Electronic Supplementary Information

Titanium Complexes of Pyrrolylaldiminate Ligands and Their Exploitation for the Ring-Opening Polymerization of Cyclic Esters

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Fig. S1 ¹H NMR spectrum of **1** in CDCl₃ at 298 K.



Fig. S2 ¹H NMR spectrum of 1 in toluene-d₈ at 298 K.



Fig. S3 ¹H NMR spectrum of **1** in toluene- d_8 at 363 K.





Fig. S4 ¹H NMR spectra of 1 in toluene- d_8 at different temperatures.



Fig. S5 Relative free energies of five possible stereoisomers of complex 1 calculated using the SMD(toluene)/M062X method with the 6-311+G(d,p) basis set for nonmetal atoms and the def2-tzvpp for Ti.

С

T	R
4	

Ti	-0.3211	-0.6790	-1.2592	Ti	-0.609668	-0.740380	0.417992	Ti	-0.2013	-0.8132	-1.2476
N	-2 3716	-0 3447	-1.0908	N	-0.624811	-1.085510	-1 726669	N	-2 1922	-0.2427	-1 0874
Ċ	-2 8582	0.0982	0.1175	C	-0.421916	-2 407275	-2 059597	C	-2 6230	0.2207	0.1355
C	2 4208	0.0902	1 0125	C	0.929700	0.420868	2.057577	C C	2 2448	0.1026	1 0240
Č	4 2454	-0.4636	-1.9123	C	-0.838709	-0.429808	-2.809203	C	-3.2440	-0.1930	-1.9249
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Č	2 7900	-1 0987	-0 7688	Č	2 634565	-1 321408	0.096614	Č	-0.6503	2 4355	-1 4019
C C	3 5086	1.0260	0.6515	C C	3 311014	0.734001	0.504604	Č	1 3074	3 1200	0.7752
c	2.0220	0.2102	-0.0513	C	3.311014	0.734001	-0.304004	C	1.39/4	2 5745	1.0676
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C	1 1002	2 5 9 2 0	1 2607	C II	1.252527	2.007727	0.011010	C II	2 0005	1 5010	1 0054
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Č	1 6870	-0.3363	-4 4200	Č	1 020458	-1 268///8	4.038851	č	1 2073	0.1466	-1 1718
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	0.1930 0.1970	27640	0.2370		1 200575	0.024005	0.574103		7 5270	2 5 407	0.3420
C C	-2.18/3	-3./040	-0.1322	C	-4.3805/5	0.034005	-0.029109	C	-2.55/9	-3.540/	-0.1019
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н	-0.1224	-5.0995	1.1711	н	-5.112575	-2.630319	-0.291192	н	-0.9105	-5,1743	1.3003
**		2.0775		11	2	_	J	11	0.7105	2.27.13	1.0000

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Ti 0.057797	-0.219898	-0.021294	Ti	-0.159192	-0.527454	0.156698
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0 0 205891	-1 820644	-0.832559	Ő	-0.450697	-2 180209	-0 529/3/
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C -0.292869	-2.098382	-1.81/8/2	C	-0.940989	-2.995925	-1.563269
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н 1704005	3 025067	2 501222	11 LT	1 017001	3 805272	2 021472
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п 1.1616/4	-4.238634	-1.421552	H	0.075397	-4./0303/	-0.8/2114
н 0.521159	-4.260809	-3.079105	Н	-0.333913	-4.798184	-2.600887



Fig. S6 ¹H NMR spectrum of 2 in CDCl₃ at 298 K.



Fig. S7 ¹H NMR spectrum of **3** in CDCl₃ at 298 K.



Fig. S8 ¹H NMR spectrum of 4 in CDCl₃ at 298 K.



Fig. S9 ¹H NMR spectrum of **5** in CDCl₃ at 298 K.





Fig. S10 ¹H NMR spectrum of 6 in CDCl₃ at 298 K.

Empirical formula	$C_{26}H_{40}N_4O_2Ti$
Formula weight	488.52
Temperature/K	100
Crystal system	monoclinic
Space group	P2 ₁ /c
$a/{ m \AA}$	8.9560(9)
$b/{ m \AA}$	16.7630(17)
$c/{ m \AA}$	18.1781(19)
$\alpha/^{\circ}$	90
$eta/^{\circ}$	75.196(3)
γ/°	90
Volume/Å ³	2638.5(5)
Ζ	4
$ ho_{ m calc} { m g/cm}^3$	1.230
μ/mm^{-1}	2.966
<i>F</i> (000)	1048.0
Crystal size/mm ³	$0.04 \times 0.02 \times 0.02$
Radiation	$CuK\alpha$ ($\lambda = 1.54178$)
2Θ range for data collection/°	7.288 to 137.29
Index ranges	$-10 \le h \le 10, -20 \le k \le 20, -21 \le l \le 21$
Reflections collected	42047
Independent reflections	4688 [$R_{int} = 0.0705, R_{sigma} = 0.0426$]
Data/restraints/parameters	4688/144/357
Goodness-of-fit on F^2	1.219
Final <i>R</i> indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.1454, wR_2 = 0.3409$
Final R indexes [all data]	$R_1 = 0.1465, wR_2 = 0.3410$
Largest diff. peak/hole / e Å ⁻³	0.73/-0.58

 Table S1 Crystal data and structure refinement for the complex 2.



Fig. S11 A view of the C–H…C interactions (C16–H16A…C16 and C12–H12…C16) forming between two cyclopentyl moieties representing weak Van der Waals forces between molecules of **2**.



Fig. S12 A view of the chain of the molecules of **2** related by a translational symmetry illustrating the C–H···*C*_g interactions (C3–H3B···*C*_g). Note that, *C*_g is the centroid of the pyrrole ring (N1 and C7 to C10).



Fig. S13 Concentration versus time profile for the ¹H NMR resonance decay of *rac*-LA (\blacktriangle) and the growth of PLA (\blacktriangledown) for the ROP of *rac*-LA by complex **2** at 70 °C along with the fits (blue and red lines) and errors determined by COPASI.



Fig. S14 Concentration versus time profile for the ¹H NMR resonance decay of *rac*-LA (\blacktriangle) and the growth of PLA (\blacktriangledown) for the ROP of *rac*-LA by complex **3** at 70 °C along with the fits (blue and red lines) and errors determined by COPASI.



Fig. S15 Concentration versus time profile for the ¹H NMR resonance decay of *rac*-LA (\blacktriangle) and the growth of PLA (\blacktriangledown) for the ROP of *rac*-LA by complex **4** at 70 °C along with the fits (blue and red lines) and errors determined by COPASI.



Fig. S16 Concentration versus time profile for the ¹H NMR resonance decay of *rac*-LA (\blacktriangle) and the growth of PLA (\blacktriangledown) for the ROP of *rac*-LA by complex **5** at 70 °C along with the fits (blue and red lines) and errors determined by COPASI.



Fig. S17 Concentration versus time profile for the ¹H NMR resonance decay of *rac*-LA (\blacktriangle) and the growth of PLA (\blacktriangledown) for the ROP of *rac*-LA by complex **6** at 70 °C along with the fits (blue and red lines) and errors determined by COPASI.



Fig. S18 Concentration versus time profile for the ¹H NMR resonance decay of *rac*-LA (\blacktriangle) and the growth of PLA (\blacktriangledown) for the ROP of *rac*-LA by complex **1** at 100 °C along with the fits (blue and red lines) and errors determined by COPASI.



Fig. S19 Concentration versus time profile for the ¹H NMR resonance decay of *rac*-LA (\blacktriangle) and the growth of PLA (\triangledown) for the ROP of *rac*-LA by complex **2** at 100 °C along with the fits (blue and red lines) and errors determined by COPASI.



Fig. S20 Concentration versus time profile for the ¹H NMR resonance decay of *rac*-LA (\blacktriangle) and the growth of PLA (\triangledown) for the ROP of *rac*-LA by complex **3** at 100 °C along with the fits (blue and red lines) and errors determined by COPASI.



Fig. S21 Concentration versus time profile for the ¹H NMR resonance decay of *rac*-LA (\blacktriangle) and the growth of PLA (\triangledown) for the ROP of *rac*-LA by complex **4** at 100 °C along with the fits (blue and red lines) and errors determined by COPASI.



Fig. S22 Concentration versus time profile for the ¹H NMR resonance decay of *rac*-LA (\blacktriangle) and the growth of PLA (\blacktriangledown) for the ROP of *rac*-LA by complex **5** at 100 °C along with the fits (blue and red lines) and errors determined by COPASI.



Fig. S23 Concentration versus time profile for the ¹H NMR resonance decay of *rac*-LA (\blacktriangle) and the growth of PLA (\triangledown) for the ROP of *rac*-LA by complex **6** at 100 °C along with the fits (blue and red lines) and errors determined by COPASI.



Fig. S24 Homonuclear decoupled ¹H NMR spectrum of the methine region of the PLA prepared from *rac*-LA at 70 °C in toluene (400 MHz, CDCl₃) using complex **1**.



Fig. S25 Homonuclear decoupled ¹H NMR spectrum of the methine region of the PLA prepared from *rac*-LA at 70 °C in toluene (400 MHz, CDCl₃) using complex **2**.



Fig. S26 Homonuclear decoupled ¹H NMR spectrum of the methine region of the PLA prepared from *rac*-LA at 70 °C in toluene (400 MHz, CDCl₃) using complex **3**.



Fig. S27 Homonuclear decoupled ¹H NMR spectrum of the methine region of the PLA prepared from *rac*-LA at 70 °C in toluene (400 MHz, CDCl₃) using complex **4**.



Fig. S28 Homonuclear decoupled ¹H NMR spectrum of the methine region of the PLA prepared from *rac*-LA at 70 °C in toluene (400 MHz, CDCl₃) using complex **5**.



Fig. S29 Homonuclear decoupled ¹H NMR spectrum of the methine region of the PLA prepared from *rac*-LA at 70 °C in toluene (400 MHz, CDCl₃) using complex **6**.



Fig. S30 Homonuclear decoupled ¹H NMR spectrum of the methine region of the PLA prepared from *rac*-LA at 100 °C in toluene (400 MHz, CDCl₃) using complex **1**.



Fig. S31 Homonuclear decoupled ¹H NMR spectrum of the methine region of the PLA prepared from *rac*-LA at 100 °C in toluene (400 MHz, CDCl₃) using complex **2**.



Fig. S32 Homonuclear decoupled ¹H NMR spectrum of the methine region of the PLA prepared from *rac*-LA at 100 °C in toluene (400 MHz, CDCl₃) using complex **3**.



Fig. S33 Homonuclear decoupled ¹H NMR spectrum of the methine region of the PLA prepared from *rac*-LA at 100 °C in toluene (400 MHz, CDCl₃) using complex **4**.



Fig. S34 Homonuclear decoupled ¹H NMR spectrum of the methine region of the PLA prepared from *rac*-LA at 100 °C in toluene (400 MHz, CDCl₃) using complex **5**.



Fig. S35 Homonuclear decoupled ¹H NMR spectrum of the methine region of the PLA prepared from *rac*-LA at 100 °C in toluene (400 MHz, CDCl₃) using complex **6**.



Fig. S36 Plot of PCL $M_n(\bullet)$ (versus polystyrene standards) and PDI (O) as a function of monomer conversion for a ε -CL polymerisation using **5** ([ε -CL]₀/[Ti] = 100, toluene, 70 °C).



Fig. S37 Concentration versus time profile for the ¹H NMR resonance decay of ε -CL (\blacktriangle) and the growth of PCL (\triangledown) for the ROP of ε -CL by complex **1** at 70 °C along with the fits (blue and red lines) and errors determined by COPASI.



Fig. S38 Concentration versus time profile for the ¹H NMR resonance decay of ε -CL (\blacktriangle) and the growth of PCL (\triangledown) for the ROP of ε -CL by complex **2** at 70 °C along with the fits (blue and red lines) and errors determined by COPASI.



Fig. S39 Concentration versus time profile for the ¹H NMR resonance decay of ε -CL (\blacktriangle) and the growth of PCL (\triangledown) for the ROP of ε -CL by complex **3** at 70 °C along with the fits (blue and red lines) and errors determined by COPASI.



Fig. S40 Concentration versus time profile for the ¹H NMR resonance decay of ε -CL (\blacktriangle) and the growth of PCL (\triangledown) for the ROP of ε -CL by complex **4** at 70 °C along with the fits (blue and red lines) and errors determined by COPASI.



Fig. S41 Concentration versus time profile for the ¹H NMR resonance decay of ε -CL (\blacktriangle) and the growth of PCL (\triangledown) for the ROP of ε -CL by complex **5** at 70 °C along with the fits (blue and red lines) and errors determined by COPASI.



Fig. S42 Concentration versus time profile for the ¹H NMR resonance decay of ε -CL (\blacktriangle) and the growth of PCL (\triangledown) for the ROP of ε -CL by complex **6** at 70 °C along with the fits (blue and red lines) and errors determined by COPASI.



Fig. S43 Semilogarithmic plots of $\ln[CL]_0/[CL]_t$ versus time for ϵ -CL polymerization using complex **5** as an initiator at different temperatures in toluene ([ϵ -CL]_0/[Ti] = 100, [ϵ -CL]_0 = 1.25 M. **I**: T = 90 °C, $k_{app} = (32.32 \pm 0.26) \times 10^{-4} \text{ s}^{-1}$; **II**: T = 80 °C, $k_{app} = (18.71 \pm 0.13) \times 10^{-4} \text{ s}^{-1}$; **III**: T = 70 °C, $k_{app} = (9.12 \pm 0.68) \times 10^{-4} \text{ s}^{-1}$; **IV**: T = 60 °C, $k_{app} = (32.32 \pm 0.26) \times 10^{-4} \text{ s}^{-1}$).

Table S2 Kinetic data for the ε -CL	polymerization	s using	complex 5
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Τ (°C)	T (K)	1/T (K ⁻¹)	k _{app} (10 ⁴ s ⁻¹)	$k_{ m p}$	$\ln(k_p/T)$
60	333	0.003003	5.04 ± 0.22	0.040316	-9.01915
70	343	0.002915	9.12 ± 0.68	0.073680	-8.44575
80	353	0.002833	18.71 ± 0.13	0.149675	-7.76576
90	363	0.002755	32.32 ± 0.26	0.258578	-7.24696