Supplementary Information

Mononuclear Co(II) polypyridyl complexes: synthesis, molecular structure, DNA binding/cleavage, radical scavenging, docking studies and anticancer activities

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Fig. S1. ¹H NMR of N-(pyridin-2-ylmethyl)aniline in CDCl₃ at room temperature.



Fig. S2. ¹³C NMR of N-(pyridin-2-ylmethyl)aniline in CDCl₃ at room temperature.



Fig. S3. ¹H NMR of L in $CDCl_3$ at room temperature.



Fig. S4. ¹³C NMR of \mathbf{L} in CDCl₃ at room temperature.



Fig. S5. FT–IR (Solid) spectrum of L.



Fig. S6. FT–IR (Solid) spectrum of 1.



Fig. S7. FT–IR (Solid) spectrum of 2.



Fig. S8. FT–IR (Solid) spectrum of 3.



Fig. S9. FT–IR (Solid) spectrum of 4.



Fig. S10. HRMS spectrum of 1 (top) with their simulated mass spectra (bottom).



Fig. S11. HRMS spectrum of 2 (top) with their simulated mass spectra (bottom).



Fig. S12. HRMS spectrum of 3 (top) with their simulated mass spectra (bottom).



Fig. S13. HRMS spectrum of 4 (top) with their simulated mass spectra (bottom).



Fig. S14. UV–vis spectrum of L and complexes 1 in acetonitrile.



Fig. S15. (a-d) Time dependent UV–vis spectrum of complexes 1-4 in 5 mM Tris-HCl buffer (pH = 7.2) at room temperature.



Fig. S16. Schematic drawing of trigonal bipyramidal geometry and polyhedron view of 1.



Fig. S17. $(a-c)\pi - \pi$ interaction between phenyl group of ligand L and bidentate ligands (bpy, phen, pic) in complexes 2-4, respectively.



Fig. S18. Schematic drawing of 2-4 with the dihedral angles (°) benzene ring of **L** and the plane of the bidentate ligand system (bpy, phen and pic).



Fig. S19. Cyclic voltammogram of 1-4 (bottom to top) in acetonitrile using 0.1 M TBAP as supporting electrolyte versus SCE, scan rate 100 mV s⁻¹.



Fig. S20. Absorption spectra of **1** (66.7 μ M) in the absence (black) and presence (colour) of ss-DNA (12.5, 25.0, 37.5, 50.0, 62.50, 75.0, 87.5 μ M, respectively) in 5 mM Tris-HCl buffer (pH =7.2). The arrow shows the absorbance changes on increasing DNA concentration. Inset: the plot of A₀/A–A₀ versus 1/[DNA] for the calculation of binding constant of complex **1**.



Fig. S21. Absorption spectra of **2** (66.7 μ M) in the absence (black) and presence (colour) of ss-DNA (12.5, 25.0, 37.5, 50.0, 62.50, 75.0, 87.5 μ M, respectively) in 5 mM Tris-HCl buffer (pH =7.2). The arrow shows the absorbance changes on increasing DNA concentration. Inset: the plot of A₀/A–A₀ versus 1/[DNA] for the calculation of binding constant of complex **2**.



Fig. S22. Absorption spectra of **4** (20 μ M) in the absence (black) and presence (colour) of ss-DNA (10.06, 15.09, 20.12, 25.15, 30.18, 35.21 μ M, respectively) in 5 mM Tris-HCl buffer (pH =7.2). The arrow shows the absorbance changes on increasing DNA concentration. Inset: the plot of A₀/A–A₀ versus 1/[DNA] for the calculation of binding constant of complex **4**.



Fig. S23. Emission spectra of EB bound DNA in the presence of **1**. ([DNA] = 1×10^{-4} M, [EB] = 1×10^{-5} M, [Quencher](μ M) for **1**: (a) 0, (b) 4, (c) 8, (d) 12, (e) 16, (f) 20, (g) 24, (h) 28, (i) 32. Inset: the plot of I_0/I versus [Quencher].



Fig. S24. Emission spectra of EB bound DNA in the presence of **2**. ([DNA] = 1×10^{-4} M, [EB] = 1×10^{-5} M, [Quencher](μ M) for **1**: (a) 0, (b) 4, (c) 8, (d) 12, (e) 16, (f) 20, (g) 24, (h) 28, (i) 32. Inset: the plot of I_0/I versus [Quencher].



Fig. S25. Emission spectra of EB bound DNA in the presence of 4. ([DNA] = 1×10^{-4} M, [EB] = 1×10^{-5} M, [Quencher](μ M) for 1: (a) 2, (b) 4, (c) 6, (d) 8, (e) 10, (f) 12, (g) 14, (h) 16, (i) 20, (j) 24, (k) 28, (m) 32. Inset: the plot of I_0/I versus [Quencher].



Fig. S26: Agarose gel showing the cleavage of plasmid DNA (pDNA 200 ng/ μ L) at different concentrations upon incubation at 37 °C for 3 h in 50 mM Tris-HCl buffer (pH 7.2). Lane 1: Control; Lane 2-6: pDNA + Complexes [(a) Complex 1: 50 μ M, 100 μ M, 200 μ M, 400 μ M, 500 μ M; (b-c) Complexes 2 and 3: 1 μ M, 5 μ M, 10 μ M, 25 μ M, 50 μ M; (d) Complex 4: 100 μ M, 250 μ M, 500 μ M, 750 μ M, 1 mM].

Complex 1					
S.No.	Concentrations	Form I	Form II		
	(µM)	(%)	(%)		
1	Control	83	17		
2	50	80	20		
3	100	73	27		
4	200	80	20		
5	400	75	25		
6	500	80	20		

Table S1: Respective Cleavage Concentrations (1-4) and Percentage of Cleaved Forms of pDNA.

Complex 2					
S.No.	Concentrations	Form I	Form II		
	(µM)	(%)	(%)		
1	Control	79	21		
2	1	78	22		
3	5	73	27		
4	10	62	38		
5	25	30	70		
6	50	2	98		

Complex 3					
S.No.	Concentrations	Form I	Form II		
	(µM)	(%)	(%)		
1	Control	85	15		
2	1	82	18		
3	5	76	24		
4	10	65	35		
5	25	26	74		
6	50	5	95		

Complex 4					
S.No.	Concentrations	Form I	Form II		
	(µM)	(%)	(%)		
1	Control	93	7		
2	100	85	15		
3	250	70	30		
4	500	43	57		
5	750	15	85		
6	1000	6	94		















Fig. S27. (a-d) Interactions between complexes (1-4) and DNA, along with the van der Waals forces are represented in a 2-Dimentional format.





Fig. S28. Crystal violet staining images of A549 cells (top) and MDA-MB-231 cells (bottom) treated with 10 μ M and 30 μ M of **1-4** for 72 h. The purple colour indicates the number of live cancer cells, whereas its loss signifies the decrease in live cells survival.