

## Supporting Information

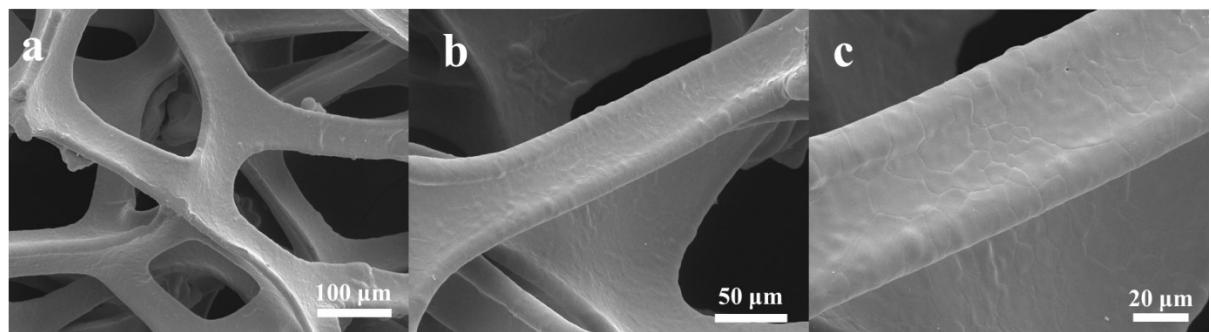
### Spherical V-doped Nickel-iron LDH Decorated on Ni<sub>3</sub>S<sub>2</sub> as High-efficiency Electrocatalyst for Oxygen Evolution Reaction

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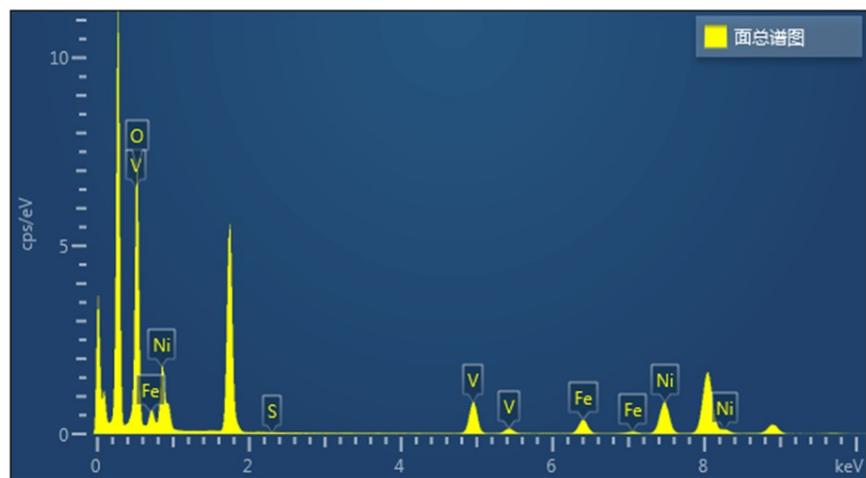
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**Fig. S1.** SEM images of commercial Ni foam at low and high magnifications.



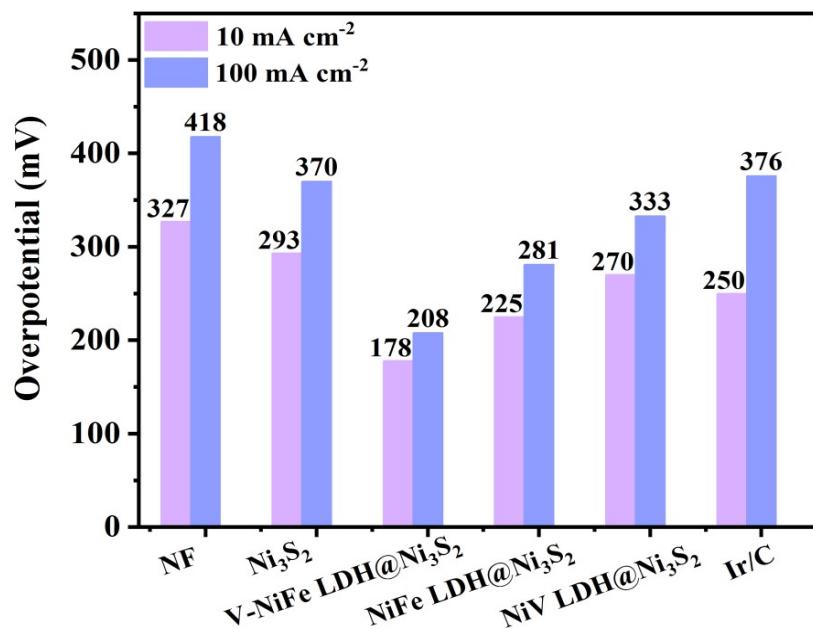
**Fig. S2.** Element composition diagram corresponding to EDX spectrum

**Table S1** Elemental composition of V, Fe, Ni and S determined by EDX spectrum

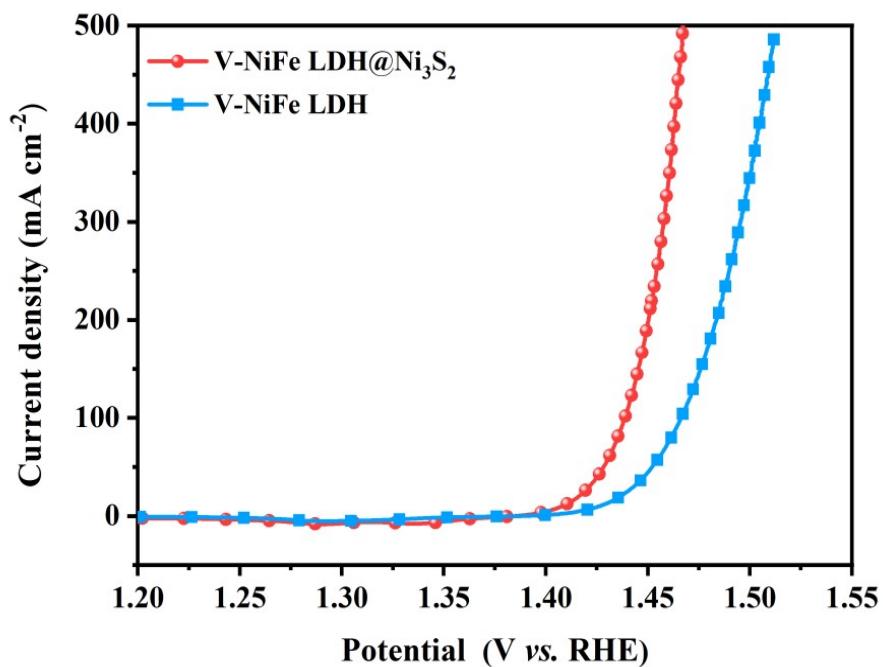
Element	V	Fe	Ni	S	O
Atom%	5.85	2.75	6.35	0.02	85.03

**Table S2** XPS quantization ratio of three valence states V

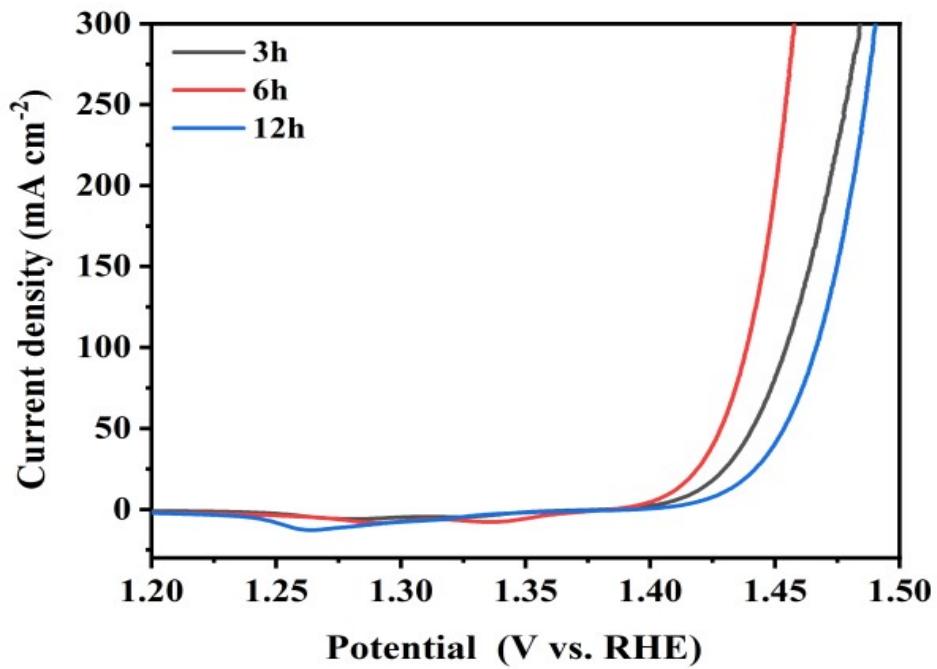
Name	V <sup>3+</sup>	V <sup>4+</sup>	V <sup>5+</sup>
%content	26.93	49.96	23.11



**Fig. S3.** comparison of overpotential required at  $10 \text{ mA cm}^{-2}$  and  $100 \text{ mA cm}^{-2}$



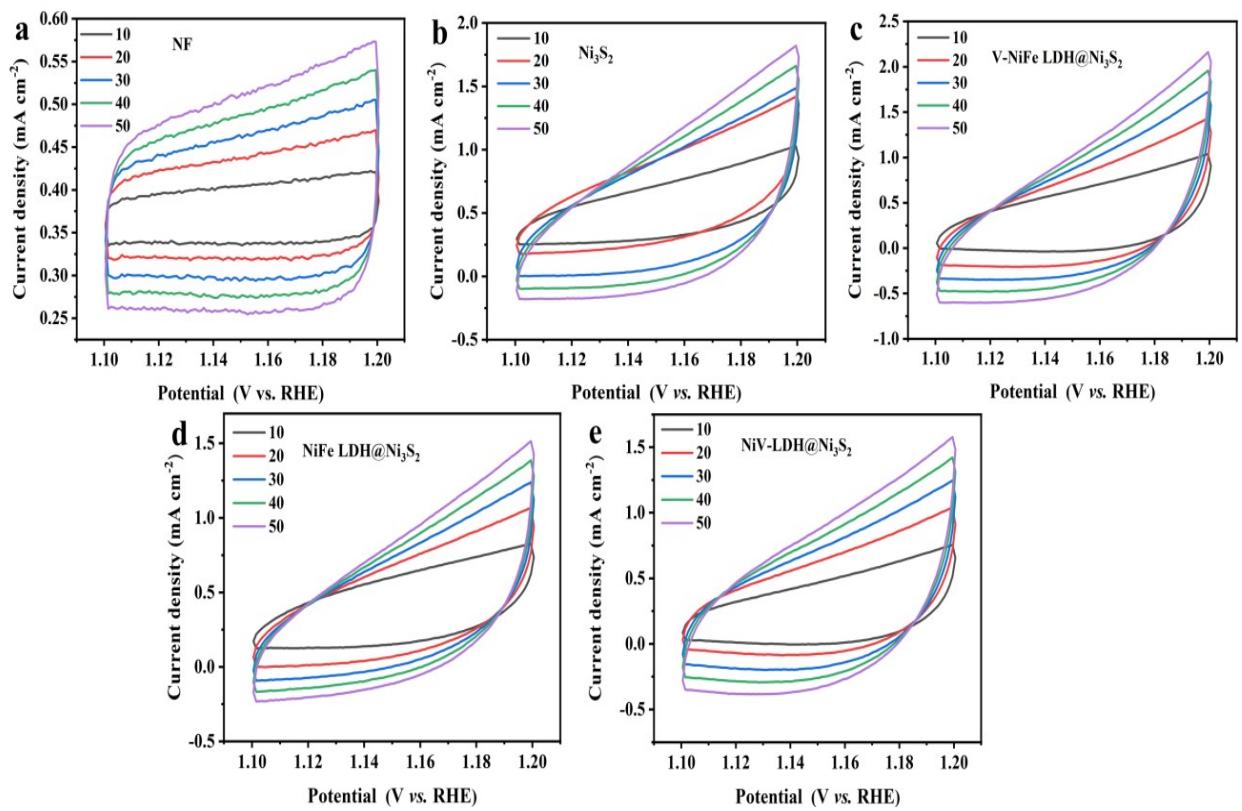
**Fig. S4.** OER polarization curves of V-NiFe LDH@ $\text{Ni}_3\text{S}_2$  and V-NiFe LDH.



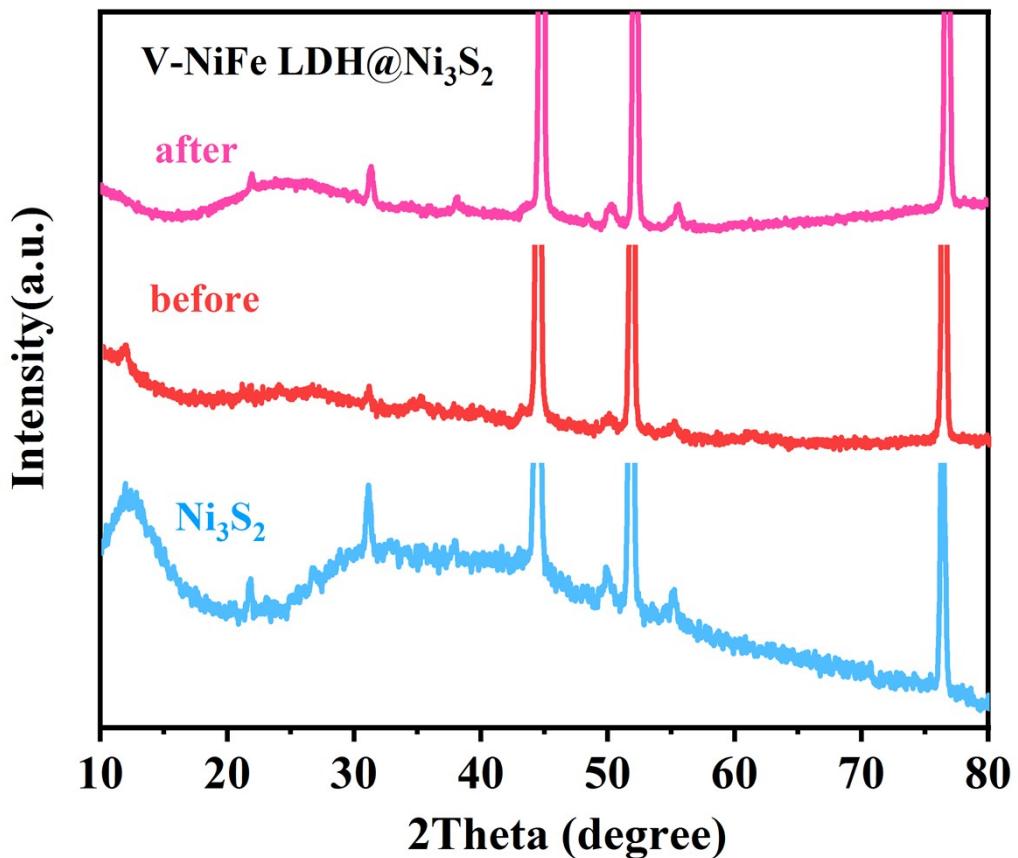
**Fig. S5.** OER polarization curves of V-NiFe LDH@Ni<sub>3</sub>S<sub>2</sub>-3h, V-NiFe LDH@Ni<sub>3</sub>S<sub>2</sub>-6h, and V-NiFe LDH@Ni<sub>3</sub>S<sub>2</sub>-12h.

**Table S3.** Comparison of the OER performance for the V-NiFe LDH@Ni<sub>3</sub>S<sub>2</sub> catalyst

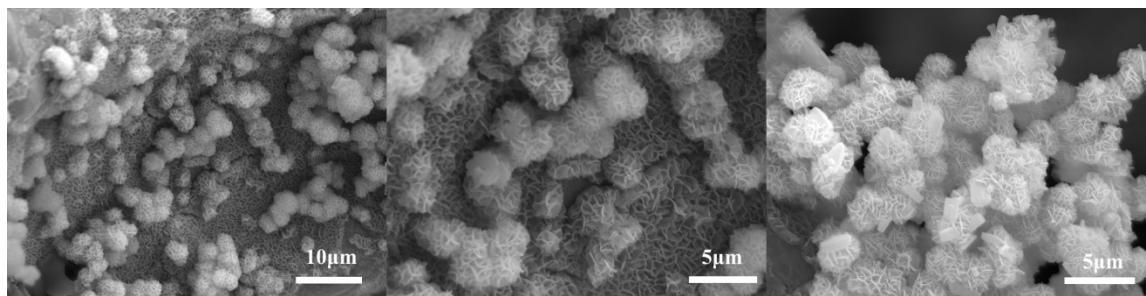
Electrocatalysts	Overpotential (mV) at 10 mA cm <sup>-2</sup>	Tafel slope (mV dec <sup>-1</sup> )	Electrolyte	Reference
<b>V-NiFe-LDH@Ni<sub>3</sub>S<sub>2</sub></b>	178	27.31	1M KOH	This work
<b>NiFe LDH/NF</b>	219	33	1M KOH	J. Mater. Chem. A, 2019, 7, 22889.
<b>Pt-NiFe LDH</b>	195	31.3	1M KOH	Nano Energy 2017, 39, 30-43.
<b>Ni<sub>5</sub>P<sub>4</sub>/Ni<sub>5</sub>P<sub>2</sub>/NiFe LDH</b>	197	46.6	1M KOH	J. Mater. Chem. A. 2018, 6, 13619-13623.
<b>Cu@NiFe LDH</b>	199	27.8	1M KOH	Energy Environ. Sci. 2017, 10 , 1820-1827.
<b>NiFe LDH/(NiFe)S<sub>x</sub></b>	210	105	1M KOH	Electrochim. Acta, 2020, 348, 136339.
<b>NiFe- LDH@NiCoP/NF</b>	220	48.6	1M KOH	Adv. Funct. Mater. 2018, 28, 1706847.
<b>NiFeRu LDH/NF</b>	225	32.4	1M KOH	Adv. Mater. 2018, 30, 1706279.
<b>Cu@CoFe LDH</b>	240	44.4	1M KOH	Nano Energy 2017, 41, 327-336.
<b>CoSe/NiFe LDH</b>	250	57	1M KOH	Energy Environ. Sci. 2016, 9,478-483.
<b>NiO@NiFe-LDH</b>	256	72	1M KOH	ACS Sustainable Chem. Eng. 2019, 7, 2327.
<b>NiCo/NiCoOx @FeOOH</b>	278	47.5	1M KOH	Electrochim. Acta 2017, 257, 1-8.
<b>Ni<sub>2</sub>P/Ni<sub>3</sub>S<sub>2</sub>/NF</b>	210	62	1M KOH	Nano Energy 2018, 51, 26-36.
<b>MoxW<sub>1-x</sub>S<sub>2</sub>@ Ni<sub>3</sub>S<sub>2</sub>/NF</b>	285	90	1M KOH	ACS Appl. Mater. Interfaces 2017, 9, 26066-26076.



**Fig. S6.** Cyclic voltammograms at different scan rates (from 10 mV/s to 50 mV/s with an interval rate of 10 mV/s). (a) NF, (a)  $\text{Ni}_3\text{S}_2$  (b) V-NiFe LDH@ $\text{Ni}_3\text{S}_2$ , and (c) NiFe-LDH@ $\text{Ni}_3\text{S}_2$ , and (d) NiV-LDH@ $\text{Ni}_3\text{S}_2$ .



**Fig. S7.** XRD patterns



**Fig. S8.** SEM images of V-NiFe LDH@Ni<sub>3</sub>S<sub>2</sub> (anode for OER) after 20 hours OER stability measurements with other reported OER electrocatalysts.