

Dalton Transactions

Electronic Supporting information for:

Non-palindromic Anthracen-diyl Bis(alkylidynes)

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Author Contributions

AFH was responsible for funding acquisition and project administration. BJF and SSW were responsible for conducting the experiments and characterisation of the products.

All authors, listed alphabetically, contributed to the preparation of the original manuscript and subsequent drafts.



Cartesian Coordinates

1. $[Tm(CO)_2W=CC(C_6H_4)_2CC=W(CO)_2Tp]$ (5d)



Figure S1. Optimised Geometry of 5d

Ato	m x	У	z
W	-3.277594	-1.978401	0.172277
W	3.542165	4.270937	-0.972450
S	-1.579119	-3.893164	0.773741
S	-5.292282	-3.839049	0.195591
S	-3.765777	-1.639312	2.690565
0	-5.439118	0.223861	-0.467362
Ν	-6.975540	-3.297114	2.241518
0	-3.080668	-2.723407	-2.881791
Ν	3.549862	4.388941	-3.205042
С	-3.170319	-2.450892	-1.760280
Ν	4.689758	4.428200	-3.928853
Ν	-5.246197	-4.428712	2.929171
С	2.184702	3.093844	-0.729976
С	0.162445	-1.188033	-2.003320

Atom	x	У	z	
н	-0.596909	-1.946803	-1.858127	
С	-2.532715	-5.330192	0.873856	
С	0.110608	-0.007115	-1.202039	
Ν	5.298253	5.770932	-1.290375	
С	-1.920737	1.582865	1.555778	
н	-2.701342	0.840031	1.679071	
Ν	5.200047	2.841816	-1.438864	
Ν	6.119484	3.081253	-2.399136	
С	-2.674759	-2.738178	3.459161	
С	-1.890500	2.690628	2.352492	
н	-2.661829	2.836529	3.103049	
0	4.032080	4.157153	2.144421	
С	2.152422	0.748713	-2.345593	
н	2.922124	1.500567	-2.482583	
С	0.117866	3.479610	1.267903	
н	0.906863	4.214868	1.157780	
С	-6.060711	-4.186604	4.018693	
н	-5.791487	-4.536406	5.002190	
С	-0.915153	1.363169	0.564392	
С	2.232817	5.767515	-0.745941	
С	-7.132958	-3.480276	3.600021	
н	-7.989536	-3.099318	4.132284	
С	-5.816049	-3.868226	1.837874	
С	5.468142	1.638510	-0.929937	
С	-0.924365	0.203525	-0.254668	
С	6.961152	2.037361	-2.491433	
С	-7.851939	-2.520464	1.382560	
н	-8.181706	-3.126406	0.537213	
н	-7.319896	-1.645331	1.003109	
н	-8.713910	-2.203692	1.971456	
С	1.138160	0.973699	-1.364274	

Atom	x	У	z
С	3.858470	4.215060	1.002537
С	3.018558	4.432211	-5.382148
н	2.448188	4.443207	-6.297693
С	7.104255	6.570867	-2.271553
С	-0.857416	3.654049	2.205817
н	-0.845352	4.533654	2.842424
С	6.581343	1.082464	-1.564181
н	7.040966	0.124723	-1.377070
Ν	6.191362	5.585187	-2.286810
С	6.802578	7.433376	-1.232074
н	7.332815	8.326658	-0.940902
С	-4.650097	-0.587635	-0.227366
С	1.138415	2.137696	-0.552825
С	0.125876	2.332287	0.418336
С	5.658629	6.882656	-0.647696
С	4.394371	4.457595	-5.239493
Ν	-2.821572	-4.057111	3.699925
Ν	-2.547450	-6.268399	-0.101401
Ν	-1.473498	-2.363028	3.947740
Ν	-3.342219	-5.767436	1.862723
С	-0.891133	-1.043559	3.760695
н	-0.714878	-0.870384	2.695751
н	0.051900	-1.003041	4.306990
н	-1.570093	-0.275148	4.131844
С	-1.686996	-4.504802	4.349383
н	-1.581886	-5.538156	4.638068
С	-1.856157	-6.144817	-1.373028
н	-2.217837	-5.264084	-1.907496
н	-2.059238	-7.041796	-1.959673
н	-0.782407	-6.042279	-1.208790
С	-3.871546	-6.986105	1.485358

H -4.555148 -7.518398 2.126890 O 1.463303 6.620423 -0.617386 C -0.841838 -3.459886 4.496132 H 0.135864 -3.389196 4.944706	
O 1.463303 6.620423 -0.617386 C -0.841838 -3.459886 4.496132 H 0.135864 -3.389196 4.944706	
C -0.841838 -3.459886 4.496132 H 0.135864 -3.389196 4.944706	
H 0.135864 -3.389196 4.944706	
C -3.384092 -7.301990 0.265419	
H -3.544891 -8.159581 -0.367756	
C 2.158737 -0.386564 -3.102063	
H 2.939352 -0.539242 -3.841558	
B 6.065464 4.375849 -3.234978	
H 6.950789 4.386673 -4.050140	
B -3.854492 -5.074228 3.150746	
H -3.989056 -5.937074 3.991039	
C 1.153132 -1.373094 -2.922938	
H 1.174171 -2.279193 -3.521130	
C 2.535427 4.388988 -4.071865	
C -1.964026 -0.764970 -0.081039	
H 7.768092 2.052008 -3.209039	
H 4.840807 1.240391 -0.145708	
H 1.520622 4.351422 -3.703225	
H 5.185241 4.493613 -5.973960	
H 7.900480 6.587382 -3.001098	
H 5.078278 7.225332 0.197437	

2. [Tm(CO)₂W=CC(C₆H₄)₂CC=W(CO)₂Tp]⁺ [5d]⁺



Figure S2. Optimised Geometry of [5d]*			
Aton	n x	У	z
w	-3.010650	-1.915829	0.640315
W	3.420586	4.503556	-1.325056
s	-1.578261	-4.095704	0.824970
S	-5.226175	-3.314431	0.168667
S	-3.411994	-1.986993	3.117338
0	-4.950663	0.570955	0.507240
Ν	-6.945497	-2.981673	2.244639
0	-2.733033	-2.211243	-2.502062
Ν	3.533788	4.225757	-3.514361
С	-2.874227	-2.100667	-1.366407
Ν	4.710785	4.201752	-4.180262
Ν	-5.409612	-4.455062	2.712424
С	2.156018	3.256837	-0.875001
С	0.611582	-1.361522	-1.246261
н	-0.070003	-2.152450	-0.950865
С	-2.730760	-5.365608	0.571063
С	0.442559	-0.066816	-0.705925
Ν	5.048272	6.038635	-1.810056
С	-1.716260	1.812426	1.785867
н	-2.394303	1.023160	2.090205
Ν	5.181946	3.157683	-1.412401
Ν	6.137569	3.278442	-2.360449
С	-2.622678	-3.420592	3.703444

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С	-1.803545	3.058399	2.362096
н	-2.565150	3.254611	3.109821
0	3.752107	5.144058	1.762560
С	2.403773	0.653406	-1.973271
н	3.098401	1.435923	-2.257981
С	0.073504	3.825384	1.057909
н	0.769548	4.607446	0.778114
С	-6.249578	-4.351028	3.803575
н	-6.092872	-4.947612	4.687950
С	-0.730494	1.532738	0.813690
С	1.979795	5.913901	-1.439498
С	-7.203025	-3.434884	3.520563
н	-8.041219	-3.077069	4.096564
С	-5.846343	-3.602631	1.762300
С	5.522064	2.119527	-0.642826
С	-0.650210	0.234889	0.187600
С	7.069457	2.327275	-2.188142
С	-7.708857	-1.963891	1.537060
н	-8.071458	-2.359801	0.587677
н	-7.081631	-1.092259	1.344772
н	-8.553719	-1.675925	2.164323
С	1.362633	0.954661	-1.062419
С	3.643598	4.921666	0.638379
С	3.119560	3.891478	-5.690852
н	2.602007	3.732711	-6.623992
С	6.845501	6.802145	-2.831960
С	-0.907166	4.075919	1.988375
н	-0.984399	5.059395	2.439986
С	6.713941	1.553803	-1.095902
н	7.245416	0.710222	-0.684428
Ν	6.016526	5.753767	-2.711473
С	6.412518	7.809702	-1.985548
н	6.852505	8.783728	-1.840855
С	-4.255640	-0.346531	0.555234
С	1.232107	2.272654	-0.497287
С	0.193590	2.550823	0.456455

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С	5.280870	7.279235	-1.368240
С	4.484794	4.006032	-5.489332
Ν	-3.011970	-4.704700	3.608586
Ν	-2.790312	-6.090016	-0.567472
Ν	-1.472096	-3.387147	4.405216
Ν	-3.679248	-5.840206	1.404372
С	-0.699646	-2.183820	4.671125
н	-0.455604	-1.693625	3.726601
н	0.216134	-2.471484	5.187778
н	-1.274659	-1.495643	5.293798
С	-2.077336	-5.486085	4.260175
н	-2.184618	-6.557433	4.313895
С	-1.937208	-5.894304	-1.728670
н	-2.044377	-4.874103	-2.101019
н	-2.248753	-6.597985	-2.500931
н	-0.894132	-6.074632	-1.464495
С	-4.339833	-6.871951	0.766394
н	-5.145228	-7.402267	1.248417
0	1.137580	6.696293	-1.495972
С	-1.115792	-4.673585	4.753636
н	-0.230501	-4.885842	5.331258
С	-3.793929	-7.030508	-0.459508
н	-4.015584	-7.724617	-1.253933
С	2.540069	-0.613151	-2.489909
н	3.346386	-0.828669	-3.183455
в	6.044571	4.388077	-3.427718
н	6.972441	4.325309	-4.187765
В	-4.157719	-5.369905	2.800862
н	-4.487922	-6.357754	3.412573
С	1.641830	-1.629876	-2.117335
н	1.760543	-2.631406	-2.518559
С	2.567070	4.041114	-4.419041
С	-1.630387	-0.713053	0.438570
н	7.919678	2.271445	-2.851751
н	4.889059	1.838395	0.186332
н	1.533895	4.025651	-4.103624

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н	5.310881	3.962646	-6.183547
н	7.685295	6.758446	-3.509781
н	4.623528	7.721909	-0.633162

3. HC≡CC(C₆H₄)₂CC≡CH



Figure S3. Optimised Geometry of HC=CC(C₆H₄)₂CC=CH

m x	У	z
-4.599999	0.000000	1.247802
-3.656169	0.000000	0.711167
-2.475481	0.000000	2.485324
-2.480614	0.000000	1.400575
-2.480614	0.000000	-1.400575
-1.226785	0.000000	0.715336
-3.656169	0.000000	-0.711167
-1.226785	0.000000	-0.715336
0.000000	0.000000	1.412881
-4.599999	0.000000	-1.247802
-2.475481	0.000000	-2.485324
1.226785	0.000000	0.715336
2.480614	0.000000	1.400575
1.226785	0.000000	-0.715336
	 x 4.599999 3.656169 -2.475481 -2.480614 -2.480614 -1.226785 0.00000 -4.599999 -2.475481 1.226785 2.480614 1.226785 	x y 4.599999 0.00000 -3.656169 0.00000 -2.475481 0.00000 -2.480614 0.00000 -2.480614 0.00000 -1.226785 0.00000 -3.656169 0.00000 -1.226785 0.00000 -4.599999 0.00000 -2.475481 0.00000 -1.226785 0.00000 -2.475481 0.00000 -2.475481 0.00000 -2.475481 0.00000 -2.475481 0.00000 -2.475481 0.00000 -2.475481 0.00000 -2.475485 0.00000

н	2.475481	0.000000	-2.485324
С	0.000000	0.000000	-1.412881
С	3.656169	0.000000	0.711167
Н	2.475481	0.000000	2.485324
Н	4.599999	0.000000	1.247802
С	3.656169	0.000000	-0.711167
Н	4.599999	0.000000	-1.247802
С	2.480614	0.000000	-1.400575
С	0.000000	0.000000	-2.843566
Н	0.000000	0.000000	-5.117870
С	0.000000	0.000000	-4.051019
С	0.000000	0.000000	2.843566
Н	0.000000	0.000000	5.117870
С	0.000000	0.000000	4.051019

4. HC(C₆H₄)₂CH



Figure S4. Optimised Geometry of HC(C₆H₄)₂CH

Ato	m x	У	z
н	-4.594978	0.000199	1.245822
С	-3.648684	0.000080	0.713206
н	-2.470576	0.000249	2.490833
С	-2.473995	0.000117	1.403842
С	-2.473995	-0.000117	-1.403842
С	-1.218674	0.000022	0.717527
С	-3.648684	-0.000080	-0.713206
С	-1.218674	-0.000022	-0.717527

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С	0.000000	0.000000	1.398638	С	2.473995	-0.000117	1.403842	н	4.594978	-0.000199	1.245822
н	-4.594978	-0.000199	-1.245822	С	1.218674	0.000022	-0.717527	С	3.648684	0.000080	-0.713206
н	0.000000	0.000000	-2.486697	н	2.470576	0.000249	-2.490833	Н	4.594978	0.000199	-1.245822
н	-2.470576	-0.000249	-2.490833	С	0.000000	0.000000	-1.398638	С	2.473995	0.000117	-1.403842
С	1.218674	-0.000022	0.717527	С	3.648684	-0.000080	0.713206				
н	0.000000	0.000000	2.486697	н	2.470576	-0.000249	2.490833				

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Selected Spectra



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Figure S8. Cyclic Voltammogram (CH₂Cl₂/[NBu₄][PF₆] 0.1 M) at 25 °C at 100 mV s⁻¹ of [W{=CC(C₆H₄)₂CBr}(CO)₂(Tp*)] (1a).

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Figure S9b. Electronic spectrum (CH_2Cl_2) of $[W{\equiv}CC(C_6H_4)_2CBr}(CO)_2(Tp^*)]$ (1a).

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Figure S9b. Electronic spectrum (CH₂Cl₂) of $[W{\equiv}CC(C_6H_4)_2CBr}(CO)_2(Tp^*)]$ (1a).

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Figure S10. Infrared spectrum (CH₂Cl₂) of $[W{\equiv CC(C_6H_4)_2CBr}(CO)_2(Tp^*)]$ (1a).

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 $\label{eq:Figure S13.} Figure S13. Cyclic Voltammogram (CH_2Cl_2/[NBu_4][PF_6] \ 0.1 \ M) \ at \ 25 \ ^{\circ}C \ at \ 100 \ mV \ s^{-1} \ of \ [W{=}CC(C_6H_4)_2CBr{}(CO)_2(Tp)] \ (1b).$

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Figure S14a. Electronic spectrum (CH₂Cl₂) of $[W{\equiv CC(C_6H_4)_2CBr}(CO)_2(Tp)]$ (1b).

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Figure S14b. Electronic spectrum (CH₂Cl₂) of $[W{\equiv CC(C_6H_4)_2CBr}(CO)_2(Tp)]$ (1b).

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Figure S16. ¹H NMR (700 MHz, CDCl₃, 25 °C, δ) of [W{≡CC(C₆H₄)₂CBr}(CO)₂(Tm)] (**1c**).

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Figure S17. ${}^{13}C{}^{1H}$ NMR (176 MHz, CDCl₃, 25 °C, δ) of [W{=CC(C₆H₄)₂CBr}(CO)₂(Tm)] (1c).

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Figure S19. Infrared spectrum (CH_2Cl_2, cm^{-1}) of $[W{\equiv}CC(C_6H_4)_2CBr}(CO)_2(Tm)]$ (1c).

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 $\label{eq:Figure S20a.} \mbox{Electronic spectrum (CH_2Cl_2) of } [W\{ \equiv CC(C_6H_4)_2CBr\}(CO)_2(Tm)] \mbox{ (1c)}.$

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Figure S20b. Electronic spectrum (CH_2CI_2) of $[W{\equiv}CC(C_6H_4)_2CBr}(CO)_2(Tm)]$ (1c).

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Figure S25a. Electronic spectrum (CH₂Cl₂) of [W{=CC(C₆H₄)₂CC=CC₆H₄CH₃-4)(CO)₂(Tp*)] (2).

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 $\label{eq:Figure S25b. Electronic spectrum (CH_2Cl_2) of [W{=}CC(C_6H_4)_2CC{=}CC_6H_4CH_3{-}4)(CO)_2(Tp^*)] \ (2).$

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Figure S26. Infrared spectrum (CH_2Cl_2, cm^{-1}) of $[W{\equiv}CC(C_6H_4)_2CC{\equiv}CC_6H_4CH_3-4)(CO)_2(Tp^*)]$ (2).

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Figure S30. Cyclic Voltammogram (CH₂Cl₂/[NBu₄][PF₆] 0.1 M) at 25 °C at 100 mV s⁻¹ of [W₂Pd{µ-CC(C₆H₄)₂CBr}₂(CO)₄(Tp)₂] (3a).

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Figure S31. Infrared spectrum (CH₂Cl₂, cm⁻¹) of $[W_2Pd{\mu-CC(C_6H_4)_2CBr}_2(CO)_4(Tp)_2]$ (3a).
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 $\label{eq:Figure S32a} \textbf{Figure S32a}. \qquad Electronic spectrum (CH_2Cl_2) \ of \ [W_2Pd\{\mu-CC(C_6H_4)_2CBr\}_2(CO)_4(Tp)_2] \ \textbf{(3a)}.$

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 $[\]label{eq:Figure S32a} \textbf{Electronic spectrum (CH_2Cl_2) of } [W_2Pd\{\mu-CC(C_6H_4)_2CBr\}_2(CO)_4(Tp)_2] \textbf{ (3a)}.$

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Figure S33. ¹H NMR (700 MHz, CDCl₃, 25 °C, δ) of [W₂Pd{ μ -CC(C₆H₄)₂CBr}₂(CO)₄(Tm)₂] (3b)

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Figure S34. ¹H NMR (700 MHz, CDCl₃, -40 °C, δ) [W₂Pd{ μ -CC(C₆H₄)₂CBr}₂(CO)₄(Tm)₂] (3b).

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 $\label{eq:Figure S37. Electronic spectrum (CH_2Cl_2) of [W_2Pd\{\mu-CC(C_6H_4)_2CBr\}_2(CO)_4(Tm)_2] \mbox{ (3b)}.$

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 $\label{eq:Figure S37. Electronic spectrum (CH_2Cl_2) of [W_2Pd\{\mu-CC(C_6H_4)_2CBr\}_2(CO)_4(Tm)_2] \mbox{ (3b)}.$

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 $\label{eq:Figure S38. Infrared spectrum (CH_2Cl_2, cm^{-1}) \ of \ [W_2Pd\{\mu-CC(C_6H_4)_2CBr\}_2(CO)_4(Tm)_2] \ \textbf{(3b)}.$

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Figure S44. Electronic spectrum (CH₂Cl₂) of $[W_2Pt{\mu-CC(C_6H_4)_2CBr}_2(CO)_4(Tp)_2]$ (4a).

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 $\label{eq:Figure S44.} Electronic spectrum (CH_2Cl_2) \ of \ [W_2Pt\{\mu-CC(C_6H_4)_2CBr\}_2(CO)_4(Tp)_2] \ \textbf{(4a)}.$

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 $\label{eq:Figure S45.} \qquad \mbox{Infrared spectrum (CH_2Cl_2, \ cm^{-1}) of $[W_2Pt\{\mu-CC(C_6H_4)_2CBr\}_2(CO)_4(Tp)_2]$ (4a)}.$

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Figure S47. Infrared spectrum (CH₂Cl₂, cm⁻¹) of $[W_2Pt{\mu-CC(C_6H_4)_2CBr}_2(CO)_4(Tm)_2]$ (4b).

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Figure S48a. Electronic spectrum (CH_2CI_2) of $[W_2Pt{\mu-CC(C_6H_4)_2CBr}_2(CO)_4(Tm)_2]$ (4b).

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 $\label{eq:Figure S48b.} Electronic spectrum (CH_2Cl_2) \ of \ [W_2Pt\{\mu-CC(C_6H_4)_2CBr\}_2(CO)_4(Tm)_2] \ \textbf{(4b)}.$

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Figure S54a. Electronic spectrum (CH_2Cl_2) of $[(Tp)(CO)_2W{\equiv}CC(C_6H_4)_2CC{\equiv}W(CO)_2(Tp^*)]$ (5b).

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Figure S55. ¹H NMR (700 MHz, CDCl₃, -20 °C, δ) of [(Tm)(CO)₂W{=CC(C₆H₄)₂CC=}W(CO)₂(Tp^{*})] (5c).

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Figure S59a. Electronic spectrum (CH_2Cl_2) of $[(Tm)(CO)_2W{\equiv}CC(C_6H_4)_2CC{\equiv}W(CO)_2(Tp^*)]$ (5c).

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Figure S63a. Electronic spectrum (CH₂Cl₂, cm⁻¹) of [(Tp*)(CO)₂W{=CC(C₆H₄)₂CC=}Mo(CO)₂(Tp*)] (6a). Poor data quality due to extremely low solubility.

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Figure S63b. Electronic spectrum (CH₂Cl₂, cm⁻¹) of $[(Tp^*)(CO)_2W{\equiv CC(C_6H_4)_2CC\equiv}Mo(CO)_2(Tp^*)]$ (6a).

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 $\label{eq:Figure S64.} \mbox{Infrared spectrum (CH_2Cl_2, cm^{-1}) of } [(Tp^*)(CO)_2W\{ \equiv CC(C_6H_4)_2CC \equiv \} Mo(CO)_2(Tp^*)] \mbox{(6a)}.$



Figure S65. ¹H NMR (700 MHz, CDCl₃, 25 °C, δ) of [(Tp)(CO)₂W{=CC(C₆H₄)₂CC=}Mo(CO)₂(Tp*)] (6b).

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 $\label{eq:Figure S67. 13C{1}} \textbf{Figure S67. 13C{1}} \ \text{NMR} \ (176 \ \text{MHz}, \ \text{CDCl}_3, \ \text{-40 }^\circ\text{C}, \ \delta) \ of \ [(\text{Tp})(\text{CO})_2 W \{ \equiv \text{CC}(\text{C}_6\text{H}_4)_2 \text{CC} \equiv \} Mo(\text{CO})_2(\text{Tp}^*)] \ \textbf{(6b)}.$

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 $\label{eq:Figure S68. Cyclic Voltammogram (CH_2Cl_2/[NBu_4][PF_6] \ 0.1 \ M) \ at \ 25 \ ^{\circ}C \ at \ 100 \ mV \ s^{-1} \ of \ [(Tp)(CO)_2W \{ \equiv CC(C_6H_4)_2CC \equiv \} Mo(CO)_2(Tp^*)] \ (6b).$

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Figure S69. Infrared spectrum (CH_2CI_2, cm^{-1}) of $[(Tp)(CO)_2W{\equiv CC(C_6H_4)_2CC \equiv}Mo(CO)_2(Tp^*)]$ (6b).

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Figure S70a. Electronic spectrum (CH₂Cl₂, cm⁻¹) of $[(Tp)(CO)_2W{\equiv CC(C_6H_4)_2CC \equiv}Mo(CO)_2(Tp^*)]$ (6b).

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Figure S70b. Electronic spectrum (CH₂Cl₂, cm⁻¹) of $[(Tp)(CO)_2W{\equiv CC(C_6H_4)_2CC\equiv}Mo(CO)_2(Tp^*)]$ (6b).

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Figure \$74. Cyclic Voltammogram (CH₂Cl₂/[NBu₄][PF₆] 0.1 M) at 25 °C at 100 mV s⁻¹ of [(Tm)(CO)₂W{\exc{2}C(C₆H₄)₂CC\exc{3}Mo(CO)₂(Tp*)] (6c).

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 $\label{eq:Figure S75.} Infrared spectrum (CH_2Cl_2, cm^{-1}) of [(Tm)(CO)_2W\{ \equiv CC(C_6H_4)_2CC \equiv \} Mo(CO)_2(Tp^*)] \ (\textbf{6c}).$

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