

Electronic Supplementary Information

Significantly Improve the Lithium-Ion Conductivity of Solid-State Electrolytes via Fabricating Large Pore Volume Hollow ZIF-8

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References



Figure S1. Optical photographs of electrolyte membrane

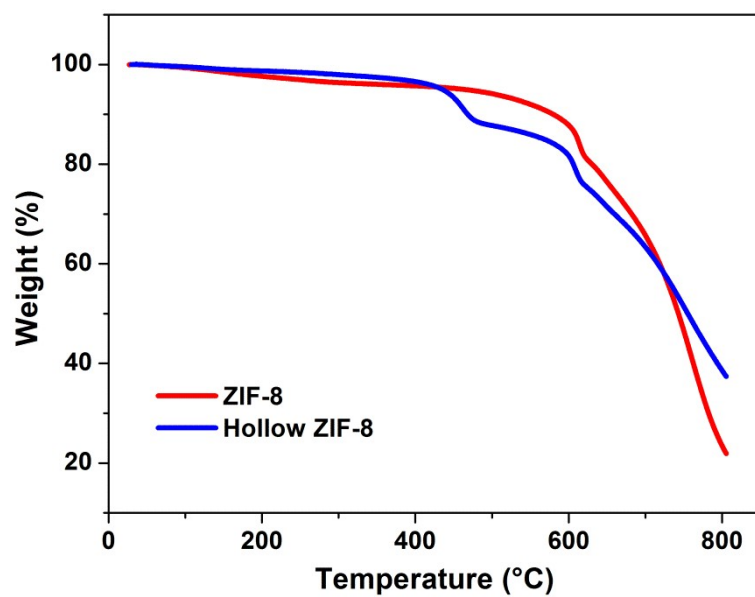


Figure S2. TGA curves of ZIF-8 and hollow ZIF-8.

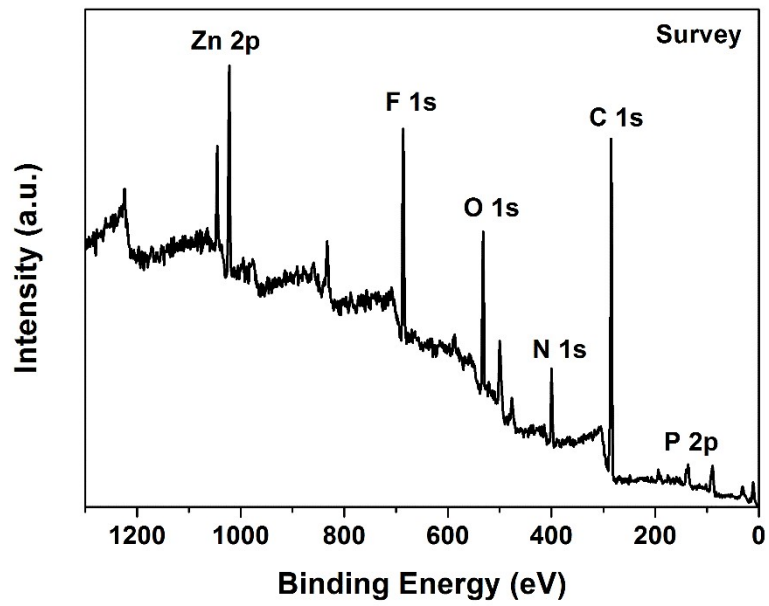


Figure S3. XPS spectra of survey of hollow ZIF-8

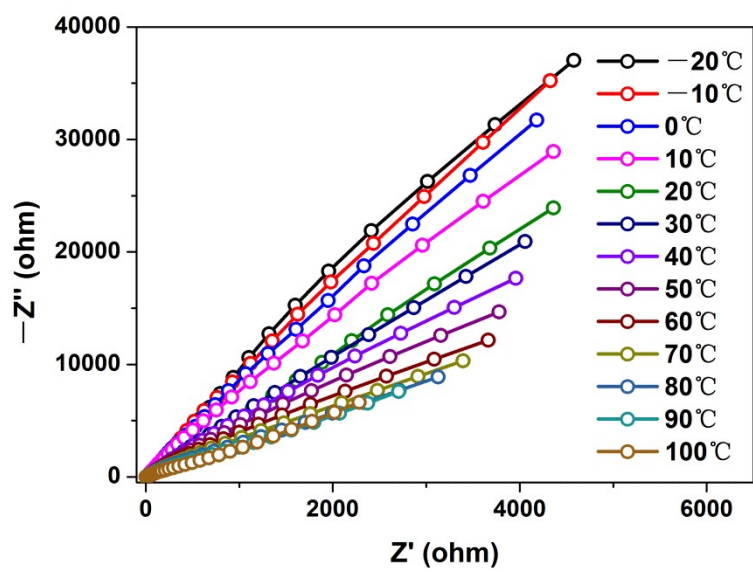


Figure S4. EIS of hollow ZIF-8 at temperature from -20 to 100°C .

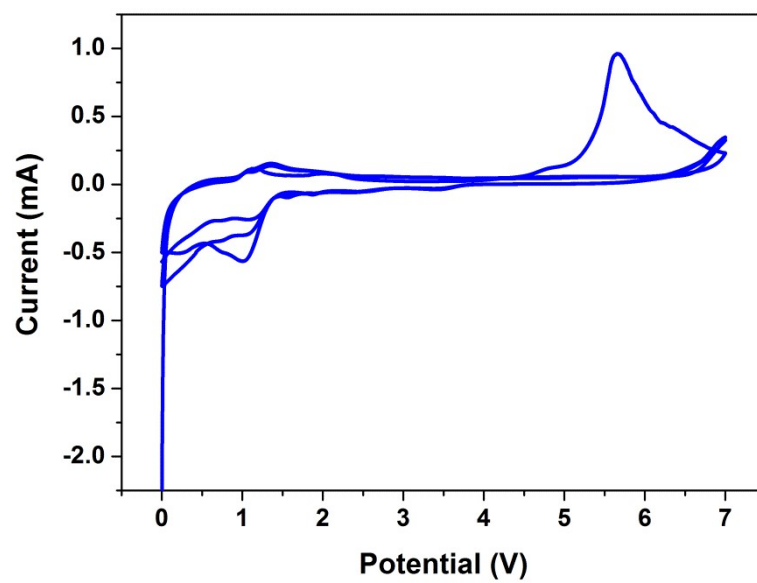


Figure S5. CV curves of ss/electrolyte membrane/Li cell of hollow ZIF-8.

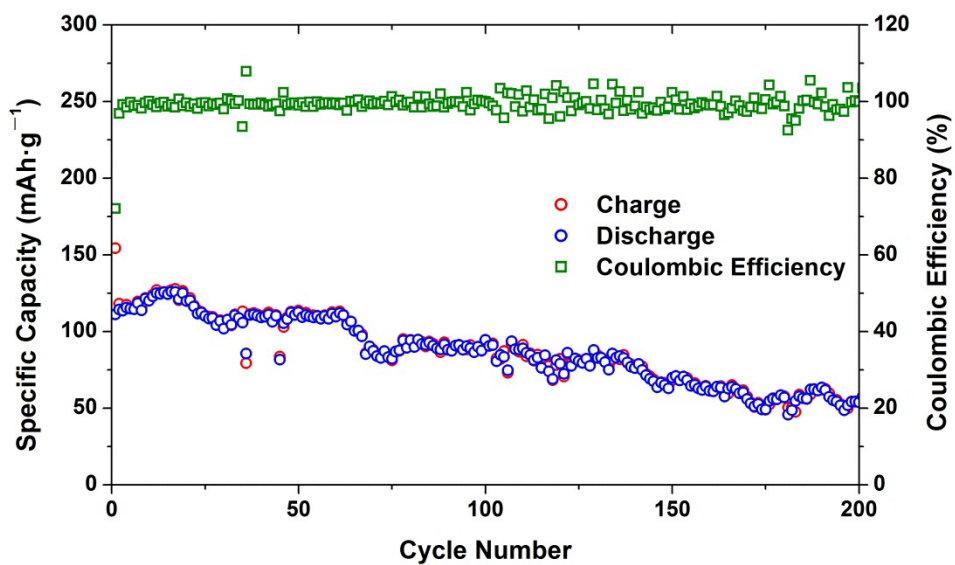


Figure S6. Specific capacity and coulombic efficiency of LiFePO₄/electrolyte membrane/Li cell of hollow ZIF-8 with 0.1 C charge/discharge rate at room temperature.

Table S1. Comparison of MOFs for solid-state lithium ion conduction at room temperature

No.	Materials	ionic conductivity (σ)	transference number (t_{Li^+})	Ref.
1	Li-IL@UIO-67-LLZO ^{a, b}	$1.0 \times 10^{-4} \text{ S} \cdot \text{cm}^{-1}$	0.13	1
2	UIO/Li-IL (15/16)	$3.2 \times 10^{-4} \text{ S} \cdot \text{cm}^{-1}$	0.33	2
3	Al-Td-MOF-1	$5.7 \times 10^{-5} \text{ S} \cdot \text{cm}^{-1}$	-	3
4	MOF-688	$3.4 \times 10^{-4} \text{ S} \cdot \text{cm}^{-1}$	0.87	4
5	Li-IL@MOF-525	$3.0 \times 10^{-4} \text{ S} \cdot \text{cm}^{-1}$	0.36	5
6	Hollow ZIF-8	$7.36 \times 10^{-4} \text{ S} \cdot \text{cm}^{-1}$	0.83	This work

^a Li-IL = lithium-containing ionic liquid

^b LLZO = $\text{Li}_7\text{La}_3\text{Zr}_2\text{O}_{12}$

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