

# Supporting Information

## Multi-function Broad-Band Emission $\text{Ba}_{4-x}\text{Sr}_y\text{La}_6\text{O}(\text{SiO}_4)_6:x\text{Eu}^{2+}$ Phosphor for White LED, and Anti-counterfeiting

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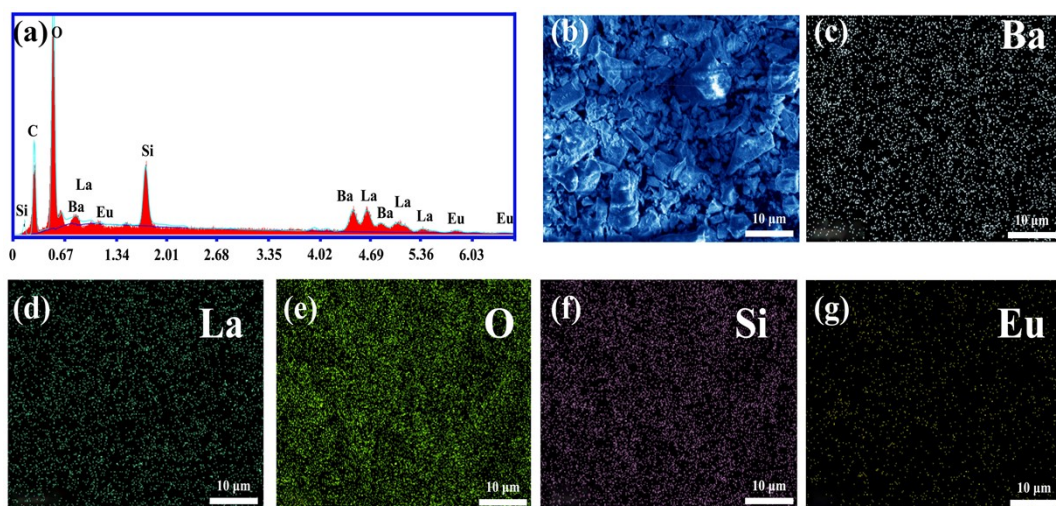


Fig.S1 (a) EDS images and (b) SEM image of BLOS:0.28Eu<sup>2+</sup>; (c-g) Elemental mapping images of Ba, La, O, Si, and Eu for the selected particle, respectively.

**Table S1. Main Crystallographic Parameters for Ba<sub>4-x</sub>La<sub>6</sub>O(SiO<sub>4</sub>)<sub>6</sub>:xEu<sup>2+</sup> ( $x = 0.016 - 0.48$ ) from the GSAS Program Rietveld Refinement**

concentr	$x=0.016$	$x=0.04$	$x=0.08$	$x=0.28$	$x=0.48$
crystal	hexagon	hexagon	hexagon	hexagon	hexagon
space	$P 6_3/m$	$P 6_3/m$	$P 6_3/m$	$P 6_3/m$	$P 6_3/m$
$a = b$ (Å)	9.807	9.8063	9.8051	9.7904	9.7795
(Å)	7.3454	7.3405	7.3388	7.3231	7.308
$V$ (Å <sup>3</sup> )	611.82	611.316	611.021	607.893	605.285
$2\theta$	10 - 75°	10 - 75°	10 - 75°	10 - 75°	10 - 75°
$R_{wp}$ (%)	10.92	9.61	11.62	10.62	11.97
$R_p$ (%)	6.95	6.49	7.15	6.75	7.52
$CHI^2$	4.476	3.489	5.086	4.342	6.156

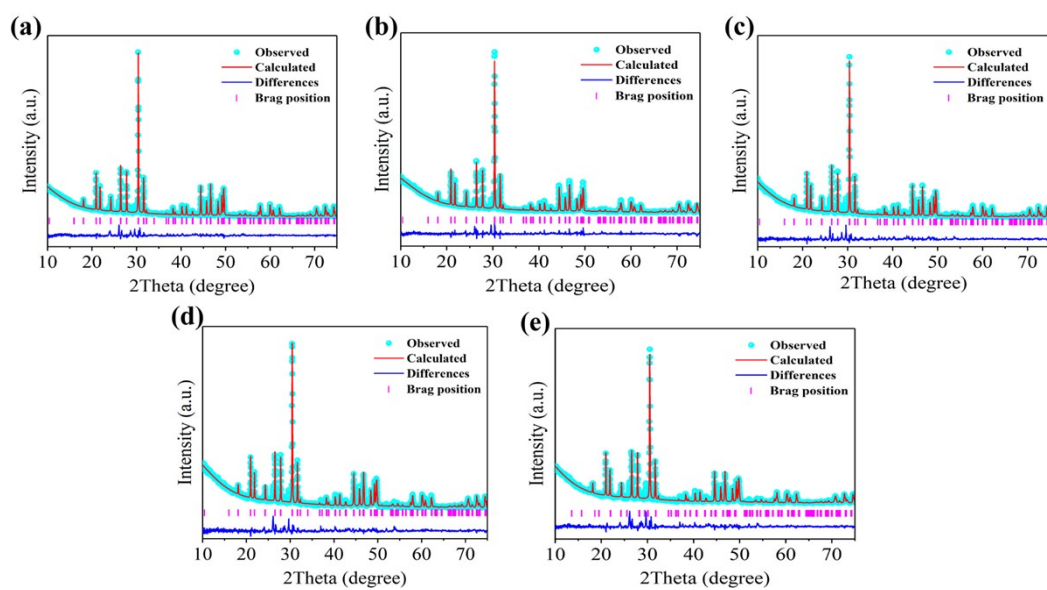


Fig.S2 (a–e) Rietveld refinement of the powder XRD profiles of BLOS: $x$ Eu ( $x = 0.016, 0.04, 0.08, 0.28, 0.48$ ), respectively.

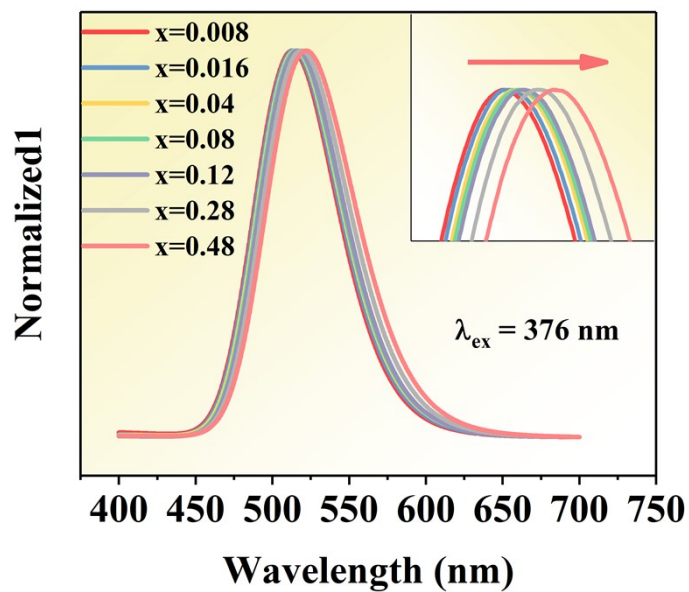


Fig.S3 Normalized PL spectrum of series phosphors BLOS:xEu.

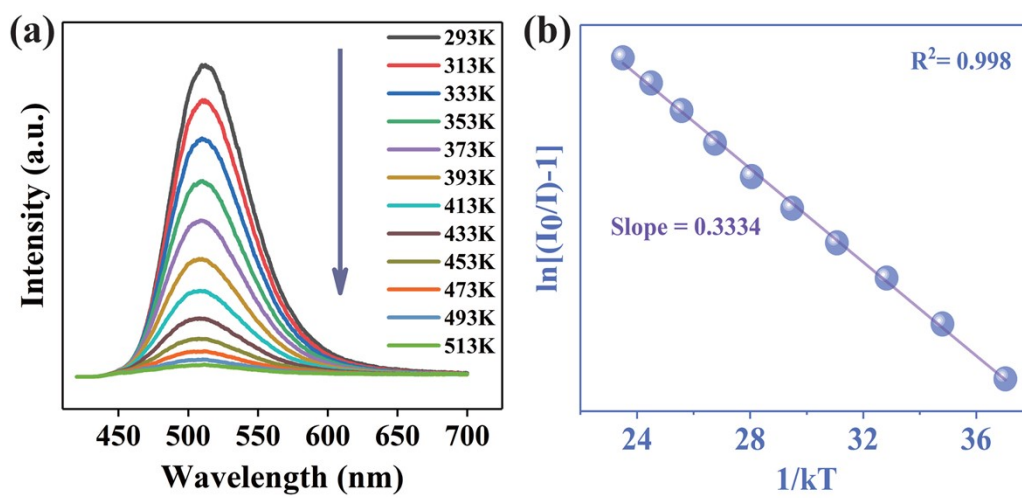


Fig.S4 (a) The temperature-dependent PL spectra of BLOS:0.08Eu phosphors at temperatures of 293–453 K ( $\lambda_{\text{ex}} = 376 \text{ nm}$ ); (b) The plot of  $\ln[I_0/I - 1]$  versus  $1/kT$  for BLOS:0.08Eu.

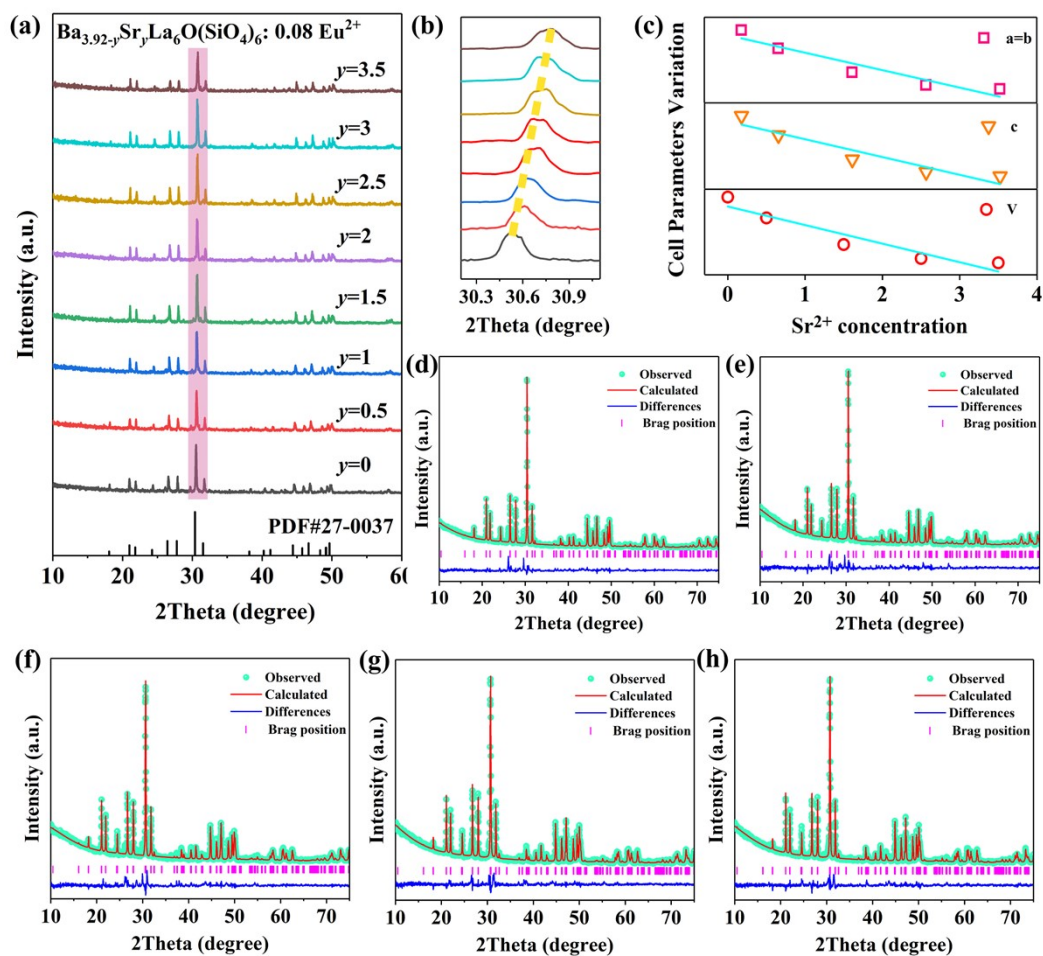


Fig.S5 (a) XRD patterns compared to the standard pattern of  $\text{Ba}_{3.92-y}\text{Sr}_y\text{La}_6\text{O}(\text{SiO}_4)_6:0.08\text{Eu}$  ( $y = 0\text{--}3.5$ ) and (b) view of the peak shift at  $30.1\text{--}31^\circ$ ; (c) evolution of lattice parameters ( $a$ ,  $b$ ,  $c$ ) and the unit cell volumes ( $V$ ) for  $\text{Ba}_{3.92-y}\text{Sr}_y\text{La}_6\text{O}(\text{SiO}_4)_6:0.08\text{Eu}$  ( $y = 0, 0.5, 1.5, 2.5, 3.5$ ); (d–h) Rietveld refinement of the powder XRD profiles of  $\text{Ba}_{3.92-y}\text{Sr}_y\text{La}_6\text{O}(\text{SiO}_4)_6:0.08\text{Eu}$ .

**Table S2. Main Crystallographic Parameters for  $\text{Ba}_{3.92-y}\text{Sr}_y\text{La}_6\text{O}(\text{SiO}_4)_6:0.08\text{Eu}^{2+}$  ( $y = 0 - 3.5$ ) from the GSAS Program Rietveld Refinement**

concentration	$y=0$	$y=0.5$	$y=1.5$	$y=2.5$	$y=3.5$
crystal system	hexagonal	hexagonal	hexagonal	hexagonal	hexagonal
space group	$P 6_3/m$	$P 6_3/m$	$P 6_3/m$	$P 6_3/m$	$P 6_3/m$
$a = b$ (Å)	9.8035	9.78	9.7493	9.7329	9.728
$c$ (Å)	7.3405	7.3112	7.2743	7.2553	7.249
$V$ (Å <sup>3</sup> )	610.971	605.615	598.777	595.209	594.088
$2\theta$ interval	10 - 75°	10 - 75°	10 - 75°	10 - 75°	10 - 75°
$R_{wp}$ (%)	10.69	10.08	10.14	8.54	7.95
$R_p$ (%)	6.66	6.57	6.61	5.78	5.46
CHI <sup>2</sup>	5.319	4.898	5.08	3.644	3.604

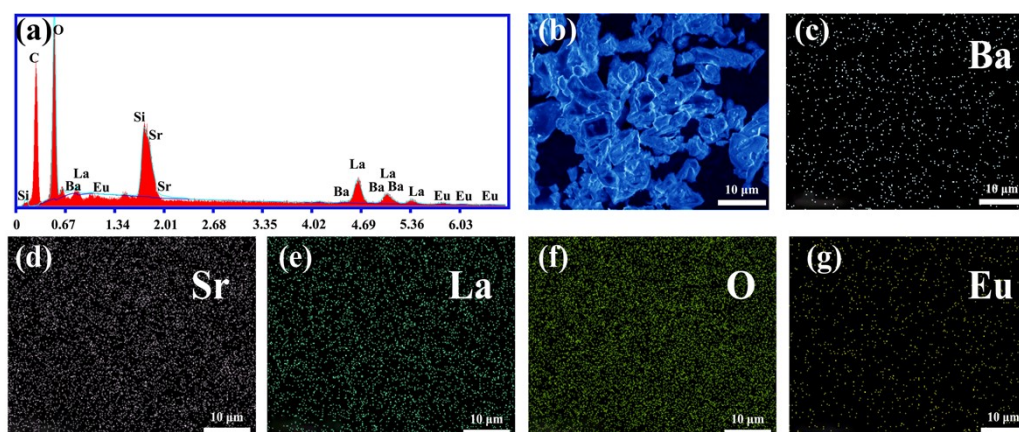


Fig.S6 (a) EDS images and (b) SEM image of  $\text{Ba}_{1.42}\text{Sr}_{2.5}\text{La}_6\text{O}(\text{SiO}_4)_6:0.08\text{Eu}$ ; (c-g) Elemental mapping images of Ba, Sr, La, O and Eu for the selected particle, respectively.

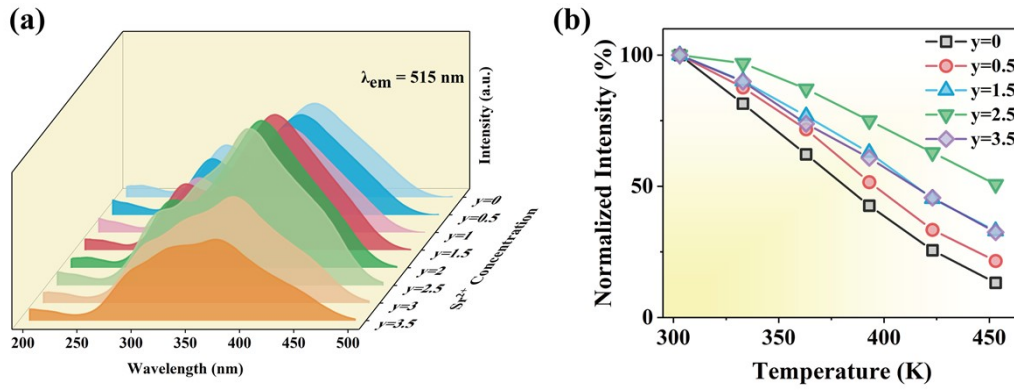


Fig.S7 (a) PL spectra of  $\text{Ba}_{3.92-y}\text{Sr}_y\text{La}_6\text{O}(\text{SiO}_4)_6:0.08\text{Eu}^{2+}$  ( $y = 0-3.5$ ) phosphor; (b) Normalized PL intensity of  $\text{Ba}_{3.92-y}\text{Sr}_y\text{La}_6\text{O}(\text{SiO}_4)_6:0.08\text{Eu}^{2+}$  ( $y = 0-3.5$ ) phosphor at different temperatures.

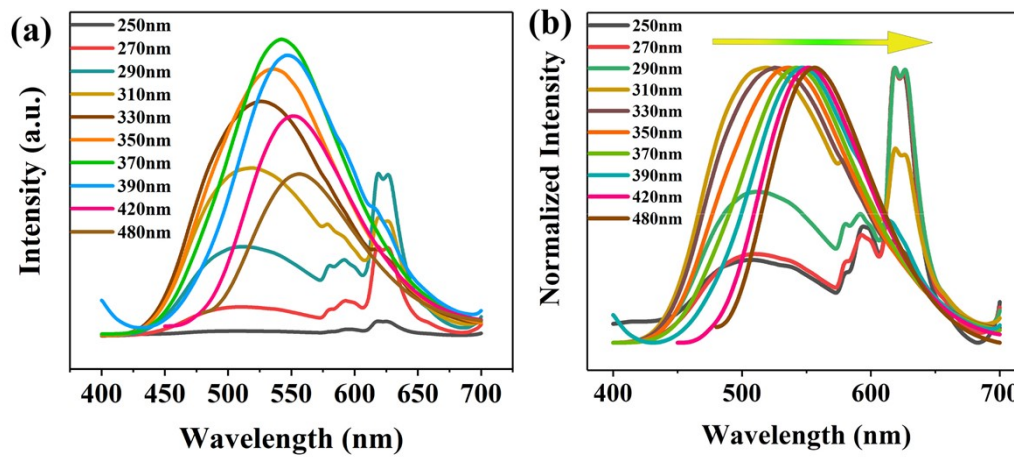


Fig.S8 (a) PL spectra of  $\text{Ba}_{0.42}\text{Sr}_{3.5}\text{La}_6\text{O}(\text{SiO}_4)_6:0.08\text{Eu}^{2+}$  phosphors at different excitation wavelengths; Normalized PL spectra of  $\text{Ba}_{0.42}\text{Sr}_{3.5}\text{La}_6\text{O}(\text{SiO}_4)_6:0.08\text{Eu}^{2+}$  phosphors at different excitation wavelengths.

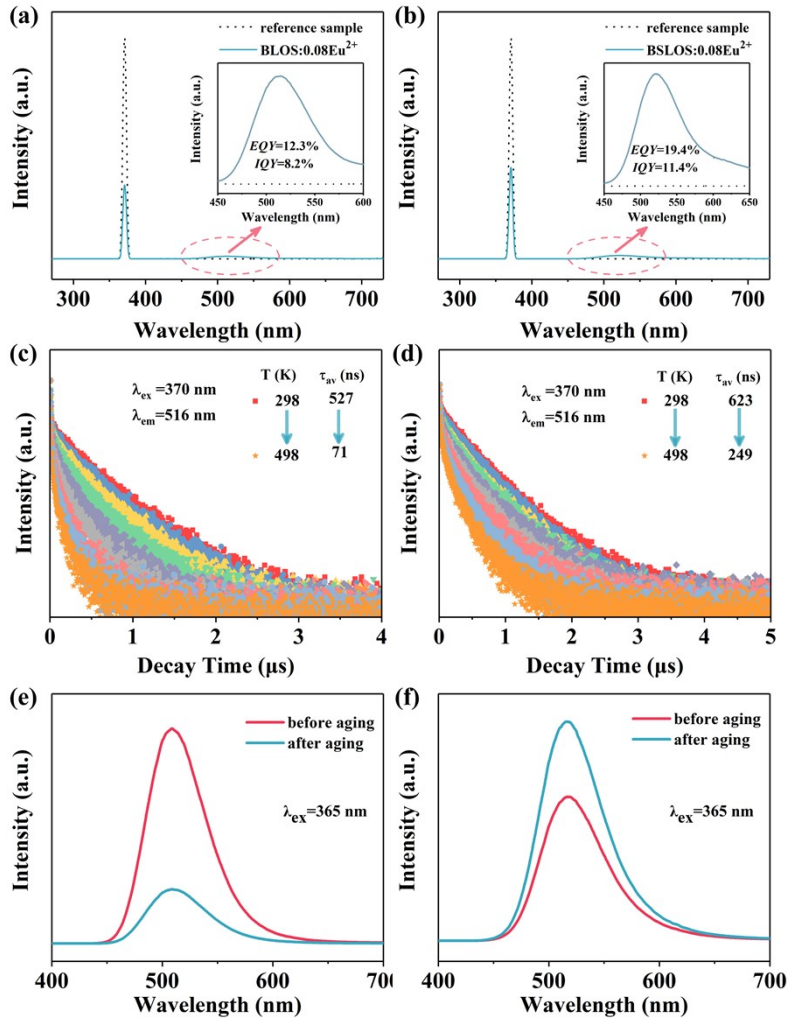


Fig.S9 The quantum efficiency of Ba<sub>3.92</sub>La<sub>6</sub>O(SiO<sub>4</sub>)<sub>6</sub>:0.08Eu phosphors(a) Ba<sub>0.42</sub>Sr<sub>3.5</sub>La<sub>6</sub>O(SiO<sub>4</sub>)<sub>6</sub>:0.08Eu phosphors(b); The excited-state lifetime dependence on temperature of Ba<sub>3.92</sub>La<sub>6</sub>O(SiO<sub>4</sub>)<sub>6</sub>:0.08Eu phosphors(c) Ba<sub>0.42</sub>Sr<sub>3.5</sub>La<sub>6</sub>O(SiO<sub>4</sub>)<sub>6</sub>:0.08Eu phosphors(d); The chemical stability of Ba<sub>3.92</sub>La<sub>6</sub>O(SiO<sub>4</sub>)<sub>6</sub>:0.08Eu phosphors(e) Ba<sub>0.42</sub>Sr<sub>3.5</sub>La<sub>6</sub>O(SiO<sub>4</sub>)<sub>6</sub>:0.08Eu phosphors(f).



Table S3 The chemical stability of  $\text{Ba}_{3.92}\text{La}_6\text{O}(\text{SiO}_4)_6:0.08\text{Eu}$  phosphors  $\text{Ba}_{0.42}\text{Sr}_{3.5}\text{La}_6\text{O}(\text{SiO}_4)_6:0.08\text{Eu}$  phosphors.

□	$\text{Ba}_{3.92}\text{La}_6\text{O}(\text{SiO}_4)_6:0.08\text{Eu}$		$\text{Ba}_{3.92}\text{La}_6\text{O}(\text{SiO}_4)_6:0.08\text{Eu}$	
	before aging	after aging	before aging	after aging
<b>Normalized intensity</b>	1	0.255	1	0.67
<b>FWHM (nm)</b>	62.14	62.91	66.27	67.3
<b>CIE</b>	(0.19,0.58)	(0.19,0.59)	(0.25,0.61)	(0.24,0.61)
<b>X at max height (nm)</b>	514.2	516.2	524.8	526.2