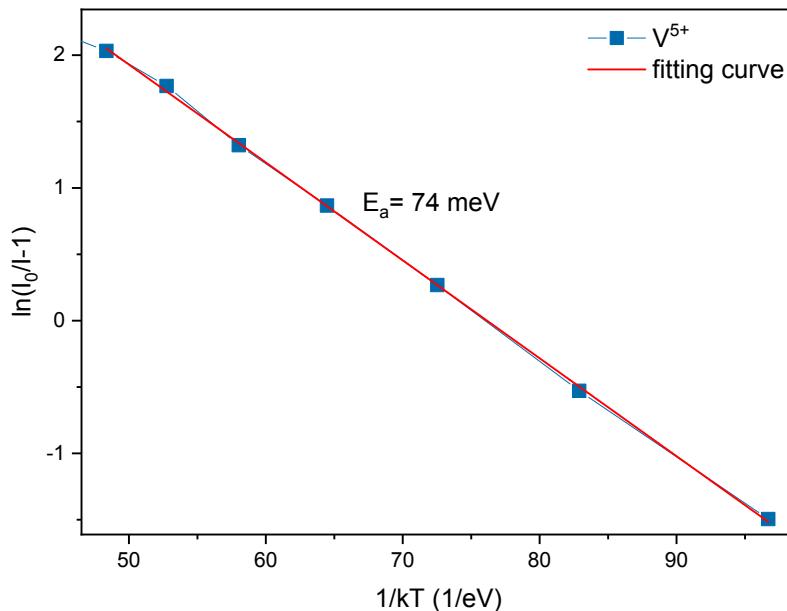


Fig. S9 Activation energies (E_a) for thermal quenching of V^{5+} emission in $\text{La}_2\text{MgTiO}_6$, calculated from a function of $\ln(I_0/I-1)$ versus $1/kT$.

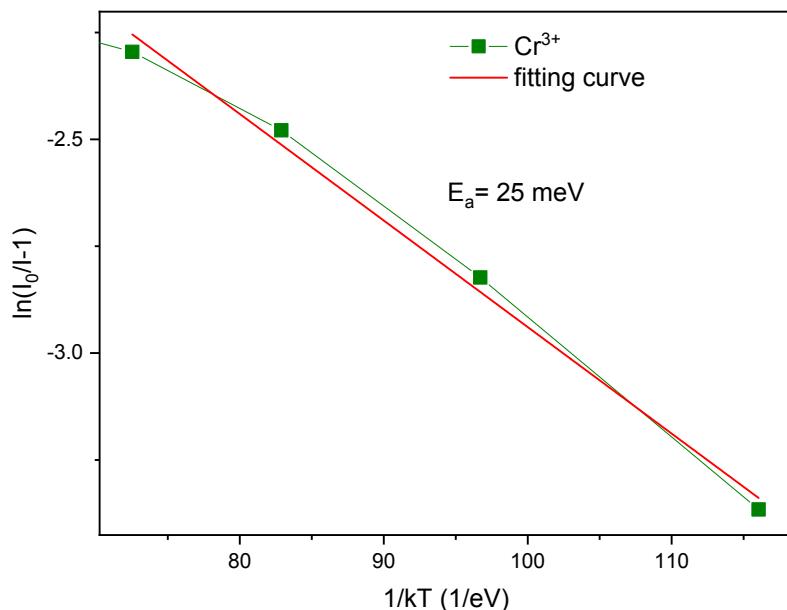


Fig. S10 Activation energies (E_a) for thermal quenching of Cr^{3+} emission in $\text{La}_2\text{MgTiO}_6$, calculated from a function of $\ln(I_0/I-1)$ versus $1/kT$.

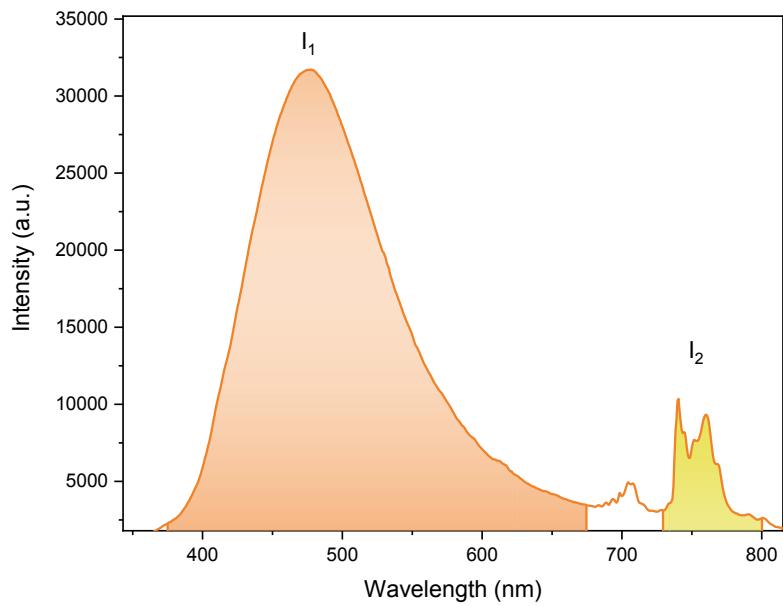


Fig. S11 Integration areas of representative sample $\text{La}_2\text{MgTiO}_6$: 0.1% V.

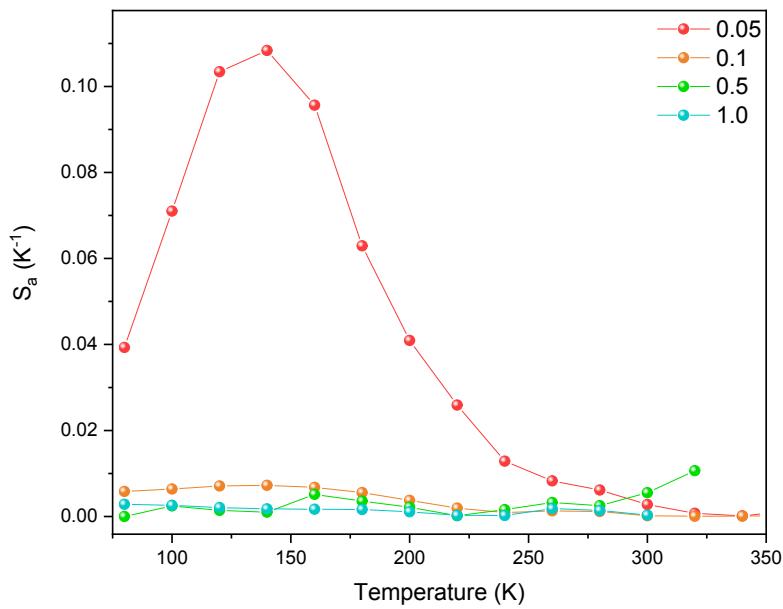


Fig. S12 Absolute sensitivity of $\text{La}_2\text{MgTiO}_6$:x% V.

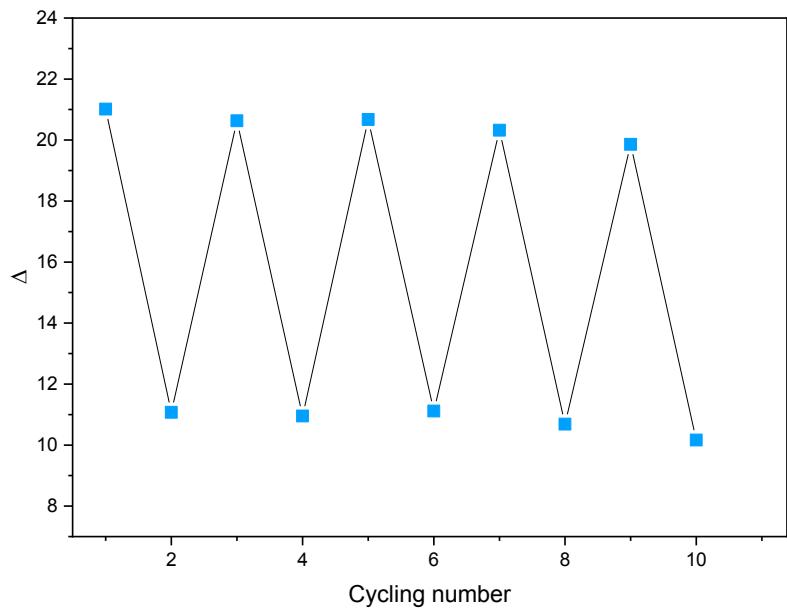


Fig. S13 Repeatability of Δ temperature parameter of I_1/I_2 emission evaluated at 80 K and 150 K during 10 heating/cooling cycles.