

Supporting Information

Robust Ionic liquid@MOF composite as a versatile superprot tonic conductor

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SEM-EDX

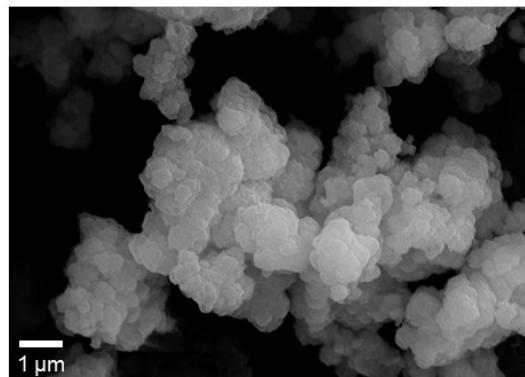


Figure S1: SEM image of MIL-101(Cr)-SO₃H

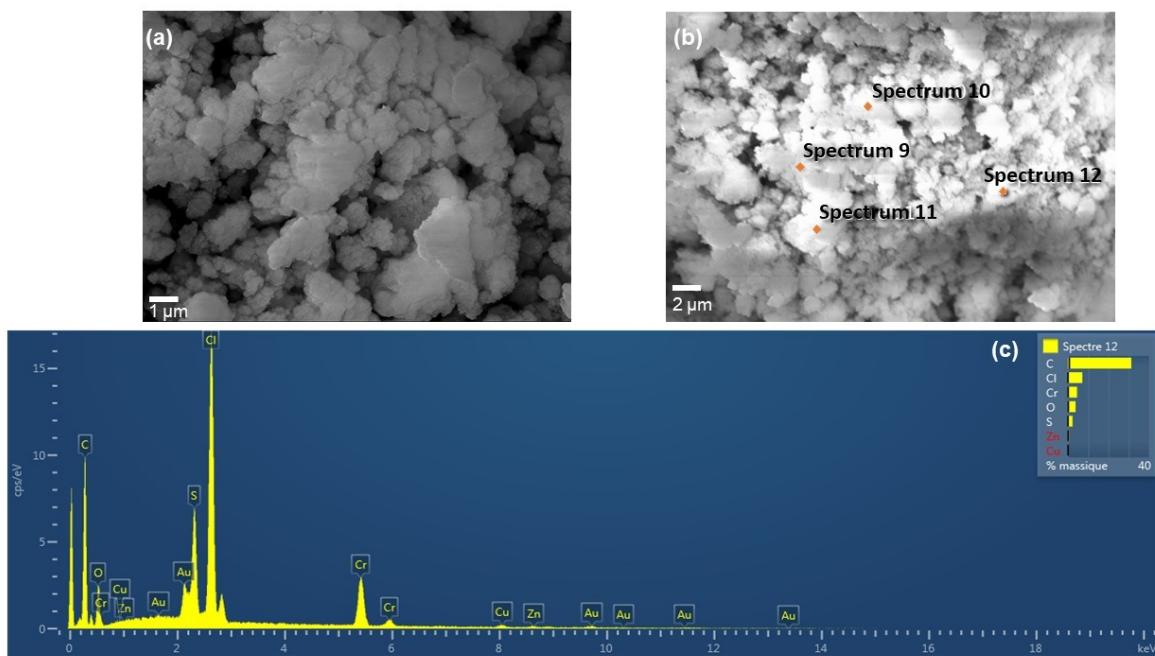


Figure S2. (a and b) SEM images of EMIM@MIL-101-SO₃H and (c) XEDS spectrum of EMIM@MIL-101-SO₃H in the case of (b).

Elemental Analysis

Table S1. EA results of MIL-101(Cr)-SO₃H and EMIM@MIL-101-SO₃H.

MIL-101(Cr)-SO₃H		
C	%exp (wt%) 27.93	Weight ratio, m(C)/m(S) = 4.34 -SO ₃ H functionalization degree= 68.6 %
S	6.43	
EMIM@MIL-101-SO₃H before PC measurements		
C	%exp (wt%) 34.7	Weight ratio, m(C)/m(N) = 3.61
N	9.6	
EMIM@MIL-101-SO₃H after PC measurements recorded at 298K-473K / 0% RH		
C	%exp (wt%) 37.4	Weight ratio, m(C)/m(N) = 3.66
N	10.2	
EMIM@MIL-101-SO₃H after PC measurements recorded at 343K / 80% RH for 6 days		
C	%exp (wt%) 35.6	Weight ratio, m(C)/m(N) = 3.67
N	9.7	

Thermogravimetric analysis (TGA)

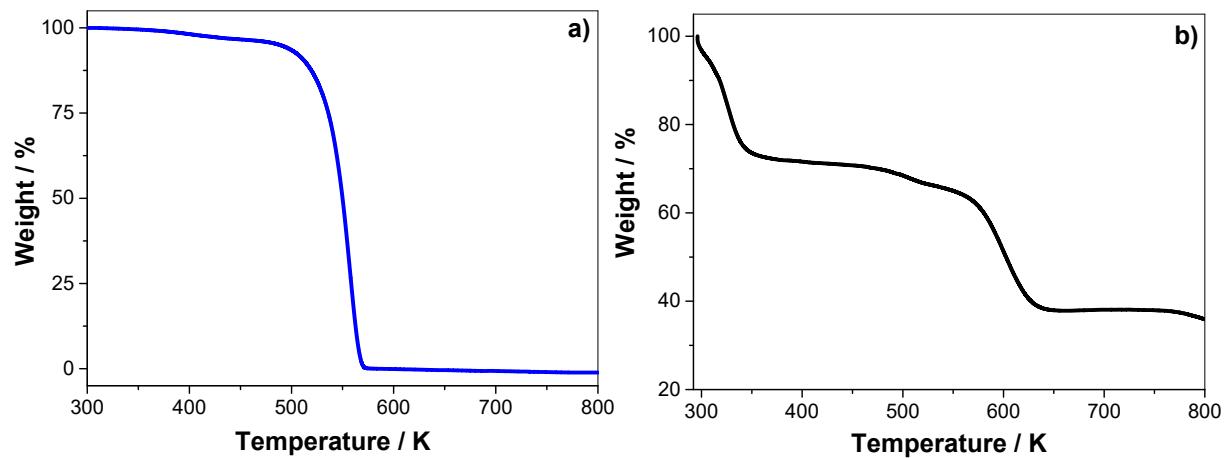


Figure S3 TGA curve (wt %) of (a) bulk EMIMCl and (b) MIL-101(Cr)-SO₃H.

FT-IR spectroscopy

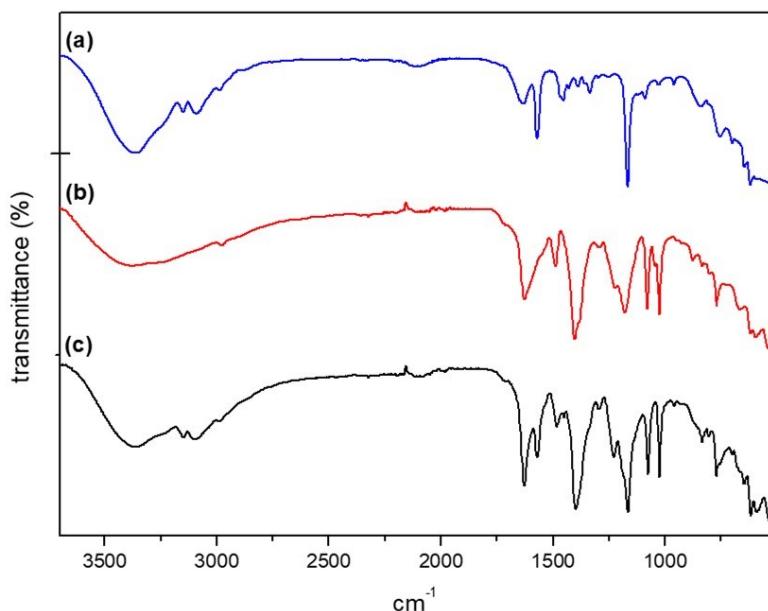


Figure S4: FT-IR spectra of (a) EMIMCl, (b) MIL-101(Cr)-SO₃H and (c) EMIM@MIL-101-SO₃H.

Table S2: Main FT-IR vibration bands of EMIMCl, MIL-101(Cr)-SO₃H and EMIM@MIL-101-SO₃H with the corresponding assignment.

MIL-101(Cr)-SO ₃ H	EMIMCl	EMIM@MIL-101-SO ₃ H	Assignment
	3151	3148	v(C-H)
	3083	3088	v (C-H)
1628	1630	1628	v(C-O) as + δ(H-O-H)
	1574	1568	Imidazole skeleton
1485		1479	v(C-O)s
	1448	1450	Imidazole skeleton
1403		1397	v(S=O) as
	1330		Imidazole skeleton
1178	1167	1166	v(S=O) s + Imidazole skeleton
1077		1077	v(S-O)
1026		1026	v(S-O)
840		840	v(Cr-O)
773		773	v(C-S)

Impedance measurements of the anhydrous EMIM@MIL-101-SO₃H

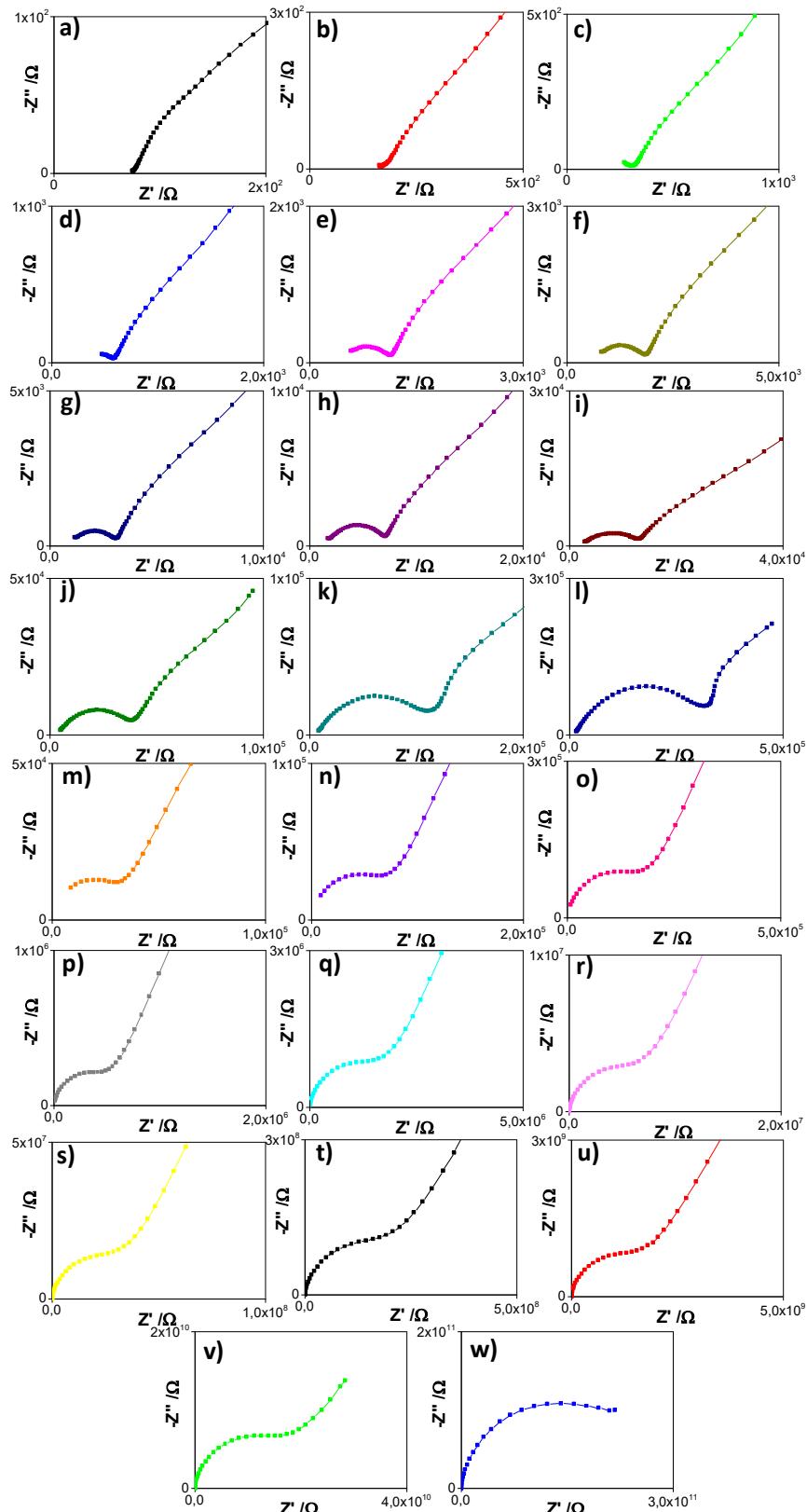


Figure S5: Nyquist plots of the impedance for the dehydrated EMIM@MIL-101-SO₃H powder recorded under 0% RH, at 473 K (a), 463 K (b), 453 K (c), 443 K (d), 433 K (e), 423 K (f), 413 K (g), 403 K (h), 393 K (i), 383 K (j), 373 K (k), 363 K (l), 353 K (m), 343 K (n), 333 K (o), 323 K (p), 313 K (q), 303 K (r), 293 K (s), 283 K (t), 273 K (u), 263 K (v) and 253 K (w).

Impedance measurements of the bulk EMIMCl

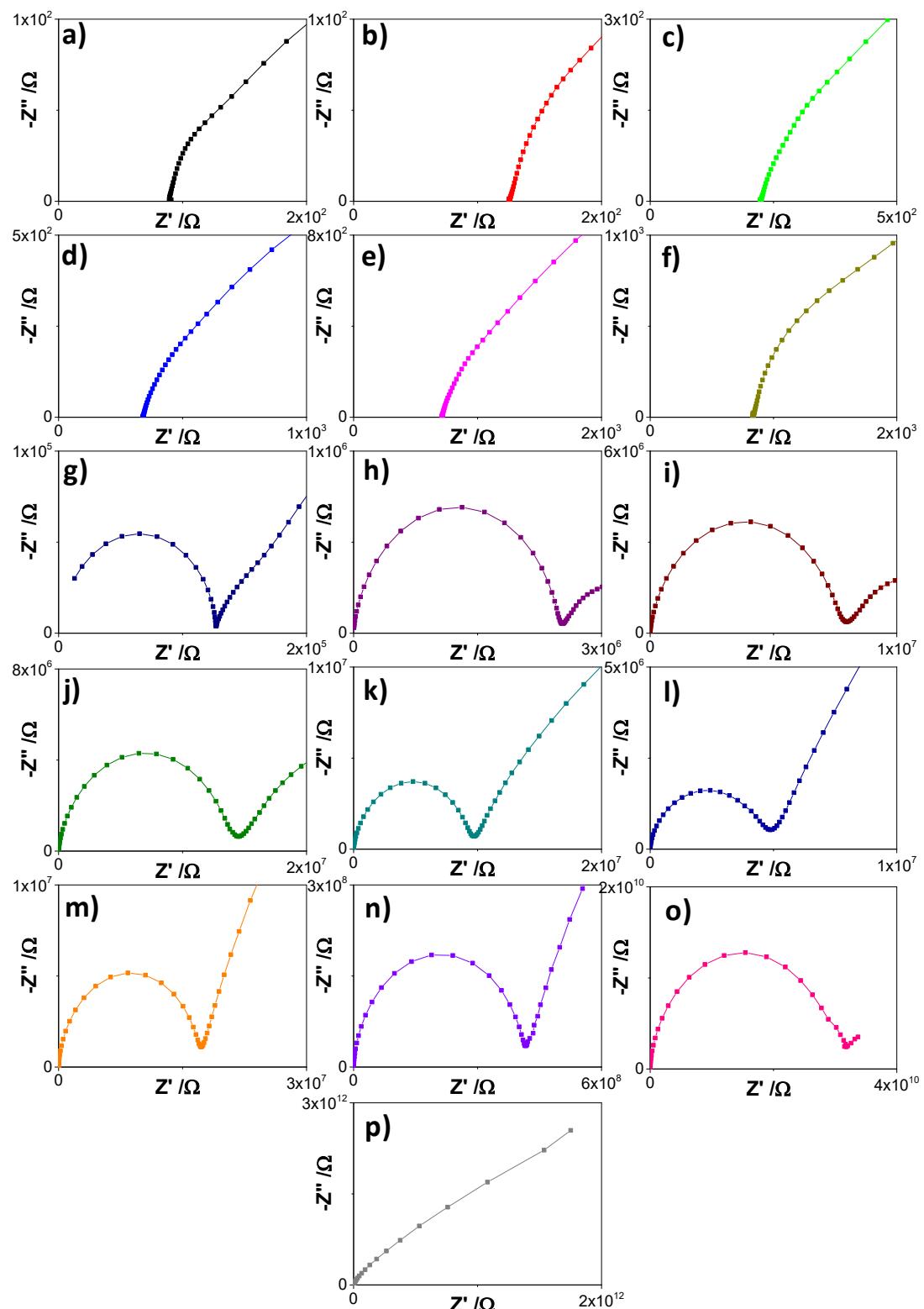


Figure S6: Nyquist plots of the impedance for the bulk EMIMCl recorded under 0% RH, at 323 K (a), 313 K (b), 303 K (c), 293 K (d), 283 K (e), 273 K (f), 263 K (g), 253 K (h), 243 K (i), 233 K (j), 223 K (k), 213 K (l), 203 K (m), 193 K (n), 183 K (o) and 173 K (p).

Impedance measurements of the anhydrous MIL-101(Cr)-SO₃H

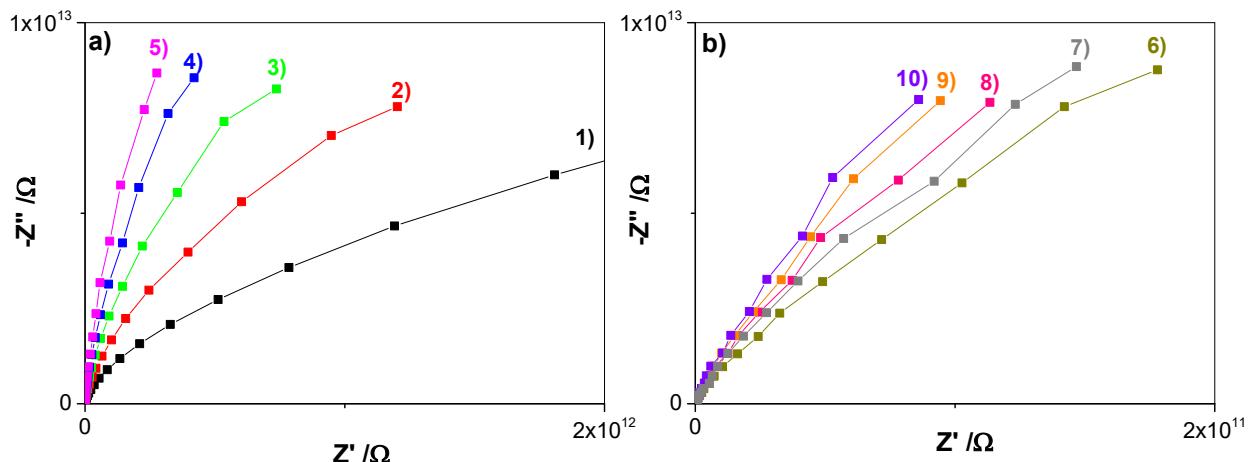


Figure S7: Nyquist plots of the impedance for the dehydrated MIL-101(Cr)-SO₃H recorded under 0% RH, at 393 K (1), 383 K (2), 373 K (3), 363 K (4), 353 K (5), 343 K (6), 333 K (7), 323 K (8), 313 K (9) and 303 K (10).

Impedance measurements of the hydrated EMIM@MIL-101-SO₃H for RH varying from 35 % to 80 %

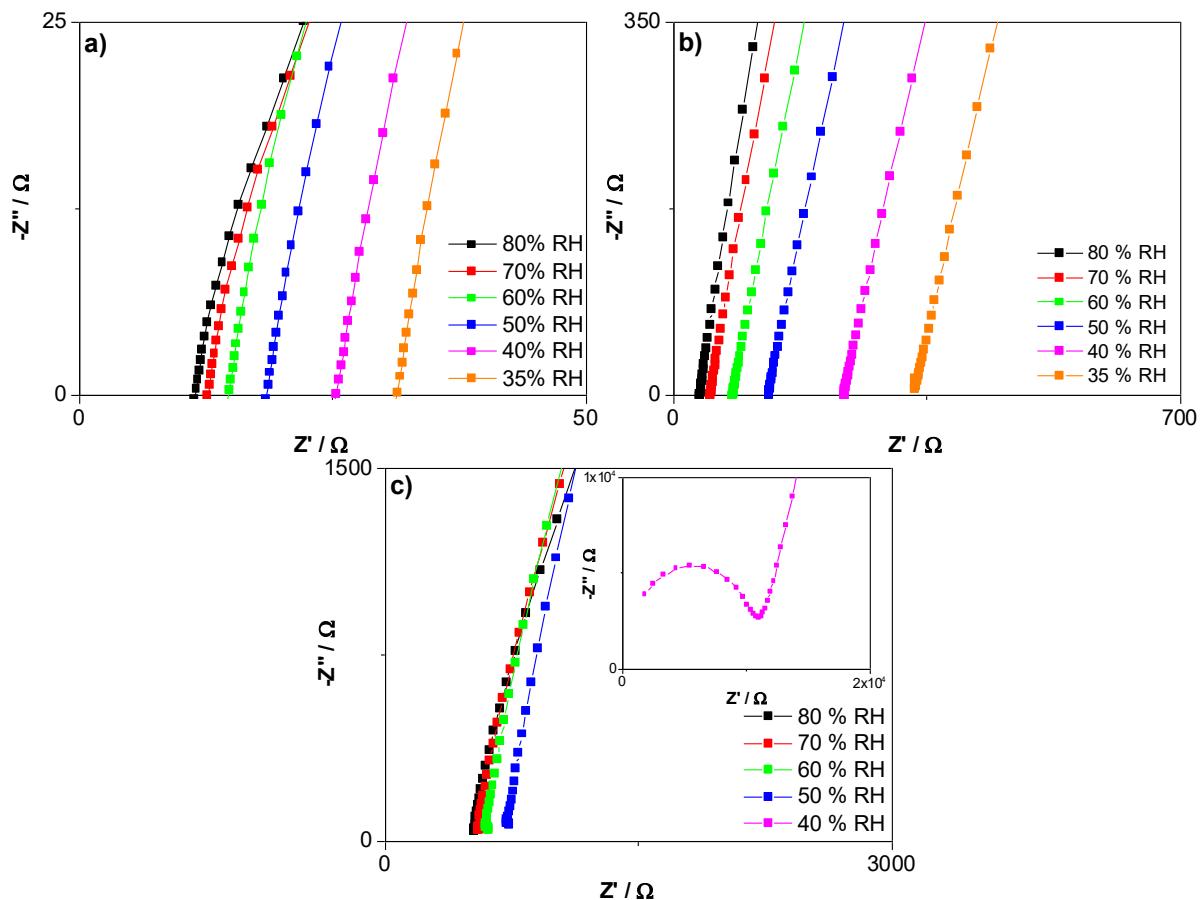


Figure S8: Nyquist plots of the impedance for EMIM@ MIL-101-SO₃H recorded at 343 K (a) and 303 K (b) and for MIL-101(Cr)-SO₃H recorded at 343 K (c) for RH varying from 35% to 80%.

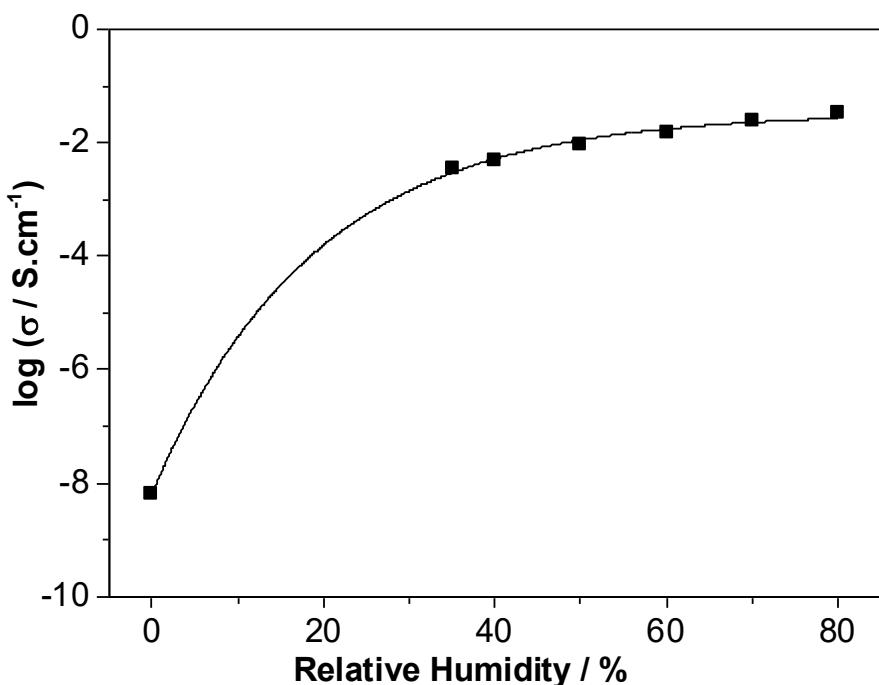


Figure S9: Humidity dependence of the conductivity for EMIM@MIL-101-SO₃H recorded at 303 K. The exponential line is a guide for the eyes.

Water sorption isotherm of EMIM@MIL-101-SO₃H

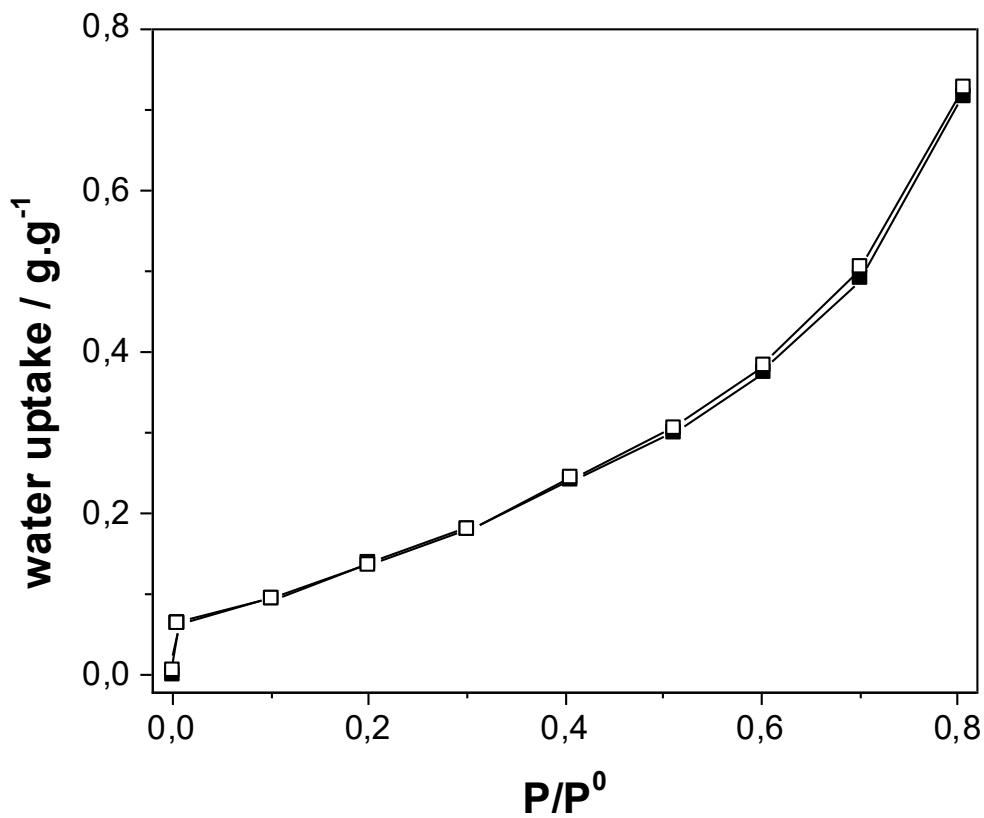


Figure S10: Water sorption isotherm of EMIM@MIL-101-SO₃H recorded at 303 K. Full and empty squares refer to the adsorption and desorption branches, respectively.

Additional impedance measurements
**Impedance measurements of the hydrated EMIM@MIL-101-SO₃H
and MIL-101(Cr)-SO₃H at RH = 80 % for T varying from 343 K to 298 K**

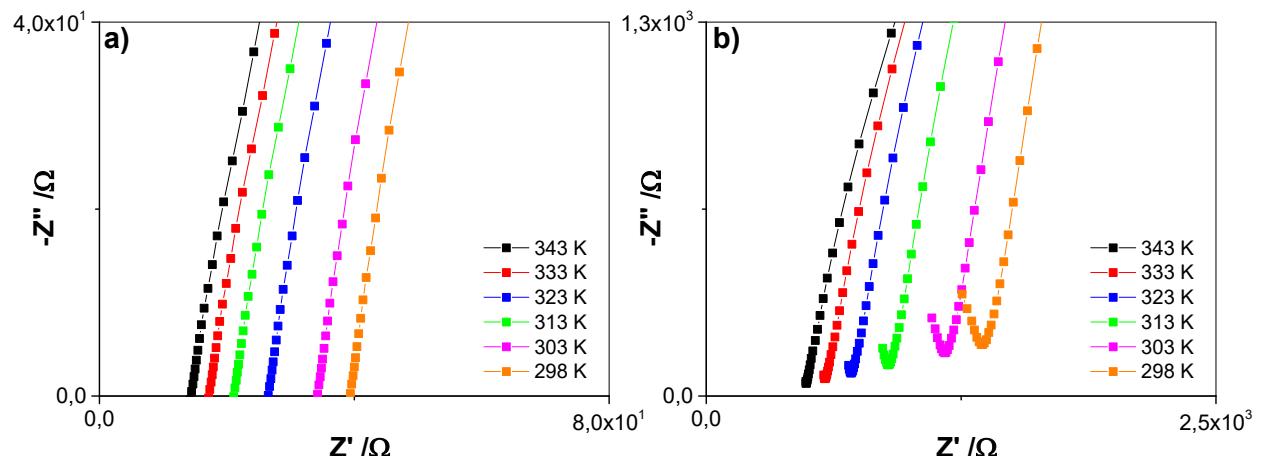


Figure S11: Nyquist plots of the impedance for EMIM@MIL-101-SO₃H (a) and MIL-101(Cr)-SO₃H (b) recorded under 80% RH, for T varying from 343 K to 298 K.

Impedance measurements of the hydrated EMIM@MIL-101-SO₃H recorded at RH = 80 % and T = 343 K

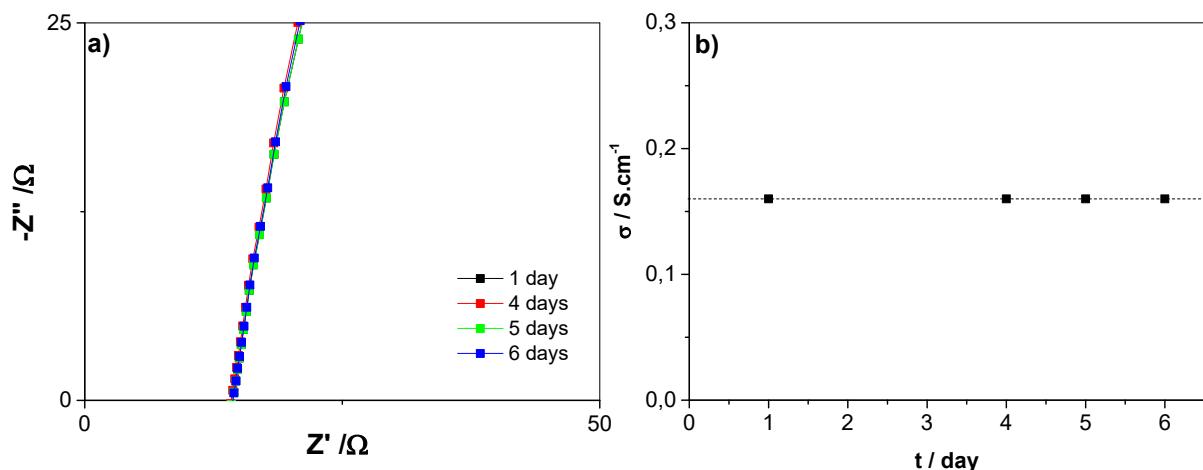


Figure S12: Time dependence of the conductivity of EMIM@MIL-101-SO₃H recorded at 343 K under 80% RH.

Table S3. Conductivity values deduced from the Nyquist plots, for EMIMCl, MIL-101(Cr)-SO₃H and EMIM@MIL-101-SO₃H powders at different RH/T conditions. L and S are the sample thickness and surface.

<i>Anhydrous EMIM@MIL-101-SO₃H</i>				RH = 0%		(L = 0.044 cm, S = 0.3631 cm ²)	
T / K	σ / S cm ⁻¹	T / K	σ / S cm ⁻¹	T / K	σ / S cm ⁻¹	T / K	σ / S cm ⁻¹
473	1.6 × 10 ⁻³	463	7.2 × 10 ⁻⁴	453	4.0 × 10 ⁻⁴	443	3.0 × 10 ⁻⁴
433	2.0 × 10 ⁻⁴	423	1.4 × 10 ⁻⁴	413	9.8 × 10 ⁻⁵	403	6.4 × 10 ⁻⁵
393	4.1 × 10 ⁻⁵	383	2.4 × 10 ⁻⁵	373	1.4 × 10 ⁻⁵	363	7.4 × 10 ⁻⁶
353	3.5 × 10 ⁻⁶	343	1.4 × 10 ⁻⁶	333	5.1 × 10 ⁻⁷	323	2.0 × 10 ⁻⁷
313	6.6 × 10 ⁻⁸	303	1.5 × 10 ⁻⁸	293	2.9 × 10 ⁻⁹	283	4.8 × 10 ⁻¹⁰
273	6.0 × 10 ⁻¹¹	263	6.0 × 10 ⁻¹²	253	5.1 × 10 ⁻¹³	243	5.9 × 10 ⁻¹⁴
<i>Bulk EMIMCl</i>				RH = 0%		(L = 0.098 cm, S = 0.3631 cm ²)	
T / K	σ / S cm ⁻¹	T / K	σ / S cm ⁻¹	T / K	σ / S cm ⁻¹	T / K	σ / S cm ⁻¹
323	3.0 × 10 ⁻³	313	2.1 × 10 ⁻³	303	1.2 × 10 ⁻³	293	7.9 × 10 ⁻⁴
283	5.0 × 10 ⁻⁴	273	3.2 × 10 ⁻⁴	263	2.1 × 10 ⁻⁶	253	1.4 × 10 ⁻⁷
243	2.8 × 10 ⁻⁸	233	2.5 × 10 ⁻⁸	223	2.8 × 10 ⁻⁸	213	5.5 × 10 ⁻⁸
203	1.9 × 10 ⁻⁸	193	6.5 × 10 ⁻¹⁰	183	8.5 × 10 ⁻¹²		
<i>Hydrated EMIM@MIL-101-SO₃H</i>							
T = 343 K (L = 0.186 cm, S = 0.1256 cm ²)				T = 298 K (L = 0.148 cm, S = 0.1256 cm ²)			
RH / %	σ / S cm ⁻¹	RH / %	σ / S cm ⁻¹	RH / %	σ / S cm ⁻¹	RH / %	σ / S cm ⁻¹
80	1.4 × 10 ⁻¹	70	1.2 × 10 ⁻¹	80	3.2 × 10 ⁻²	70	2.3 × 10 ⁻²
60	1.0 × 10 ⁻¹	50	8.4 × 10 ⁻²	60	1.4 × 10 ⁻²	50	9.0 × 10 ⁻³
40	6.0 × 10 ⁻²	35	4.8 × 10 ⁻²	40	5.0 × 10 ⁻³	35	3.5 × 10 ⁻³
<i>Hydrated MIL-101(Cr)-SO₃H</i>				T = 343 K (L = 0.376 cm, S = 0.1256 cm ²)			
RH / %	σ / S cm ⁻¹	RH / %	σ / S cm ⁻¹	RH / %	σ / S cm ⁻¹	RH / %	σ / S cm ⁻¹
80	5.6 × 10 ⁻³	70	5.4 × 10 ⁻³	60	5.0 × 10 ⁻³	50	4.2 × 10 ⁻³
40	2.7 × 10 ⁻⁴						
<i>Hydrated EMIM@MIL-101-SO₃H</i>				RH = 80% (L = 0.270 cm, S = 0.1256 cm ²)			
T / K	σ / S cm ⁻¹	T / K	σ / S cm ⁻¹	T / K	σ / S cm ⁻¹	T / K	σ / S cm ⁻¹
343	1.4 × 10 ⁻¹	333	1.3 × 10 ⁻¹	323	1.0 × 10 ⁻¹	313	8.3 × 10 ⁻²
303	6.4 × 10 ⁻²	298	5.5 × 10 ⁻²				
<i>Hydrated MIL-101(Cr)-SO₃H</i>				RH = 80% (L = 0.313 cm, S = 0.1256 cm ²)			
T / K	σ / S cm ⁻¹	T / K	σ / S cm ⁻¹	T / K	σ / S cm ⁻¹	T / K	σ / S cm ⁻¹
343	5.1 × 10 ⁻³	333	4.3 × 10 ⁻³	323	3.5 × 10 ⁻³	313	2.8 × 10 ⁻³
303	2.1 × 10 ⁻³	298	1.8 × 10 ⁻³				