

Nickel Hydroxide/Sulfide Hybrids: Halide Ion Controlled Synthesis, Structural Characteristics, and Electrochemical Performance

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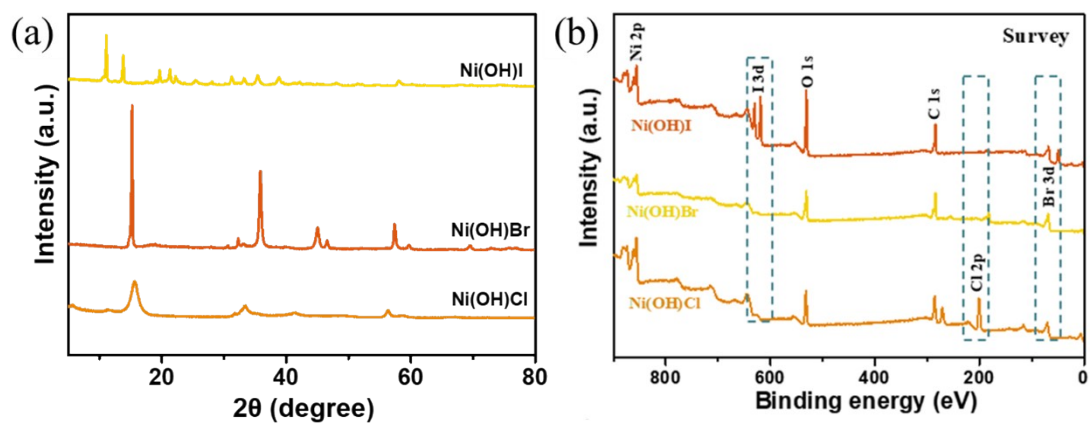


Figure S1. (a) XRD patterns of Ni(OH)Cl, Ni(OH)Br and Ni(OH)I. (b) XPS spectra of Ni(OH)Cl, Ni(OH)Br and Ni(OH)I.

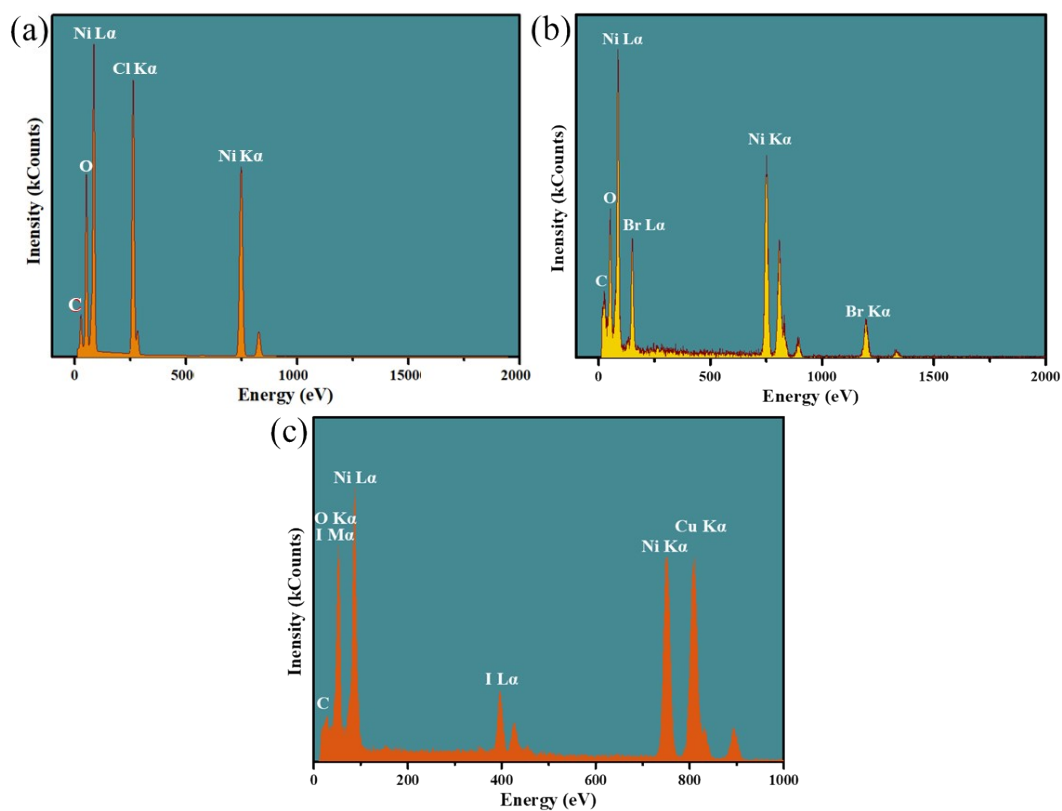


Figure S2. EDS spectra of (a) Ni(OH)Cl, (b) Ni(OH)Br, and (c) Ni(OH)I.

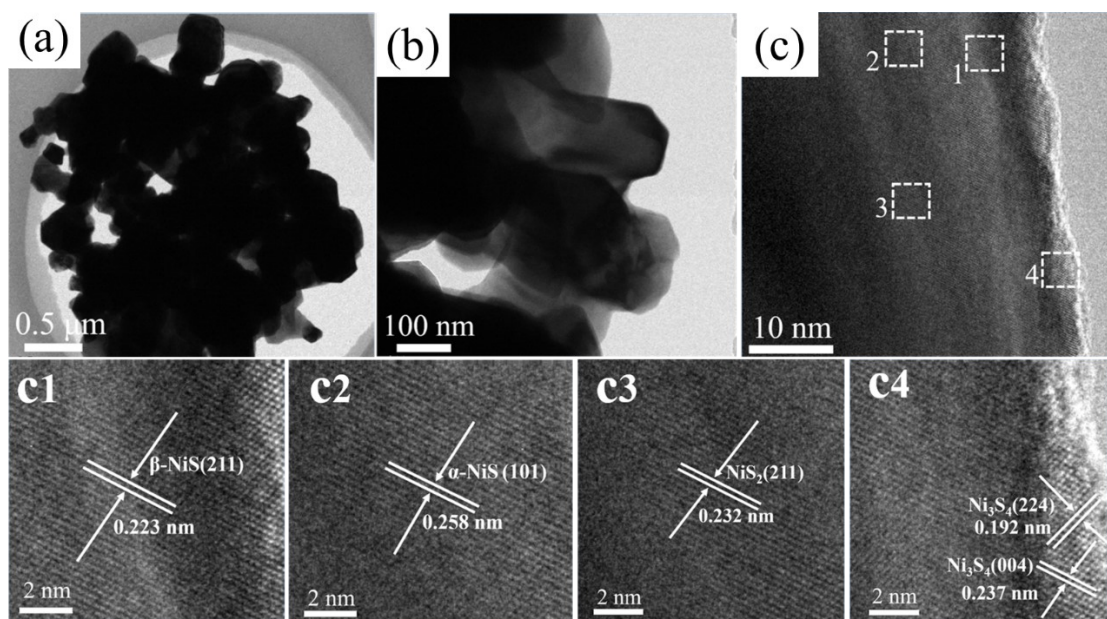


Figure S3. TEM images of S1.

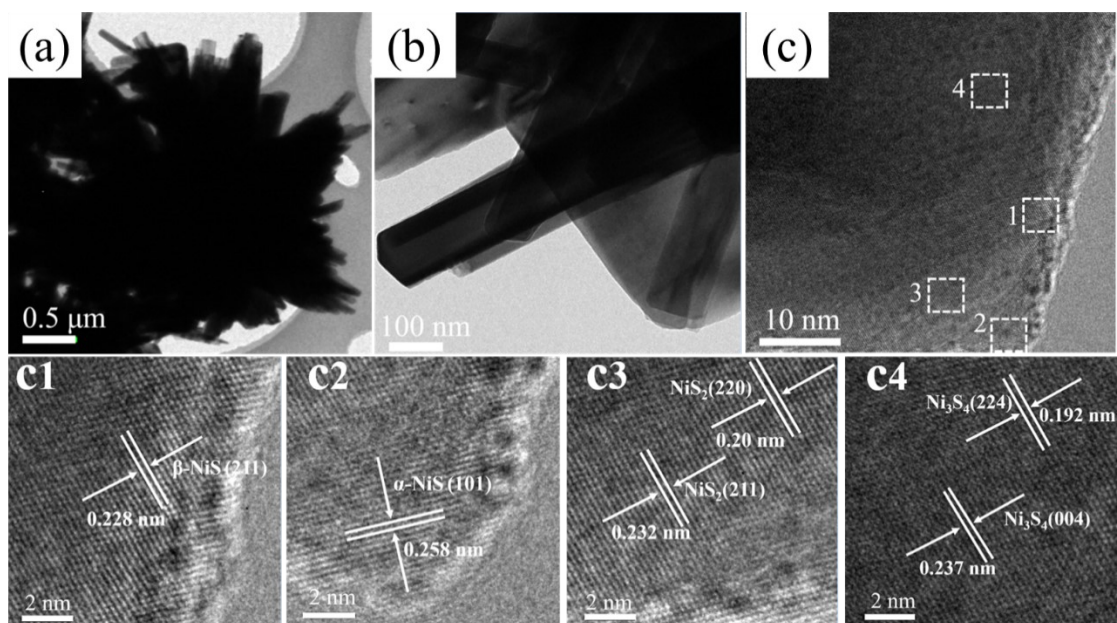


Figure S4. TEM images of S2.

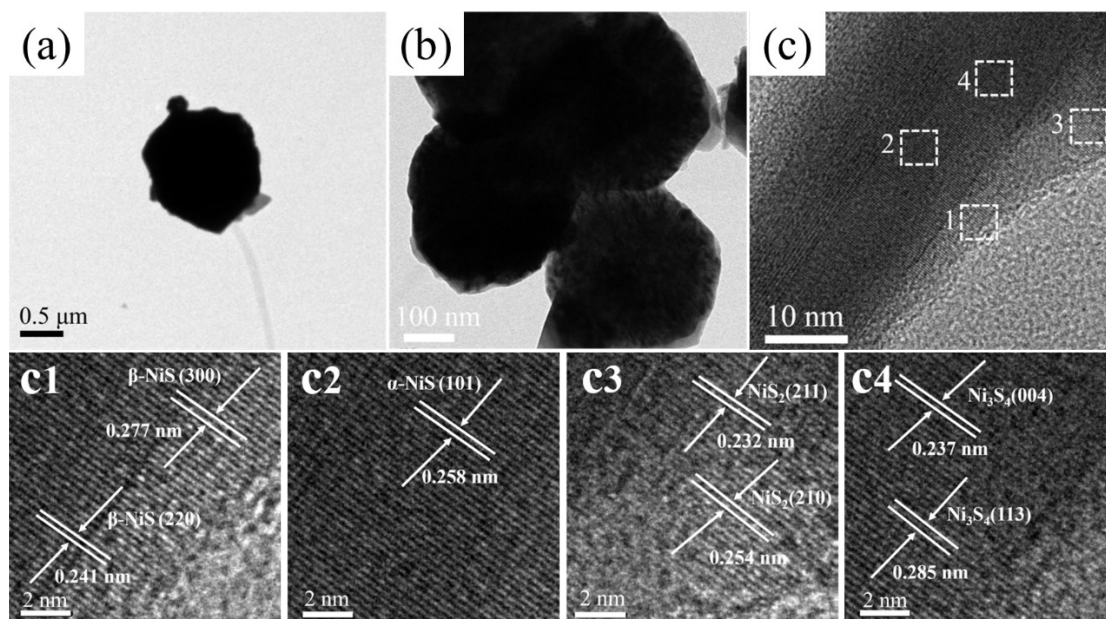


Figure S5. TEM images of S3.

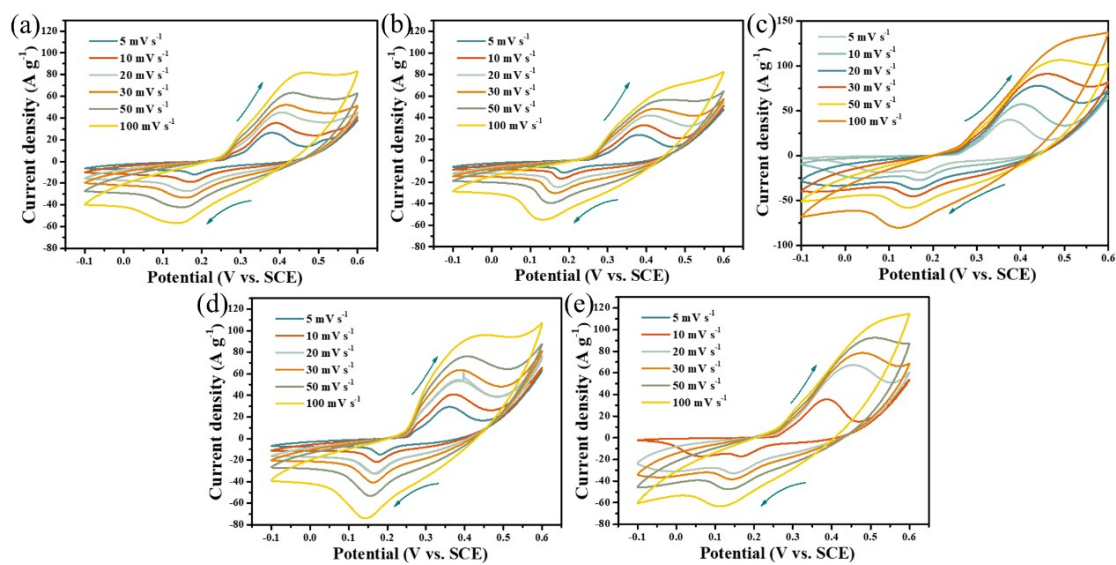


Figure S6. (a-e) CV curves of NS-1h, NS-3h, NS-6h, NS-12h, and NS-20h at different scan rates, successively.

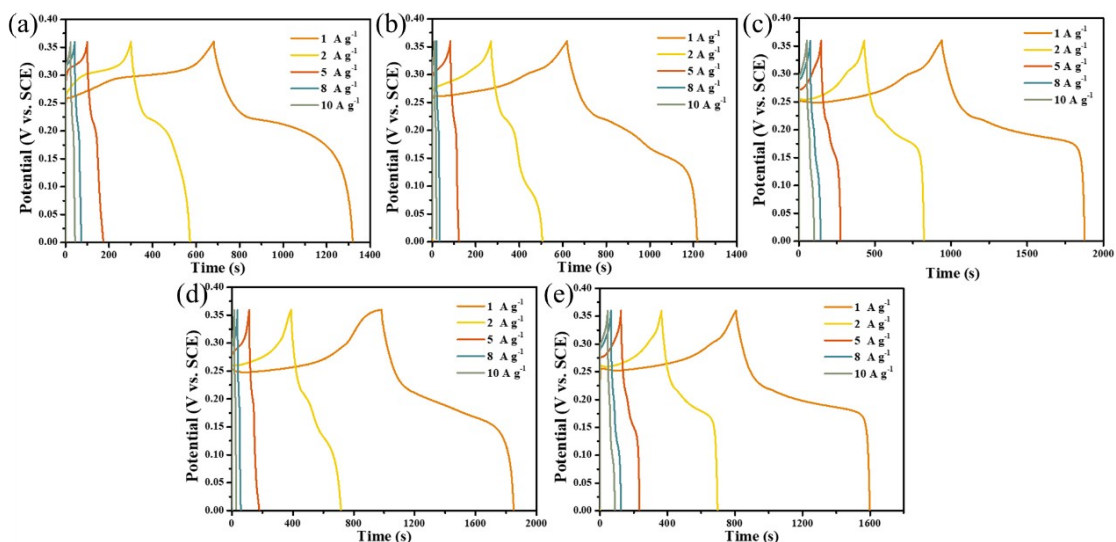


Figure S7. (a-e) GCD curves of NS-1h, NS-3h, NS-6h, NS-12h, and NS-20h at different scan rates.

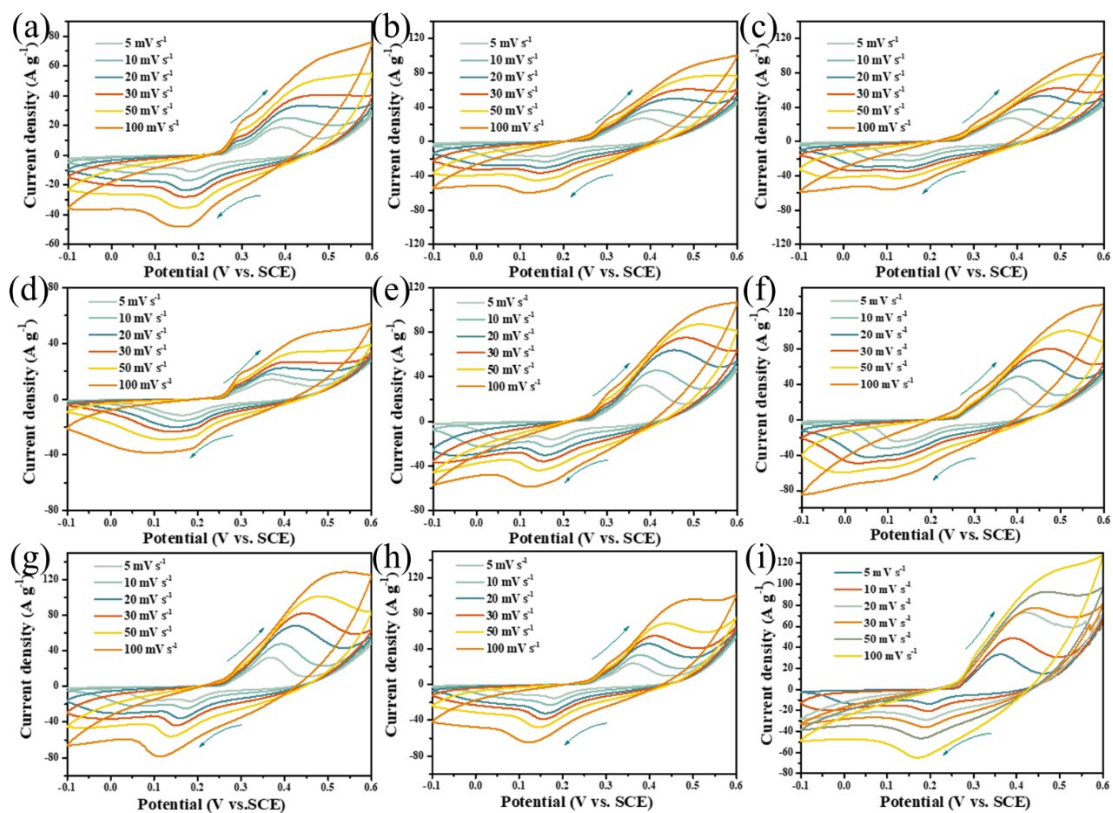


Figure S8. CV curves of N-1 to N-9.

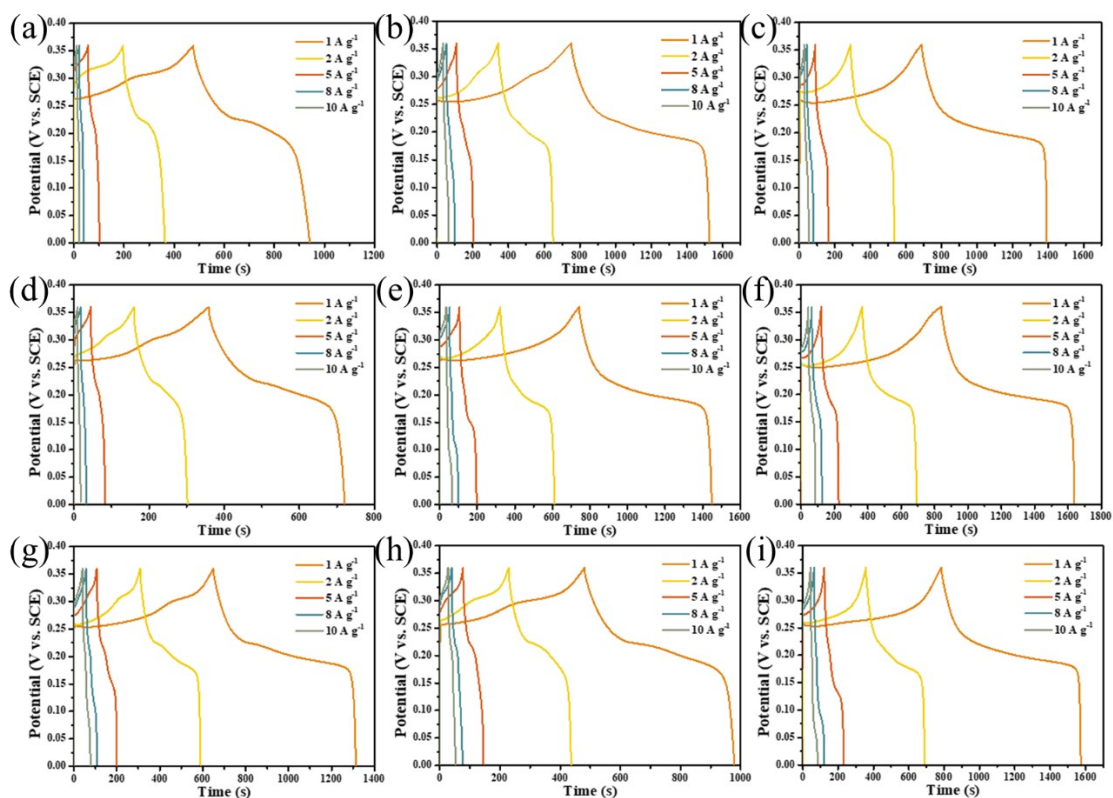


Figure S9. GCD curves of N-1 to N-9.

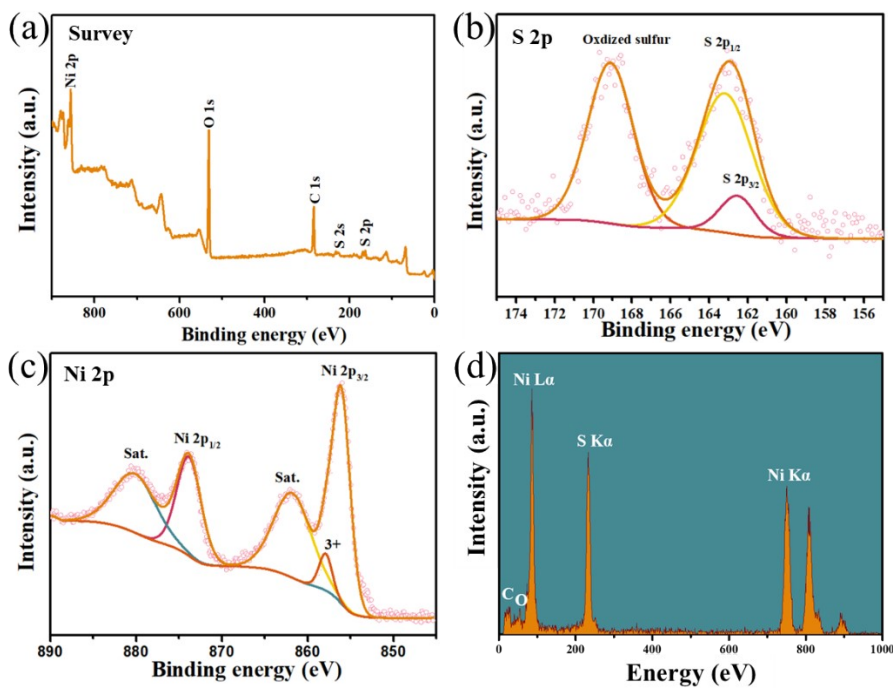


Figure S10. (a) Survey XPS spectrum of NCS. (b) S2p XPS spectrum of NCS. (c) Ni2p XPS spectrum of NCS. (d) EDS spectrum of NCS.

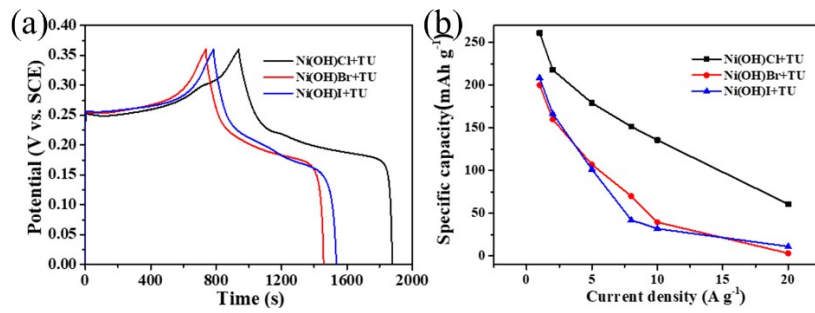


Figure S11. (a) GCD curves of NCS, NBS, and NIS. (b) Specific capacities of NCS, NBS, and NIS at diverse current densities.

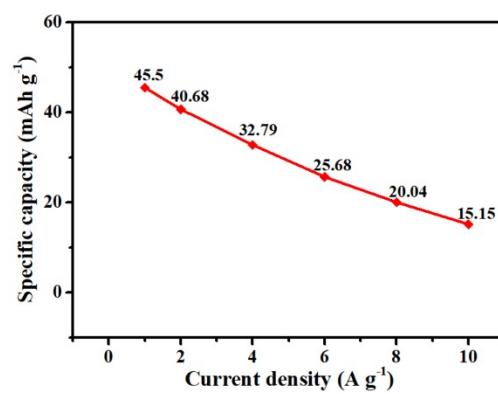


Figure S12. Specific capacities of the fabricated supercapacitor at different current densities.

Table S1. L9(3⁴) orthogonal experimental data.

No.	Col. t (h)	A T (°C)	B -	C n(Ni(OH)Cl)/n(TU)	D C _m mAh·g ⁻¹
1	1(6)	1(120)	1	1(0.15)	129.5
2	1	2(160)	2	2(0.3)	214
3	1	3(200)	3	3(1)	196.3
4	2(12)	1	2	3	100.7
5	2	2	3	1	196.1
6	2	3	1	2	222.4
7	3(20)	1	3	2	184.5
8	3	2	1	3	138.3
9	3	3	2	1	220.7
K _{1j}	539.8	414.7	-	546.3	-
K _{2j}	519.2	548.4	-	620.9	-
K _{3j}	543.5	639.4	-	435.3	-
k _{1j}	179.9	138.2	-	182.1	-
k _{2j}	173.1	182.8	-	207.0	-
k _{3j}	181.2	213.1	-	145.1	-
R	8.1	74.9	-	61.9	-
Factor Order	T > n(Ni(OH)Cl)/n(TU) > t				
Optimal Scheme	B ₃ C ₂ A ₃				

Table S2. Quantitative analysis of S1, S2, and S3 by XRD data.

Sample	Phase	Weight Percentage (%)
S1	β -NiS	66.1
	Ni_3S_4	23.5
	α -NiS	8.2
	NiS_2	2.2
S2	β -NiS	75.6
	Ni_3S_4	15.6
	α -NiS	5.7
	NiS_2	3.1
S3	β -NiS	67.7
	Ni_3S_4	16.6
	α -NiS	7.4
	NiS_2	8.3

Table S3. A comparison of the specific capacity of nickel-based materials reported previously.

Sample	Capacity (mAh·g ⁻¹)	Current density (A·g ⁻¹)	Electrolyte	Ref.
5-NiS@CoS	168.1	1	2 M KOH	1
NiS-C	168.3	0.5	3 M KOH	2
NiCo ₂ S ₄ /Co ₉ S ₈ hollow nanospheres	126.0	1	1 M KOH	3
NiS nanostructures	120.5	1	2 M KOH	4
NiS hierarchical hollow cubes	133.6	1	2 M KOH	5
RHAC/NiCo ₂ S ₄	233.3	1	2 M KOH	6
Ni ₃ S ₄ @rGO	203.3	2	2 M KOH	7
NiS/NiO	118.1	2	2 M KOH	8
NiS/NTA-2	191.3	1	2 M KOH	9
NiS@C QDs-CNTs-rGO	241.0	1	1 M KOH	10
Ni ₃ S ₄ -NiS/rGO	219.2	1	2 M KOH	11
H-NiS _{1-x} /C-50	216.0	1	6 M KOH	12
R-NiS/rGO	206.6	1	2 M KOH	13
C@MoS ₂ /Ni ₃ S ₄	132.1	2	2 M KOH	14
Ni _{1-x} Co _{2-x} S/Co(OH) ₂	195.1	1	3 M KOH	15
Cu-doped Ni ₃ S ₂	105.9	1	3 M KOH	16
Ni ₃ S ₄ -MoS ₂	136.8	1	3 M KOH	17
pyrite NiS ₂ nanoparticles	163.9	2	6 M KOH	18
β-NiS@Ni	191.1	2	1 M KOH	19
NCS	261.2	1	2 M KOH	This work

Table S4. A comparison of some supercapacitors reported previously.

ECs	Energy density (Wh·kg ⁻¹)	Power density (W·kg ⁻¹)	Ref.
5-NiS@CoS//AC	24.1	752.15	1
NiS-C//AC	35.07	420.0	2
NiCo ₂ S ₄ /Co ₉ S ₈ hollow nanospheres//AC	36.7	800.0	3
NiS nanostructures//AC	16.5	250.0	4
NiS hierarchical hollow cubes//AC	34.9	387.5	5
RHAC/NiCo ₂ S ₄ /RHAC	41.6	150.0	6
Ni ₃ S ₄ @rGO//AC	37.3	398	7
NiS/NTA-2//AC	30.8	4187.2	9
NiS@C QDs-CNTs- rGO//Graphene	21.0	811	10
H-NiS _{1-x} /C-50//AC	36.88	750	12
Ni ₁ -Co ₂ -S/Co(OH) ₂ //AC	48.8	800	15
Cu-doped Ni ₃ S ₂ //AC	33.7	850.1	16
Ni ₃ S ₄ -MoS ₂ //AC	58.43	385.95	17
pyrite NiS ₂ nanoparticles//AC	54.2	940.0	18
β-NiS@Ni//AC	40.0	720.0	19
NCS//rGO	36.4	800.0	This work

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