Supporting Information

Cobalt Coordination Controlled Carbon Nanospheres Formation and Inclusion of Amorphous Co₃O₄ and AuNPs: Strongly Enhanced Oxygen Evolution Reaction with Excellent Mass Activity

Pandi Muthukumar,^a Shreya Narasimhan,^a Arunprasanth Panneer Selvam,^b Mariappan Mariappan,^b Mohammed A. Assiri^c and Savarimuthu Philip Anthony^{*a}

^{a)}Department of Chemistry, School of Chemical & Biotechnology, SASTRA Deemed

University, Thanjavur-613401, Tamil Nadu, India. Fax: +914362264120;

Tel: +914362264101, E-mail: philip@biotech.sastra.edu.

^{b)}Department of Chemistry, SRM IST, Kattankulathur, Chennai-603203, Tamil Nadu, India.

^{c)}Department of Chemistry, King Khalid University, Abha 61413, Saudi Arabia.



Figure S1. Absorption spectra of HBA, HBA@AuNPs, Co²⁺ ions and coordination with HBA ligand.



Figure S2. PXRD pattern of 1, 2 and 3.



Figure S3. HR-TEM images of AuNPs synthesised using HBA.



Figure S4. HR-TEM images (a, b) and EDX spectra (c) of **1**.



Figure S5. HR-TEM images of **2**.



Figure S6. HR-TEM images of **3**.



Figure S7. FE-SEM images of **2**.



Figure S8. XPS survey spectra of 1-3.



Figure S9. High resolution XPS spectra of C, N and O of 1.



Figure S10. High resolution XPS spectra of C, N and O of **2**.



Figure S11. High resolution XPS spectra of C, N and O of **3**.



Figure S12. Elemental mapping of **1**.



Figure S13. Elemental mapping of **2**.



Figure S14. Elemental mapping of **3**.



Figure S15. EDX spectra of (a) 1, (b) 2 and (c) 3.

Catalyst	Element	wt% by EDAX	
1	Au	0.01	
	Со	0.46	
2	Au	0.04	
	Со	1.74	
3	Au	0.05	
	Со	2.86	



Figure S16. Linear sweep OER polarization curve of 2 before and after iR correction.

Catalyst	Overpotential (mV)	Tafel slope (mV/dec)	Stability	Reference	
2	256	52.4	1000 cycle & 48 h	Present work	
Au/NiCo ₂ O ₄	360	63	1000 cycle & 120 min	1	
Au/m Co ₃ O ₄	440	46	2000 cycle	2	
MnO ₂ / AuNP-4.4	390	~200	NA	3	
CNTs-Au@Co ₃ O ₄	350	68	1000 cycle & 25 h	4	
AuNCs@Ni(OH) ₂	375 73		600 cycle & 2 h	5	
AuNCs@Co(OH) ₂	350	72	600 cycle	6	
AuNDs@LDH/GCE	530	53 2000 seconds		7	
NiCeOOH/Au	259		NA	8	
NiFeOOH/Au	267			8	
ZnCo ₂ O ₄ /Au/CNTs	440	46.2	10 h	9	
AuCuCo	596	160	4 h	10	
$Au_{0.89}Fe_{0.11} NPs$	800	163	NA	11	
Au@NiO _x	394	117	2 h	12	

Table S2	OER	data	of	different	catalysts
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Au/FeO _x	439	43	2 h	12
Au@CoFeO _x	328	58	2 h	12
Au-Ni ₁₂ P ₅	340	49	2.78 h	13
Au-Co(OH) ₂	320	119	6 h	14
Au25/CoSe ₂	430	NA	1000 cycle	15
Au-Ru NPs	220	62	NA	16
AuNPs@LDH	510	61	1.5 h	17
Au/NiFe LDH	237	36	2000 cycle & 20	18
			h	
AuNi-Cu ₂ O	532	NA	10 h	19
(after 2 nd OER test)				
AuNi HDs	350	45.9	2 h	20
Au–Fe _x O _y 12	450	132	5 h	21
$Au_{0.10}Ir_{0.90}O_{y}$ -50	241	55.2	5000 cycle	22
Au/Ir NCs	300	52.94	36 h	23
Au-CoFe ₂ O ₄	312	35	8 h	24
3DG-Au-Ni3S2	370 (91.15 j)	106	NA	25
Au-vanadate nanoflute	310	127	NA	26
GMN@Co _x S _y	$MN@Co_xS_y$ 345		6000 seconds	27



Figure S17. Comparison of linear sweep OER polarization curves of **2** synthesised at three different batches.



Figure S18. Comparison of linear sweep OER polarization curves of 2,4 and 5.



Figure S19. PXRD of 2 after heating at 500 °C.



Figure S20. Linear sweep OER polarization curve of **2** after heating at 500 °C.



Figure S21. Linear sweep OER polarization curve of 2.



Figure S22. (a) Chronopotentiometric response and (b) OER polarization curve of RuO₂.



Figure S23. High resolution XPS spectra of (a) Co and (b) Au of 2 after catalytic studies.



Figure S24. (a) CV of **1**, **2** and **3**.



Figure S25. (a) CV of **4** and **5** and capacitive currents as a function of the scan rate for **2** and **4**.

Catalyst	Loading (mg)	Overpotential (mV)	Current density (mA/cm ²)	Mass activity (mA mg ⁻¹)	Ref
2	0.05	300	67.62	1352.5	This work
α-Ni(OH) ₂	0.2	350	30.02	150.1	28
γ-CoOOH	0.15	300	10	66.6	29
CuCo ₂ S ₄	0.7	310	10	14.29	30
NiFe-LDH/CuO NRs/CF	0.70	300	~105	150	31
Ba2CoMo0.5Nb0.5O6-8	0.232	440	~10	~43	32
LaCo _{0.8} Fe _{0.2} O _{3-δ} -700	0.245	293	10	40.80	33
Electrochemically activated Co _x Ni _{1-x} S ₂	0.285	340	60	217	34
CeO ₂ /Co ₃ O ₄	0.35	340	~45	128.6	35
Au/NiFe LDH	2	280	129.8	64.9	18
Co-S-130	0.17	350	~13	76	36
CoOOH-NS	2	320	~45	~22.5	37

Table S3. Comparison of mass activities of different catalysts.

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